HETEROCYCLES, Vol. 74, 2007, pp. 849 - 853. © The Japan Institute of Heterocyclic Chemistry Received, 26th September, 2007, Accepted, 15th November, 2007, Published online, 16th November, 2007. COM-07-S(W)77

SERAGADINE A, A β-CARBOLINE ALKALOID FROM MARINE SPONGE

Kohei Nozawa,^a Masashi Tsuda,^a Takaaki Kubota,^a Jane Fromont,^b and Jun'ichi Kobayashi^{a,*}

a: Graduate School of Pharmaceutical Sciences, Hokkaido University, Sapporo 060-0812, Japan; b: Western Australian Museum, Locked Bag 49, Welshpool DC, WA 6986, Australia. E-mail: jkobay@pharm.hokudai.ac.jp.

Abstract - A new quaternary β -carboline alkaloid, seragadine A (1), and three known related alkaloids (2 ~ 4) have been isolated from an Okinawan Haplosclerida sponge (SS-1022). The structure of 1 was determined by the spectroscopic data and its synthesis.

INTRODUCTION

Many β -carboline alkaloids from natural sources are known to have a variety of bioactivity¹. In our search for bioactive substances from marine invertebrates, some different types of β -carboline alkaloids have been isolated from a tunicate *Eudistoma glaucus*², and sponges of *Amphimedon* sp.³ and family Thorectidae⁴. In our continuing search for new metabolites from marine sponges, we have recently isolated a new β -carboline alkaloid with 2-*N*-methyl and 1-ethyl group, seragadine A (1), together with three known related alkaloids⁵⁻⁹ (2 ~ 4) from an Okinawan Haplosclerida marine sponge (SS-1022). Here we describe the isolation and structure elucidation of 1 and its synthesis.



RESULTS AND DISCUSSION

The sponge (SS-1022) collected off Seragaki, Okinawa, was extracted with MeOH. The MeOH extract was partitioned between EtOAc and H₂O, and subsequently the aqueous layer was extracted with *n*-BuOH. *n*-BuOH-soluble materials of the extract were subjected to SiO₂ gel and C₁₈ column chromatographies followed by C₁₈ HPLC to afford seragadine A (1, 0.00004%, wet weight) together with known related alkaloids, 2-methyl-9*H*- β -carbolin-2-ium^{5,6} (2), 2,9-dimethyl-9*H*- β -carbolin-2-ium^{6,7} (3), and melinonine F^{8,9} (4).

The molecular formula, $C_{14}H_{15}N_2$, of **1** was established by HRESIMS [*m*/*z* 211.1241 (M)⁺, Δ +0.5 mmu]. IR absorptions indicated the existence of OH and/or NH (3243 cm⁻¹) functionalities. The ¹H NMR (Table 1) spectrum of **1** was similar to that of **4**, except for the presence of a methylene (δ_H 3.55, q) and methyl (δ_H 1.51, t) resonances in **1**. The molecular weight of **1** was larger than that of **4** by 14 mass units, indicating that a methyl group at C-1 in **4** was replaced by an ethyl one in **1**, which was also supported by the COSY correlation for H₂-11/H₃-12. Thus, seragadine A (**1**) was elucidated to be 1-ethyl-2-methyl-9*H*- β -cabolin-2-ium.

	Seragadine A (1)	synthetic 1
position	δ _H [ppm]	$\delta_{\rm H}$ [ppm]
3	8.53 (d, 6.9)	8.48 (d, 6.6)
4	8.48 (d, 6.9)	8.47 (d, 6.6)
5	7.80 (d, 6.9)	7.80 (d, 8.4)
6	7.51 (t, 7.5)	7.50 (t, 7.4)
7	7.85 (t, 8.1)	7.83 (t, 8.3)
8	8.42 (d, 8.1)	8.40 (d, 8.1)
10	4.47 (s) 3H	4.50 (s)
11	3.55 (q, 7.6) 2H	3.58 (q, 7.7)
12	1.51 (t, 7.7) 3H	1.54 (t, 7.7)

Table 1. ¹H NMR Data for Seragadine A (1) and Synthetic 1 in CD₃OD

Because of a small amount of isolated sample of **1**, we planned to prepare **1** by synthesis from tryptamine (Scheme 1). A TFA-catalyzed Picter–Spengler reaction¹⁰ was utilized to form the tetrahydro- β -carboline (**5**). Treatment of **5** with palladium black in boiling toluene for 2 h produced the ring *C*-dehydrogenated product¹¹ (**6**) in 83% yield. The quaternary iminium (**1**) was prepared from **6** by using CH₃I in DMF at room temperature for 12 h¹². The ¹H NMR data (Table 1) and C₁₈ HPLC profile (Figure 1) of synthetic **1** were identical with those of natural specimen. Thus, the structure of seragadine A (**1**) was confirmed to be 1-ethyl-2-methyl- β -carbolin-2-ium.

Seragadine A (1) showed antibacterial activity and a weak cytotoxicity against L1210 murine leukemia cells in vitro.



Scheme 1. Synthesis of seragadine A (1)



Figure 1. HPLC profiles of seragadine A (1) (a), synthetic 1 (b), and co-injection of them (c).

EXPERIMENTAL

General Experimental Procedures

IR and UV spectra were recorded on JASCO FT/IR-230 and Shimadzu UV-1600PC spectrophotometer, respectively. NMR spectra were recorded on a Bruker AMX-600 spectrometer using 2.5 mm micro cells (Shigemi Co., Ltd). The 3.35 and 49.8 ppm resonances of residual CD₃OD were used as internal references for ¹H and ¹³C NMR spectra, respectively. Positive-mode ESIMS was obtained on a JEOL JMS 700-TZ spectrometer using a sample dissolved in MeOH.

Collection, Extraction, and Isolation

The dark brown Haplosclerida sponge (SS-1022) was collected off Seragaki, Okinawa Island and kept frozen until used. The sponge (2.6 kg, wet weight) was extracted three times with MeOH (2 L) and then evaporated to give a residue (66.12 g). A portion of the residue (24.78 g) was partitioned between *n*-hexane (500 mL X 2) and 90% aqueous MeOH (500 mL). The 90% aqueous MeOH-soluble portion was evaporated under reduced pressure and partitioned between EtOAc (500 mL) and H₂O (500 mL).

Water-soluble portion was extracted with n-BuOH (500 mL). A part (257 mg) of the n-BuOH-soluble fraction (513 mg) was subjected to a silica gel column chromatography (Wako gel C-300, Wako Pure Chemical, 2 X 30 cm) with CHCl₃/MeOH. The fraction eluted with CHCl₃/MeOH (3:2) was subjected to a C₁₈ Sep-Pak column with 30 % aqueous MeCN and then reversed-phase HPLC (Mightysil RP-18, KANTO CHEMICAL CO., INC., 10 X 250 mm; flow rate 2.5 mL/min; UV detection at 240 nm; eluent MeCN /H₂O/TFA, 18:82:0.1) to afford seragadine A (1, 0.2 mg, 0.00004%, wet weight, t_R 25.2 min), 2-methyl-9*H*- β -carbolin-2-ium (2,0.6 mg, 0.00012%, weight, wet t_R 13.6 min), 2,9-dimethyl-9H- β -carbolin-2-ium (3, 0.4 mg, 0.00008%, t_R 20.4), and melinonine F (4, 0.5 mg, $0.00010\%, t_R 21.0)$

Seragadine A (1): colorless oil; IR (neat) v_{max} 3243 cm⁻¹; UV (MeOH) λ_{max} 308 nm (17000); ¹H NMR (see Table 1); ESIMS *m/z* 211 (M)⁺, HRESIMS *m/z* 211.1241 [calcd. for C₁₄H₁₅N₂, (M)⁺, 211.1236].

Synthesis of Seragadine A (1): To a stirred solution of 1-ethyl- β -carboline (6, 145 mg, 0.74 mmol) in DMF (2 mL) was added methyl iodide (120 mL) under an argon atmosphere. After stirring at rt for 48 h, the reaction mixture was concentrated *in vacuo*. The residue was purified by an amino silica gel column chromatography (CHCl₃/MeOH, 9 : 1) to give 1-ethyl-2-methyl- β -carboline (140 mg, 0.66 mmol, 90%). : colorless amorphous solid; IR (film) v_{max} 3156, 3075, 2981, 2902, 2787, 2682, 1683, 1633, 1198, 1126 cm⁻¹; UV (MeOH) λ_{max} 308 (ϵ 17000); ¹H NMR (see Table 1); ¹³C NMR (CD₃OD, 500 MHz) δ 146.9, 146.4, 136.8, 134.7, 133.9, 125.0, 123.9, 122.3, 120.2, 114.6, 97.2, 45.6, 24.1, 12.8. ESIMS *m/z* 211 (M)⁺, HRESIMS *m/z* 211.1231 [calcd. for C₁₄H₁₅N₂, (M)⁺, 211.1236].

HPLC Analysis of Synthetic 1 and Seragadine A (1). The synthetic 1 and seragadine A (1) were subjected to C18 HPLC analyses (Mightysil RP-18, KANTO CHEMICAL CO., INC., 10 X 250 mm; flow rate 1.0 mL/min; UV detection at 240 nm; eluent MeCN/H₂O (20:80)). The retention times of synthetic 1 and seragadine A (1) were both 13.3 min (Figure 1).

ACKNOWLEDGEMENTS

Authors thank Ms. S. Oka, Center for Instrumental Analysis, Hokkaido University, for measurements of ESIMS. This work was partly supported by a grant from the Yakult Bio-Science Foundation and a Grant-in-Aid for Scientific Research from the Ministry of Education, Culture, Sports, Science and Technology of Japan.

REFERENCES

- 1. Y. Nakamura, J. Kobayashi, J. Gilmore, M. Mascal, K. L. Rinehart, Jr., H. Nakamura, and Y. Ohizumi, *J. Biol. Chem.*, 1985, **261**, 4139.
- 2. J. Kobayashi, G. C. Harbour, J. Gilmore, and K. L. Rinehart, Jr., J. Am. Chem. Soc., 1984, 106, 1526.
- 3. M. Tsuda, N. Kawasaki, and J. Kobayashi, *Tetrahedron Lett.*, 1994, 35, 4387.
- T. Kubota, T. Nishi, E. Fukushi, J. Kawabata, J. Fromont, and J. Kobayashi, *Tetrahedron Lett.*, 2007, 48, 4983.
- 5. R. Speitel and E. Schlittler, Helv. Chim. Acta, 1949, 32, 860.
- 6. A. Ruiz, P. Rocca, F. Marsais, A. Godard, and G. Queguiner, Tetrahedron Lett., 1997, 38, 6205.
- 7. S. Ghosal, S. K. Bhattacharya, and R. Mehta, J. Pharm. Sci., 1972, 61, 808.
- 8. A. Gray, E. Spinner, and C. Cavallito, J. Am. Chem. Soc., 1954, 76, 2792.
- 9. D. Nunes, L. Koike, J. Taveira, and F. Reis, *Phytochemistry*, 1992, **31**, 2507.
- (a) J. Zhang, G. Wang, P. Xie, S. Chen, and X. Liang, *Tetrahedron Lett.*, 2000, **41**, 2211. (b) R. Tschesche and H. Jenssen, *Chem. Ber.*, 1960, **93**, 271.
- 11. E. Spath and E. Lederer, Ber. Chem. Ges., 1930, 63, 2102.
- T. V. Stupnikova, A. R. Kirilash, N. A. Klyuev, and A. A. Perov, *Khimi. Geterotsikl. Soed.*, 1985, 4, 530.