HETEROCYCLES, Vol. 68, No. 4, 2006, pp. 801 - 806. © The Japan Institute of Heterocyclic Chemistry Received, 9th January, 2006, Accepted, 28th February, 2006, Published online, 3rd March, 2006. COM-06-10669

SYNTHESIS OF 1*H*-PYRAZOLO [4', 3':5, 6] PYRIMIDO [2, 1-*a*]-ISOINDOL-4(10*H*)-ONES. DERIVATIVES OF A NEW RING SYSTEM

Abolghasem Davoodnia,^a Mehdi Bakavoli,^{a*} Ali Vahedinia,^a Mohammad Rahimizadeh,^b and Mina Roshani ^a

^aDepartment of Chemistry, School of Sciences, Islamic Azad University, Mashhad, Iran. ^bDepartment of Chemistry, School of Sciences, Ferdowsi University of Mashhad, Iran

Corresponding author, email: mbakavoli@yahoo.com

Abstract –Several derivatives of the new pyrazolo[4,3:5,6]pyrimido[2,1-a]isoindolone ring system (4) have been prepared through heterocyclization of 2-chloromethylbenzoyl chloride (1) with appropriately substituted 5-amino-1*H*-pyrozole-4-carbonitriles (2). The reaction intermediate inden-2-yl-1-phenylpyrazole-4-carbonitrile (3), on subjection to acid or base, underwent hydrolysis and cyclocondensation simultaneously to afford the tetracyclic compound (4).

INTRODUCTION

The preparation of novel polycyclic *N*-heterocyclic compounds and the exploration of their synthetic pathways have recently received much attention in our group.¹ In connection with these studies, in a previous paper,² we reported a new synthetic route to new tetracyclic isoindolo[2,1-*a*]quinazoline derivatives through heterocyclization of 2-aminobenzonitrile with 2-chloromethylbenzoyl chloride and we gave an account of the synthetic pathways to these compounds. As an extension of this work we now report our investigations for the synthesis of 1*H*-pyrazolo[4,3:5,6]pyrimido[2,1-*a*]isoindol-4(10*H*)ones, members of a new heterocyclic ring system.

RESULTS AND DISCUSSION

The synthetic utility of 2-chloromethylbenzoyl chloride (1), as a reactive bifunctional reagent in producing polycyclic *N*-heterocycles is well known.³⁻⁹ In our hands, we reacted this reagent with 5-amino-1*H*-pyrazole-4-carbonitriles (2a-d) in the presence of four equivalents of potassium *t*-butoxide in

t-butanol at reflux temperature to afford the corresponding $5-(1-\infty - 1,3-dihydro-2H-isoindol-2-yl)-1H$ pyrazole-4-carbonitriles (**3**a-d). These products were respectively cyclised to the corresponding1H-pyrazolo[4,3:5,6]pyrimido[2,1-*a*]isoindol-4(10H)-ones (**4**) either on subjection to acid or base atelevated temperatures. A tentative mechanism to explain the formation of this heterocyclic ring system(Scheme 1) primarily involves the formation of the intermediate (**3**) through cyclocondensation of2-chloromethylbenzoyl chloride with the amino moiety of the pyrazole ring (**2**) followed by hydrolysis ofthe cyano group into amide with subsequent cyclocondesation to the novel tetracyclic compound (**4**). It isworth noting that in this multi-step synthesis the key intermediate (**5**) was not isolated. It seems likely thathydrolysis of the cyano group into amide and ring closure step occurs simultaneously. All our effort toisolate this intermediate was not successful.

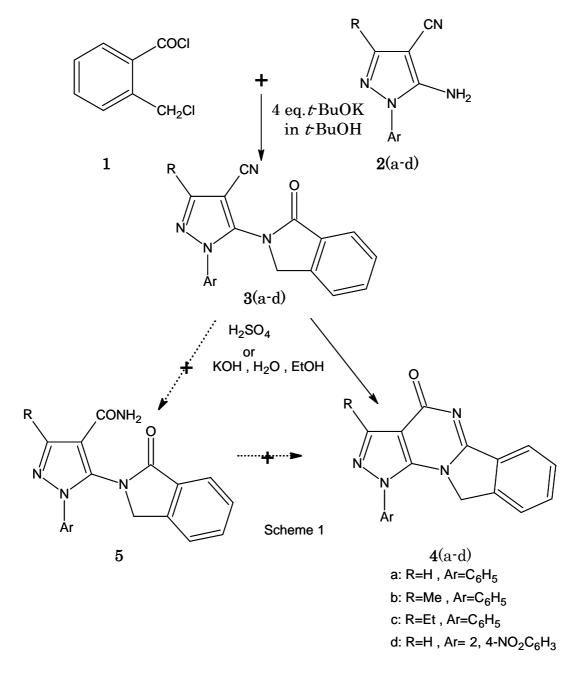


Table 1 Physical, Spectral, and Microanalytical Data of5-(1-Oxo-1,3-dihydro-2H-isoindol-2-yl)-1H-pyrazole-4-
carbonitriles (3a-d)

Entry	Yield (%)	mp (° C)	Spectral data
3a	69	156-158	¹ H NMR: δ (CDCl ₃), 4.66 (s, 2H, CH ₂), 7.43-8.04 (m,10H, Aromatic rings);IR (KBr, disc), v, CN, 2212 cm ⁻¹ ; MS m/z, M ⁺ 300; Anal. Calcd for C ₁₈ H ₁₂ N ₄ O : C, 71.99; H, 4.03; N, 18.66. Found: C, 72.05; H, 4.01 N, 18.62.
3b	83	138-140	¹ H NMR: δ (CDCl ₃), 2.48 (s, 3H, Me), 4.62 (s, 2H, CH ₂), 7.41-7.86 (m, 9H, Aromatic rings); IR (KBr, disc), v , CN, 2210 cm ⁻¹ , MS m/z, M ⁺ 314; Anal. Calcd for C ₁₉ H ₁₄ N ₄ O: C, 72.60; H, 4.49; N, 17.82. Found: C, 72.54; H, 4.44 N, 17.85.
3c	85	60-62	¹ H NMR: δ (CDCl ₃), 2.47 (t, 3H, j=7.8 Hz, Me), 3.97 (q, 2H, j=7.8 Hz, CH ₂), 4.57 (s, 2H, CH ₂), 7.42-7.86 (m, 9H, Aromatic rings); IR (KBr, disc), v, CN, 2148 cm ⁻¹ ; MS m/z, M ⁺ 328; Anal. Calcd for C ₂₀ H ₁₆ N ₄ O: C, 73.15; H, 4.91; N, 17.06. Found: C, 73.12; H, 4.96 N, 17.04.
3d	65	272-274	¹ H NMR: δ (CDCl ₃), 5.25 (s, 2H, CH ₂),7.36-8.82 (m, 8H, Aromatic rings); IR (KBr, disc), v, CN, 2217 cm ⁻¹ ; MS m/z, M ⁺ 390; Anal. Calcd for C ₁₈ H ₁₀ N ₆ O ₅ : C, 55.39; H, 2.58; N, 21.53. Found: C, 55.35; H, 2.56 N, 21.56.

Entry	Yield (%)	mp (°C)	Spectral data
4a	81	182-184	 ¹H NMR: δ (DMSO-d₆), 4.16 (s, 2H, CH₂), 6.90-8.41 (m, 10H, Aromatic rings); MS m/z, M⁺ 300; Anal. Calcd for C₁₈H₁₂N₄O : C, 71.99; H, 4.03; N, 18.66. Found: C, 71.96; H, 3.99 N, 18.67.
4b	74	258-260	 ¹H NMR: δ (DMSO-d₆), 1.29 (s, 3H, CH₃), 4.86 (s, 2H, CH₂), 7.70-8.07 (m, 9H, Aromatic rings); MS m/z, M⁺ 314; Anal. Calcd for C₁₉H₁₄N₄O : C, 72.60; H, 4.49; N, 17.82. Found: C, 72.57; H, 4.53 N, 17.76.
4c	79	270-272	 ¹H NMR: δ (DMSO-d₆), 1.29 (t, 3H, j=8.0 Hz, CH₃), 2.92 (q, 2H, j=8.0 Hz, CH₂), 4.79 (s, 2H, CH₂), 7.65-8.03 (m, 9H, Aromatic rings); MS m/z, M⁺ 328; Anal. Calcd for C₂₀H₁₆N₄O : C, 73.15; H, 4.91; N, 17.06. Found: C, 73.10; H, 4.93; N, 17.05.
4d	61	350(decomp)	 ¹H NMR: δ (DMSO-d₆), 4.56 (s, 2H, CH₂), 7.27-9.04 (m, 8H, Aromatic rings); MS m/z, M⁺ 390; Anal. Calcd for C₁₈H₁₀N₆O₅: C, 55.39; H, 2.58; N, 21.53. Found: C, 55.43; H, 2.54 N, 21.51.

Table 2 Physical, Spectral, and Microanalytical Data of
1H-Pyrazolo[4',3':5,6]pyrimido[2,1-a]isoindol-4(10H)-
ones (4a-d)

The structure of the novel compounds (4a-d) was deduced from their spectral data. The MS of these compounds displayed molecular ion peaks at the appropriate m/z values.

The ¹H NMR spectrum of **4**a exhibited one single sharp line readily recognizable as arising from CH₂ of the isoindole ring protons ($\delta = 4.16$ ppm). The aromatic hydrogens give rise to charactristic signals in the aromatic region of the spectrum ($\delta = 6.90$ -8.41 ppm). The IR spectrum was devoid of the CN absorption band at 2212 cm⁻¹ of the precurser, which shows the inclusion of nitrile moiety in cyclocondensation process.

In conclusion, we have developed a method for preparing in moderate to good yields of tetracyclic pyrazolo[4,3:5,6]pyrimido[2,1-*a*]isoindoles.

EXPERIMENTAL

Melting points were recorded on an Electrothermal type 9100 melting point apparatus. The IR spectra were obtained on a 4300 Shimadzu spectrophotometer as KBr disks. The ¹H NMR (100 MHz) spectra were recorded on a Bruker AC 100 spectrometer. The MS were scanned on a Varian CH-7 instrument at 70 ev. Elemental analysis was performed on a Thermofinnigan Flash EA microanalyzer.

General Procedure for the Preparation of 5-(1-Oxo-1,3-dihydro-2*H*-isoindol-2-yl)-1*H*-pyrazole-4-carbonitriles (3a-d).

To a solution of the 5-amino-1*H*-pyrazole-4-carbonitriles (2a-d) (5 mmol) and potassium *t*-butoxide (20 mmol) in *t*-butanol (40 mL), 2-chloromethylbenzoyl chloride (1) (1.13 g, 6 mmol) was added. The reaction mixture was heated under reflux for 5.0 h. After the completion of the reaction, the mixture was cooled to rt and water was added. The precipitate was collected and recrystallised from *n*-hexane-toluene to give compounds (3a-d) in 69, 83, 85 and 65% yield, respectively (see Table 1).

General Procedure for the Preparation of 1*H*-pyrazolo[4,3:5,6]pyrimido[2,1-*a*]isoindol-4(10*H*)-one (4a and 4d).

5-(1-Oxo-1,3-dihydro-2*H*-isoindol-2-yl)-1*H*-pyrazole-4-carbonitriles (**3a or 3d**) (5 mmol) was dissolved in EtOH (20 mL)-H₂O (10 mL) containing KOH (0.37 g, 6 mmol). The mixture was refluxed for 12 h. The reaction mixture was cooled to rt and neutralized by 1M HCl. The crude product was collected and washed with water and chloroform respectively, then recrystallized from ethanol to give compounds (**4a**) and (**4d**) in 81 and 61% yield, respectively (see Table 2).

General Procedure for the Preparation of 1*H*-Pyrazolo[4,3:5,6]pyrimido[2,1-*a*]isoindol-4(10*H*)-one (4b and 4c).

5-(1-Oxo-1,3-dihydro-2*H*-isoindol-2-yl)-1*H*-pyrazole-4-carbonitriles (**3b** or **3c**) (5 mmol) was dissolved in conc. H_2SO_4 (15 mL). The mixture was stirred at 180 °C for 4 h, then cooled and neutralized by potassium hydroxide. The precipitate was collected and washed with water and methanol respectively, then recrystallized from ethanol to give compounds (**4b**) and (**4c**) in 74 and 79% yield respectively (see Table 2).

REFERENCES

1. M. Bakavoli, A. Davoodnia, M. Rahimizadeh, M. M. Heravi, and M. Ghassemzadeh, J. Chem. Res.

(s), 2002, 178.

- M. Bakavoli, A. Davoodnia, M. Rahimizadeh, and M. M. Heravi, *Mendeleyev Commun.*, 2006, 16, 29.
- 3. J. Kant, F. D. Popp, and B. C. Uff, J. Heterocycl. Chem., 1985, 22, 1313.
- 4. W. Letwanawatana, S. Thianpatanagul, J. L. Cashaw, and V. E. Davis, *Tetrahedron Lett.*, 1984, **25**, 3485.
- J. B. Doherty, C. P. Dorn, B. E. Witzel, D. L. Allison, T. Y. Shen, and P. E. Finke, Eur. Pat. Appl. EP 68,460 (*Chem. Abstr.*, 1983, 98, P198190b).
- 6. C. R. Dalton, J. M. Kane, and D. Rampe, *Tetrahedron Lett.*, 1992, 33, 5713.
- 7. Y. Sato and H. Fujita, Japan Kokai 75,117,790 (Chem. Abstr., 1976, 84, P105670f).
- 8. H. Fujita and Y. Sato, Chem. Pharm. Bull., 1975, 23, 1764.
- H. W. Heine, D. W. Ludovici, J. A. Pardoen, R. C. Weber, E. Bonsall, and K. R. Oserhout, J. Org. Chem., 1979, 44, 3843.