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SULFUR-MEDIATED SYNTHESIS OF SUBSTITUTED TETRAHYDROFURANS: APPLICATION TO THE SYNTHESIS OF GONIOFUFURONE

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Abstract – An efficient strategy for the synthesis of densely functionalized 2,5-*trans*-disubstituted-tetrahydrofurans (4), from sulfinyldienols is described. Further transformations of this substrate have resulted in a new synthesis of goniofufurone (1) by means of silylated lactone (11), also readily available from D-xylose.

INTRODUCTION

Asian trees of the genus *Goniothalamus* (Annonaceae) have been recognized as a source of therapeutic agents in folk medicine. In 1972, Geran *et al.* found that the ethanolic extract of the stem bark of *Goniothalamus giganteus* Hook F. & Thomas showed cytotoxic activity during an in vivo antileukemic screening.¹ Subsequent bioactivity-directed studies by McLaughlin's group on the ethanolic extracts of these plants led to the discovery of novel styryllactones which were found to possess significant cytotoxicities against several human tumors.² Among these styryllactones, (+)-goniofufurone (1),^{2a} and its stereoisomer, (+)-7-*epi*-goniofufurone (2),^{2b} (Scheme 1) have attracted much attention from the biological and synthetic points of view due to their significant activity and selectivity toward several human tumor cell lines such as human lung carcinoma, and they have become interesting synthetic targets for a number of groups. Thus, Shing completed the first synthesis of (+)-goniofufurone (1) starting from D-glycero-D-glyco-heptono- γ -lactone.^{3a} Most of the subsequent synthesis relied on a Sharpless asymmetric epoxidation,^{4a} whereas Roberts used a Julia-Colonna asymmetric epoxidation,^{4b} and Hanaoka used a highly diastereoselective aldol reaction.⁵ In connection with our involvement in organosulfur

chemistry,⁶ we report herein substantial extensions of our methodologies to prepare substituted tetrahydrofurans from α -hydroxysulfinyldienes,⁷ that have resulted in a synthesis of (+)-goniofufurone (1) and (+)-7-*epi*-goniofufurone (2).



Scheme 1

RESULTS AND DISCUSSION

Our retrosynthetic analysis (Scheme 2) was designed with introduction of the aryl ring at the end of the sequence to facilitate the preparation of analogs at the aryl moiety for preliminary biological testing. Another key point was to develop a selective route to 2,5-*trans*-disubstituted-tetrahydrofurans functionalized at C-3 and C-4 to complement our prior efforts on this field.⁷ Thus, **1** could derive from aldehyde (**3**), similarly to a previous report,^{3j} and **3** should be available from tetrahydrofuran (**4**) through an unprecedented sulfonyloxirane cleavage with concurrent lactonization to render a ketolactone, reduction of the ketone and selective oxidation of the primary alcohol. Sulfonyltetrahydrofuran (**4**) could be obtained by stereoselective intramolecular Michael addition of oxiranylenoate (**5**), that could derive from α -hydroxysulfinyldiene (**6b**) by our metal-catalyzed oxidation/epoxidation to produce a vinyl oxirane, followed by oxidative-cleavage and a Wittig reaction. Diene (**6b**) could be prepared through a condensation of the corresponding aldehyde and metalated 1-sulfinylbutadiene, available in a single step by the protocol of Craig.⁸





To establish the viability and stereoselectivity of the sequence, it was decided to use initially racemic materials. Thus, lithiation of the racemic mixture of sulfinyldienes (7) (E/Z mixture),⁸ (Scheme 3), with concurrent isomerization to the *E* geometry,⁷ followed by reaction with

t-butyldiphenylsilyloxyacetaldehyde,⁹ yielded the readily separable mixture of diastereoisomers (**6a**) and (**6b**) (66:34) that could be interconverted by a Mitsunobu protocol.¹⁰ At this stage, the synthesis was pursued with the mixture and the treatment of the **6a/6b** mixture with *t*-BuOOH and VO(acac)₂ led to oxidation at sulfur and hydroxyl-directed epoxidation at the more electron deficient double bond with complete regio- and stereoselectivity.¹¹ Ozonolysis of the labile vinyloxirane intermediate gave lactol (**8**) that readily reacted with stabilized ylide Ph₃PCHCO₂*t*-Bu to give the desired *E*-enoate (**5**). Optimal conditions to trigger the desired intramolecular Michael addition of **5** entailed the use of Triton B to form 2,5-*trans*-tetrahydrofuran (**4**) as a single isomer.¹² Finally, we were pleased to find that the treatment of **4** with MgBr₂ gave lactone (**9**), with the fused furano-furone core of (+)-goniofufurone (**1**), through epoxysulfone cleavage and lactonization, thus validating our original hypothesis. To our knowledge this process is unprecedented, with related cyclizations in the literature using more nucleophilic alcohols, thus generating cyclic ethers, instead of lactones.¹³

Scheme 3. Reagents and conditions: *1*. (a) LDA, THF, -78 °C. (b) TBDPSOCH₂CHO, THF, -78 °C, 30 min, 76%. 2. (a) 5% VO(acac)₂, *t*-BuOOH, C₆H₆, rt, 5 h 30 min. (b) O₃, CH₂Cl₂-MeOH, 1:1, -78 °C, 15 min; then, SMe₂, -78 to 0 °C, 2 h, 42% two steps. *3*. Ph₃PCHCO₂*t*-Bu, C₆H₆, 0 °C to rt, 3 h, 30 min, 82%. *4*. Triton B, MeOH, rt, 16 h, 72%. *5*. MgBr₂, Et₂O, rt, 72%.

Reduction of ketone (9) under a variety of conditions, afforded the undesired epimer at the carbinol center as major product.¹⁴ After considerable experimentation, lactone opening to afford dimethylamide (10), followed by hydroxyl-directed ketone reduction¹⁵ and acid-catalyzed cyclization led to the desired alcohol (11) as single isomer (Scheme 4). Seeking additional structural confirmation for 11, furanolactone (+)-(12) was prepared by the methodology described by Galbis,^{3j,16} that entails the condensation of D-xylose and Meldrum's acid. Selective protection of the primary hydroxyl group as a *t*-butyldiphenylsilyl ether gave (+)-**11**, that had identical spectral features to the racemic material. It was then decided to continue the synthetic sequence with the optically pure material.

After considerable experimentation with several protecting groups for the secondary alcohol of **11** that then allowed for selective deprotection of the primary alcohol, protection of the secondary alcohol as a TBDPS ether was examined. Subsequent selective desilylation of the primary silyl group with HF-pyridine complex followed by Swern oxidation gave aldehyde (+)-(3). Treatment of this aldehyde with excess phenylmagnesium bromide led to the separable 95:5 mixture of diastereomeric alcohols (+)-(13) and (+)-(14) in moderate yield.¹⁷ From the major product (+)-(13), (+)-7-*epi*-goniofufurone (2), was easily obtained by deprotection of the hydroxyl group. On the other hand, (+)-goniofufurone (1) was synthesized via a three step sequence, oxidation of the 95:5 mixture followed by reduction and cleavage of the silyl protecting group. Synthetic **1** and **2** showed all spectroscopic data identical to those reported in the literature,^{2a,b} including melting points and the sign of the optical rotation.¹⁸

Scheme 4. Reagents and conditions: *1*. (a) Me₂NH, EtOH, rt, 2 h; then, Bu₄NBH₄, CH₂Cl₂, 0 °C to rt, 6 h 15 min, 53%. (b) PPTS, toluene, 80 °C, 3 h, 63%. *2*. (a) TBDPSCl, imidazole, DMAP, CH₂Cl₂, rt, 3 days, 86% and 14% of **11**. (b) HF·Pyr, THF, 0 °C, 3 h, 67% and 13% of starting material. (c) (COCl)₂, DMSO, Et₃N, CH₂Cl₂, -60 °C, 35 min, 79%. *3*. PhMgBr, THF, -20 to 0 °C, 2 h 45 min, 62%. *4*. HF·Py, THF, 0 °C, 3 h, 30 min, 78%. *5*. (a) PCC, 4 Å MS, CH₂Cl₂, rt, 70%. (b) LiAlH(O-*t*-Bu)₃, THF, -20 °C, 2 h, 80%, 56% of (+)-**14** and 24% of (+)-**13**. (c) HF·Py, THF, 0 °C, 4 h, 71%. *6*. (a) Meldrum's acid, *t*-BuNH₂, DMF, 40 °C, 5 days, 61%. *7*. TBDPSCl, imidazole, DMAP, CH₂Cl₂, rt, 24 h, 90%.

To summarize, we have developed a novel entry to 2,5-*trans*-disubstituted-tetrahydrofurans functionalized at C-3 and C-4. This methodology relies on our metal-catalyzed oxidation/epoxidation of hydroxysulfinyldienes, followed by a stereoselective Michael cyclization of sulfonyloxiranylenoates and an unprecedented sulfonyloxirane cleavage with concurrent lactonization. These protocols have been

applied to the preparation of goniofufurone (1) and 7-*epi*-goniofufurone (2) through intermediate (11) that is also readily available from D-xylose. Further applications of these methodologies, as well as the preparation of analogs of goniofufurone (1), are being examined in our laboratories.

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REFERENCES AND NOTES

- 1. R. I. Geran, N. H. Greenberg, M. M. McDonald, A. M. Schumacher, and B. J. Abbott, *Cancer Chemother. Rep.*, 1972, **3**, 1.
- (a) X.-P. Fang, J. E. Anderson, C.-J. Chang, P. E. Fanwick, and J. L. McLaughlin, *J. Chem. Soc., Perkin Trans. 1*, 1990, 1655. (b) X.-P. Fang, J. E. Anderson, C.-J. Chang, J. L. McLaughlin, and P. E. Fanwick, *J. Nat. Prod.*, 1991, **54**, 1034. (c) For a recent review on the cytotoxic activity of styryllactones and derivatives, see: H. B. Mereyala and M. Joe, *Curr. Med. Chem.; Anti-Cancer Agents*, 2001, **1**, 293.
- (a) T. K. M. Shing and H.-C. Tsui, J. Chem. Soc., Chem. Commun., 1992, 432. (b) T. K. M. Shing, H.-C. Tsui, and Z.-H. Zhou, J. Org. Chem., 1995, **60**, 3121. (c) T. Gracza and V. Jäger, Synlett, 1992, 191. (d) P. J. Murphy, J. Chem. Soc., Chem. Commun., 1992, 1096. (e) K. R. C. Prakash, and S. P. Rao, Tetrahedron, 1993, **49**, 1505. (f) M. Tsubuki, K. Kanai, and T. Honda, Synlett, 1993, 653. (g) J. Ye, R. K. Bhatt, and J. R. Falck, Tetrahedron Lett., 1993, **34**, 8007. (h) J. P. Surivet and J. M. Vatèle, Tetrahedron Lett., 1996, **37**, 4373. (i) X. H. Yi, Y. Meng, and C. J. Li, J. Chem. Soc., Chem. Commun., 1998, 449. (j) R. Bruns, A. Wernicke, and P. Köll, Tetrahedron, 1999, **55**, 9793. (k) Y.-L. Su, C-S. Yang, S-J. Teng, G. Zhao, and Y. Ding, Tetrahedron, 2001, **57**, 2147. (l) P. Ruiz, J. Murga, M. Carda, and J. A. Marco, J. Org. Chem., 2005, **70**, 713. (m) K. R. Prasad and S. L. Gholap, Synlett, 2005, 2260, erratum 3018.
- (a) Z. C. Yang and W. S. Zhou, J. Chem. Soc., Perkin Trans. 1, 1994, 3231. (b) W. P. Chen and S. M. Roberts, J. Chem. Soc., Perkin Trans. 1, 1999, 103.
- 5. C. Mukai, I. J. Kim, and M. Hanaoka, *Tetrahedron Lett.*, 1993, 34, 6081.
- 6. (a) R. Fernández de la Pradilla, S. Castro, P. Manzano, J. Priego, and A. Viso, *J. Org. Chem.*, 1996,
 61, 3586. (b) R. Fernández de la Pradilla, M. V. Buergo, P. Manzano, C. Montero, J. Priego, A. Viso,
 F. H. Cano, and M. P. Martínez-Alcázar, *J. Org. Chem.*, 2003, 68, 4797. (c) R. Fernández de la
 Pradilla, M. V. Martínez, C. Montero, and A. Viso, *Tetrahedron Lett.*, 1997, 38, 7773. (d) R.
 Fernández de la Pradilla, C. Montero, M. Tortosa, and A. Viso, *Chem. Eur. J.*, 2005, 11, 5136. (e) A.

Viso, R. Fernández de la Pradilla, C. Guerrero-Strachan, M. Alonso, M. Martínez-Ripoll, and I. André, *J. Org. Chem.*, 1997, **62**, 2316. (f) A. Viso, R. Fernández de la Pradilla, M. L. López-Rodríguez, A. García, A. Flores, and M. Alonso, *J. Org. Chem.*, 2004, **69**, 1542.

- (a) R. Fernández de la Pradilla, C. Montero, J. Priego, and L. A. Martínez-Cruz, *J. Org. Chem.*, 1998,
 63, 9612. (b) R. Fernández de la Pradilla, P. Manzano, C. Montero, J. Priego, M. Martínez-Ripoll, and L. A. Martínez-Cruz, *J. Org. Chem.*, 2003, 68, 7755. (c) R. Fernández de la Pradilla, C. Alhambra, A. Castellanos, J. Fernández, P. Manzano, C. Montero, M. Ureña, and A. Viso, *J. Org. Chem.*, 2005, 70, 10693.
- 8. D. Craig, K. Daniels, and A. R. MacKenzie, *Tetrahedron*, 1993, 49, 11263.
- W.-B. Choi, L. J. Wilson, S. Yeola, D. C. Liotta, and R. F. Schinazi, J. Am. Chem. Soc., 1991, 113, 9377.
- The stereochemistry of these dienols was established by comparison of their spectral and chromatographic features with those of closely related systems of known stereochemistry (see ref. 7). These and other related substrates may be interconverted readily (1. *p*-NO₂C₆H₄CO₂H, PPh₃, DIAD, THF, rt; 2. K₂CO₃, MeOH, rt, 65-80% two steps). For reviews on the Mitsunobu reaction, see: (a) D. L. Hughes, *Org. React.*, 1992, **42**, 335. (b) D. L. Hughes, *Org. Prep. Proced. Int.*, 1996, **28**, 127.
- (a) R. Fernández de la Pradilla, P. Méndez, J. Priego, and A. Viso, *J. Chem. Soc., Perkin Trans.1*, 1999, 1247. (b) R. Fernández de la Pradilla, A. Castellanos, J. Fernández, M. Lorenzo, P. Manzano, P. Méndez, J. Priego, and A. Viso, *J. Org. Chem.*, 2006, **71**, 1569.
- 12. The use of other ester substrates and/or other bases and solvents gave inferior results. The details will be reported elsewhere.
- 13. (a) H. Furuta, M. Hase, R. Noyori, and Y. Mori, *Org. Lett.*, 2005, 7, 4061. (b) H. Furuta, T. Takase, H. Hayashi, R. Noyori, and Y. Mori, *Tetrahedron*, 2003, 59, 9767 and previous papers by this group.
- A variety of reducing agents were tested (NaBH₄, NaBH₄•CeCl₃, Zn(BH₄)₂, DIBAL-H, L-selectride, Bu₄NBH₄). Complete details will be published in a full account of this work.
- 15. S. C. Gatling and J. E. Jackson, J. Am. Chem. Soc., 1999, 121, 8655.
- (a) F. Zamora Mata, M. Bueno Martínez, and J. A. Galbis Pérez, *Carbohydr. Res.*, 1990, 201, 223.
 (b) F. Zamora Mata, M. Bueno Martínez, and J. A. Galbis Pérez, *Carbohydr. Res.*, 1992, 225, 159.
- 17. The stereochemical outcome of this addition can be understood by considering the Felkin-Anh model and/or chelation control of the tetrahydrofuran oxygen; both models lead to the observed diastereoisomer in this case.
- 18. (1), mp 154-155 °C and $[\alpha]^{20}{}_{D} = +8.9$ ° (c = 0.235 EtOH), lit.,^{2a} mp 152-154 °C and $[\alpha]^{20}{}_{D} = +9.0$ ° (c = 0.5 EtOH). (2), mp 192-193 °C and $[\alpha]^{20}{}_{D} = +103.1$ ° (c = 0.32 EtOH), lit.,^{2b} mp 190-192 °C and $[\alpha]^{20}{}_{D} = +108$ ° (c = 0.2 EtOH).