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**OXIDATIVE AROMATIZATION OF 3,5-DISUBSTITUTED ISOXAZOLINES TO THE CORRESPONDING ISOXAZOLES WITH *N,N,N',N'*-TETRABROMOBENZENE-1,3-DISOLPHONAMIDE (TBBDS) AND 1,3-DIBROMO-5,5-DIMETHYLHYDANTOIN (DBH) AS EFFICIENT REAGENTS UNDER MILD REACTION CONDITIONS**

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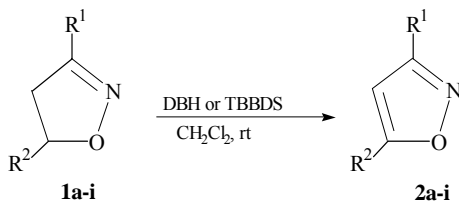
**Abstract-** The oxidation of 3,5-disubstituted isoxazolines to the corresponding isoxazoles has been carried out using *N,N,N',N'*-tetrabromobenzene-1,3-disolphonamide and 1,3-dibromo-5,5-dimethylhydantoin (DBH) as effective oxidizing agents under mild conditions at rt with good yields.

Among five-membered heterocycles, isoxazolines and their aromatic derivatives are biologically active natural products and synthetic compounds of medicinal interest.<sup>1</sup> The biological and medicinal importance of these heterocycles stem from their various chemotherapeutic properties.<sup>2</sup> Isoxazoles are also versatile intermediates in organic synthesis,<sup>3</sup> used in the synthesis of a wide variety of natural products and pharmacophores in medicinal chemistry.<sup>4</sup> Isoxazolines and isoxazoles are biologically active as fungicides, herbicides, antibacterial, ant-tubercular, antiviral and nicotinic acetylene receptor ligands.<sup>5,6</sup> Isoxazolines, easily prepared from the cyclization of related chalcones with hydroxylamine, can be conveniently oxidized to isoxazoles.<sup>7,8</sup> Oxidative aromatization of some five-membered heterocycles has been accomplished by various reagents including Pd/C/AcOH,<sup>9</sup> activated carbon-O<sub>2</sub> system,<sup>10</sup> cobalt soap of fatty acids,<sup>11</sup> Pb(OAc)<sub>4</sub>,<sup>12</sup> mercury or lead oxide,<sup>13</sup> MnO<sub>2</sub>,<sup>7</sup> potassium permanganate,<sup>14,15</sup> silver nitrate,<sup>16</sup> iodobenzene diacetate,<sup>17</sup> zirconium nitrate,<sup>18</sup> nickel peroxide,<sup>19</sup> cromite,<sup>20</sup> NBS,<sup>21</sup> Mn(OAc)<sub>3</sub>,<sup>22</sup> NaHCO<sub>3</sub>/DMF,<sup>23</sup> DDQ,<sup>24</sup> Na<sub>2</sub>CO<sub>3</sub>/CH<sub>3</sub>OH,<sup>25</sup> and tetrakispyridinenickel(II) dichromate.<sup>26</sup> However, most of these methods suffer from certain limitations such as long reaction times, high temperatures, side

product formation, and low yields. In addition, the presence of metal cations like Co(II), Pd(IV), Hg(II), Mn(IV or VII), Ag(I) and Zr(IV) added as catalysts, leave residual toxicity in the products. To tackle these drawbacks, we have previously reported some more convenient and easily available reagents for the oxidative aromatization of certain azolines including pyrazolines and isoxazolines.<sup>27-32</sup>

In continuation of our research to develop more robust and efficient oxidants, we were prompted to examine a simple, cheap and safe method for the oxidative aromatization of isoxazolines to isoxazoles. In this work, we have employed 1,3-dibromo-5,5-dimethylhydantoin (DBH) and *N,N,N',N'*-tetrabromobenzene-1,3-disulphonamide (TBBDS),<sup>33</sup> as efficient oxidizing agents to convert the isoxazolines **1a-i** to isoxazoles **2a-i** under mild conditions (Table 1).

**Table 1.** Oxidative aromatization of 3,5-disubstituted isoxazolines (**1a-i**) to the corresponding isoxazoles (**2a-i**) with DBH and TBBDS in CH<sub>2</sub>Cl<sub>2</sub> at rt

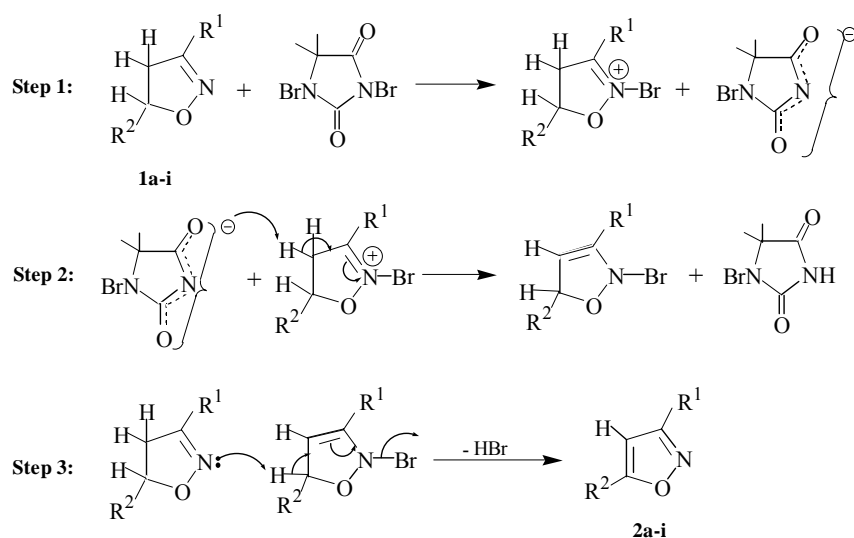


Substrate <sup>a</sup>	Product	R <sup>1</sup>	R <sup>2</sup>	Yield (%) <sup>b</sup>		Time (h)		Mp (°C)
				DBH	TBBDS	DBH	TBBDS	
<b>1a</b>	<b>2a</b>	2-naphthyl	2-ClC <sub>6</sub> H <sub>4</sub>	85	82	2.0	2.5	94-95
<b>1b</b>	<b>2b</b>	2-naphthyl	2-thienyl	80	68	2.2	2.8	118-120
<b>1c</b>	<b>2c</b>	2-naphthyl	4-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	76	73	2.4	3.0	107-109
<b>1d</b>	<b>2d</b>	1-naphthyl	2-ClC <sub>6</sub> H <sub>4</sub>	98	82	2.5	2.5	64-65
<b>1e</b>	<b>2e</b>	2-thienyl	4-ClC <sub>6</sub> H <sub>4</sub>	84	72	2.5	2.3	192-194
<b>1f</b>	<b>2f</b>	2-naphthyl	3-ClC <sub>6</sub> H <sub>4</sub>	70	65	2.0	3.2	90-92
<b>1g</b>	<b>2g</b>	2-furyl	4-ClC <sub>6</sub> H <sub>4</sub>	75	67	2.8	3.5	124-125
<b>1h</b>	<b>2h</b>	2-pyrrolyl	4-ClC <sub>6</sub> H <sub>4</sub>	92	85	2.3	2.8	185-187
<b>1i</b>	<b>2i</b>	2-naphthyl	2-styryl	94	88	2.0	3.0	165-167

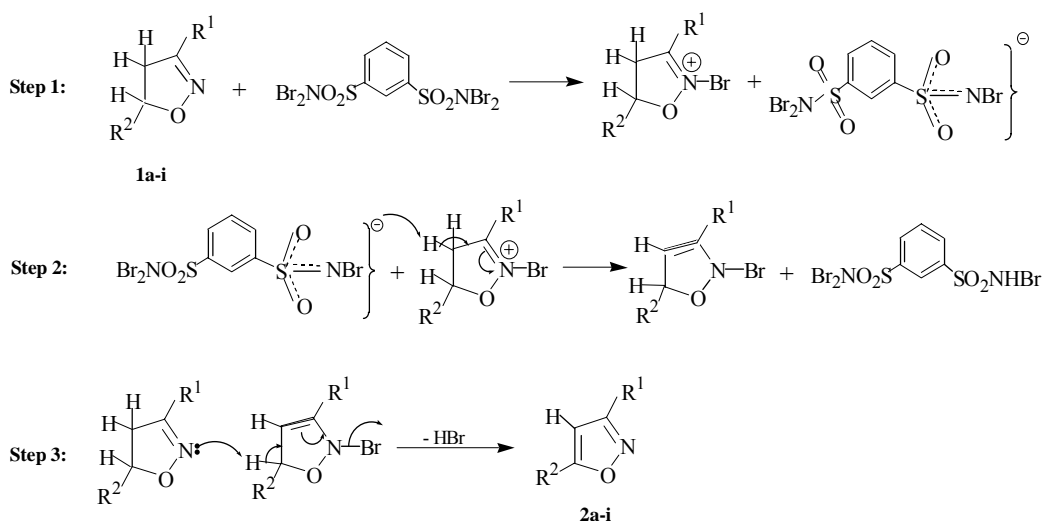
<sup>a</sup> Isoxazolines were prepared according to the literature<sup>8</sup> and characterized by their elemental analysis and IR, <sup>1</sup>H-NMR, <sup>13</sup>C-NMR spectral and physical data. <sup>b</sup> Isolated yields.

The experimental results indicate that the reactions proceed relatively rapidly within few hours at rt to afford the products **2a-i** in good yields (65-98%, Table 1). Numerous repetitions of the reactions under different molar conditions indicated that, the most effective conversions occur when the molar ratios of isoxazolines to DBH and TBBDS are 3 and 1 respectively. The possible mechanisms proposed for these transformations are presented in Schemes 1 and 2.

In conclusion, the advantages allocated to the DBH and TBBDS as oxidants include: low cost, easy accessibility, high stability, easy recovery of the reagents, mild reaction conditions and high yields of the products. These properties render them as efficient and convenient reagents to oxidize the isoxazolines to isoxazoles.



Scheme 1



Scheme 2

## EXPERIMENTAL

Chemicals were purchased from Fluka and Merck companies. IR spectra were recorded using a Shimadzu 435-U-04 spectrophotometer (KBr pellets) and NMR spectra were obtained using 90 MHz JEOL FT NMR spectrometer. Isoxazolines were all prepared as reported.<sup>8</sup> The reagent *N,N,N',N'*-tetrabromobenzene-1,3-disulfonamide (TBBDS) was prepared as reported method.<sup>33</sup>

### Oxidation of 3,5-Disubstituted Isoxazolines; *General procedure:*

To a solution of isoxazolines **2a-i** (0.1 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL), was added DBH (0.3 mmol) or TBBDS (0.1 mmol), and the mixture was vigorously stirred for 2-3.5 h at rt (Table 1). The progress of the reaction was monitored by TLC using EtOAc/n-hexane (1/4), and slow liberation of hydrogen bromide was observed during the reaction. After the complete conversion of the substrate, the reaction mixture was filtered to remove the solid materials. The filtrate was then treated with saturated Na<sub>2</sub>CO<sub>3</sub>, washed with

water, dried over anhydrous sodium sulphate and filtered. After removal of the solvent under reduced pressure, the remaining crude solid product was recrystallized from ethanol to give the pure isoxazoles (**2a-i**) in 65-98% yields (Table 1). The isoxazoles **2a-I** were all characterized on the basis of their microanalysis and IR,  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR spectral analysis as given below.

**5-(2-Chlorophenyl)-3-(2-naphthyl)isoxazole (2a):** Yield: 85%; mp 94-95 °C (recrystallization from EtOH). IR (KBr): 3059, 1603, 1572, 1474, 1440, 1340, 751  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (90 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm 7.13 (1H, s, C4-H isoxazole), 7.21-7.87 (10H, m, Ar-H), 8.25 (1H, s, C1-H naphth.).  $^{13}\text{C}$  NMR (22.5 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm 105.68 (C4 isoxazole), 123.88, 125.05, 125.58, 126.23, 126.80, 127.19, 127.74, 128.12, 129.12, 129.91, 130.61 ( $\text{CH}_{\text{arom}}$ ), 132.12, 133.24, 134.26, 134.77, 136.08 ( $\text{C}_{\text{arom}}$ ), 156.66 (C3 isoxazole), 165.80 (C5 isoxazole). Anal. Calcd for  $\text{C}_{19}\text{H}_{12}\text{NOCl}$ : C 74.63; H, 3.93; N, 4.58. Found: C 74.58; H, 3.90; N, 4.54.

**3-(2-Naphthyl)-5-(2-thienyl)isoxazole (2b):** Yield: 80%; mp 118-120 °C. IR (KBr): 3056, 1594, 1499, 1447, 1410, 792  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (90 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm 7.19 (1H, s, C4-H isoxazole), 7.32-7.95 (9H, m, Ar-H), 8.41 (1H, s, C1-H naphth.).  $^{13}\text{C}$  NMR (22.5 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm 98.76 (C4 isoxazole), 122.19, 123.75, 124.21, 125.70, 126.33, 126.99, 127.12, 127.42, 128.95, 129.98 ( $\text{CH}_{\text{arom}}$ ), 133.03, 134.10, 140.51, 142.20 ( $\text{C}_{\text{arom}}$ ), 156.51 (C3 isoxazole), 168.32 (C5 isoxazole). Anal. Calcd for  $\text{C}_{17}\text{H}_{11}\text{NOS}$ : C 73.65; H 3.97; N 5.05. Found: C 73.62; H, 3.95; N, 4.98.

**5-(Methoxyphenyl)-3-(2-naphthyl)isoxazole (2c):** Yield: 76%; mp 107-109 °C. IR (KBr): 3052, 2963, 1611, 1505, 1420, 1396, 803  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (90 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm 3.79 (3H, s,  $\text{OCH}_3$ ), 7.04 (1H, s, C4-H isoxazole), 7.20-7.87 (10H, m, Ar-H), 8.31 (1H, s, C1-H naphth.).  $^{13}\text{C}$  NMR (22.5 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm 54.97 ( $\text{OCH}_3$ ), 110.34 (C4 isoxazole), 120.87, 123.43, 124.57, 125.54, 126.23, 127.06, 128.43, 129.21, 130.67, 131.92 ( $\text{CH}_{\text{arom}}$ ), 133.21, 137.71, 138.92, 142.50, 142.85 ( $\text{C}_{\text{arom}}$ ), 150.28 (C3 isoxazole), 162.56 (C5 isoxazole). Anal. Calcd for  $\text{C}_{20}\text{H}_{15}\text{NO}_2$ : C 79.73; H, 4.98; N, 4.65. Found: C 79.68; H, 4.92; N, 4.63.

**5-(2-Chlorophenyl)-3-(1-naphthyl)isoxazole (2d):** Yield: 98%; mp 64-65 °C. IR (KBr): 3053, 1602, 1580, 1471, 1428, 1375, 774  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (90 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm 6.94 (1H, s, C4-H isoxazole), 7.28-8.46 (11H, m, Ar-H).  $^{13}\text{C}$  NMR (22.5 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm 110.32 (C4 isoxazole), 122.76, 123.43, 124.44, 125.89, 126.23, 126.95, 127.45, 128.32, 129.36, 129.99, 130.87 ( $\text{CH}_{\text{arom}}$ ), 131.14, 133.65, 134.56, 139.67, 143.12 ( $\text{C}_{\text{arom}}$ ), 150.09 (C3 isoxazole), 160.32 (C5 isoxazole). Anal. Calcd for  $\text{C}_{19}\text{H}_{12}\text{NOCl}$ : C 74.63; H, 3.93; N, 4.58. Found: C 74.62; H, 3.85; N, 4.56.

**5-(4-Chlorophenyl)-3-(2-thienyl)isoxazole (2e):** Yield: 84%; mp 192-194 °C. IR (KBr): 3087, 1607, 1570, 1488, 1430, 1391, 781  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR (90 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm 6.59 (1H, s, C4-H isoxazole), 7.11-7.70 (7H, m, Ar-H).  $^{13}\text{C}$  NMR (22.5 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm 103.11 (C4 isoxazole), 126.43, 127.14, 127.98, 128.61, 129.45, 130.17, 131.27 ( $\text{CH}_{\text{arom}}$ ), 132.65, 133.69, 138.88 ( $\text{C}_{\text{arom}}$ ), 151.81 (C3 isoxazole), 164.46

(C5 isoxazole). Anal. Calcd for  $C_{13}H_8NOCl$ : C 59.65; H, 3.06; N, 5.35. Found: C 59.48; H, 3.10; N, 5.25.

**5-(3-Chlorophenyl)-3-(2-naphthyl)isoxazole (2f):** Yield: 70%; mp 90-92 °C. IR (KBr): 3113, 3062, 1585, 1562, 1483, 1448, 1408, 788  $cm^{-1}$ .  $^1H$  NMR (90 MHz,  $CDCl_3$ )  $\delta$  ppm 6.85 (1H, s, C4-H isoxazole), 7.17-7.85 (10H, m, Ar-H), 8.17 (1H, s, C1-H naphth.).  $^{13}C$  NMR (22.5 MHz,  $CDCl_3$ )  $\delta$  ppm 108.96 (C4 isoxazole), 123.47, 123.95, 125.97, 126.79, 127.02, 127.15, 128.31, 128.53, 129.38, 130.06, 131.52 ( $CH_{arom}$ ), 132.89, 134.61, 135.76, 137.62, 138.43 ( $C_{arom}$ ), 156.17 (C3 isoxazole), 159.68 (C5 isoxazole). Anal. Calcd for  $C_{19}H_{12}NOCl$ : C 74.63; H, 3.93; N, 4.58. Found: C 74.61; H, 3.90; N, 4.45.

**5-(4-Chlorophenyl)-3-(2-furyl)isoxazole (2g):** Yield: 75%; mp 124-125 °C. IR (KBr): 3131, 1663, 1594, 1486, 1460, 1396, 774  $cm^{-1}$ .  $^1H$  NMR (90 MHz,  $CDCl_3$ )  $\delta$  ppm 5.96 (1H, s, C4-H isoxazole), 7.24-7.71 (7H, m, Ar-H).  $^{13}C$  NMR (22.5 MHz,  $CDCl_3$ )  $\delta$  ppm 110.12 (C4 isoxazole), 112.57, 113.45, 126.67, 127.11, 129.85, 131.74, 132.71 ( $CH_{arom}$ ), 136.98, 138.70, 144.24 ( $C_{arom}$ ), 148.17 (C3 isoxazole), 169.11 (C5 isoxazole). Anal. Calcd for  $C_{13}H_8NO_2Cl$ : C 63.54; H, 3.26; N, 5.70. Found: C 63.48; H, 3.22; N, 5.68.

**5-(4-Chlorophenyl)-3-(2-pyrrolyl)isoxazole (2h):** Yield: 92%; mp 185-187 °C. IR (KBr): 3259, 3049, 1602, 1494, 1446, 1350, 756  $cm^{-1}$ .  $^1H$  NMR (90 MHz,  $CDCl_3$ )  $\delta$  ppm 7.06 (1H, s, C4-H isoxazole), 7.21-8.01 (7H, m, Ar-H), 9.87 (1H, s br, N-H).  $^{13}C$  NMR (22.5 MHz,  $CDCl_3$ )  $\delta$  ppm 103.23 (C4 isoxazole), 109.66, 112.82, 126.96, 127.38, 128.99, 130.67, 131.83 ( $CH_{arom}$ ), 134.42, 136.78, 139.37 ( $C_{arom}$ ), 151.81 (C3 isoxazole), 167.41 (C5 isoxazole). Anal. Calcd for  $C_{13}H_9N_2OCl$ : C 63.80; H, 3.68; N, 11.45. Found: C 63.75; H, 3.65; N, 11.38.

**3-(2-Naphthyl)-5-(2-styryl)isoxazole (2i):** Yield: 94%; mp 165-167 °C. IR (KBr): 3052, 1636, 1599, 1542, 1492, 1477, 1363, 714  $cm^{-1}$ .  $^1H$  NMR (90 MHz,  $CDCl_3$ )  $\delta$  ppm 6.96 (1H, s, C4-H isoxazole), 7.12-7.92 (13H, m, Ar-H and vinyl), 8.35 (1H, s, C1-H naphth.).  $^{13}C$  NMR (22.5 MHz,  $CDCl_3$ )  $\delta$  ppm 104.32 (C4 isoxazole), 123.37, 124.21, 126.29, 126.55, 127.12, 128.94, 129.32, 131.28, 131.74, 132.38, 133.06, 133.67, 134.74, 135.31 ( $CH_{arom}$ ), 136.95, 141.51, 143.30, 144.20 ( $C_{arom}$  and vinyl), 156.48 (C3 isoxazole), 162.34 (C5 isoxazole). Anal. Calcd for  $C_{21}H_{15}NO$ : C 84.85; H, 5.05; N, 4.71. Found: C 84.78; H, 5.03; N, 4.68.

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