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## REACTION OF 2*H*-CYCLOHEPTA[*b*]FURAN-2-ONES WITH PYRIDINIUM SALTS OF TRIFLUOROMETHANESULFONIC ANHYDRIDE\*

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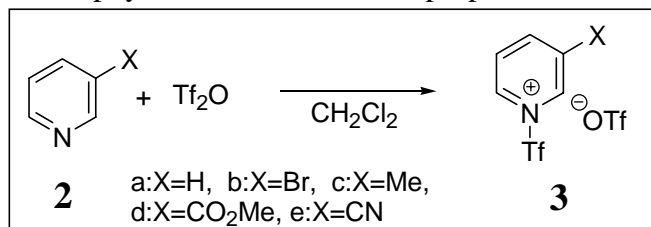
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**Abstract-** 2*H*-Cyclohepta[*b*]furan-2-one (**1**) reacted with TPT, electrophilically to give 3-dihydropyridinyl-2*H*-cyclohepta[*b*]furan-2-one and subsequent treatment with sodium hydroxide in ethanol gave 3-(4-pyridinyl)-2*H*-cyclohepta[*b*]furan-2-one in excellent yields. Furthermore, it reacted with methyl iodide give a corresponding pyridinium salt easily.

2*H*-Cyclohepta[*b*]furan-2-one (**1**)<sup>1</sup> and its derivatives have been well known as precursors of azulens.<sup>2</sup>

Their physical<sup>3</sup> and chemical properties<sup>4</sup> have received considerable attention due to their similar



acetylation, bromination and nitration.<sup>4</sup>

Recently, it was demonstrated that 2-oxo-2*H*-cyclohepta[*b*]furan-3-yl groups could stabilize some carbocations by extension of their  $\pi$ -electronic systems.<sup>5</sup> Although some chemical properties of **1** have been clarified,<sup>4,5</sup> it is not enough for the construction of new extended  $\pi$ -electronic systems. To develop a new binding procedure of two  $\pi$ -electronic systems is very important for functionalization and extension of  $\pi$ -electronic systems. We have reported the reactivities<sup>6</sup> of azulenes with trifluoromethanesulfonylpyridinium trifluoromethanesulfonate (TPT) (**3a**),<sup>7</sup> which is readily available from pyridine with trifluoromethanesulfonic anhydride (Tf<sub>2</sub>O) and has a high electrophilicity. We have also reported about our original synthetic way for **1** and its derivatives.<sup>8</sup> In continuation of these studies, we investigated the reactivities of **1** with trifluoromethanesulfonic anhydride in the presence of pyridine and its derivatives, consistently. Here we report a new type of cross-coupling reaction for introduction to pyridinyl group to

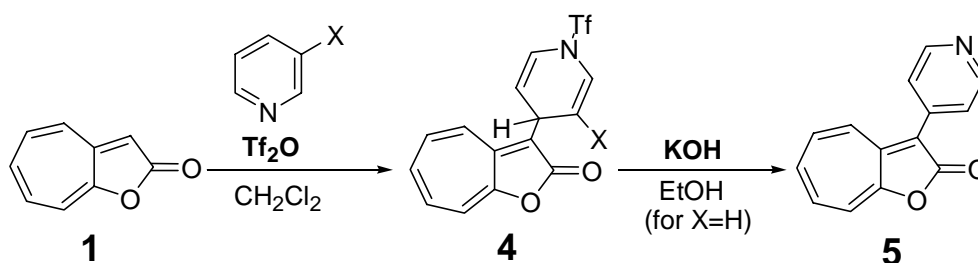
\* † Dedicated to Professor Satoshi Omura occasion of his 70<sup>th</sup> birthday.

3-position of **1** without transition-metal catalysis.

After the reaction conditions of **1** with **3a** were examined, we found a reaction product (**4a**) was obtained in good yields when the reaction was carried out as follows. A solution of **1** (1.521 g, 10.0 mmol) in dry dichloromethane (20 mL) was added to a solution of trifluoromethanesulfonic anhydride (5.645 g, 20.0 mmol) and dry pyridine (8.104 g, 101.8 mmol) in dry dichloromethane (10 mL) at rt in the period of 10 min under nitrogen atmosphere. It was stirring for 10 h, reaction mixture was passed through column packed with dry MgSO<sub>4</sub> and the solvent was removed under reduced pressure. The residue was separated by column chromatography on silica gel using the dichloromethane as an eluant to give a orange colored product **4a** (3.241 g, 9.1 mmol, 91% yield). The structure of **4a** was established as 3-(*N*-trifluoromethanesulfonyl-1,4-dihydro-4-pyridyl)-2*H*-cyclohepta[*b*]furan-2-one by following spectra. Mass spectrum exhibits molecular ion peak at *m/z* 357. The characteristic band for carbonyl group is observed at 1725(s) cm<sup>-1</sup> in the IR spectrum. Dihydropyridine ring protons appear at 6.64 (d, *J*=7.8 Hz, 2'-, 6'-H), 5.07 (dd, *J*=7.8, 3.5 Hz, 3'-5'-H), 4.45 (bs, 4'-H) in <sup>1</sup>H NMR spectrum. The ring protons of 2*H*-cyclohepta[*b*]furan-2-one appear at 7.42 (d, *J*=11.6 Hz, 4-H), 7.07-6.98 (m, 5-, 7-, and 8-H), 6.85 (t, *J*=9.8 Hz, 6-H). The reaction proceed regioselectively at 4-position of the pyridinium salt. We could not find a 2-position attack product.<sup>6</sup> Another 3-substituted pyridinium salts (**3b-e**) react similarly with **1** to give corresponding 2*H*-cyclohepta[*b*]furan-2-ones (**4b-e**) as shown in Table 1.

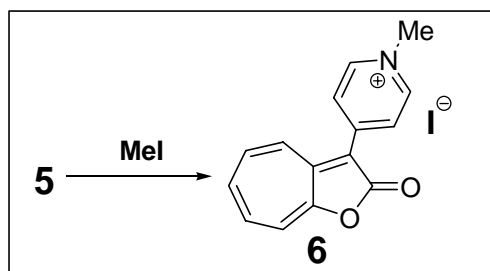
Table 1. Reaction of **1** with **2a-e** and trifluoromethanesulfonic anhydride

Entry	Substituted pyridines X	Yields of <b>4</b> (%)	mp (°C)
1	H	91	136-137
2	Br	40	134-135
3	CH <sub>3</sub>	49	109-110
4	CO <sub>2</sub> CH <sub>3</sub>	78	167 (decomp.)
5	CN	100	176 (decomp.)



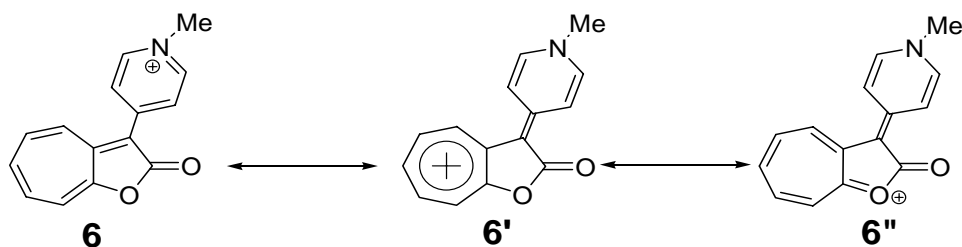
The dihydropyridine (**4a**) was treated with 3 equiv. of potassium hydroxide in ethanol to give 3-(4-pyridyl)-2*H*-cyclohepta[*b*]furan-2-one (**5**) in 64.3% yield as orange solid (mp 243.8-245.4 °C (decomp.)) through the dissociation of trifluoromethane sulfinic acid. The methine proton of **4a** at 4.45 ppm was disappeared in <sup>1</sup>H NMR of **5** and pyridyl ring protons of **5** appeared at δ 8.71 (dd, 2H, *J* = 4.4, 1.6 Hz) and 7.60 (dd, 2H, *J* = 4.4, 1.6 Hz). Another 2*H*-cyclohepta[*b*]furan-2-one ring protons appear at

7.71 (d, 1H,  $J = 11.6$  Hz, 4-H), 7.19 (dd, 1H,  $J = 11.6, 8.4$  Hz, 5-H), 7.15-7.11(m, 2H), 6.97 (dt, 1H,  $J = 8.4, 6.4$  Hz, 6-H). The  $\lambda_{\max}$  (log  $\epsilon$ ) of the longest wavelength of **5** is 404 (4.12) in  $\text{CH}_2\text{Cl}_2$ , which is shifted to longer wavelength due to the contribution of pyridine ring as compared with **1**. Reaction of **5** with methyl iodide gave a pyridinium salt (**6**) in 93% as orange solid (mp 272.0-272.5 °C (decomp.)), whose  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{CN}$ ) shows at  $\delta$  8.53 (d, 2H,  $J = 6.8$  Hz, 2', 6'-H), 8.36 (d, 2H,  $J = 6.8$  Hz, 3', 5'-H), 8.18 (d, 1H,  $J = 11.2$  Hz, 4-H), 7.74 (dd, 1H,  $J = 11.2, 9.2$  Hz, 5-H), 7.66-7.65 (m, 2H), 7.50-7.45



(m, 1H), and 4.22 (s, 3H). The UV-vis spectrum of **6** shows  $\lambda_{\max}$ , nm (log  $\epsilon$ ) at 236 (4.38), 251 (4.43), 324 (3.81) and 454 (4.44). To compare the positions of  $\lambda_{\max}$  of **1**, **4**, **5** and **6** in the longest wavelengths, their positions change to longer wavelength in order of  $1 < 4 < 5 < 6$ . On the basis of these results, the cationic charge of **6**

was delocalized to oxazalanone ring by the contribution of **6'** and **6''** structures.



In summary, a simple new procedure for the preparation of **5** and its salt (**6**) from **1** has been succeeded. Investigation into the scope and application of this methodology for the synthesis of 3-heteroaryl-2H-cyclohepta[b]furan-2-one is currently under way in our laboratories.

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