HETEROCYCLES, Vol. 74, 2007, pp. 185 - 189. © The Japan Institute of Heterocyclic Chemistry Received, 29th August, 2007, Accepted, 28th September, 2007, Published online, 28th September, 2007. COM-07-S(W)46

SYNTHESIS OF (±)-LAUTHISAN[†]

Naoki Miyakoshi, Yuki Ohgaki, Kosuke Masui, and Chisato Mukai*

Division of Pharmaceutical Sciences, Graduate School of Natural Science and Technology, Kanazawa University, Kakuma-machi, Kanazawa 920-1192, Japan cmukai@kenroku.kanazawa-u.ac.jp

Abstract – A new total synthesis of (±)-lauthisan was accomplished based on the endo-mode ring-closing reaction of 1-(5-hydroxyhept-1-yl)-3-pentyl-1-phenylsulfonylallene.

As a part of our programs¹ on the development of the efficient methods for the preparation of various kinds of ring-closed products, we reported a novel method for the synthesis of five- to medium-sized oxacycles **2** based on the endo-mode ring-closing reaction of 1,1-disubstituted allenes **1**, which possess both an electron-withdrawing group and the ω -hydroxyalkyl appendage at C₁-position of the allenyl moiety (Scheme 1).² This paper deals with several experiments regarding the scope of the endo-mode ring-closing reaction using tri-and tetrasubstituted allenes and its application to synthesis of (±)-lauthisan³ as a preliminary examination for total syntheses of marine natural products, laurencin and laurefucin.



Scheme 1. Endo-mode ring-closing reaction of allenyl alcohols

The tri- and tetrasubstituted allenes **3** and **4** for the ring-closing reaction were prepared by the similar procedure described for the preparation of 1.² The ring-closing reaction of **3** was carried out according to the previously optimized conditions² for the ring-closing reaction of 1,1-disubstituted allenes **1**. Thus, compound **3a** was exposed to *t*-BuOK (1.5 equiv) in *t*-BuOH at room temperature for less than 5 min to

[†] This paper is dedicated to Professor Dr. Ekkehard Winterfeldt on the occasion of his 75th birthday.

produce the *endo*-double bond derivative, *endo*-**5a**, along with the *exo*-double bond isomer, *exo*-(Z)-**5a** (87%: *endo*-**5a** : *exo*-(Z)-**5a** = 73 : 27) (Table 1, Entry 1).⁴ Similar treatment of the one carbon homologated **3b** predominantly furnished the *exo*-double bond derivative, *exo*-(E)-**5b**, along with *endo*-**5b** (89%: *endo*-**5b** : *exo*-(E)-**5b** = 19 : 81) (Entry 2). The exclusive formation of the *exo*-double bond derivatives, *exo*-(E)-**5c** and *exo*-(E)-**5d**, were recorded when **3c** and **3d** were submitted to the ring-closing conditions (Entries 3,4).

он	SO ₂ Ph Me 3	<i>t-</i> BuOK ^a <i>t-</i> BuOH (0.1 M) rt	SO ₂ Ph O Me endo-5	+ 0-	() exo-5	SO₂Ph _∖ , Me
Entry	Substrate	n	Products and Yield (%)	endo	: e	хо
1	3a	1	5a (87)	73	: 2	27 (<i>Z</i>) ^b
2	3b	2	5b (89)	19	: 8	31 (<i>E</i>) ^b
3	3c	3	5c (84)	0	: 10)0 (<i>E</i>)
4	3d	4	5d (84)	0	: 10)0 (<i>E</i>)

^a The starting **3** was completely consumed within 5 min.

^b The ratio was determined on the basis of isolated amounts of each isomer.

Table 1. Ring-closing reaction of 1,1,3-trisubstituted allenes

Independent exposure of *endo*-**5a** and *exo*-**5a** to the ring-closing conditions for a longer reaction time (1 h) provided an equilibrium mixture of *endo*-**5a** and *exo*-**5a** in a ratio of 90 to 10. The six-membered *endo*-**5b** was found to be stable under the ring-closing conditions, while exo-(*E*)-**5b** was completely isomerized to *endo*-**5b** within 5 h. These experiments revealed that *endo*-**5a**, **b** must be thermodynamically controlled products. It should be mentioned that both seven- and eight-membered products *exo*-**5c**, **d** were inactive toward base-catalyzed isomerization.

We next investigated the endo-mode ring-closing reaction of tetrasubstituted allenes **4**. Exposure of **4** to the standard basic conditions consistently gave the *exo*-double bond isomers, *exo*-**6**, in high yields and the corresponding *endo*-double bond derivatives could not be detected in the reaction mixture (Table 2). The *exo*-**6** was shown to be stable under the basic conditions and no isomerization to the corresponding *endo*-derivatives could be observed. Thus, it can be concluded that the endo-mode ring-closing reaction of phenylsulfonylallenes, leading to the formation of five- to eight-membered oxacycles, proceeds irrespective of the substitution pattern at the allenic terminus.

	∠SO₂Ph	<i>t-</i> BuOK	- í	⟨-)SO₂Ph				
Сп Ц	N	t-BuOH	Ö	Me				
Me Me		(0.1M) rt	 Me					
4				<i>exo</i> -6				
Entry	Substrate	n	Time ^a (min)	Product and Yield (%)				
1	4a	1	> 5	6a (91)				
2	4b	2	10	6b (90)				
3	4c	3	20	6c (94)				
4	4d	4	180	6d (84)				
^a The reaction mixture was guenched when the								

complete consumption of 4 was detected by TLC.

Table 2. Ring-closing reaction of tetrasubstituted allenes

Synthesis of (±)-lauthisan (11) by taking advantage of the endo-mode ring-closing reaction of 1,1,3-trisubstituted allene was the next subject in this paper. Thus, protection of the secondary hydroxyl moiety of the known non-1-yn-7-ol (7)⁵ with the TBS was followed by treatment with *n*-BuLi in THF at -78 °C and the resulting acetylide was quenched by hexanal to give the propargyl alcohol 8⁶ in 77% yield. Compound 8 was then converted into the trisubstituted allene 9⁶ in 94% yield. Exposure of 9 to the standard ring-closing conditions (*t*-BuOK in *t*-BuOH at room temperature for 40 min) furnished (*E*)-8-ethyl-2-hexylidene-3-(phenylsulfonyl)oxocane (10)⁷ in 81% yield. The final stage of this synthesis was the removal of a phenylsulfonyl group as well as reduction of enol ether moiety. Treatment of 10 with Mg in MeOH at 50 °C⁸ effected dephenylsulfonylation to produce the corresponding eight-membered oxacycle, which was subsequently exposed to Et₃SiH in THF in the presence of TMSOTf at -78 °C⁹ to afford (±)-lauthisan (11)^{10,11} in 73% yield (Scheme 2).



Scheme 2. Synthesis of (±)-lauthisan

In conclusion, we have described here that (i) tri- and tetrasubstituted allenes were suitable substrates in the endo-mode ring-closing reaction and (ii) successful application of our method to completion of the total synthesis of (\pm) -lauthisan (11) from the known non-1-yn-7-ol (7).

ACKNOWLEDGEMENTS

This work was supported in part by a Grant-in-Aid for Scientific Research from the Ministry of Education, Culture, Sports, Science and Technology, Japan, for which we are grateful.

REFERENCES

- (a) C. Mukai, R. Ukon, and N. Kuroda, *Tetrahedron Lett.*, 2003, 44, 1583. (b) C. Mukai, M. Kobayashi, S. Kubota, Y. Takahashi, and S. Kitagaki, *J. Org. Chem.*, 2004, 69, 2128. (c) C. Mukai, N. Kuroda, R. Ukon, and R. Itoh, *J. Org. Chem.*, 2005, 70, 6282. (d) C. Mukai and Y. Takahashi, *Org. Lett.*, 2005, 7, 5793. (e) N. Kuroda, Y. Takahashi, K. Yoshinaga, and C. Mukai, *Org. Lett.*, 2006, 8, 1843. (f) S. Kitagaki, S. Teramoto, and C. Mukai, *Org. Lett.*, 2007, 9, 2549.
- (a) C. Mukai, H. Yamashita, and M. Hanaoka, *Org. Lett.*, 2001, **3**, 3385. (b) C. Mukai, M. Ohta, H. Yamashita, and S. Kitagaki, *J. Org. Chem.*, 2004, **69**, 6867. (c) S. Kitagaki, D. Shibata, and C. Mukai, *Tetrahedron Lett.*, 2007, **48**, 1735.
- For synthesis of lauthisan, see: (a) R. W. Carling and A. B. Holmes, J. Chem. Soc., Chem. Commun., 1986, 565. (b) H. Kotsuki, Y. Ushio, I. Kadota, and M. Ochi, J. Org. Chem., 1989, 54, 5153. (c) K. Tsushima and A. Murai, Chem. Lett., 1990, 761. (d) K. C. Nicolaou, D. G. McGarry, P. K. Somers, B. H. Kim, W. W. Ogilvie, G. Yiannikouros, C. V. C. Prasad, C. A. Veale, and R. R. Hark, J. Am. Chem. Soc., 1990, 112, 6263. (e) L. A. Paquette and T. J. Sweeney, J. Org. Chem., 1990, 55, 1703. (f) J. U. Udding, J. P. M. Giesselink, H. Hiemstra, and W. N. Speckamp, J. Org. Chem., 1994, 59, 6671. (g) H. Kim, C. Ziani-Cherif, J. Oh, and J. K. Cha, J. Org. Chem., 1995, 60, 792. (h) Y.-G. Suh, B.-A. Koo, E.-N. Kim, and N.-S. Choi, Tetrahedron Lett., 1995, 36, 2089. (i) M. J. Coster and J. J. De Voss, Org. Lett., 2002, 4, 3047. (j) H. J. Rhee, H. Y. Beom, and H.-K. Kim, Tetrahedron Lett., 2004, 45, 8019. (k) M. C. Carreño, R. Des Mazery, A. Urbano, F. Colobert, and G. Solladie, Org. Lett., 2005, 7, 2039. (l) N. Ortega, T. Martín, and V. S. Martín, Org. Lett., 2006, 8, 871.
- 4. The stereochemistry of *exo-5* was determined by NMR spectral consideration.
- 5. D. E. Ames and G. M. R. Davison, Chem. Phys. Lipids, 1974, 13, 223.
- Compounds 8 and 9 must be a mixture of two diastereoisomers judging from the reaction conditions used. However, their ¹H and ¹³C NMR spectra appeared as if they were a single isomer.
- 7. Compound 10 was obtained as a single isomer. The (E)-stereochemistry was tentatively determined by comparison with the related compounds, although the relative stereochemistry of two chiral

centers was not determined yet.

- 8. T. Luker and R. J. Withby, *Tetrahedron Lett.*, 1996, 37, 7661.
- 9. Y. Zhao and P. Quayle, *Tetrahedron Lett.*, 1994, 35, 4179.
- 10. The structure of synthetic lauthisan was confirmed by comparison with spectral data in the literatures.
- 11. The stereoselective reduction of **10** might be rationalized in terms of the attack of the hydride species from the sterically less hindered face (α -face) on the basis of the literature precedent.^{3k}