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SYNTHESIS OF ENANTIOMERICALLY PURE (+)- AND (-)-3-METHYL-3-(2-NAPHTHYL)-2*H*-BENZO[*e*][1,2]THIAZINE 1,1,4-TRIONES

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Abstract – Treatment of *N*-*t*-butylbenzenesulfonamide with an excess of BuLi, followed by the reaction with methyl 2-naphthylpropanoate, gave the corresponding 2-carboxybenzenesulfonamide, which underwent cyclization under TMSCI-NaI-MeCN reflux conditions to afford the *N*-sulfonylimine. Bromination of the *N*-sulfonylimine and ring expansion mediated by 20% KOH (aq.), formed the 3-methyl-3-(2-naphthyl)benzo[*e*][1,2]thiazine 1,1,4-trione. Optical resolution of the racemic benzosultam using (–)-menthoxyacetyl chloride, furnished the (+)-and (–)-3-methyl-3-(2-naphthyl)benzo[*e*][1,2]thiazine 1,1,4-triones.

1. INTRODUCTION

1,2-Benzisothiazoline-3-one 1,1-dioxide (1), known as saccharin, is a cheap and versatile starting material for the synthesis of related heterocyclic compounds.¹ *N*-Alkylated saccharin derivatives act as agonists of 5-HT_{1A} receptors² and have therefore found applications as neuroprotectants³ or anxiolytics (e.g., Ipsapirone 2).⁴ They also selectively inhibit human leukocyte elastase⁵ or human mast cell tryptase.⁶ The oxicam ring system 3, which can be regarded as the enol form of one carbon-extended homologue of saccharin derivatives, is a common substructure of certain pharmaceutically important antiinflammatory agents.⁷ Oxicams can conveniently be derived from 1,2-benzisothiazoline-3-one 1,1-dioxide (saccharin) (1) system by ring expansion.⁸⁻¹⁰ Surprisingly, 3,3-disubstituted-2*H*-benzo[*e*][1,2]thiazine 1,1,4-triones, the one carbon-extended homologue of saccharin derivatives, have remained mostly unexplored both biologically and chemically. There are few reports on the synthesis of *N*-alkylated 3,3-disubstituted-2*H*-benzo[*e*][1,2]thiazine 1,1,4-triones,¹¹⁻¹² but the difficulty in removing the alkyl protective group limits their further derivatization. To our best knowledge, the literature gives three references¹³⁻¹⁴ to the preparation of 3,3-disubstituted-2*H*-benzo[*e*][1,2]thiazine 1,1,4-triones, and the only

(4)¹³⁻¹⁴ 3-ethyl-3-methyl-2*H*-benzo[*e*][1,2]thiazine 1.1.4-trione known and compounds are 6-methoxy-3,3-dimethyl-2*H*-benzo[*e*][1,2]thiazine 1,1,4-trione.¹⁵ We proved previously that introducing a carbonyl functional group in the 4-position made the N-fluorobenzosultam (5) more reactive for the electrophilic fluorination of carbanions.¹⁶ However, the little difference in steric bulkiness of ethyl and methyl group in sultam 4 made it unsuitable as a template for the developing of chiral electrophilic fluorinating agent. Therefore, it is of interest to synthesis chiral 2H-benzo[e][1,2]thiazine 1,1,4-triones having a bulky substituent and a small one at the 3-position. Also considering that Oppolzer's saccharin derived benzosultams are useful chiral auxiliaries in asymmetric organic reactions,¹ plus the importance of chirality in pharmacologically and biologically active compounds, we report here the first synthesis of enantiomerically pure 3-methyl-3-(2-naphthyl)benzo[*e*][1,2]thiazine 1,1,4-triones.





2. RESULTS AND DISCUSSION

The sulfonamide functionality is regarded as a powerful Directed Metalation Group (DMG) because of its ability to direct metalation to specific positions on the aryl ring. The resulting lithiated species also couple excellently with a variety of electrophiles.¹⁷ The carboxylation of *o*-lithiobenzenesulfonamides is a convenient method for preparing 2-carboxybenzenesulfonamides that function as the standard intermediates for the synthesis of saccharin-related heterocycles.¹⁸ By adopting a similar strategy, *N-t*-butylbenzenesulfonamide was treated with an excess of BuLi (2.5 eq) in THF solution at 0 °C, ortho-lithiation occurred smoothly with the solution changing to deep red. The formed anion reacted with methyl 2-(2-naphhtyl)propanoate to give the corresponding 2-carboxybenzenesulfonamide 6 in 65% yield. The application of our TMSCI-NaI-MeCN system for deprotective-cyclization¹⁹⁻²¹ of the sulfonamide **6** under reflux conditions, successfully produced 3-[2-(2-naphtyl)ethyl]benzo[d]isothiazole 1,1-dione (7)in 81% vield (Scheme 1). The *N*-sulfonylimine 7 was converted to the racemic 3,3-disubstituted-2*H*-benzo[e][1,2]thiazine 1,1,4-trione according to the Abramovitch's procedures.¹³⁻¹⁴ Bromination of the N-sulforylimine 7 in benzene gave the desired product 8 quantitatively, which 20% 3-methyl-3underwent ring expansion mediated by KOH to form (aq.) (2-naphthyl)-2*H*-benzo[*e*][1,2]thiazine 1,1,4-trione **9** in excellent yield.





Optical resolution of sultam 9 was carried out by derivatization with (–)-menthoxyacetyl chloride followed by the separation of the two diastereomers (+)-10a (more polar) and (–)-10b (less polar) using column chromatography on silica gel. Removal of the chiral auxiliaries of (+)-10a and (–)-10b was achieved smoothly with LiOH in aqueous THF to furnish (+)-9 and (–)-9 respectively in an optically pure state (Scheme 2). Unfortunately, due to the poor crystal properties of compounds (+)-10a, (–)-10b and (+)-9, (–)-9, the absolute stereochemistry of these compounds could not be determined by X-ray crystallography.



In conclusion, a convenient method has been developed for the synthesis of (+)- and (-)-3-methyl-3-(2-naphthyl)-2*H*-benzo[*e*][1,2]thiazine 1,1,4-triones. It is the first report on the synthesis of chiral 3,3-disubstituted-2*H*-benzo[*e*][1,2]thiazine 1,1,4-triones. Optically pure benzosultams (+)-9 and (-)-9 would be useful chiral auxiliaries for asymmetric synthesis.

EXPERIMENTAL

Melting points were determined on an X-6 micro-melting point apparatus (Beijing Tech. Co., Ltd) and are uncorrected. IR spectra (cm⁻¹) were recorded on a Perkin-Elmer 1600 spectrometer. ¹H NMR (300 MHz) spectra were recorded at rt for CDCl₃ solutions. All chemical shifts were reported as δ values (ppm) relative to Me₄Si (0.00 ppm) as internal standards for ¹H spectra. Mass spectra and high-resolution spectra were recorded on a JEOL JMS D-300 mass spectrometer. Optical rotations were determined by using a Jasco DIP-370 digital polarimeter. Microanalyses were performed with a YANAKO CHN-coder MT-5. Column chromatography was performed on silica gel (200-300 mesh). All reactions involving oxygen- or moisture-sensitive compounds were carried out under a dry N₂ atmosphere. Unless otherwise noted, reagents were added by syringe. THF was distilled from sodium/benzophenone immediately prior to use.

N-t-Butyl-2-(2-naphthalen-2-yl-propionyl)benzenesulfonamide (6). To a stirred solution of *N-t*-butylbenzenesulfonamide (5.8 g, 27.2 mmol) in THF (150 mL) was slowly added a 1.60 M solution of BuLi in hexane (42.5 mL, 68 mmol) under nitrogen at 0 °C. The reaction mixture was stirred at 0 °C for an additional 30 min. A solution of methyl 2-naphthylpropanoate (6.11 g, 28.5 mmol) in 20 mL THF was added. After 2 h, saturated aqueous NH₄Cl was added. The mixture was extracted with EtOAc, the combined organic layers were washed with brine, dried (MgSO₄) and concentrated in vacuo. The residue was chromatographed (30% EtOAc in hexane) to give compound **6** (6.97 g, 65%) as a white solid: mp 141 °C (EtOAc/hexane); IR (KBr) 3309, 1693 cm⁻¹; ¹H NMR δ 1.26 (s, 9H), 1.66 (d, *J* = 6.9 Hz, 3H), 4.55 (q, *J* = 6.9 Hz, 1H), 5.24 (s, 1H), 7.02 (d, *J* = 7.6 Hz, 1H), 7.32 (t, *J* = 7.6 Hz, 1H), 7.42–7.54 (m, 4H), 7.75 (s, 1H), 7.79–7.90 (m, 3H), 8.02 (d, *J* = 7.9 Hz, 1H); MS *m*/z 395 (M⁺); HRMS calcd for C₂₃H₂₅NO₃S 395.1556, found 395.1566. Anal. Calcd for C₂₃H₂₅NO₃S: C, 69.84; H, 6.37; N, 3.54. Found: C, 69.71; H, 6.31; N, 3.47.

3-(2-Naphthylethyl)benzo[*d*]isothiazole 1,1-dione (7). To a stirred solution of **6** (6.97 g, 17.6 mmol) in MeCN (80 mL) was added under nitrogen sodium iodide (5.28 g, 35.2 mmol) and chlorotrimethylsilane (4.50 mL, 35.2 mmol). The reaction mixture was refluxed for 1 h. It was cooled to rt and 10% sodium thiosulfate aqueous solution was added. The mixture was extracted with EtOAc, and the combined organic layers were washed with water, brine, dried (MgSO₄) and concentrated in vacuo. Crystallization from hexane gave *N*-sulfonylimine **7** (4.56 g, 81%) as colorless prisms: mp 156 °C (EtOAc/hexane); IR (KBr) 3021, 1314, 1217, 1174 cm⁻¹; ¹H NMR δ 1.81 (d, *J* = 6.9 Hz, 3H), 4.61 (q, *J* = 6.6 Hz, 1H), 7.42–7.50 (m, 5H), 7.58–7.61 (m, 1H), 7.79–7.88 (m, 5H); MS *m*/z 321 (M⁺); HRMS calcd for C₁₉H₁₅NO₂S 321.0824, found 321.0813. Anal. Calcd for C₁₉H₁₅NO₂S: C, 71.00; H, 4.70; N, 4.36. Found: C, 70.72; H, 4.64; N, 4.15.

3-[1-Bromo-1-(2-naphthyl)ethyl]benzo[*d*]isothiazole 1,1-dione (8). To a stirred solution of **7** (2.94 g, 9.16 mmol) in benzene (60 mL) was added dropwise a solution of bromine (0.47mL, 9.16 mmol) in benzene (10 mL) at rt. The reaction mixture was stirred for 1 h. 10% sodium sulfite aqueous solution was added. The mixture was extracted with EtOAc, the combined organic layers were washed with brine, dried (MgSO₄) and concentrated in vacuo. The residue was chromatographed (30% EtOAc in hexane) to give 8 (3.6 g, 98%) as a white solid: mp 65 °C (EtOAc/hexane); IR (KBr) 3023, 1346, 1275, 1178 cm⁻¹; ¹H NMR δ 2.51 (s, 3H), 7.21 (d, *J* = 7.9 Hz, 1H), 7.34 (t, *J* = 7.6 Hz, 1H), 7.49–7.61 (m, 4H), 7.81–7.92 (m, 4H), 8.04 (br s, 1H); MS *m*/z 320 (M⁺–HBr). Anal. Calcd for C₁₉H₁₄NO₂SBr: C, 57.01; H, 3.53; N, 3.50. Found: C, 57.02; H, 3.67; N, 3.40.

3-Methyl-3-(2-naphthyl)-2*H***-benzo[***e***][1,2]thiazine 1,1,4-trione (9). To a stirred solution of 8** (3.26 g, 8.15 mmol) in THF (40 mL) was added 20% KOH aqueous solution (4.6 mL). The reaction mixture was refluxed for 1 h. After cooling to rt, 10% HCl (aq.) was added. The mixture was extracted with EtOAc, the combined organic layers were washed with brine, dried (MgSO₄) and concentrated in vacuo. The residue was chromatographed (30% EtOAc in hexane) to give sultam **9** (2.73 g, 99%) as colorless glasses: mp 55 °C (EtOAc/hexane); IR (KBr) 3264, 1694 cm⁻¹; ¹H NMR δ 2.11 (s, 3H), 5.54 (s, 1H), 7.46–7.50 (m, 2H), 7.58 (dd, *J* = 6.9, 1.5 Hz, 1H), 7.76–7.88 (m, 7H), 8.16 (dd, *J* = 6.9, 1.5 Hz, 1H); MS *m/z* 337 (M⁺); HRMS calcd for C₁₉H₁₅NO₂S 337.0773, found 337.0768. Anal. Calcd for C₁₉H₁₅NO₃S: C, 67.64; H, 4.48; N, 4.15. Found: C, 67.60; H, 4.57; N, 3.91.

N-Menthoxyacetyl-3-methyl-3-(2-naphthyl)-2*H*-benzo[*e*][1,2]thiazine 1,1,4-trione (10). A solution of 9 (2.82 g, 8.36 mmol) in THF (40 mL) was treated with NaH (60% oil dispersion, 0.5 g, 12.5 mmol) at 0 °C and stirred at rt for 1.5 h. (–)-Menthoxyacetyl chloride (2.34 g, 10 mmol) in THF (5 mL) was added to the mixture and stirred for overnight. Cold saturated aqueous NH₄Cl was added, and the resulting mixture was extracted with EtOAc. The combined organic layer was washed with water, brine, dried (MgSO₄), and concentrated in vacuo. The residue was chromatographed (25% CH₂Cl₂ in hexane) to give the less polar isomer (–)-10b (2.04 g, 46%), the more polar isomer (+)-10a (1.79 g, 40%) as a white powder.

(-)-*N*-Menthoxyacetyl-3-methyl-3-(2-naphthyl)-2*H*-benzo[*e*][1,2]thiazine 1,1,4-trione [(-)-10b]: colorless prisms; mp 160 °C; $[\alpha]^{26}{}_{D}$ –69.3 (*c* 0.88, CHCl₃); IR (KBr) 1710 cm⁻¹; ¹H NMR δ 0.69 (d, *J* = 6.9 Hz, 3H), 0.80 (d, *J* = 10.2 Hz, 3H), 0.82 (d, *J* = 9.9 Hz, 1H), 0.77–0.98 (m, 2H), 1.18–1.23 (m, 2H), 1.51–1.60 (m, 3H), 1.92–1.96 (m, 1H), 2.09–2.12 (m, 1H), 2.58 (s, 3H), 3.06 (td, *J* = 10.5, 3.9 Hz, 1H), 4.46, 4.66 (ABq, *J* = 16.1, 2H), 7.05 (dd, *J* = 6.9, 1.9 Hz, 1H), 7.40–7.45 (m, 2H), 7.56–7.73 (m, 6H), 7.83 (t, *J* = 7.6 Hz, 1H), 8.02 (d, *J* = 7.3 Hz, 1H); MS *m*/z 533 (M⁺); HRMS calcd for C₃₁H₃₅NO₅S

533.2236, found 533.2242. Anal. Calcd for C₃₁H₃₅NO₅S: C, 69.77; H, 6.61; N, 2.62. Found: C, 69.87; H, 6.53; N, 2.59.

(+)-*N*-Menthoxyacetyl-3-methyl-3-(2-naphthyl)-2*H*-benzo[*e*][1,2]thiazine 1,1,4-trione [(+)-10a]: mp 59-61 °C; $[\alpha]^{27}_{D}$ +3.6 (*c* 1.46, CHCl₃); IR (KBr) 1711 cm⁻¹; ¹H NMR δ 0.51 (d, *J* = 7.0 Hz, 3H), 0.77 (d, *J* = 7.0 Hz, 3H), 0.88 (d, *J* = 6.6 Hz, 1H), 0.76–0.89 (m, 3H), 1.18–1.22 (m, 1H), 1.50–1.60 (m, 3H), 1.95–1.99 (m, 1H), 2.12–2.15 (m, 1H), 2.57 (s, 3H), 2.99 (td, *J* = 10.5, 3.9 Hz, 1H), 4.55, 4.66 (ABq, *J* = 16.8, 2H), 7.07 (dd, *J* = 7.0, 1.9 Hz, 1H), 7.40–7.44 (m, 2H), 7.58–7.74 (m, 6H), 7.84 (t, *J* = 7.6 Hz, 1H), 8.04 (d, *J* = 7.6 Hz, 1H); MS *m*/*z* 533 (M⁺); HRMS calcd for C₃₁H₃₅NO₅S 533.2236, found 533.2227. Anal. Calcd for C₃₁H₃₅NO₅S: C, 69.77; H, 6.61; N, 2.62. Found: C, 70.06; H, 6.64; N, 2.52.

(+)-3-Methyl-3-(2-naphthyl)-2*H*-benzo[*e*][1,2]thiazine 1,1,4-trione [(+)-9]. A solution of (–)-10b (2.31 g, 4.33 mmol) in THF (20 mL) was treated with 2 M LiOH (11 mL) at rt and stirred for 2 h. The mixture was poured into 10% HCl aqueous solution and extracted with EtOAc. The combined organic layer was washed with water, brine, dried (MgSO₄), and concentrated in vacuo. The residue was chromatographed (30% EtOAc in hexane) to give (+)-9 (1.45 g, 99%) as colorless glasses: mp 65–67 °C; $[\alpha]^{27}_{D}$ +42.1 (*c* 0.84, CHCl₃). Anal. Calcd for C₁₉H₁₅NO₃S: C, 67.64; H, 4.48; N, 4.15. Found: C, 67.44; H, 4.55; N, 3.96. The optical pure (+)-9 exhibited the same spectral properties (IR, MS, ¹H NMR) as the racemic 9.

(-)-3-Methyl-3-(2-naphthyl)-2*H*-benzo[*e*][1,2]thiazine 1,1,4-trione [(-)-9]. In the same way, 1.31 g (4.33 mmol) of (+)-10a was treated with 2 M LiOH aqueous solution (11 mL) to give (-)-9 (1.39 g, 95%) as colorless glasses: $[\alpha]_D^{26}$ -42.8 (*c* 1.04, CHCl₃). The optical pure (-)-9 exhibited the same spectral properties (IR, MS, ¹H NMR) as (+)-9.

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REFERENCES

- 1. See recent review: Z.-P. Liu and Y. Takeuchi, *Heterocycles* 2002, 56, 693.
- G. Caliendo, F. Fiorino, E. Perissutti, B. Severino, S. Gessi, E. Cattabriga, P. A. Borea, and V. Santagada, *Eur. J. Med. Chem.*, 2001, 36, 873.
- 3. Drugs Future, 1997, 22, 341.
- 4. Drugs Future, 1986, **11**, 565.
- K. D. Combrink, H. B. Gulgeze, N. A. Meanwell, B. C. Pearce, P. Zulan, G. S. Bisacchi, D. G. Roberts, P. Stanley, and S. M. Seiler, *J. Med. Chem.*, 1998, 41, 4854.

- D. J. Hlasta, C. Subramanyam, M. R. Bell, P. M. Carabateas, J. J. Court, R. C. Desai, M. L. Drozd, W. M. Eickhoff, E. W. Ferguson, R. J. Gordon, J. A. Johnson, V. Kumar, A. L. Maycock, K. R. Mueller, E. D. Pagani, D. T. Robinson, M. T. Saindane, P. J. Silver, and S. Subramanian, *J. Med. Chem.*, 1995, **38**, 739.
- 7. J. G. Lombardino and E. H. Wiseman, Med. Res. Rev., 1982, 2, 127.
- 8. P. Catsoulacos and C. Camoutsis, J. Heterocycl. Chem., 1979, 16, 1503.
- 9. J. G. Lombardino and D. E. Kuhla, Adv. Heterocycl. Chem., 1982, 28, 73.
- 10. P. D. Weeks, F. J. Vinick, R. Breitenbach, and S. Jung, J. Org. Chem., 1983, 48, 3601.
- 11. D. C. Piero, D. M. Demetrio, and L. R. Concetta, J. Chem. Soc., Perkin Trans. 1, 1986, 299.
- 12. H. Zinnes and J. Shavel, Jr., J. Heterocycl. Chem., 1973, 10, 956.
- 13. R. A. Abramovitch, E. M. Smith, M. Humber, B. Purtschert, P. C. Srinivasan, and G. M. Singer, J. *Chem. Soc., Perkin Trans. 1*, 1974, 2589.
- 14. R. A. Abramovitch, K. M. More, I. Shinkai, and P. C. Srinivasan, J. Chem. Soc., Chem. Commun., 1976, 771.
- 15. W. I. Bakker, O. B. Familoni, J. Padfield, and V. Snieckus, Synlett, 1997, 1019.
- 16. Y. Takeuchi, Z.-P. Liu, E. Suzuki, N. Shibata, and K. L. Kirk, J. Fluorine Chem., 1999, 97, 65.
- 17. O. B. Familoni, Synlett, 2002, 1181.
- 18. M. Takahashi, K. Ohtsuki, T. Taga, and Y. Chohnan, Heterocycles, 1998, 48, 1643.
- 19. Z.-P. Liu, N. Shibata, and Y. Takeuchi, J. Org. Chem., 2000, 65, 7583.
- 20. Z.-P. Liu, N. Shibata, and Y. Takeuchi, J. Chem. Soc., Perkin Trans. 1, 2002, 302.
- 21. Z.-P. Liu, T. Toyoshi, and Y. Takeuchi, Synth. Commun., 2004, 34, 471.