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CYCLIC ACETALS AS PRECURSORS OF SUBSTITUTED ISOCHROMANS AND NAPHTHOXEPINES

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Abstract – The reaction of 6,8-dioxabenzocycloheptenes **3** or 8,10-dioxacycloocta[*de*]naphthalenes **5** [easily prepared from the dibenzylic diols **1** and **4**, respectively, and a carbonyl compound] with an excess of lithium and a catalytic amount of DTBB (2.5 mol %) in THF at temperatures ranging between -78 and -60 °C leads, after hydrolysis with water, to the corresponding homobenzylic alcohols **6** and **7**, respectively. Cyclization of compounds **6** and **7** under acidic conditions (85% H₃PO₄ for diols **6** and *p*-TsOH for diols **7**) affords the expected isochromans **8** and naphthoxepines **9**, respectively.

INTRODUCTION

Benzylic organolithium compounds can be prepared by carbon-oxygen bond cleavage from a benzyl ether by means of lithium metal, through a single electron transfer (SET) process.¹ In the case of cyclic benzyl ethers,² the reductive opening lithiation of the corresponding heterocycles³ leads to a functionalized organolithium compounds. These intermediates show a wide applicability in organic synthesis⁴ upon reaction with electrophiles, leading to polyfunctionalized molecules. The most commonly used lithiating reagents are lithium metal itself or in the presence of a stoichiometric or catalytic⁵ amount of an arene, mainly naphthalene and 4,4'-di-*tert*-butylbiphenyl (DTBB). More recently, polymer supported naphthalene, biphenyl⁶ and also polyphenylene⁷ have been used as electron transfer reagents in these processes.⁸ In this paper we report on the application of the mentioned arene-catalyzed lithiation methodology to the reductive ring opening of dibenzylic cyclic acetals and the study of the synthetic utility of the resulting functionalized organolithium compounds.

This paper is dedicated to Professor Dr. Ekkehard Winterfeldt on the occasion of his 75th birthday.

RESULTS AND DISCUSSION

Cyclic acetals **3** were prepared in a one-pot two steps process starting from commercially available 1,2-benzenedimethanol (**1**). Thus, the reaction of diol **1** with trimethyl orthoformate in the presence of a catalytic amount of *p*-toluenesulfonic acid for 5 h at room temperature in 1,2-dimethoxyethane (DME) led to the cyclic orthoester **2**. Further addition of a carbonyl compound $[R^1R^2C=O : CH_3(CH_2)_7CHO, Ph(CH_2)_2CHO, PhCHO, EtCOMe, Et_2CO, (CH_2)_5CO, (-)-menthone] gave the expected cyclic acetals$ **3**in good yields (Scheme 1).⁹



Scheme 1. Reagents and conditions: (i) $HC(OMe)_3$, TsOH (cat.), DME, 25 °C, 5 h; (ii) R^1R^2CO , 25 °C, 15 h.

It was not possible to prepare cyclic acetals **5** derived from 1,8-naphthalenedimethanol (**4**) by using the same methodology as for acetals **3**, because under those reaction conditions the cyclic ether resulting from the dehydration of diol **4** was the main reaction product. However, the desired 7H,11H-8,10-dioxacycloocta[*de*]naphthalenes **5** were prepared in good yields by treatment of diol **4** (easily prepared by reduction with LiAlH₄ in the presence of ZnCl₂ of commercially available 1,8-naphthalic anhydride)¹⁰ with the dimethyl acetal¹¹ of the corresponding carbonyl compound [R₂C(OMe)₂ : Me₂C(OMe)₂, (CH₂)₅C(OMe)₂] in the presence of a catalytic amount of *p*-toluenesulfonic acid in DME at room temperature for 5 h (Scheme 2).



Scheme 2. Reagents and conditions: (i) R₂C(OMe)₂, TsOH (cat.), DME, 25 °C, 5 h.

The reaction of cyclic acetals **3** or **5** with an excess of lithium powder (1:14 molar ratio) and a catalytic amount of DTBB (1:0.1 molar ratio; 5.0 mol %) in THF at temperatures ranging between -78 and -60 °C for 5 h led, after hydrolysis with water, in moderate yields to the corresponding diols **6** and **7**, respectively (Scheme 3 and Table 1).



Scheme 3. Reagents and conditions: (i) Li, DTBB (2.5 mol %), THF, -78 to -60 °C, 5 h; (ii) H₂O, -78 to 25 °C; (iii) H₃PO₄ (85%), PhMe, 115 °C, 2-6 h; (iv) TsOH (cat.), MS 4 Å, PhMe, 110 °C, 2 h.

Concerning a possible mechanistic pathway for the formation of compounds **6** and **7**, it could be possible that in the first step, a benzylic cleavage takes place giving dianionic intermediates of the type **I** and **IV**, respectively. These intermediates either could afford directly the dialcoholates **II** and **V** which are the precursors, after hydrolsys with water, of the final diols **6** and **7**, respectively, or could give complexes of type **III** and **VI** between a benzylic dianion and a carbonyl compound which is generated by elimination from intermediates **I** and **IV**, respectively (Chart 1).



Chart 1.

	Starting		Diols 6 and 7^a			Isochromans 8 and oxepines 9 ^a		
Entry	acetal	No.	Structure	Yield $(\%)^{b}$	No.	Structure	Yield (%) ^c	
1	3 a	6a	OH OH ()7	62	8 a	()7	>95	
2	3b	6b	OH OH OH OH	43	8b	O ()2 Ph	92	
3	3c	6c		65	8c	Ph	90	
4	3d	6d	OH	51	8d	€	48	
5	3e	6e	OH	40	8e	C C	83	
6	3f	6f	OH	74	8f		86	
7	3g	6g	НО	40^{d}	8g	C C	78 ^{d,e}	
8	5a	7a	ОН	45	9a	o o	>95	
9	5b	7b	ОН	48	9b	Co Co Co Co Co Co Co Co Co Co Co Co Co C	>95	

 Table 1. Preparation of diols 6 and 7 and oxygen containing heterocycles 8 and 9

^a All products were >95% pure (GLC and 300 MHz ¹H RMN). ^b Yield based on the starting material **3** or **5**. ^c Yield based on the starting material **6** or **7**. ^d Only the shown diastereomer was obtained. ^e For cyclization conditions see reference 12.

The cyclization of diols **6** and **7** under acidic conditions was studied in the last part of this work. Thus, 3-substituted isochromans **8a-f** were obtained in high yields (except in the case of compound **8d**, for which some other by-products were also obtained; Table 1, entry 4) by treatment of diols **6a-f** with 85% phosphoric acid in refluxing toluene (Scheme 3, Table 1, entries 1-6).^{2a} All attempts to perform the dehydration of diol **6g** (Table 1, entry 7) under acidic conditions gave a mixture of reaction products. However, spiro isochroman derivative **8g** (Table 1, entry 7) was obtained in moderate yield when diol **6g** was treated with an excess of methanesulfonyl chloride in the presence of triethylamine in CH_2Cl_2 at

room temperature.¹² Finally, intramolecular dehydration of diols **7a,b** by treatment with a catalytic amount of *p*-toluenesulfonic acid in the presence of 4 Å molecular sieves in toluene at 110°C gave the corresponding oxygen-containing seven membered heterocycles **9a,c** in very high yields (Scheme 3, Table 1, entries 8-9).¹³

In summary, we have described in this paper a methodology which allows the transformation of cyclic acetals **3** and **5** into diols **6** and **7**, respectively, through a DTBB-catalyzed lithiation. Moreover, these diols are cyclized under acidic conditions to give isochromans **8** and naphthoxepines **9**.

EXPERIMENTAL

All reactions were carried out under an argon atmosphere in oven-dried glassware. All reagents were commercially available (Acros, Aldrich) and were used without further purification. Commercially available anhydrous THF (99.9%, water content $\leq 0.006\%$, Acros) was used as solvent in all the lithiation reactions. IR spectra were measured (film) with a Nicolet Impact 400 D-FT Spectrometer. NMR spectra were recorded with a Bruker AC-300 or a Bruker ADVANCE DRX-500 using CDCl₃ as the solvent. LRMS and HRMS were measured with Shimadzu GC/HS QP-5000 and Finingan MAT95 S spectrometers, respectively. The purity of volatile products and the chromatographic analyses (GLC) were determined with a flame ionisation detector and a 12 m capillary column (0.2 mm diam., 0.33 µm film thickness), using nitrogen (2 mL/min) as carrier gas, $T_{injector} = 275$ °C, $T_{detector} = 300$ °C, $T_{column} = 60$ °C (3 min) and 60-270 °C (15 °C/min), P = 40 kPa. Specific rotations were determined with a Perkin Elmer 341 digital polarimeter.

Preparation of 5,9-dihydro-6,8-dioxabenzenecycloheptenes 3. General procedure.

To a solution of 1,2-benzenedimethanol (1.2 mmol, 168 mg) and TsOH (5 mg) in DME (1 mL) was added trimethyl orthoformate (1.3 mmol, 138 mg) at 25 °C. After 5 h, the corresponding carbonyl compound (1.3 mmol) was added and the reaction mixture was stirred for 15 additional h. Then, all volatile compounds were evaporated (15 Torr). The resulting residue was purified by column chromatography (silica gel, hexane/EtOAc) to yield pure products **3**. Yields are given on Scheme 1. Physical and spectroscopic data as well as literature references follow.

7-Octyl-5,9-dihydro-6,8-dioxabenzocycloheptene (**3a**): White solid; R_f 0.33 (hexane/EtOAc: 30/1); mp 58-59 °C (hexane/CH₂Cl₂); IR *v* (KBr) 3027 (ArH), 1025 cm⁻¹ (COC); ¹H NMR (400 MHz, CDCl₃) δ 0.88 (3H, t, *J* = 6.8 Hz, CH₃), 1.27-1.29 (10H, m, 5xCH₂), 1.40-1.42 (2H, m, CHCH₂CH₂), 1.68-1.73 (2H, m, CHCH₂), 4.84-4.89 (5H, m, CH₂O, CH), 7.15-7.21 (4H, m, ArH); ¹³C NMR (100 MHz, CDCl₃) δ 14.2 (CH₃), 22.8, 24.7, 29.4, 29.6, 29.7, 32.0, 34.7 (CH₂), 71.6 (CH₂O), 108.7 (CH), 127.4, 127.5, 139.4

(ArC); LRMS (EI) *m*/*z* 262 (M⁺, 1%), 150 (10), 149 (100), 121 (24), 119 (11), 93 (17), 91 (35); HRMS (EI) calcd for C₉H₉O₂ (M⁺-C₈H₁₇) 149.0603, found 149.0598.

7-(2-Phenylethyl)-5,9-dihydro-6,8-dioxabenzocycloheptene (**3b**): Colourless oil; R_f 0.25 (hexane/EtOAc: 30/1); IR ν (film) 3063, 3020 (ArH), 1129 cm⁻¹ (COC); ¹H NMR (400 MHz, CDCl₃) δ 2.01-2.04 (2H, m, CH₂CH), 2.75 (2H, t, J = 8.0 Hz, CH₂Ph), 4.82-4.84 (5H, m, CH₂O, CH), 7.13-7.21 (7H, m, ArH), 7.25-7.29 (2H, m, ArH); ¹³C NMR (100 MHz, CDCl₃) δ 30.9, 36.2 (CH₂), 71.8 (CH₂O), 107.8 (CH), 126.0, 127.5, 127.6, 128.5, 128.6, 139.3, 141.7 (ArC); LRMS (EI) *m*/*z* 254 (7%), 192 (44), 179 (5), 149 (34), 133 (10), 121 (28), 120 (10), 119 (22), 118 (14), 105 (57), 104 (83), 93 (30), 92 (23), 91 (100), 89 (10), 77 (22), 65 (17); HRMS (EI) calcd for C₁₇H₁₈O₂ 254.1307, found 254.1308.

7-Phenyl-5,9-dihydro-6,8-dioxabenzocycloheptene (**3c**)¹⁴: White solid; R_f 0.27 (hexane/EtOAc: 30/1); mp 78-79 °C (hexane/CH₂Cl₂); IR ν (KBr) 3063, 3031 (ArH), 1109 cm⁻¹ (COC); ¹H NMR (300 MHz, CDCl₃) δ 4.94 (4H, s, CH₂O), 5.92 (1H, s, (CHPh), 7.13-7.20 (4H, m, ArH), 7.21-7.53 (3H, m, ArH), 7.55 (2H, d, J = 6.4 Hz, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 70.2 (CH₂O), 105.0 (CH), 126.5, 127.1, 127.3, 128.3, 128.7, 138.9, 139.0 (ArC); LRMS (EI) *m*/*z* 226 (M⁺, 7%), 120 (41), 119 (81), 105 (61), 104 (100), 92 (21), 91 (92), 89 (19), 77 (28), 65 (17), 51 (10).

7-Ethyl-7-methyl-5,9-dihydro-6,8-dioxabenzocycloheptene (**3d**)⁹: Colourless oil; R_f 0.29 (hexane/EtOAc: 30/1); IR *v* (film) 3076 (ArH), 1088 cm⁻¹ (COC); ¹H NMR (300 MHz, CDCl₃) δ 0.98 (3H, t, *J* = 7.6 Hz, CH₂CH₃), 1.42 (3H, s, CH₃CO), 1.83 (2H, q, *J* = 7.5 Hz, CH₂CH₃), 4.81 (2H, d, *J* = 15.3 Hz, 2xCHHO), 4.86 (2H, d, *J* = 15.3 Hz, 2xCHHO), 7.04-7.07 (2H, m, ArH), 7.15-7.18 (2H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 8.8, 20.6 (CH₃), 29.0 (CH₂), 64.8 (CH₂O), 104.4 (C), 126.2, 126.8, 138.4 (ArC); LRMS (EI) *m*/*z* 192 (M⁺, 1%), 164 (10), 163 (100), 121 (36), 120 (14), 119 (44), 93 (11), 92 (12), 91 (51).

7,7-Diethyl-5,9-dihydro-6,8-dioxabenzocycloheptene (**3e**)⁹: White solid; R_f 0.31 (hexane/EtOAc: 30/1); mp 77-78 °C (hexane/CH₂Cl₂); IR ν (KBr) 3018 (ArH), 1054 cm⁻¹ (COC); ¹H NMR (400 MHz, CDCl₃) δ 0.92 (6H, t, J = 7.5 Hz, 2xCH₃), 1.78 (4H, q, J = 7.5 Hz, 2xCH₂CH₃), 4.83 (4H, s, 2xCH₂O), 7.05-7.07 (2H, m, ArH), 7.15-7.26 (2H, m, ArH); ¹³C NMR (100 MHz, CDCl₃) δ 8.4 (CH₃), 24.3 (CH₂), 64.5 (CH₂O), 106,5 (C), 126.2, 126.7, 138.6 (ArC); LRMS (EI) *m*/*z* 177 (M⁺-Et, 100%), 120 (12), 119 (36), 91 (36), 57 (94).

Spiro[cyclohexane-1,7'-(5',9'-dihydro-6',8'-dioxabenzocycloheptene)] (**3f**)⁹: White solid; R_f 0.33 (hexane/EtOAc: 30/1); mp 83-84 °C (hexane/CH₂Cl₂); IR ν (KBr) 3066, 3006 (ArH), 1115 cm⁻¹ (COC); ¹H NMR (400 MHz, CDCl₃) δ 1.43-1.48 (2H, m, CH₂), 1.57-1.62 (4H, m, 2xCH₂), 1.79-1.82 (4H, m, 2xCH₂), 4.86 (4H, s, 2xCH₂O), 7.04-7.07 (2H, m, ArH), 7.14-7.24 (2H, m, ArH); ¹³C NMR (100 MHz, CDCl₃) δ 23.3, 25.9, 32.6 (CH₂), 64.2 (CH₂O), 102.4 (C), 126.2, 126.7, 138.5 (ArC); LRMS (EI) *m/z* 218 (M⁺, 53%), 175 (32), 131 (12), 119 (30), 105 (29), 104 (100), 91 (36), 78 (10), 55 (16).

(2*S*,5*R*)-2-Isopropyl-5-methylspiro[cyclohexane-1,7'-(5',9'-dihydro-6',8'-dioxabenzocycloheptene)]

(**3g**): White solid; $[\alpha]^{22}_{D}$ +23.0 (*c* 0.62, CH₂Cl₂); R_f 0.40 (hexane/EtOAc: 30/1); mp 62-63 °C (hexane/CH₂Cl₂); IR *v* (KBr) 3069, 3027 (ArH), 1118 cm⁻¹ (COC); ¹H NMR (400 MHz, CDCl₃) δ 0.93 (3H, d, *J* = 6.9 Hz, CH₃CH), 0.96 (3H, d, *J* = 6.6 Hz, CH₃CH), 1.01-1.05 (1H, m, CH), 1.11 (3H, d, *J* = 6.6 Hz, CH₃CH), 1.30-1.36 (1H, m, CH), 1.42-1.49 (1H, m, CH), 1.67-1.82 (4H, m, 4xCH), 1.86-1.95 (1H, m, CH), 1.96-2.03 (1H, m, CH), 4.57 (1H, d, *J* = 14.8 Hz, CHHO), 4.68 (1H, d, *J* = 15.1 Hz, CHHO), 4.97 (1H, d, *J* = 15.1 Hz, CHHO), 5.00 (1H, d, *J* = 14.8 Hz, CHHO), 7.07 (2H, s, ArH), 7.14-7.26 (2H, m, ArH); ¹³C NMR (100 MHz, CDCl₃) δ 22.3, 22.4, 23.6 (CH₃), 25.3, 27.0, 29.5 (CH₂), 29.9, 36.2, 44.0 (CH), 63.9, 64.5 (CH₂O), 105.9 (C), 126.0, 126.3, 126.6, 126.8, 138.55, 138.6 (ArC); LRMS (EI) *m/z* 274 (M⁺, 13%), 189 (20), 105 (15), 104 (100), 91 (12), 69 (14); HRMS (EI) calcd for C₁₈H₂₆O₂ 274.1933, found 274.1937.

Preparation of 7H,11H-8,10-dioxacycloocta[de]naphthalenes 5. General procedure.

To a solution of 1,8-naphthalenedimethanol (1.5 mmol, 282 mg) and TsOH (5 mg) in DME (4 mL) was added the dimethyl acetal of the corresponding carbonyl compound (1.1 mmol; 5 mmol, 504 mg in the case of the dimethyl acetal of acetone) at 25 °C and the reaction mixture was stirred for 5 h at the same temperature. Then, all volatile compounds were evaporated (15 Torr). The resulting residue was purified by column chromatography (silica gel, hexane/EtOAc) to yield pure products **5**. Yields are given on Scheme 2. Physical and spectroscopic data follow.

9,9-Dimethyl-7*H***,11***H***-8,10-dioxacycloocta**[*de*]**naphthalene** (**5a**): White solid; R_f 0.11 (hexane/EtOAc: 30/1); mp 109-110 °C (hexane/CH₂Cl₂); IR ν (KBr) 3035, 2990, 2945, 2890 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.57 (6H, s, 2×CH₃), 5.01 (4H, s, 2×OCH₂), 7.37-7.42 (4H, m, ArH), 7.79-7.82 (2H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 25.2 (CH₃), 66.8 (CH₂), 100.9 (CO₂), 125.2, 130.2, 130.3, 132.9, 134.8, 135.6 (ArC); LRMS (EI) *m*/*z* 228 (M⁺, 43%), 171 (31), 170 (93), 169 (77), 155 (24), 142 (57), 141 (100), 139 (33), 115 (37); HRMS (EI) calcd for C₁₅H₁₆O₂ 228.1150, found 228.1143.

Spiro[cyclohexane-1,9'-(7'*H***,11'***H***-8',10'-dioxacycloocta[***de***]naphthalene)] (5b): White solid; R_f 0.18 (hexane/EtOAc: 30/1); mp 115-116 °C (pentane/CH₂Cl₂); IR** *v* **(KBr) 3045, 2935, 2855 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) \delta 1.44-1.47 (2H, m, CH₂), 1.50-1.61 (4H, m, 2×CH₂), 1.81-1.84 (4H, m, 2×CH₂), 4.94 (4H, s, 2×OCH₂), 7.32-7.37 (4H, m, ArH), 7.74-7.77 (2H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) \delta 23.3, 25.8, 34.0 (CH₂), 65.7 (OCH₂), 101.0 (CO₂), 125.1, 130.1, 130.2, 132.7, 134.9, 135.5 (ArC); LRMS (EI)** *m/z* **268 (M⁺, 12%), 170 (100), 169 (49), 154 (23), 153 (66), 152 (21), 142 (59), 141 (74), 115 (34); HRMS (EI) calcd for C₁₈H₂₀O₂ 268.1463, found 268.1469.**

Lithiation of cyclic acetals 3 and 5. Preparation of diols 6 and 7. General procedure.

To a blue suspension of lithium powder (100 mg, 14 mmol) and a catalytic amount of DTBB (27 mg, 0.1 mmol; 5 mol%) in dry THF (2.5 mL) under argon was added dropwise a solution of the corresponding acetal **3** or **5** (1 mmol) in THF (0.5 mL) at -78 °C, and the resulting mixture was stirred for 5 h at temperatures ranging between -78 and -60 °C. After that, the reaction mixture was hydrolyzed with water (4 mL), extracted with EtOAc (3×10 mL), dried over anhydrous MgSO₄ and evaporated (15 Torr). The residue was purified by column chromatography (silica gel, hexane/EtOAc) to yield pure products **6** and **7**. Yields are given in Table 1, physical, analytical and spectroscopic data follow as well as literature references follow.

1-(2-Hydroxymethylphenyl)decan-2-ol (6a): Colourless oil; R_f 0.25 (hexane/EtOAc: 2/1); IR ν (film) 3540-3140 (OH), 3020, 2950, 2925, 2854 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.89 (3H, t, J = 6.8 Hz, CH₃), 1.22-1.57 (14H, m, 7×CH₂), 2.71-2.79 (2H, m, ArCH₂CH), 3.55 (2H, br s, 2×OH), 3.70-3.77 (1H, m, CHOH), 4.41 (1H, d, J = 11.7 Hz, CHHOH), 4.69 (1H, d, J = 11.7 Hz, CHHOH), 7.16-7.21 (2H, m, ArH), 7.24-7.29 (2H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 14.2 (CH₃), 22.8, 25.9, 29.4, 29.7, 29.8, 32.0, 37.9, 39.8 (CH₂), 63.2 (CH₂OH), 73.2 (CHOH), 126.8, 128.4, 130.1, 130.5, 138.3, 139.5 (ArC); LRMS (EI) m/z 246 (M⁺-H₂O, 3%), 207 (15), 133 (10), 105 (31), 104 (100); HRMS (EI) calcd for C₁₅H₁₆O (M⁺-H₂O) 246.1984, found 246.2005.

1-(2-Hydroxymethylphenyl)-4-phenylbutan-2-ol (**6b**)¹⁵: Colourless oil; R_f 0.31 (hexane/EtOAc: 2/1); IR ν (film) 3583-3120 (OH), 3060, 3024, 1494, 1453, 1015 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.91-1.97 (2H, m, CHC*H*₂CH₂), 2.59-2.80 (4H, m, 2×CH₂), 3.52 (2H, br s, 2×OH), 3.81-3.84 (1H, m, CHOH), 4.50 (1H, d, J = 11.7 Hz, CHHOH), 4.76 (1H, d, J = 11.7 Hz, CHHOH), 7.17-7.31 (9H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 32.3, 39.4, 40.0 (CH₂), 63.5 (CH₂OH), 72.6 (CHOH), 126.1, 127.0, 128.5, 128.6, 130.1, 130.5, 137.9, 139.5, 141.9 (ArC); LRMS (EI) *m*/*z* 238 (M⁺-H₂O, 2%), 117 (13), 106 (100), 105 (19), 91 (84), 77 (10).

2-(2-Hydroxymethylphenyl)-1-phenylethanol (**6c**)^{2a}: White solid; R_f 0.44 (hexane/EtOAc: 1/1); mp 70-71 °C (pentane/CH₂Cl₂); IR ν (KBr) 3600-3080 cm⁻¹ (OH); ¹H NMR (300 MHz, CDCl₃) δ 2.95 (1H, dd, J = 14.0, 3.7 Hz, CHHCHOH), 3.05 (1H, dd, J = 14.0, 9.1 Hz, CHHCHOH), 3.75 (2H, br s, 2×OH), 4.37 (1H, d, J = 11.8 Hz, CHHOH), 4.68 (1H, d, J = 11.8 Hz, CHHOH), 4.78 (1H, dd, J = 9.1, 7.3 Hz, CHOH), 7.14-7.36 (9H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 42.2 (CH₂), 63.1 (CH₂OH), 75.3 (CHOH), 125.7, 126.8, 127.5, 128.3, 128.4, 130.0, 130.5, 137.4, 139.4, 144.3 (ArC); LRMS (EI) *m/z* 210 (M⁺-H₂O, 9%), 105 (11), 104 (100), 103 (11), 77 (10).

1-(2-Hydroxymethylphenyl)-2-methylbutan-2-ol (6d): Colourless oil; R_f 0.24 (hexane/EtOAc: 2/1); IR ν (film) 3470-3180 (OH), 3070, 3020, 2968, 2925, 2860 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.98 (3H, t, J = 7.2 Hz, CH₃CH₂), 1.17 (3H, s, CCH₃), 1.54-1.60 (2H, m, CH₂CH₃), 2.74 (1H, d, J = 13.9 Hz, CHHAr), 2.94 (1H, d, J = 13.9 Hz, CHHAr), 3.68 (2H, br s, 2×OH), 4.49 (1H, d, J = 11.8 Hz, CHHOH),

4.61 (1H, d, J = 11.8 Hz, CHHOH), 7.12-7.32 (4H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 8.5 (CH₃), 26.4 (CH₂), 35.5 (CH₃), 43.2 (CH₂), 63.3 (CH₂OH), 72.8 (COH), 126.9, 127.7, 130.7, 132.3, 136.4, 140.2 (ArC); LRMS (EI) m/z 176 (M⁺-H₂O, 1%), 147 (13), 119 (13), 105 (21), 104 (100), 91 (19), 73 (25); HRMS (EI) calcd for C₁₂H₁₈O₂ (M⁺-H₂O) 176.1201, found 176.1191.

2-Ethyl-1-(2-hydroxymethylphenyl)butan-2-ol (**6e**)^{2a}: Colourless oil; R_f 0.31 (hexane/EtOAc: 2/1); IR v (film) 3600-3060 cm⁻¹ (OH); ¹H NMR (300 MHz, CDCl₃) δ 0.92 (6H, t, J = 7.5 Hz, 2×CH₃), 1.44-1.60 (6H, m, 2×OH, 2×CH₂CH₃), 2.83 (2H, s, ArCH₂), 4.52 (2H, s, CH₂OH), 7.10-7.31 (4H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 8.0 (CH₃), 30.7, 40.5 (CH₂), 63.1 (CH₂OH), 74.5 (COH), 126.7, 127.5, 130.5, 131.9, 136.1, 140.3 (ArC); LRMS (EI) m/z 190 (M⁺-H₂O, 9%), 161 (12), 105 (14), 104 (100), 91 (13), 87 (11), 77 (12), 57 (29), 45 (17), 41 (10).

1-[(2-Hydroxymethylphenyl)methyl]cyclohexanol (6f)^{2a}: White solid; R_f 0.48 (hexane/EtOAc: 2/1); mp 70-71 °C (pentane/CH₂Cl₂); IR ν (KBr) 3600-3060 cm⁻¹ (OH); ¹H NMR (300 MHz, CDCl₃) δ 1.19-1.61 (10H, m, 5×CH₂), 2.86 (2H, s, ArCH₂C), 3.45 (2H, br s, 2×OH), 4.56 (2H, s, CH₂OH), 7.12-7.34 (4H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 22.1, 25.6, 37.9, 44.5 (CH₂), 63.3 (CH₂OH), 71.2 (COH), 126.8, 127.5, 130.5, 132.1, 135.8, 140.3 (ArC); LRMS (EI) *m/z* 202 (M⁺-H₂O, 9%), 105 (11), 104 (100).

(1*R*,2*S*,5*R*)-2-Isopropyl-5-methyl-1-[(2-hydroxymethylphenyl)methyl]cyclohexanol (6g)¹²: Colourless oil; $[α]^{22}_{D}$ -19.5 (*c* 1.0, CH₂Cl₂); R_f 0.13 (hexane/EtOAc: 5/1); IR *ν* (film) 3500-3180 cm⁻¹ (OH); ¹H NMR (300 MHz, CDCl₃) δ 0.81 (3H, d, *J* = 6.2 Hz, CH₃CH), 0.87-1.04 (2H, m, 2×CH), 0.98 [6H, d, *J* = 7.0 Hz, (CH₃)₂CH], 1.21-1.51 (4H, m, 2×CH₂), 1.59-1.67 (1H, m, CH), 1.73-1.78 (1H, m, CH), 2.35-2.45 (1H, m, CH), 2.43 (1H, d, *J* = 13.9 Hz, ArCHHC), 2.71 (2H, br s, 2×OH), 3.61 (1H, d, *J* = 13.9 Hz, ArCHHC), 4.42 (1H, d, *J* = 11.9 Hz, ArCHHOH), 4.81 (1H, d, *J* = 11.9 Hz, ArCHHOH), 7.14-7.27 (3H, m, ArH), 7.33-7.38 (1H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 18.0 (CH₃), 21.0 (CH₂), 22.3, 23.8 (CH₃), 25.9, 28.0 (CH), 34.8, 42.4, 46.8 (CH₂), 51.1 (CH), 63.3 (CH₂OH), 74.7 (COH), 126.9, 127.3, 130.6, 133.0, 136.0, 140.5 (ArC); LRMS (EI) *m*/*z* 258 (M⁺-H₂O, 3%), 104 (100), 95 (11), 81 (29), 69 (21), 57 (29), 45 (17), 41 (10); HRMS (EI) calcd for C₁₈H₂₆O (M⁺-H₂O) 258.1984, found 258.1971.

1-[(8-Hydroxymethyl)-1-naphthyl]-2-methylpropan-2-ol (**7a**)¹³: Colourless oil; R_f 0.14 (hexane/EtOAc: 2/1); IR ν (film) 3560-3275 (OH), 3055, 3033 (ArH), 1095 cm⁻¹ (COC); ¹H NMR (300 MHz, CDCl₃) δ 1.25 [6H, s, (CH₃)₂C], 2.17 (2H, br s, 2×OH), 3.57 (2H, s, CH₂COH), 5.18 (2H, s, CH₂OH), 7.33-7.45 (3H, m, ArH), 7.52-7.55 (1H, m, ArH), 7.78-7.83 (2H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 29.5 (CH₃), 47.7 (CH₂), 66.5 (CH₂OH), 72.7 (COH), 124.6, 125.0, 129.2, 129.5, 130.5, 131.4, 132.6, 133.6, 135.8, 137.0 (ArC); LRMS (EI) *m*/*z* 212 (M⁺-H₂O, 9%), 179 (18), 172 (22), 154 (92), 153 (100), 141 (17), 128 (27), 115 (17), 59 (47); HRMS (EI) calcd for C₁₅H₁₆O (M⁺-H₂O) 212.1201, found 212.1196.

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1-{[(8-Hydroxymethyl)-1-naphthyl]methyl}cyclohexanol (7b)¹³: Colourless oil; R_f 0.36

(hexane/EtOAc: 2/1); IR ν (film) 3535-3140 (OH), 3058, 3036 (ArH), 1045 cm⁻¹ (COC); ¹H NMR (300 MHz, CDCl₃) δ 1.27-1.93 (12H, m, 5×CH₂, 2×OH), 3.54 (2H, s, ArCH₂COH), 5.16 (2H, s, CH₂OH), 7.30 (1H, d, J = 5.9 Hz, ArH), 7.36-7.42 (2H, m, ArH), 7.52 (1H, d, J = 5.8 Hz, ArH), 7.76-7.80 (2H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 22.0, 25.8, 37.4, 47.2 (CH₂), 66.3 (CH₂OH), 73.2 (COH), 124.4, 125.0, 129.1, 130.4, 131.6, 132.4, 132.5, 132.9, 135.7, 137.1 (ArC); LRMS (EI) m/z 252 (M⁺-H₂O, 5%), 172 (19), 165 (20), 155 (15), 154 (100), 153 (75), 152 (36), 141 (12), 128 (19), 115 (15), 81 (18), 55 (30); HRMS (EI) calcd for C₁₈H₂₀O (M⁺-H₂O) 252.1514, found 252.1503.

Cyclization of diols 6. Preparation of isochromans 8. General procedure.

To a solution of the corresponding diol **6** (1 mmol) in toluene (5 mL) was added 85% phosphoric acid (0.4 mL). The reaction mixture was heated at 110 °C for 4 h. Then toluene was removed by distillation and the resulting residue was hydrolyzed with water (5 mL), extracted with EtOAc (3×10 mL), dried over anhydrous Na₂SO₄ and evaporated (15 Torr). The residue was purified by column chromatography (silica gel, hexane) to yield pure products **8**. Compound **8g** was prepared according to reference 12. Yields are given in Table 1, physical, analytical and spectroscopic data follow as well as literature references follow.

3-Octylisochroman (8a): Colourless oil; R_f 0.31 (hexane/EtOAc: 30/1); IR ν (film) 3069, 3014, 2955, 2925, 2850 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.88 (3H, t, J = 7.0 Hz, CH₃), 1.23-1.70 (14H, m, 7×CH₂), 2.70 (2H, d, J = 9.7 Hz, CH₂CH), 3.57-3.66 (1H, m, CHO), 4.80 (1H, d, J = 12.3 Hz, CHHO), 4.83 (1H, d, J = 12.3 Hz, CHHO), 6.96-6.99 (1H, m, ArH), 7.00-7.16 (3H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 14.1 (CH₃), 22.6, 25.5, 29.3, 29.5, 29.7, 31.8, 34.1, 36.0 (CH₂), 68.2 (CH₂O), 74.9 (CHO), 124.1, 125.8, 126.2, 128.8, 133.6, 134.9 (ArC); LRMS (EI) *m*/*z* 246 (M⁺, 3%), 133 (12), 105 (39), 104 (100); HRMS (EI) calcd for C₁₇H₂₆O 246.1984, found 246.2001.

3-(2-Phenylethyl)isochroman (8b): Colourless oil; R_f 0.20 (hexane/EtOAc: 30/1); IR ν (film) 3063, 3020 (ArH), 1129 cm⁻¹ (COC); ¹H NMR (400 MHz, CDCl₃) δ 1.87-2.02 (2H, m, CH₂CH₂CH), 2.70-2.88 (4H, m, ArCH₂CHO, PhCH₂), 3.60-3.66 (1H, m, CHO), 4.79 (1H, d, J = 15.1 Hz, CHHO), 4.87 (1H, d, J = 15.1 Hz, CHHO), 7.00-7.31 (9H, m, ArH); ¹³C NMR (100 MHz, CDCl₃) δ 31.8, 34.2, 37.7 (CH₂), 68.3 (CH₂O), 74.0 (CHO), 124.3, 125.9, 126.1, 126.5, 128.5, 128.6, 129.0, 133.5, 135.0, 142.1 (ArC); LRMS (EI) *m*/*z* 238 (M⁺, 12%), 117 (16), 116 (60), 105 (66), 104 (100), 91 (30); HRMS (EI) calcd for C₁₇H₁₈O 238.1358, found 238.1350.

3-Phenylisochroman (8c)^{16,2a}: White solid; R_f 0.26 (hexane); mp 69-70 °C (hexane/CH₂Cl₂); IR ν (KBr) 3040, 1600, 1500, 740, 700 cm⁻¹ (ArH); ¹H NMR (300 MHz, CDCl₃) δ 2.94 (1H, dd, J = 16.4, 3.6 Hz, CHHCHO), 3.06 (1H, dd, J = 16.4, 10.6 Hz, CHHCHO), 4.70 (1H, dd, J = 10.6, 3.6 Hz, CHO), 4.98 (2H, s, CH₂O), 7.00-7.45 (9H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 36.0 (CH₂), 68.6 (CH₂O), 76.8 (CHO),

124.2, 125.8, 126.1, 126.4, 127.6, 128.4, 128.7, 133.4, 134.5, 142.1 (ArC); LRMS (EI) *m*/*z* 210 (M⁺, 20%), 105 (11), 104 (100), 103 (16), 78 (15), 77 (12).

3-Ethyl-3-methylisochroman (8d): Colourless oil; R_f 0.24 (hexane/EtOAc: 30/1); IR v (film) 3063, 3020, 2965, 2928, 2836 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.96 (3H, t, J = 7.5 Hz, CH₃CH₂), 1.21 (3H, s, CCH₃), 1.43-1.68 (2H, m, CH₂CH₃), 2.64 (1H, d, J = 16.1 Hz, CHHAr), 2.74 (1H, d, J = 16.1 Hz, CHHAr), 4.76 (2H, s, CH₂O), 6.98-7.01 (1H, m, ArH), 7.06-7.09 (1H, m, ArH), 7.12-7.17 (2H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 7.9 (CH₃), 22.5 (CH₂), 32.1 (CH₃), 38.0 (CH₂), 62.8 (CH₂O), 73.0 (CO), 123.8, 125.7, 126.3, 129.2, 132.9, 134.2 (ArC); LRMS (EI) *m*/*z* 176 (M⁺, 1%), 147 (52), 119 (24), 105 (16), 104 (100), 78 (12); HRMS (EI) calcd for C₁₂H₁₆O 176.1201, found 176.1206.

3,3-Diethylisochroman (**8e**)^{2a}: Colourless oil; R_f 0.35 (hexane); IR ν (film) 3040, 3000, 1580, 740, 720 cm⁻¹ (ArH); ¹H NMR (300 MHz, CDCl₃) δ 0.90 (6H, t, J = 7.5 Hz, 2×CH₃), 1.42-1.54 (2H, m, 2×CHHCH₃), 1.60-1.72 (2H, m, 2×CHHCH₃), 2.67 (2H, s, ArCH₂CO), 4.72 (2H, s, CH₂O), 6.96-7.15 (4H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 7.6 (CH₃), 27.5, 36.1 (CH₂), 62.5 (CH₂O), 75.1 (CO), 123.8, 125.6, 126.3, 129.2, 132.8, 134.4 (ArC); LRMS (EI) *m/z* 190 (M⁺, 2%), 162 (11), 161 (100), 105 (20), 104 (52), 57 (13).

(1*R*,2*S*,5*R*)-2-Isopropyl-5-methylspiro[cyclohexane-1,3'-isochroman] (8f)¹²: Colourless oil; $[α]^{22}_{D}$ -29.0 (*c* 0.42, CH₂Cl₂); R_f 0.39 (hexane); IR *v* (film) 2951, 2918, 2853, 1732, 1454, 1072 1363 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.67 (1H, dd, *J* = 14.0, 12.5 Hz, CH), 0.81 (3H, d, *J* = 6.7 Hz, CH₃CH), 0.87 (3H, d, *J* = 6.9 Hz, CH₃CH), 0.93 (3H, d, *J* = 7.0 Hz, CH₃CH), 1.17-1.32 (2H, m, CH₂), 1.50-1.68 (3H, m, CH, CH₂), 1.70-1.82 (1H, m, CH), 1.95-2.05 (1H, m, CH), 2.16-2.22 (1H, m, CH), 2.17 (1H, d, *J* = 16.0 Hz, ArCHHC), 3.33 (1H, d, *J* = 16.0 Hz, ArCHHC), 4.60-4-72 (2H, m, ArCH₂O), 6.99-7.02 (1H, m, ArH), 7.07-7.16 (3H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) δ 18.3 (CH₃), 20.9 (CH₂), 22.3, 24.0 (CH₃), 26.2, 27.9 (CH), 35.5, 35.9, 42.0 (CH₂), 50.3 (CH), 62.0 (CH₂OH), 75.2 (CO), 123.9, 125.4, 126.3, 129.2, 134.0, 134.9 (ArC); LRMS (EI) *m*/*z* 258 (M⁺, 27%), 173 (54), 146 (17), 145 (23), 117 (10), 105 (16), 104 (100), 103 (12), 78 (15), 69 (27); HRMS (EI) calcd for C₁₈H₂₆O 258.1984, found 258.1978.

Cyclization of diols 7. Preparation of naphthoxepines 9. General procedure.

To a solution of the corresponding diol 7 (0.1 mmol) in toluene (1.5 mL) a catalytic amount of p-toluenesulfonic acid (30 mg) and 4 Å molecular sieves (30 mg) were added. The reaction mixture was heated at 110 °C for 2 h. Then toluene was removed by distillation and the resulting residue was hydrolyzed with water (5 mL), extracted with EtOAc (3 × 10 mL), dried over anhydrous Na₂SO₄ and evaporated (15 Torr). The residue was purified by column chromatography (silica gel, hexane) to yield pure products **9**. Yields are given in Table 1, physical, analytical and spectroscopic data as well as literature references follow.

9,9-Dimethyl-9,10-dihydro-7*H***-8-oxacyclohepta[***de***]naphthalene (9a)¹³: Pale yellow oil; R_f 0.37 (hexane/EtOAc: 10/1); IR** *v* **(film) 3054, 3035 (ArH), 1065 cm⁻¹ (COC); ¹H NMR (300 MHz, CDCl₃) \delta 1.34 [6H, s, C(CH₃)₂], 3.32 (2H, s, CH₂CO), 5.03 (2H, s, CH₂O), 7.16-7.40 (4H, m, ArH), 7.71 (2H, d,** *J* **= 8.4 Hz, ArH); ¹³C NMR (75 MHz, CDCl₃) \delta 27.6 (CH₃), 46.9 (CH₂), 67.2 (CH₂O), 77.5 (COCH₂), 124.8, 124.9, 125.4, 125.5, 127.5, 128.2, 128.3, 135.1, 135.6, 139.6 (ArC); LRMS (EI)** *m***/***z* **212 (M⁺, 11%), 165 (12), 154 (71), 153 (100), 152 (32), 151 (11); HRMS (EI) calcd for C₁₅H₁₆O 212.1201, found 212.1217.**

Spiro[cyclohexane-1,9'-(9',10'-dihydro-7'*H***-8'-oxacyclohepta[***de***]naphthalene)] (9b)¹³: Pale yellow oil; R_f 0.46 (hexane/EtOAc: 10/1); IR \nu (film) 3052, 3035 (ArH), 1064 cm⁻¹ (COC); ¹H NMR (300 MHz, CDCl₃) \delta 1.27-1.76 (10H, m, 5×CH₂), 3.29 (2H, s, CH₂CO), 5.05 (2H, s, CH₂O), 7.17-7.36 (4H, m, ArH), 7.68-7.72 (2H, m, ArH); ¹³C NMR (75 MHz, CDCl₃) \delta 22.4, 25.9, 36.1, 45.5 (CH₂), 66.7 (CH₂O), 78.0 (CO), 124.9, 125.3, 125.4, 127.4, 128.0, 128.1, 132.4, 135.1, 135.4, 140.0 (ArC); LRMS (EI)** *m/z* **252 (M⁺, 5%), 155 (15), 154 (100), 153 (63), 152 (21); HRMS (EI) calcd for C₁₈H₂₀O 252.1514, found 252.1537.**

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REFERENCES

- (a) U. Azzena, S. Carta, G. Melloni, and A. Sechi, *Tetrahedron*, 1997, **53**, 16205. (b) E. Alonso, D. J. Ramón, and M. Yus, *J. Org. Chem.*, 1997, **62**, 417. (c) U. Azzena, *J. Chem. Soc., Perkin Trans. 1*, 2002, 360. (d) U. Azzena, G. Dettori, G. Sforazzini, M. Yus, and F. Foubelo, *Tetrahedron*, 2006, **62**, 1557.
- (a) J. Almena, F. Foubelo, and M. Yus, *Tetrahedron*, 1995, **51**, 3351. (b) J. Almena, F. Foubelo, and M. Yus, *Tetrahedron*, 1995, **51**, 3365. (c) U. Azzena, S. Demartis, and G. Melloni, *J. Org. Chem.*, 1996, **61**, 4913. (d) U. Azzena and L. Pilo, *Synthesis*, 1999, 664. (e) U. Azzena, S. Demartis, L. Pilo, and E. Piras, *Tetrahedron*, 2000, **56**, 8375. (f) F. Foubelo, B. Moreno, and M. Yus, *Tetrahedron Lett.*, 2004, **45**, 8983. (g) F. Foubelo, B. Moreno, T. Soler, and M. Yus, *Tetrahedron*, 2005, **61**, 9082. (h) F. Foubelo, D. García, B. Moreno, and M. Yus, *Tetrahedron Lett.*, 2007, **48**, 3379.
- For reviews on the reductive opening lithiation of heterocycles, see: (a) F. Foubelo and M. Yus, *Rev. Heteroatom Chem.*, 1997, 17, 73. (b) M. Yus, *Pure Appl. Chem.*, 2003, 75, 1453. (c) F. Foubelo and M. Yus, *Targets Heterocycl. Syst.*, 2002, 6, 136.

- For reviews, see: (a) C. Nájera and M. Yus, *Trends Org. Chem.*, 1991, 2, 155. (b) C. Nájera and M. Yus, *Curr. Org. Chem.*, 2003, 7, 867. (c) C. Nájera, J. M. Sansano, and M. Yus, *Tetrahedron*, 2003, 59, 9255. (d) R. Chinchilla, C. Nájera, and M. Yus, *Tetrahedron*, 2005, 61, 3139. (e) M. Yus and F. Foubelo, 'Handbook of Functionalized Organometallics', ed. by P. Knochel, Wiley-VCH, Weinheim, 2005, chapter 2.
- For reviews, see: (a) M. Yus, *Chem. Soc. Rev.*, 1996, 25, 155. (b) D. J. Ramón and M. Yus, *Eur. J. Org. Chem.*, 2000, 225. (c) M. Yus, *Synlett*, 2001, 1197. (d) M. Yus, 'The Chemistry of Organolithium Compounds', ed. by Z. Rappoport and I. Marek, Wiley, Chichester, 2004, chapter 11.
- 6. (a) T. R. van den Ancker, G. R. Hanson, F.-C. Lee, and C. L. Raston, *Chem. Commun.*, 1997, 125.
 (b) C. Gómez, S. Ruiz, and M. Yus, *Tetrahedron Lett.*, 1998, **39**, 1397. (c) C. Gómez, S. Ruiz, and M. Yus, *Tetrahedron*, 1999, **55**, 7017. (d) T. Arnauld, A. G. M. Barret, and B. T. Hopkins, *Tetrahedron Lett.*, 2002, **43**, 1081.
- 7. M. Yus, C. Gómez, and P. Candela, *Tetrahedron*, 2002, **58**, 6207.
- For the studies on the mechanism of this reaction, see: (a) M. Yus, R. P. Herrera, and A. Guijarro, *Tetrahedron Lett.*, 2001, 42, 3455. (b) M. Yus, R. P. Herrera, and A. Guijarro, *Chem. Eur. J.*, 2002, 8, 2574.
- 9. N. Machinaga and C. Kibayashi, Tetrahedron Lett., 1989, 30, 4165.
- 10. H. Luo, Q. Zeng, Z. Liu, Y. Wei, B. Li, and F. Wang, Synth. Commun., 2004, 34, 2269.
- 11. E. Napolitano, R. Fiaschi, and E. Mastrorilli, Synthesis, 1986, 122.
- 12. M. Yus, B. Moreno, and F. Foubelo, Synthesis, 2004, 1115.
- 13. F. Foubelo, B. Moreno, and M. Yus, *Tetrahedron*, 2004, **60**, 4655.
- 14. H. K. Patney, *Tetrahedron Lett.*, 1991, **32**, 413.
- 15. I. M. Pastor and M. Yus, Tetrahedron, 2001, 57, 2365.
- E. J. Eisenbraum, J. R. Mattox, R. C. Bansal, M. A. Wilhelm, P. W. K. Flanagan, A. B. Carel, R. E. Laramy, and M. C. Hamming, *J. Org. Chem.*, 1968, 33, 2000.