HETEROCYCLES, Vol. 68, No. 8, 2006, pp. 1699 - 1703. © The Japan Institute of Heterocyclic Chemistry Received, 27th April, 2006, Accepted, 20th June, 2006, Published online, 23rd June, 2006. COM-06-10775

AN ACYLATED CYANIDIN 3-RUTINOSIDE-7-GLUCOSIDE WITH *p*-HYDROXYBENZOIC ACID FROM THE RED-PURPLE FLOWERS OF *CAMPANULA MEDIUM*

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Abstract – A new acylated anthocyanin was isolated from the red-purple flowers of Campanula medium as a major pigment together with a known anthocyanin. The new pigment was determined to be cvanidin 3-O-[6-O-(α-L-rhamnopyranosyl)-β-D-glucopyranosyl]-7-O-{4-[6-O-(4-(6-O- $(p-hydroxybenzoyl)-\beta-D-glucopyranosyl)-oxybenzoyl)-\beta-D-glucopyranosyloxy$ benzoyl]- β -D-glucopyranoside} based on spectroscopic analyses. The known pigment was also identified to be rubrocampanin, pelargonidin 3-rutinoside-7-(p-hydroxybenzoylglucopyranosyl-p-hydroxybenzoylglucopyranosyl-p-hydroxybenzoylglucopyranoside).

In 1990, we reported the isolation and structure determination of the flower anthocyanins of *Campanula medium*, and described that its purple and pink cultivars contained campanin and rubrocampanin as their main anthocyanins, respectively.¹ Both pigments were determined to be 3-rutinoside-7-(p-hydroxybenzoylglucopyranosyl-p-hydroxybenzoylglucopyranosyl-p-hydroxybenzoylglucopyranosides) of delphinidin (campanin) and pelargonidin (rubrocampanin), respectively.¹

Two analogous of campanin, such as monodeacylcampanin lacking of a molecule of *p*-hydroxybenzoic acid, and violdelphin lacking of one molecule of glucosyl-*p*-hydroxybenzoic acid, were also found in the flowers of *C. isophyla*, *C.carpatica* and *C. proskarshyana* by Goto and his coworkers in 1993.² As a part of our continuing studies on flower color variation,³ we report here the isolation and structure elucidation of a new anthocyanin along with a known anthocyanin, rubrocampanin, from the red-purple flowers of *C. medium* cv. 'May Purple Margin'.⁴ For the sake of convenience, we named this new compound as

purprocampanin (PC).

Air-dried petals (50g) of *C. medium* were extracted with 50% AcOH. Two major and at least 8 minor peaks of anthocyanins were observed by HPLC analysis of the crude extract. Among these anthocyanins, two main compounds (1) (*ca*.50mg) and (2) (*ca* 70mg) were obtained as scarlet and dark red powders, respectively, by the process described previously.¹

On acid hydrolysis,⁵ both pigments gave glucose, rhamnose, and *p*-hydroxybenzoic acid as their sugar and hydroxyacid moieties. As their aglycones, pelargonidin and cyanidin were observed in pigments (1) and (2), respectively. On alkaline hydrolysis,⁵ pigments (1) and (2) gave *p*-hydroxybenzoic acid, 4-glucosylhydroxybenzoic acid,⁶ and a deacylanthocyanin. The deacylanthocyanin of pigment (1) was identified as pelargonidin 3-rutinoside-7-glucoside by direct comparison of TLC and HPLC⁷ with authentic specimen,⁸ which was derived from rubrocampanin on the alkaline hydrolysis.¹ By HPLC analysis, pigment (1) was identified to be rubrocampanin, and its structure was further confirmed based on the analysis of FAB mass and NMR spectra.¹⁰ On the other hand, the deacylanthocyanin⁹ of pigment (2) were not identical with both 3-rutinoside-7-glucosides of pelargonidin and delphinidin, obviously. Based on the examination of chemical and spectral properties of pigment (2),¹¹ it was presumed to be a new acylated cyanidin glycoside, an analogue of campanin and rubrocampanin.

The HR FAB mass spectrum of pigment (2) gave its molecular ion $[M]^+$ at m/z 1441.3906, in agreement with the mass calculated for C₆₆H₇₃O₃₆, which was composed of cyanidin with four molecules of glucose, three molecules of *p*-hydroxybenzoic acid and one molecule of rhamnose.



Figure 1. Purprocampanin from *Campanula medium*. Observed NOE's are indicated by arrows.

In order to determine the structure of pigment (2), its ¹H and ¹³C NMR spectroscopic measurements including 2D COSY, negative difference NOE (NOEDIF), HMQC, HMBC and NOESY spectra were carried out in CF₃CO₂D-DMSO- d_6 (1:9). The chemical shifts of its protons were assigned as shown in Table 1. Regarding the sugar moieties of pigment (2), their signals were observed in the region of δ 5.39-1.20, and five anomeric proton signals were assigned to be δ 5.37 (d, *J*=7.7 Hz, Glc-A), 5.39 (d, *J*=7.7 Hz, Glc-B), 5.13 (d, *J*=6.7 Hz, Glc-C), 5.00 (d, *J*=7.7 Hz, Glc-D), and 4.70 (s, Rhamnose). The coupling constants of the glucose moieties were in the region of *J*=6.7-7.7 Hz, suggesting that all glucose units must be β -glucopyranose form (Figure1). Furthermore, six characteristic proton signals shifted to the lower magnetic field at δ 4.73 and 4.38 (Glc-B, H-6a and –6b), δ 4.66 and 4.60 (Glc-C, H-6a and –6b), and δ 4.43 and 4.15 (Glc-D, H-6a and –6b), were observed and assigned to the methylene protons of glucose moieties by the analysis of 2D COSY and NOESY spectra, indicating that Glc-B, Glc-C, and Glc-D were acylated with *p*-hydroxybenzoic acids (I, II, and III) at OH-6 groups of three glucose moieties.

Courling constants (I	:	-									
Coupling constants (J in Hz) in parentheses.											
Cyanidin		se A	Glucose D								
8.68 s	1	5.37 d (7.7) 2.70 m	1	5.00 d (7.7)							
7.45 brs	2))	2 3	3.39 m							
8.19 d (2.2) 7 10 d (8 9)	4 5	} 3.20-3.50	4 5	3.20-3.60 3.80 m							
8.42 dd (2.2, 8.9)	6a 6b	4.01 d (10.1) 3 54 m	6a 6b	4.43 d (10.7) 4.15 dd (7.5, 10.7)							
p-Hydroxybenzoic acid (I)		se B	Rhamnose								
7.87 d (8.2) 7.16 d (8.2)	1 2	5.39 d (7.7) 3.59 m	1 2	4.70 s							
p-Hydroxybenzoic acid (II)		3.20-3.75	3 ⊿	3.20-3.70							
7.61d (8.6) 6.84m	5 6a 6b	4.10 m 4.73 d (12.5) 4.38 dd (7.3, 12.5)	5 Me	3.48 s 1.20 s							
p-Hydroxybenzoic acid (III)		ose C									
7.51 d (8.6) 6.63 d (8.6)	1 2 3 4 5 6a	5.13 d (6.7) 3.43 m 3.36 m 3.89 t (7.8) 4.27 m 4.66 m									
	Coupling constants (J in 8.68 s 6.80 brs 7.45 brs 8.19 d (2.2) 7.10 d (8.9) 8.42 dd (2.2, 8.9) oxybenzoic acid (I) 7.87 d (8.2) 7.16 d (8.2) 7.16 d (8.2) oxybenzoic acid (II) 7.61d (8.6) 6.84m oxybenzoic acid (III) 7.51 d (8.6) 6.63 d (8.6)	Coupling constants $(J \text{ in Hz})$ ininGluco8.68 s16.80 brs27.45 brs38.19 d (2.2)47.10 d (8.9)58.42 dd (2.2, 8.9)6a6b6boxybenzoic acid (I)Gluco7.87 d (8.2)17.16 d (8.2)2oxybenzoic acid (II)47.61d (8.6)56.84m6a6b0oxybenzoic acid (III)47.61d (8.6)56.84m6a6b3456a256a	Coupling constants (J in Hz) in parentheses. Glucose A 8.68 s 1 $5.37 d (7.7)$ 6.80 brs 2 $3.70 m$ 7.45 brs 3 $3.20-3.50$ 7.10 d (8.9) 5 $3.20-3.50$ 7.10 d (8.9) 5 $3.20-3.50$ 8.42 dd (2.2, 8.9) 6a $4.01 d (10.1)$ 6b $3.54 m$ $6a$ oxybenzoic acid (I) $6b$ $3.54 m$ oxybenzoic acid (I) $7.87 d (8.2)$ 1 $5.39 d (7.7)$ $7.16 d (8.2)$ 1 $5.39 d (7.7)$ 2 $7.61 d (8.6)$ 5 $4.10 m$ $6a$ $4.73 d (12.5)$ $6b$ $4.38 d (7.3, 12.5)$ oxybenzoic acid (II) $7.51 d (8.6)$ 1 $5.13 d (6.7)$ $7.51 d (8.6)$ 1 $5.13 d (6.7)$ 2 $7.51 d (8.6)$ 1 $5.13 d (6.7)$ 2 $7.51 d (8.6)$ 1 $5.13 d (6.7)$ 2 $6a$ $4.389 t (7.8)$ 5 $4.27 m$ $6a$ $4.66 m$ $6a$ $4.60 m$ <	Coupling constants (J in Hz) in parentheses. Glucose in Glucose A Glucose 8.68 s 1 $5.37 d (7.7)$ 1 6.80 brs 2 $3.70 m$ 2 7.45 brs 3 3 3 $8.19 d (2.2)$ 4 $3.20-3.50$ 4 $7.10 d (8.9)$ 5 5 5 $8.42 dd (2.2, 8.9)$ 6a $4.01 d (10.1)$ 6a $6b$ $3.54 m$ 6b $3.54 m$ 6b oxybenzoic acid (I) Glucose B Rham 2 $7.87 d (8.2)$ 1 $5.39 d (7.7)$ 1 $7.87 d (8.2)$ 1 $5.39 d (7.7)$ 1 $7.61 d (8.6)$ 5 $4.10 m$ 5 $6.84m$ 6a $4.73 d (12.5)$ Me oxybenzoic acid (III) 4 $3.20-3.75$ 4 $7.51 d (8.6)$ 1 $5.13 d (6.7)$ 6 $6.84m$ $6a$ $4.73 d (12.5)$ Me oxybenzoic acid (III) $5.13 d (6$							

Table 1. ¹H NMR data for Purprocampanin of Campanura medium cv 'May Purple Margin'.(500 MHz in DMSO-d₆-CF₃CO₂D, TMS as an internal standard)

The	linkages	and/or	the	positions	of	attachment	of	glucoses	and	<i>p</i> -hydroxylbenzoic	acids	were
detei	mined by	the me	asure	ements of 2	NO	EDIF and N	OES	SY spectra	ı. By	irradiation at H-1 o	f Glc-A	A, the

appearance of a strong NOE signal at H-4 of cyanidin indicated Glc-A to be attached to OH-3 of cyanidin through a glucosidic bond. Moreover, a rather weak NOE signal was observed at H-1 of rhamnose as well as at those of other four proton signals of Glc-A, supporting that Glc-A bonded with rhamnose at OH-6. The presence of this binding system was confirmed by the detection of rutinose in the products of H_2O_2 degradation of pigment (2).⁵ By irradiation at H-1 of Glc-B, the strong NOE signals of H-6 and H-8 in cyanidin were observed. Thus, Glc-B was confirmed to attach to OH-7 of cyanidin. Furthermore, by irradiation of both H-1 protons of Glc-C and Glc-D, strong NOE signals were observed at H-3 and H-5 signals of p-hydroxybenzoic acid I, and also H-3 and H-5 signals of p-hydroxybenzoic acid II, supporting that p-hydroxybenzoic acid I is glycosylated with Glc-C at OH-4 of p-hydroxybenzoic acid I and also *p*-hydroxybenzoic acid II is glycosylated with Glc-D at OH-4 of *p*-hydroxybenzoic acid II (Figure 1). Similarly, rather weak NOEs were observed at H-2 and H-6 signals of *p*-hydroxybenzoic acid I by the irradiation at H-1 and H-6 of Glc-B indicating that Glc-B is acylated with p-hydroxybenzoic acid I at OH-6 of Glc-B. However, NOE signals were not observed at any proton signals of *p*-hydroxybenzoic acid III by above irradiations. Consequently, the structure of pigment (2), purprocampanin, was determined to be cyanidin $3-O-[6-O-(\alpha-L-rhamnopyranosyl)-\beta-D-glucopyranosyl]-7-O-{4-[6-O-(4-(6-O-(p-hydroxy$ benzoyl)- β -D-glucopyranosyl)oxybenzoyl)- β -D-glucopyranosyloxybenzoyl]- β -D-glucopyranoside, which is a new anthocyanin in plants.

Purprocampanin occurs mainly in the margin of *Campanula* 'May Purple Margin' flower petals, accompanying with rubrocampanin. The contents and distribution ratios of the pigments in the flowers are dependent on the flower stages and/or climate circumstances, and the flower color of C. 'May Purple Margin' is usually varied in the range of red to red-purple. Based on the above results, it is considered that increased distribution ratios of purprocampanin affect the change of the flower color of C. 'May Purple Margin' from red to purple.

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- 2. K. Brandt, T. Kondo, H. Aoki, and T. Goto, *Phytochemistry*, 1993, 33, 209.
- N. Saito, K. Toki, Y. Morita, A. Hoshino, S. Iida, A. Shigihara, and T. Honda, *Phytochemistry*, 2005, 66, 1852.
- 4. The flowers of *Campanula medium* cv 'May Purple Margin' were obtained from the plants growing in the green house of Minami-Kyushu University, Takanabe, Miyazaki.
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- 6. 4-Glucosyloxybenzoic acid: UV-VIS λ_{max} : 246nm: Rf values (TLC): *n*-butanol-acetic acid-water (4 :

1 : 5, BAW) 70%; 6% AcOH 76%; H₂O 91%; *n*-butanol-ethanol-water (4 : 1 : 2.2, BEW) 66%: HPLC: Rt (min) 5.4.

- 7. HPLC was run in the same condition as described previously.³
- Pelargonidin 3-rutinoside-7-glucoside; UV-VIS λ_{max} (0.1%HCl-MeOH) 268, 504nm: Rf values (TLC): BAW 27%; 1%HCl 51%; BuH(2N HCl-*n*-BuOH, 1 : 1) 17%; AcOH-HCl (H₂O-acetic acid-water, 15 : 3 : 82) 75%: HPLC: Rt (min) 10.08.
- Cyanidin 3-rutinoside-7-glucoside; UV-VIS λ_{max} (0.1%HCl-MeOH) 283, 526 nm: Rf values (TLC): BAW 17%; 1%HCl 29%; BuH 9%; AcOH-HCl 59%: HPLC: Rt(min) 7.58.
- Rubrocampanin: UV-VIS λ_{max} (0.1%HCl-MeOH) 518 nm, Eacyl/Evismax=163%: Rf values (TLC): BAW 29%; 1%HCl 24%; BuH 36%; AcOH-HCl 59%: HPLC: Rt (min) 24.2: High resolution FABMS calc. For C₆₆H₇₃O₃₅: 1425.3932 *m/z*. Found: 1425.3893. ¹H NMR (500MHz, CF₃CO₂D-DMSO-*d*₆, 1 : 9): pelargonidin; δ 8.76 (1H, s, H-4), 8.74 (2H, d, *J*=9.2 Hz, H-2', 4'), 7.51 (1H, m, H-8), 7.10 (2H, d, *J*=9.2 Hz, H-3', 5'), *p*-hydroxybenzoic acid (I); δ 7.86 (2H, d, *J*=8.9 Hz, H-2, 6), 7.18 (2H, d, *J*=8.9 Hz, H-3, 5), *p*-hydroxybenzoic acid (II); δ 7.62 (2H, d, *J*=8.9 Hz, H-2, 6), 6.86 (2H, d, *J*=8.9 Hz, H-3, 5), *p*-hydroxybenzoic acid (III); δ 7.61 (2H, d, *J*=8.9 Hz, H-2, 6), 6.63 (2H, d, *J*=8.9 Hz, H-3, 5), *s*ugars: Glc-A; δ 5.37 (1H, d, *J*=8.0 Hz, H-1), 4.00 (1H, d, *J*=10.4 Hz, H-6a), 3.69 (1H, m, H-5), 3.63 (1H, d, *J*=8.3 Hz, H-2), Glc-B; δ 5.37 (1H, d, *J*=8.0 Hz, H-1), 4.75 (1H, m, H-6a), 4.35 (1H, dd, *J*=7.4 Hz, 11.9 Hz, H-6b), 4.12 (1H, m, H-5), 3.55 (1H, m, H-2), Glc-C; 5.13 (1H, d, *J*=7.3 Hz, H-1), 4.62 (1H, d, *J*=12.2 Hz, H-6a), 4.27 (1H, m, H-6b), 3.87 (1H, t, *J*=9.5 Hz, H-5), 3.41 (1H, m, H-2), Glc-D; δ 5.00 (1H, d, *J*=8.0 Hz, H-1), 4.43 (1H, d, *J*=10.4 Hz, H-6a), 4.15 (1H, m, H-6b), 3.81 (1H, t, *J*=9.2 Hz, H-5), 3.39 (1H, m, H-2), Rhamnose; δ 4.69 (1H, s, H-1), 3.74 (1H, d, *J*=1.2 Hz, H-2), 3.42 (1H, m, H-5), 1.16 (3H, s, -CH3).
- Purprocampanin (pigment 2): UV-VIS λ_{max} (0.1%HCl-MeOH) 537 nm, Eacyl/Evismax=159%: Rf values (TLC): BAW 23%; 1%HCl 18%; BuH 26%; AcOH-HCl 50%: HPLC: Rt (min) 22.9: High resolution FABMS calc. For C₆₆H₇₃O₃₆; 1441.3882 *m/z*. Found: 1441.3906. ¹³C NMR (125.65MHz, CF₃CO₂D-DMSO-*d*₆, 1 : 9): cyanidin; δ 162.3 (C-2), 146.0 (C-3), 133.2 (C-4), 155.2 (C-5), 103.1 (C-6), 165.2 (C-7), 96.0 (C-8), 156.9 (C-9), 112.5 (C-10), 119.6 (C-1'), 118.2 (C-1'), 141.0 (C-3'), 156.0 (C-4'), 116.0 (C-5'), 128.4 (C-6'), *p*-hydroxybenzoic acid (I); δ 123.5 (C-1), 131.1 (C-2), 115.3 (C-3), 146.3 (C-4), 115.6 (C-5), 131.1(C-6), 161.1 (COOH), *p*-ηψδροξψβενζοιχ αχιδ (II); δ 123.3(C-1), 131.0 (C-2), 115.3 (C-3), 146.6 (C-4), 115.3 (C-5), 131.0 (C-6), 160.7(COOH), *p*-hydroxybenzoic acid (III); δ 123.1(C-1), 131.4 (C-2), 114.5 (C-3), 145.1 (C-4), 114.5 (C-5), 131.4 (C-6), 162.0 (COOH), Sugars, Glc-A; δ 102.3 (C-1), 66.6 (C-6); Glc-B; δ 99.9 (C-1), 74.3 (C-5), 64.2 (C-6), Glc-C; 99.5 (C-1), 64.0 (C-6); Glc-D; δ 99.0 (C-1), 73.7 (C-5), 64.0 (C-6).