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# SULFURIC ACID MEDIATED HETEROCYCLIZATION OF *ORTHO*-CYANOMETHYLNITROARENES TO BENZO[*C*]ISOXAZOLES AND FUSED BENZO[*C*]ISOXAZOLES

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**Abstract** – The vicarious nucleophilic substitution of hydrogen (VNS) is used as the key step to convert substituted 5-nitrobenzimidazoles **1a-b** into their *ortho*cyanomethyl derivatives **2a-b**. Conc. sulfuric acid mediated heterocyclization of these intermediates gave the novel 3*H*-imidazo[4',5':3,4]benzo[*c*]isoxazole-8carboxamides **3a-b**. To generalize this synthetic strategy for benzo[*c*]isoxazoles syntheses, the VNS products of *para*-substituted nitrobenzene **4a-d** were successfully converted to the new benzo[*c*]isoxazoles derivatives **5a-d**.

### **INTRODUCTION**

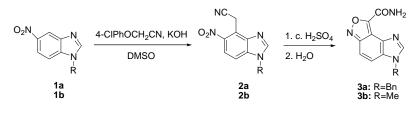
Benzo[*c*]isoxazoles (2,1-benzisoxazoles) derivatives are prescribed as antipsychotic risperidone drugs<sup>1</sup> and play a key role in many organic reactions,<sup>2</sup> notably those leading to anthranilic acids. There are several methods for the synthesis of these compounds. The formation of benzo[*c*]isoxazoles from *ortho* nitrobenzene derivatives is generally catalyzed both by acids and by bases and can also be initiated thermally or photochemically. A variety of *ortho*-nitrobenzylcarbonyl derivatives cyclize under both acidic and basic conditions and thermally, to afford the simple benzo[*c*]isoxazoles derivatives.<sup>3-5</sup> The sole structural requirement for the success of these cyclizations appears to be depended upon the presence of a moderately acidic benzylic C-H group. Conversion of *ortho*-nitrobenzylcarboxylic acids in hot concentrated sulfuric acid to benzo[*c*]isoxazoles derivatives has been reported in early literature.<sup>6</sup> To the best of our knowledge, heterocyclization of *ortho*-cyanomethylnitroarenes to benzo[*c*]isoxazoles have not

been investigated in conc. sulfuric acid. Owing to our growing interest in the synthesis of bioactive heterocycles and exploration of their synthetic pathways,<sup>7</sup> we became interested in the transformation of *ortho* cyanomethyl nitroarenes to the new benzo[*c*]isoxazoles and fused benzo[*c*]isoxazoles. The vicarious nucleophilic substitution of hydrogen<sup>8</sup> (VNS) is a highly efficient synthetic tool for the introduction of various functionalized substituents into the *ortho*-position relative to nitro group. Nitroarenes containing functionalized substituents in the *ortho*-position are valuable synthones in a variety of cyclocondensation reactions yielding potentially biologically active heterocycles such as indoles, benzimidazoles, quinolines, fused pyrimidines, fused pyrazines, etc (see a recent review and references cited therein).<sup>9</sup>

In this work, we have introduced the CH<sub>2</sub>CN moiety into the *ortho*-position of nitro group in nitrobenzene derivatives (**4a-d**) and *N*-alkyl substituted 5-nitrobenzimidazoles (**1a-b**) with the aim to synthesize the new derivatives of benzo[c]isoxazoles (**5a-d**) and fused benzo[c]isoxazoles (**3a-b**) *via* conc. sulfuric acid catalysis at room temperature.

#### **RESULTS AND DISCUSSION**

The key intermediates 2-(1-benzyl-5-nitro-1*H*-benzo[*d*]imidazol-4-yl)acetonitrile (**2a**) and 2-(1-methyl-5-nitro-1*H*-benzo[*d*]imidazol-4-yl)acetonitrile (**2b**) were obtained *via* the VNS reaction of *N*-benzyl- and *N*-methyl-5-nitrobenzimidazoles **1a-b** with 4-chlorophenoxyacetonitrile in basic DMSO solution. The compounds **2a-b** were cyclized to fused benzo[*c*]isoxazoles **3a-b** in conc. sulfuric acid at room temperature (Scheme 1).

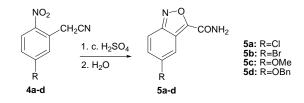


Scheme 1

The structural assignments of compounds **3a-b** were based on the analytical and spectral data. For example, in the <sup>13</sup>C NMR spectrum of 3-benzyl-3*H*-imidazo[4',5':3,4]benzo[*c*]isoxazole-8-carboxamide (**3a**), the signal at 17.39 ppm attributed to CH<sub>2</sub>CN carbon atom of compound **2a** is absent but instead there is a signal at 131.57 ppm for an aromatic carbon atom which is a clear indication of the cyclization step which led to the formation of the third aromatic ring. In the <sup>1</sup>H NMR spectrum of **3a** the signal at 4.64 ppm assignable to cyanomethyl protons of compound **2a** is not present but instead two signals attributed to two exchangeable protons (amide NH<sub>2</sub> group) appeared at 8.45 and 9.55 ppm which is a

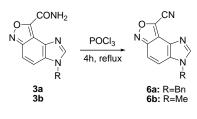
further indication of the heterocyclization step. Moreover, the FT-IR spectrum of **3a** in KBr showed two different absorption bands at 3170 cm<sup>-1</sup> and 3380 cm<sup>-1</sup> assignable to amide  $NH_2$  group, 1698 cm<sup>-1</sup> and 1670 cm<sup>-1</sup> assignable to C=O group. All this evidence plus the molecular ion peak at m/z 292 and microanalytical data strongly support the cyclic structure of compound **3a**.

Heterocyclization of *ortho*-cyanomethylnitrobenzenes **4a-d** in conc. sulfuric acid afforded the new derivatives of benzo[c]isoxazoles **5a-d** (Scheme 2).



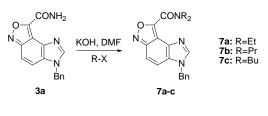
Scheme 2

Dehydration of **3a-b** compounds in boiling POCl<sub>3</sub> gave the new derivatives of 3-benzyl-3*H*-imidazo-[4',5':3,4]benzo[*c*]isoxazol-8-yl cyanide (**6a**) and 3-methyl-3*H*-imidazo[4',5':3,4]benzo[*c*]isoxazol-8-yl cyanide (**6b**).



Scheme 3

Alkylation of **3a** with different alkyl halides in DMF and KOH gave the dialkyl derivatives **7a-c** (Scheme 4).



Scheme 4

In summary, this research has demonstrated that the sequential VNS reaction of nitroarenes and heterocylization of the *ortho* substituted products in the presence of conc. sulfuric acid is a reliable strategy for the synthesis of new benzo[c]isoxazoles and fused benzo[c]isoxazoles. This work can be extended to the synthesis of other novel heterocyclic compounds.

### **EXPERIMENTAL**

Melting points were recorded on an Electrothermal type 9100 melting point apparatus. The IR spectra were obtained on a 4300 Shimadzu spectrometer and only noteworthy absorptions are listed. The <sup>13</sup>C NMR (125MHz) spectra were recorded on a Bruker Avance DRX-500 Fouriertransformer spectrometer. The <sup>1</sup>H NMR (100MHz) spectra were recorded on a Bruker AC 100 spectrometer. Chemical shifts are reported in ppm downfield from TMS as internal standard; coupling constance *J* are given in Hertz. The mass spectra were scanned on a Varian. Mat CH-7 at 70 ev. Elemental analysis was performed on a Thermo Finnigan Flash EA microanalyzer.

4-Chlorophenoxyacetonitrile,<sup>10</sup> compounds  $1a-b^{11}$  and  $4a-d^{12}$  were obtained according to the published methods. Other reagents were commercially available.

## 1. General procedure for the synthesis of *ortho* disubstituted benzo[d]imidazol-4-yl)acetonitrile 2a-b.

To a solution of the *N*-substituted 5-nitrobenzimidazoles **1a-b** (19.7 mmol) and 4-cholrophenoxyacetonitrile (20 mmol) in DMSO (90 mL), powdered KOH (5.6 g, 100 mmol) was added. The reaction mixture was stirred for 3 h at rt, and then poured onto crushed ice containing (10 mL) hydrochloric acid. The precipitate was collected by filtration and recrystallized from EtOH to give the pure products **2a-b**.

**1.1. 2-(1-Benzyl-5-nitro-1***H***-benzo[***d***]imidazol-4-yl)acetonitrile (2a).** Compound 2a was obtained as bright yellow crystals, yield (72%), mp 161-163 °C; <sup>1</sup>H NMR (100 MHz, CDCl<sub>3</sub>) δ 4.64 (s, 2H), 5.43 (s, 2H), 7.2-7.4 (m, 6H), 8.15 (s, 1H), 8.23 (d, *J*=10.9 Hz, 1H); <sup>13</sup>C NMR (125MHz, DMSO-*d*<sub>6</sub>): δ 149.15, 143.50, 143.00, 137.61, 136.97, 129.72, 128.92, 128.36, 120.93, 119.92, 118.23, 112.24, 49.07, 17.34; IR (KBr): 1350, 1525 (NO<sub>2</sub>), 2250 cm<sup>-1</sup> (CN). MS, m/z (%): 292 (M<sup>+</sup>, 33), 273 (42), 251(12), 207 (38), 127 (34), 91 (100), 65 (53). Anal. Calcd for C<sub>16</sub>H<sub>12</sub>N<sub>4</sub>O<sub>2</sub> (292.3): C, 64.74; H, 3.62; N, 20.13. Found: C, 64.92; H, 3.59; N, 20.09.

**1.2. 2-(1-Methyl-5-nitro-1***H***-benzo[***d***]imidazol-4-yl)acetonitrile (2b). Compound 2b was obtained as bright yellow crystals, yield (70%), mp 188-190 °C; <sup>1</sup>H NMR (100 MHz, DMSO-***d***<sub>6</sub>) \delta 3.92 (s, 3H), 4.58 (s, 2H), 7.9 (d,** *J* **=14 Hz, 1H), 8.2 (d,** *J* **=14 Hz, 1H), 8.55 (s, 1H); IR (KBr): 1350, 1525 (NO<sub>2</sub>), 2250 cm<sup>-1</sup> (CN). MS (m/z) 216 (M<sup>+</sup>). Anal. Calcd for C<sub>10</sub>H<sub>8</sub>N<sub>4</sub>O<sub>2</sub> (216.2): C, 53.47; H, 2.99; N, 27.71. Found: C, 53.38; H, 2.85; N, 27.54.** 

# 2. General procedure for the synthesis of fused benzo[c]isoxazoes (3a-b) and benzo[c]isoxazoes (5a-d).

To concentrated sulfuric acid (6 mL), which was kept in an ice-bath, compounds **2a-b** and **4a-d** (6.8 mmol) was gradually added, with stirring. The inside temperature was kept between 15-20 °C. The

addition was accomplished over a period of 1 h, the solution was stirred at rt for further 4 h, then water (8 mL) was added to the solution in an ice-bath and it was stirred for further 2 h before it was poured onto crushed ice, and finally it was neutralized with dilute aqueous NaOH. The reaction mixture which was allowed to reach 50-70 °C during the neutralization, was cooled to rt, filtered and washed with water and then  $CH_2Cl_2$ , to give the pure products **3a-b**.

**2.1. 3-Benzyl-3***H***-imidazo[4',5':3,4]benzo[***c***]isoxazole-8-carboxamide (3a). Compound 3a was obtained as colorless crystals (1,4-dioxane), yield (90%), mp 260-263 °C; <sup>1</sup>H NMR (100MHz, DMSO-***d***<sub>6</sub>) \delta 5.65 (s, 2H), 7.34 (s, 5H), 7.61 (d,** *J* **= 12.0 Hz, 1H), 7.81 (d,** *J* **= 12.0 Hz, 1H), 8.45 (br s, 1H), 8.61 (s, 1H), 9.55 (br s, 1H); <sup>13</sup>C NMR (125MHz, DMSO-***d***<sub>6</sub>): \delta 157.93, 157.79, 155.39, 142.85, 137.36, 131.57, 130.31, 129.74, 128.92, 128.26, 120.74, 111.97, 111.13, 49.24; IR (KBr): 3170, 3380 cm<sup>-1</sup> (NH<sub>2</sub>), 1698, 1670 cm<sup>-1</sup> (C=O). MS, m/z (%): 292 (M<sup>+</sup>, 31), 290 (99), 264 (26), 248 (40), 173 (45), 91 (100), 65 (62). Anal. Calcd for C<sub>16</sub>H<sub>12</sub>N<sub>4</sub>O<sub>2</sub> (292.3): C, 65.75; H, 4.14; N, 19.17. Found: C, 65.54; H, 4.11; N, 18.94.** 

**2.2. 3-Methyl-3***H***-imidazo[4',5':3,4]benzo[***c***]isoxazole-8-carboxamide (3b). Compound 3b was obtained as colorless crystals (1,4-dioxane), yield (86%), mp 270-272 °C; <sup>1</sup>H NMR (100MHz, DMSO-***d***<sub>6</sub>) \delta 3.97 (s, 3H), 7.6 (d,** *J***=10.0 Hz, 1H), 7.9 (d,** *J***=10.0 Hz, 1H), 8.35 (br s, 1H), 8.45 (s, 1H), 9.55 (br s, 1H); IR (KBr): 3170, 3380 cm<sup>-1</sup> (NH<sub>2</sub>), 1698, 1670 cm<sup>-1</sup> (C=O). MS (m/z) 216 (M<sup>+</sup>). Anal. Calcd for C<sub>10</sub>H<sub>8</sub>N<sub>4</sub>O<sub>2</sub> (216.2): C, 55.56; H, 3.73; N, 25.91. Found: C, 55.95; H, 3.57; N, 25.77.** 

**2.3. 5-Chlorobenzo**[*c*]isoxazole-3-carboxamide (5a). Compound 5a was obtained as colorless crystals (acetone), yield (75%), mp 226-227 °C; <sup>1</sup>H NMR (100MHz, DMSO-*d*<sub>6</sub>) δ 7.47 (d, *J*=9.0 Hz, 1H), 7.86 (d, *J*=9.0 Hz, 1H), 7.95 (s, 1H), 8.25 (br s, 1H), 8.69 (br s, 1H); IR (KBr): 3185, 3390 cm<sup>-1</sup> (NH<sub>2</sub>), 1690, 1665 cm<sup>-1</sup> (C=O). MS (m/z) 197 (M<sup>+</sup>). Anal. Calcd for C<sub>8</sub>H<sub>5</sub>ClN<sub>2</sub>O<sub>2</sub>(196.6): C, 48.88; H, 2.56; N, 14.25. Found: C, 48.49; H, 2.51; N, 14.03.

**2.4. 5-Bromobenzo**[*c*]isoxazole-3-carboxamide (5b). Compound 5b was obtained as colorless crystals (acetone), yield (55%), mp 235-237 °C; <sup>1</sup>H NMR (100MHz, DMSO-*d*<sub>6</sub>) δ 7.60 (d, *J*=10.0 HZ, 1H), 7.80 (d, *J*=10.0 Hz, 1H), 8.16 (s, 1H), 8.24 (br s, 1H), 8.67 (br s, 1H); IR (KBr): 3185, 3390 cm<sup>-1</sup> (NH<sub>2</sub>), 1690, 1665 cm<sup>-1</sup> (C=O). MS (m/z) 241 (M<sup>+</sup>). Anal. Calcd for C<sub>8</sub>H<sub>5</sub>BrN<sub>2</sub>O<sub>2</sub> (241.0): C, 39.86; H, 2.09; N, 11.62. Found: C, 39.64; H, 1.99; N, 11.45.

**2.5. 5-Methoxybenzo**[*c*]isoxazole-3-carboxamide (5c). Compound 5c was obtained as colorless crystals (acetone), yield (59%), mp 232-234 °C; <sup>1</sup>H NMR (100MHz, DMSO-*d*<sub>6</sub>) δ 3.96 (s, 3H), 7.29 (d, *J*=10.0 HZ, 1H), 7.35 (s, 1H),7.50 (d, *J*=10.0 Hz, 1H), 8.15 (br s, 1H), 8.78 (br s, 1H); IR (KBr): 3185, 3390 cm<sup>-1</sup> (NH<sub>2</sub>), 1690, 1665 cm<sup>-1</sup> (C=O). MS (m/z) 192 (M<sup>+</sup>). Anal. Calcd for C<sub>9</sub>H<sub>8</sub>N<sub>2</sub>O<sub>3</sub> (192.2): C, 56.25; H, 4.20; N, 14.58. Found: C, 55.95; H, 4.16; N, 14.42.

**2.6. 5-(Benzyloxy)benzo**[*c*]isoxazole-3-carboxamide (5d). Compound 5d was obtained as colorless crystals (acetone), yield (69%), mp 228-230 °C; <sup>1</sup>H NMR (100MHz, DMSO-*d*<sub>6</sub>)  $\delta$  5.45 (s, 2H), 7.21 (s,

5H), 7.29 (s, 1H), 7.37 (d, J=10.0 Hz, 1H), 7.54 (d, J=10.0 Hz, 1H), 8.26 (br s, 1H), 8.77 (br s, 1H); IR (KBr): 3185, 3390 cm<sup>-1</sup> (NH<sub>2</sub>), 1690, 1665 cm<sup>-1</sup> (C=O). MS (m/z) 268 (M<sup>+</sup>). Anal. Calcd for C<sub>15</sub>H<sub>12</sub>N<sub>2</sub>O<sub>3</sub> (268.3): C, 67.16; H, 4.51; N, 10.44. Found: C, 66.88; H, 4.49; N, 10.37.

## 3. General procedure for the synthesis of 6a-b from 3a-b.

The mixture of **3a-b** (0.2 g, 0.68 mmol) and POCl<sub>3</sub> (3 mL) was refluxed with stirring for 3 h. After cooling to rt the product poured into crushed ice and neutralized with ammonia solution. The product was extract with EtOAC ( $2 \times 50$  mL). The extract was dried, and evaporated to give pure **6a-b**.

**3.1. 3-Benzyl-3***H***-imidazo[4',5':3,4]benzo[***c***]isoxazol-8-yl cyanide (6a). Compound 6a was obtained as pale yellow crystals, yield (75%), mp 182-184 °C; 1H NMR (100MHz, DMSO-***d6***) δ 5.65 (s, 2H), 7.33 (s, 5H), 7.67 (d,** *J***=10.0 Hz, 1H),7.95 (d,** *J***=10.0 Hz, 1H), 8.56 (s, 1H); <sup>13</sup>C NMR (125MHz, DMSO-***d***<sub>6</sub>): δ 157.71, 144.51, 137.39, 133.22, 131.92, 130.53, 129.79, 128.89, 128.20, 121.96, 119.00, 111.73, 110.58, 49.09; IR (KBr): 2220 cm<sup>-1</sup> (CN). MS (m/z) 274 (M+). Anal. Calcd for C<sub>32</sub>H<sub>20</sub>N<sub>8</sub>O<sub>2</sub> (274.3): C, 70.07; H, 3.67; N, 20.43. Found: C, 70.33; H, 3.72; N, 20.56.** 

**3.2. 3-Methyl-3***H***-imidazo[4',5':3,4]benzo[***c***]isoxazol-8-yl cyanide (6b). Compound 6b was obtained as pale yellow crystals, yield (72%), mp 128-130 °C; 1H NMR (100MHz, CDCl<sub>3</sub>) δ 3.95 (s, 3H), 7.65 (d,** *J***=9.5 Hz, 1H), 7.85 (d,** *J***=9.5 Hz, 1H), 7.92 (s, 1H); IR (KBr): 2220 cm<sup>-1</sup> (CN). MS (m/z) 198 (M+). Anal. Calcd for C<sub>20</sub>H<sub>12</sub>N<sub>8</sub>O<sub>2</sub> (198.18): C, 60.61; H, 3.05; N, 28.27. Found: C, 60.55; H, 3.05; N, 28.39.** 

# 4. General procedure for the synthesis of 7a-c from 3a.

To a solution of **3a** (0.68 mmol) in DMF (5 mL), alkyl halide (1.4 mmol) and KOH (0.34 g, 6 mmol) was added. The mixture was stirred for 1 day and then poured into water. The precipitate was collected by filtration, washed with water and air-dried to give **7a-e**.

**4.1.** *N*,*N*-Diethyl-3-benzyl-3*H*-imidazo[4',5':3,4]benzo[*c*]isoxazole-8-carboxamide (7a). Compound 7a was obtained as yellow crystals (EtOH), yield (65%), mp 184-187 °C; <sup>1</sup>H NMR (100MHz, CDCl<sub>3</sub>)  $\delta$  1.4 (t, *J*=8.0 Hz, 3H), 1.6 (t, *J*=8.0 Hz, 3H), 3.6 (q, *J*=6.7 Hz, 2H) 4 (q, *J*=6.7 Hz, 2H), 5.45 (s, 2H), 7.1-7.5 (m, 7H), 7.95 (s, 1H). MS (m/z) 348 (M<sup>+</sup>). Anal. Calcd for C<sub>20</sub>H<sub>20</sub>N<sub>4</sub>O<sub>2</sub> (348.4): C, 68.95; H, 5.79; N, 16.08. Found: C, 68.73; H, 5.65; N, 15.92.

# 4.2. N,N-Dipropyl-3-benzyl-3H-imidazo[4',5':3,4]benzo[c]isoxazole-8-carboxamide (7b).

Compound **7b** was obtained as yellow crystals (EtOH), yield (67%), mp 129-132 °C; <sup>1</sup>H NMR (100MHz, CDCl<sub>3</sub>)  $\delta$  0.9 (t, *J*=8.0 Hz, 3H), 1.15 (t, *J*=8.0 Hz, 3H), 1.6-2 (m, 4H), 3.4-3.8 (m, 4H), 5.45 (s, 2H), 7.15-7.6 (m, 7H), 7.95 (s, 1H). MS (m/z) 376 (M<sup>+</sup>). Anal. Calcd for C<sub>22</sub>H<sub>24</sub>N<sub>4</sub>O<sub>2</sub> (376.5): C, 70.19; H, 6.43; N, 14.88. Found: C, 70.45; H, 6.20; N, 15.16.

Compound **7c** was obtained as yellow crystals (EtOH), yield (60%), mp 91-94 °C; <sup>1</sup>H NMR (100MHz, CDCl<sub>3</sub>)  $\delta$  0.8-1.95 (m, 14H), 3.35-3.8 (m, 4H), 5.4 (s, 2H), 7.1-7.5 (m, 7H), 7.9 (s, 1H). MS (m/z) 404 (M<sup>+</sup>). Anal. Calcd for C<sub>24</sub>H<sub>28</sub>N<sub>4</sub>O<sub>2</sub> (404.5): C, 71.26; H, 6.98; N, 13.85. Found: C, 71.49; H, 6.79; N, 13.64.

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