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SYNTHESIS OF SPIRO[2-CYCLOPENTENE-1,3'-IMIDAZO[1,2-*a*]-PYRIDINE] DERIVATIVES AND THEIR INTERESTING BEHAVIOR IN ¹H-NMR SPECTRA IN DEUTERIOCHLOROFORM¹

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Abstract – Ethyl 2',3'-dihydro-2-methylthio-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylates were synthesized from the reactions of 3-[bis(methylthio)methylene]-2(3*H*)-imidazo[1,2-*a*]pyridinones with ethyl 4-chloroacetoacetate in the presence of a base. The 2-methylthio group in these spiro compounds was easily replaced with some primary and secondary amines to afford the corresponding 2-amino derivatives. Very interestingly, the proton signals of these spiro compounds in the ¹H-NMR spectra in deuteriochloroform (CDCl₃) changed with an increase in the sample concentration, and the analysis for the magnitude and the direction of each proton shift disclosed the conformational change of the cyclopentenone moiety in this molecule.

INTRODUCTION

For many years, imidazo[1,2-*a*]pyridine ring has been known to be core structural unit of bioactive molecules in synthetic and natural products.² Thus the development of a novel synthetic route and the structural modifications for this scaffold have been intensively investigated with the aim of developing novel active agents. Recently, 2(3H)-imidazo[1,2-*a*]pyridinone derivatives bearing a spiro ring at the 3-position were shown to possess some important pharmacological activities. For example,

spiro[imidazo[1,2-*a*]pyridine-3,2-indan]-2(3*H*)-one (**A**, ZSET1446) was reported to act as an Alzheimer's disease progression inhibitor and cognitive enhancer,³⁻⁵ and spiro[cyclopentane-1,3-imidazo[1,2-*a*]pyridine]-2'(3'*H*)-one (**B**) was shown to act as a progesterone receptor

modulator (see Figure 1).⁶

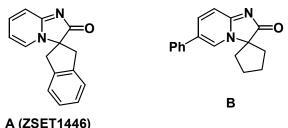


Figure 1

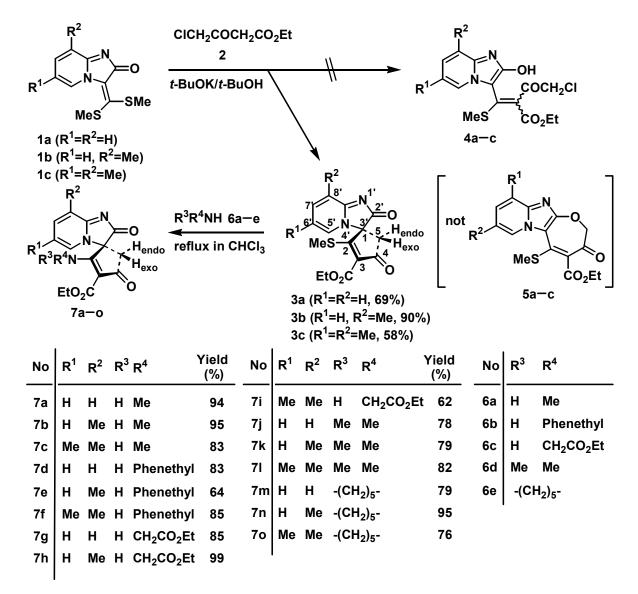
In general, these spiro compounds have been prepared from the reactions of 2(3H)-imidazo-[1,2-*a*]pyridinones, readily obtainable from the alkaline treatment of the corresponding 2-amino-1-(ethoxycarbonylmethyl)pyridinium halides, with 1,4-dihalides such as α,α -dibromo-*o*-xylene or 1,4-dibromobutane under alkaline conditions.⁷

Recently we developed a construction method for new types of 2(3H)-imidazo[1,2-a]pyridinones, 3-[bis(methylthio)methylene]-2(3*H*)-imidazo[1,2-*a*]pyridinones, from the alkaline treatment of 1-[1-carbamoyl-2,2-bis(methylthio)vinyl]pyridinium iodides, which were readily obtained from the S-alkylation of pyridinium 1-carbamoyl-1-[(methylthio)thiocarbonyl]methylides with iodomethane.⁸ We also described the first synthesis of 2*H*-pyrano[2',3':4,5]imidazo[1,2-*a*]pyridin-2-one derivatives through the electrophilic substitution of these 3-methylene-2(3H)-imidazo[1,2-a]pyridinones with the carbanion species generated from some activated ethyl acetates, followed by the intramolecular cyclization of the resulting ethyl 3-[2-hydroxyimidazo[1,2-a]pyridin-3-yl]acrylates with the elimination of an ethanol. Since a high electrophilicity of the 3(1)-position in 3-[bis(methylthio)methylene]-2(3H)-imidazo[1,2-a]pyridinones in these reactions was demonstrated, we were interested next in their reactions with other bifunctionalized reagents. In this paper we report the smooth formation of the title compounds, spiro[2-cyclopentene-1,3'-imidazo[1,2-a]pyridine] derivatives, from the reactions of 3-[bis(alkylthio)methylene]-2(3H)-imidazo[1,2-a]pyridinones with ethyl 4-chloroacetoacetate, and their transformation to the corresponding 2-amino derivatives by nucleophilic reactions with some amines. In addition we also describe the interesting behavior in the ¹H-NMR spectra of the title compounds in CDCl₃.

RESULTS AND DISCUSSION

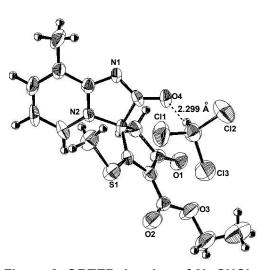
When the reaction of 3-[bis(methylthio)methylene]-2(3*H*)-imidazo[1,2-*a*]pyridinones (1a) with ethyl 4-chloroacetoacetate (2) were carried out in the presence of potassium *t*-butoxide in *t*-butanol at room temperature, the initially expected ethyl 2-chloroacetyl-3-[2-hydroxyimidazo[1,2-*a*]pyridin-3-yl]-3-(methylthio)acrylate (4a) was not formed at all but, instead of it, the colorless crystalline ethyl 2',3'-dihydro-2-methylthio-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylate (3a) was obtained in a 69 % yield. Similar treatment of 1b,c with 2 afforded the corresponding products 3b,c in 90 and 58% yields respectively. Furthermore, the 2-methylthio group in 3a—c was smoothly replaced by the treatment with primary and secondary amines such as methylamine (6a), phenethylamine (6b), ethyl glycinate (6c), dimethylamine (6d), and piperidine (6e) in chloroform at the reflux temperature to afford the corresponding 2-amino derivatives 7a—o in good yields (62—99%) with the elimination of a molecule of methanethiol. These results are shown in Scheme 1.

Elemental analyses for products 3a—c and 7a—o supported our proposed compositions and the fact that



Scheme 1

former products **3a**—**c** do not involve a chlorine atom in the molecule was also indicated by the negative Bilstein test. The IR spectra of **3a**—**c** and **7a**—**o** exhibited the lowered carbonyl band (1602—1640 cm⁻¹) characteristic of the 2(3*H*)-imidazo[1,2-*a*]pyridinones such as **1a**—**c**⁸ and α,β -unsaturated carbonyl bands at 1635—1732 cm⁻¹, together with an amino absorption band (3221—3249 cm⁻¹) for **7a**—**i** and a saturated ester carbonyl band (1738—1746 cm⁻¹) for **7g**—**i**. In addition, the AB type signals for the 5-methylene protons in the ¹H-NMR spectra (See Table 1) of **3a**—**c** and **7a**—**o** appeared near δ 2.6 (H_{endo}) and 3.2 (H_{exo}) and these values are consistent with that (near δ 3.0) expected for such a system, but they are considerably higher than that (near δ 5.1) expected for the methylene protons in the alternative structure, oxepino[2',3':2,3]imidazo[1,2-*a*]pyridine (**5**).⁹ In addition the NOE measurement of **3a** made sure the assignment of the *endo-* and *exo*-protons at the 5-position. That is, the irradiation on the 5'-H signal caused the NOE interaction to the signals of the 6'-H, 5-H_{endo}, and 3-methylthio, and that on the 5-H_{endo} to the 5-H_{exo} and 5'-H. To obtain further information for the structures of **3a**—**c** and **7a**—**o**, we carried out a single crystal X-ray analysis of compound (**3b**) obtained by the recrystallization from CHCl₃-ether and we could finally confirm this structure. ORTEP drawing of **3b** involving a molecule of CHCl₃ is shown in Figure 2.¹⁰ From this crystal data the presence of a significant interaction (hydrogen bonding) between the carbonyl oxygen in 2(3H)-imidazo[1,2-*a*]pyridinone moiety and the methine hydrogen of CHCl₃ could also be presumed, since the distance between the oxygen and the hydrogen is 2.299



Å and is considerably close. Very interestingly, we **Figure 2. ORTEP drawing of 3b:CHCl₃** observed the shifts of each proton signal with an increase in the concentration of **3a**—**c** and **7a**—**o** during the measurement of their ¹H-NMR spectra in CDCl₃. For example, the signals due to the 5-H_{endo}, 5-H_{exo}, 6'-H, 8'-H, 5'-H, and 7'-H, the methylthio and the ethoxy protons of **3a** (its concentration was 0.67×10^{-2} mol/L) appeared at δ 2.66 (d, *J*=18.0 Hz), 3.28 (d, *J*=18.0 Hz), 6.80 (ddd, *J*=6.8, 6.8, 0.8), 7.24 (br d, *J*=9.2 Hz), 7.51 (br d, *J*=6.8 Hz), 7.74 (ddd, J=9.2, 6.8, 1.2 Hz), 2.27 (s), and 1.38 (t, *J*=7.2 Hz) and 4.37 (q, *J*=7.2 Hz) respectively. When the concentration of **3a** was increased to 15.06×10⁻² mol/L, these signals appeared at δ 2.76, 3.22, 6.87, 7.23, 7.74, 7.77, 2.25, and 1.36 and 4.34 respectively. That is, the signals of the 5'-H, 6'-H, 7'-H, and 5-H_{endo} shifted to a lower magnetic field in the range of 0.04—0.23

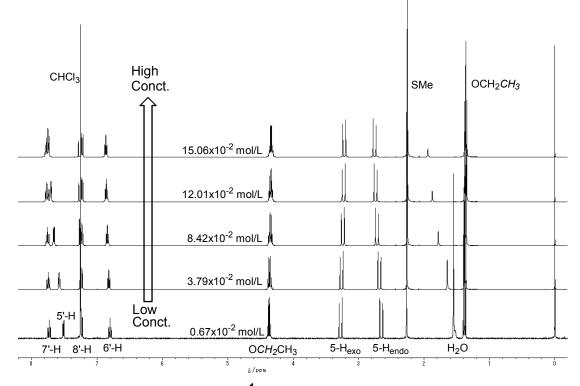


Figure 3. The signal changes in the ¹H-NMR spectra of 3a in some concentrations

ppm and the order of their shift values was 5'-H>5-H_{endo}>6'-H>7'-H, while those of 8'-H, 5-H_{exo}, methylthio, and ethoxy protons moved to a higher magnetic field but the magnitudes were not large (<0.06 ppm). In addition, the shifts in the proton signals of CHCl₃ and H₂O in the solvent were also observed and the movement distance (0.39 ppm) of the latter protons was maximum. The ¹H-NMR spectral changes for some concentrations (0.67, 3.79, 8.42, 12.01, and 15.06×10^{-2} mol/L) of **3a** are shown in Figure 3 and the plots of the shift values for the principal protons are exhibited in Figure 4. An examination of the magnitudes and the directions of their shifts for both the pyridine ring protons and the 5-methylene protons indicated that these shifts were caused by the conformational

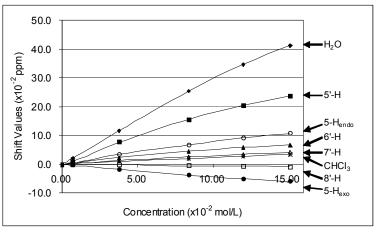
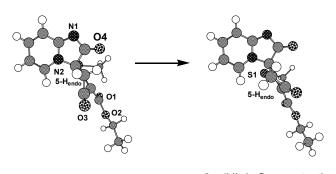


Figure 4. Plots of the shift values of principal protons



3a (Low Concentration) 3a (High Concentration)

Figure 5. Conformational change of the cyclopentenone ring

change (the movement from inside to outside, see Figure 5) of the cyclopentenone ring.¹¹ That is, the shieldings of the 5'-H, 6'-H, and 7'-H on the pyridine ring due to the enone system of the cyclopentenone ring decreased with an increase in the concentration of compound **3a**. Similarly, both the low field shift of the 5-H_{endo} and the high field shift of the 5-H_{exo} with an increase in the concentration can be well explained by considering the anisotropy effect of the imidazo[1,2-*a*]pyridinone ring accompanied with the same motion of the cyclopentenone ring. In contrast, when the ¹H-NMR spectra in some concentrations in the range of 0.55—10.58 mol/L of **3a** in DMSO-*d*₆ was measured, such signal shifts were no longer observed. These facts strongly suggest that this phenomenon was caused by the interaction between **3a** and CDCl₃ or **3a** itself. It is seemed that the partner of the interaction with **3a** is not water in solvent (CDCl₃), even though the shift quantity by the water were maximum, because 1) the participation of water is the weakest when the concentration was the highest one to **3a**, 2) the chemical shift (δ 1.55 ppm) when the concentration of **3a** in CDCl₃ is 0.67×10^{-2} mol/L is close to that of free water, 3) the addition of water (20 mg)¹² in the CDCl₃ solution (0.6 mL) of **3a** (0.66×10^{-2} mol/L) did not provide any change in the ¹H-NMR spectrum. As seen in the X-ray analysis for **3b**:CHCl₃, the interaction between the carbonyl oxygen and deuterium atom of CDCl₃ which is a solvent and present in

| No ^{a)} | Conc. ^{b)} | C-5' | C-6' | C-7' | C-8' | 5-H _{endo} | 5-H _{exo} | C-2 | CO₂Et | NH | CHCl₃ | H₂O |
|-------------------------|---------------------|------|------|------|------|---------------------|--------------------|------|----------------|------|-------|------|
| 3a | 0.67 | 7.51 | 6.80 | 7.74 | 7.24 | 2.66 | 3.28 | 2.27 | 1.38 4.37 | | 7.25 | 1.55 |
| | 15.06 | 7.74 | 6.87 | 7.77 | 7.23 | 2.76 | 3.22 | 2.25 | 1.36 4.34 | | 7.29 | 1.94 |
| $\mathbf{3a}^{c^{)}}$ | | 8.41 | 6.98 | 7.94 | 7.22 | 2.86 | 3.13 | 2.21 | 1.28 4.28 | | | |
| 3b | 0.80 | 7.39 | 6.74 | 7.54 | 2.41 | 2.65 | 3.28 | 2.24 | 1.38 4.37 | | 7.26 | 1.56 |
| | 11.64 | 7.52 | 6.78 | 7.57 | 2.41 | 2.71 | 3.24 | 2.23 | 1.37 4.35 | | 7.29 | 1.86 |
| 3с | 1.08 | 7.17 | 2.24 | 7.41 | 2.40 | 2.63 | 3.27 | 2.23 | 1.38 4.37 | | 7.26 | 1.58 |
| | 13.00 | 7.32 | 2.25 | 7.44 | 2.39 | 2.68 | 3.23 | 2.22 | 1.37 4.35 | | 7.29 | 1.93 |
| 7a ^{d)} | 0.79 | 7.52 | 6.81 | 7.75 | 7.27 | 2.54 | 3.22 | 2.74 | 1.39 4.36 | 9.21 | 7.26 | 1.54 |
| 7b | 0.83 | 7.40 | 6.74 | 7.55 | 2.43 | 2.53 | 3.21 | 2.71 | 1.39 4.36 | 9.19 | 7.26 | 1.57 |
| | 5.54 | 7.47 | 6.76 | 7.57 | 2.43 | 2.55 | 3.20 | 2.70 | 1.37 4.34 | 9.19 | 7.27 | 1.75 |
| 7c | 0.60 | 7.18 | 2.22 | 7.42 | 2.40 | 2.51 | 3.21 | 2.71 | 1.39 4.36 | 9.20 | 7.26 | 1.57 |
| | 10.75 | 7.34 | 2.24 | 7.45 | 2.40 | 2.55 | 3.18 | 2.71 | 1.37 4.33 | 9.20 | 7.29 | 2.02 |
| 7d | 0.50 | 7.38 | 6.68 | 7.72 | e) | 2.47 | 3.20 | f) | 1.38 4.26-4.43 | 9.27 | 7.26 | 1.56 |
| | 5.05 | 7.50 | 6.72 | 7.73 | e) | 2.50 | 3.18 | f) | 1.35 4.22-4.40 | 9.26 | 7.27 | 1.76 |
| 7e | 0.47 | e) | 6.64 | 7.52 | 2.41 | 2.48 | 3.21 | f) | 1.37 4.26-4.43 | 9.28 | 7.26 | 1.56 |
| | 4.00 | 7.34 | 6.66 | 7.53 | 2.41 | 2.49 | 3.19 | f) | 1.36 4.25-4.41 | 9.28 | 7.27 | 1.72 |
| 7f | 0.48 | 7.05 | 2.16 | 7.37 | 2.39 | 2.45 | 3.20 | f) | 1.38 4.27-4.42 | 9.27 | 7.26 | 1.56 |
| | 7.83 | 7.22 | 2.16 | 7.38 | 2.38 | 2.47 | 3.18 | f) | 1.35 4.20-4.40 | 9.25 | 7.27 | 1.86 |
| 7g | 0.97 | 7.55 | 6.82 | 7.75 | 7.27 | 2.54 | 3.22 | f) | 1.40 4.31-4.46 | 9.63 | 7.26 | 1.58 |
| | 9.28 | 7.74 | 6.86 | 7.78 | 7.28 | 2.59 | 3.18 | f) | 1.37 4.27-4.44 | 9.60 | 7.27 | 1.85 |
| 7h | 0.88 | 7.40 | 6.75 | 7.55 | 2.43 | 2.53 | 3.22 | f) | 1.40 4.32-4.46 | 9.63 | 7.26 | 1.58 |
| | 9.83 | 7.54 | 6.78 | 7.57 | 2.43 | 2.57 | 3.18 | f) | 1.38 4.28-4.44 | 9.59 | 7.28 | 1.86 |
| 7i | 1.10 | 7.19 | 2.21 | 7.42 | 2.41 | 2.51 | 3.22 | f) | 1.40 4.32-4.46 | 9.63 | 7.26 | 1.58 |
| | 5.86 | 7.30 | 2.23 | 7.43 | 2.41 | 2.53 | 3.19 | f) | 1.39 4.29-4.44 | 9.61 | 7.28 | 1.74 |
| 7j ^{d)} | 1.10 | 7.63 | 6.81 | 7.74 | 7.24 | 2.51 | 3.25 | 2.91 | 1.37 4.33 | | 7.26 | 1.57 |
| 7k | 0.47 | 7.46 | 6.73 | 7.53 | 2.41 | 2.49 | 3.25 | 2.89 | 1.36 4.33 | | 7.26 | 1.56 |
| | 8.17 | 7.58 | 6.77 | 7.56 | 2.41 | 2.54 | 3.21 | 2.89 | 1.35 4.31 | | 7.28 | 1.98 |
| 71 | 0.54 | 7.23 | 2.22 | 7.40 | 2.39 | 2.46 | 3.24 | 2.90 | 1.37 4.34 | | 7.26 | 1.56 |
| | 10.62 | 7.34 | 2.24 | 7.42 | 2.39 | 2.50 | 3.21 | 2.90 | 1.36 4.32 | | 7.27 | 1.99 |
| 7m | 0.56 | 7.63 | 6.82 | 7.73 | 7.25 | 2.49 | 3.22 | f) | 1.36 4.32 | | 7.26 | 1.57 |
| | 11.14 | 7.88 | 6.88 | 7.76 | 7.25 | 2.58 | 3.16 | f) | 1.34 4.29 | | 7.28 | 1.85 |
| 7n | 0.44 | 7.49 | 6.74 | 7.53 | 2.42 | 2.47 | 3.22 | f) | 1.36 4.32 4.32 | | 7.26 | 1.56 |
| | 12.03 | 7.67 | 6.79 | 7.55 | 2.41 | 2.53 | 3.18 | f) | 1.35 4.23-4.37 | | 7.29 | 1.94 |
| 7o | 0.68 | 7.28 | 2.23 | 7.40 | 2.39 | 2.45 | 3.21 | f) | 1.37 4.33 | | 7.26 | 1.57 |
| <u> </u> | 7.81 | 7.43 | 2.25 | 7.42 | 2.39 | 2.50 | 3.18 | f) | 1.36 4.31 | | 7.28 | 1.92 |

Table 1. ¹H-NMR spectral data for 3a—c and 7a—o in CDCl₃

a) The coupling constants are as follows; $J_{5(gem)}=17.6-18.1$ Hz, $J_{5',6}=J_{6',7}=6.8-7.0$ Hz, $J_{7',8}=9.0-9.2$ Hz, $J_{5',7}=1.2-2.0$ Hz, $J_{6'-8}=1.0-1.2$ Hz, $J_{Et}=7.0-7.2$ Hz, $J_{NHCH}=5.6-6.6$ Hz.. b) $\times 10^{-2}$ mol/L. c) The chemical shifts in DMSO- $d_{\hat{c}}$ d) Low solubility. e) Overlapped with the phenyl proton signals. f) These data are described in the Experimental section.

the highest concentration may be an alternative selection for such interaction, though it is not certain. Such interaction and the concomitant approach of a bulky trichloromethyl group may force the cyclopentenone ring inside.

In summary, we have synthesized some spiro[2-cyclopentene-1,3'-imidazo[1,2-a]pyridine] derivatives

and observed an interesting behavior in their ¹H-NMR spectra in CDCl₃.

EXPERIMENTAL

Melting points were measured with a Yanagimoto micromelting point apparatus and were not corrected. Microanalyses were carried out on a Perkin-Elmer 2400 elemental analyzer. The ¹H-NMR and ¹³C-NMR spectra were determined with a JEOL JNM-LA400 (¹H: 400 MHz and ¹³C: 100.4 MHz) spectrometer in deuteriochloroform with tetramethylsilane used as the internal standard; the chemical shifts are expressed in δ values. The IR spectra were taken with JASCO FT/IR-5300 IR spectrophotometers.

Preparation of ethyl 2',3'-dihydro-2-methylthio-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo-[1,2-*a*]pyridine]-3-carboxylates (3a—c). General method. A mixture of 3-[bis(methylthio)methylene]-2(3*H*)-imidazo[1,2-*a*]pyridinone (1, 1 mmol) and ethyl 4-chloroacetoacetate (2, 1.2 mmol) was stirred with potassium *t*-butoxide (0.135 g, 1.2 mmol) in *t*-BuOH (30 mL) at room temperature for the time indicated in the description for each product. The solution was then concentrated under reduced pressure, and the residue was separated by column chromatography on alumina using benzene-EtOH (9:1) for **3a,b** or CHCl₃ for **3c** as an eluent. The first pale yellow layers were collected and the combined solution was concentrated under reduced pressure. Recrystallization of the crude product from CHCl₃-ether afforded the corresponding spiro compounds (**3a**—c).

The ¹H-NMR spectral data of 3a—c are shown in Table 1 and the other data are as follows.

Ethyl 2',3'-dihydro-2-methylthio-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3carboxylates (3a); From 1a and 2, 69% (reaction time 3 h), colorless powder, mp 191—194 °C. IR (KBr): v 1624, 1711 cm⁻¹. ¹³C-NMR δ : 14.17, 14.21, 47.60, 61.81, 73.45, 113.73, 116.28, 132.75, 133.18, 143.33, 162.01, 168.32, 174.80, 182.68, 192.73. *Anal.* Calcd for C₁₅H₁₄N₂O₄S: C, 56.59; H, 4.43; N, 8.80. Found: C, 56.37; H, 4.47; N, 9.08.

Ethyl 2',3'-dihydro-8-methyl-2-methylthio-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a***]-pyridine]-3-carboxylates (3b)**; From **1b** and **2**, 90% (reaction time 11 h), colorless powder, mp 167—171 °C. IR (KBr): v 1622, 1709 cm⁻¹. ¹³C-NMR δ: 14.16, 14.20, 17.28, 47.72, 61.80, 74.05, 113.45, 126.94, 129.88, 132.73, 141.37, 162.09, 168.32, 174.81, 182.69, 192.70. *Anal.* Calcd for C₁₆H₁₆N₂O₄S: C, 57.82; H, 4.85; N, 8.43. Found: C, 57.88; H, 4.82; N, 8.40.

Ethyl 2',3'-dihydro-6,8-dimethyl-2-methylthio-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylates (3c); From 1c and 2, 58% (reaction time 26 h), colorless prisms, mp 191—193 °C. IR (KBr): v 1639, 1696 cm⁻¹. ¹³C-NMR δ: 14.07, 14.20, 17.14, 17.56, 47.88, 61.74, 74.17, 123.70, 126.28, 127.57, 144.14, 162.13, 166.73, 175.46, 182.71, 192.85. *Anal.* Calcd for C₁₇H₁₈N₂O₄S: C, 58.94; H, 5.24; N, 8.09. Found: C, 59.06; H, 5.22; N, 7.99.

Reactions of ethyl 2',3'-dihydro-2-methylthio-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo-[1,2-*a*]pyridine]-3-carboxylates (3a—c) with some amines. General method. A chloroform solution (15 mL) of 3 (1 mmol) and amine (6a,b,d,e, 3 mmol) was heated under the refluxing temperature until the TLC spot of 3 disappears or the generation of methanethiol ceases (1—12 h). After the completion of the reaction, the resulting solution was concentrated at reduced pressure and the residue was purified by the recrystallization from chloroform-ether to afford the corresponding substitution products 7a-f,j-o.

Similar reactions of **3** (1 mmol) with ethyl glycinate hydrochloride (**6c**) were performed in the presence of triethylamine (0.253g, 2.5 mmol) to yield products 7g—i, respectively.

The ¹H-NMR spectral data of 7a—o are shown in Table 1 and the other data are as follows.

Ethyl 2',3'-dihydro-2-methylamino-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3carboxylates (7a); From 3a and methylamine (6a), 94% (reaction time 1 h), colorless powder, mp 191—194 °C. IR (KBr): v 1626, 1666, 1711, 3227 cm⁻¹. *Anal.* Calcd for $C_{15}H_{15}N_3O_4$: C, 59.80; H, 5.02; N, 13.95. Found: C, 59.96; H, 4.73; N, 14.07.

Ethyl 2',3'-dihydro-8-methyl-2-methylamino-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylates (7b); From 3b and 6a, 95% (reaction time 1 h), colorless powder, mp 270—273 °C. IR (KBr): v 1620, 1670, 1705, 3227 cm⁻¹. ¹³C-NMR δ : 14.43, 17.36, 29.82, 47.52, 60.74, 69.53, 104.94, 113.52, 127.07, 129.57, 141.33, 165.85, 167.93, 174.14, 182.65, 191.04. *Anal.* Calcd for C₁₆H₁₇N₃O₄: C, 60.94; H, 5.43; N, 13.33. Found: C, 61.20; H, 5.42; N, 13.09.

Ethyl 2',3'-dihydro-6,8-dimethyl-2-methylamino-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo-[1,2-*a*]-pyridine]-3-carboxylates (7c); From 3c and 6a, 83% (reaction time 1 h), colorless powder, mp 257—260 °C. IR (KBr): v 1607, 1635, 1995, 3249 cm⁻¹. ¹³C-NMR δ : 14.37, 17.17, 17.51, 29.79, 47.49, 60.65, 69.68, 104.75, 123.86, 126.28, 127.42, 144.14, 165.80, 166.24, 174.29, 182.62, 191.33. *Anal.* Calcd for C₁₇H₁₉N₂O₄+H₂O: C, 58.78; H, 6.09; N, 12.10. Found: C, 59.06; H, 6.01; N, 11.90.

Ethyl 2',3'-dihydro-2',4-dioxo-2-(phenethylamino)spiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylates (7d); From 3a and phenethylamine (6b), 83% (reaction time 12 h), colorless powder, mp 192—196 °C. IR (KBr): v 1602, 1662, 1703, 3221 cm⁻¹. ¹H-NMR (0.50×10^{-2} mol/L) δ: 2.65 (1H, ddd, *J*=13.9, 7.6, 7.6 Hz), 2.79 (1H, ddd, *J*=13.9, 6.2, 6.2 Hz), 2.96 (1H, m), 3.46 (1H, m), 7.03 (2H, br d, *J*=8.1 Hz), 7.20—7.30 (3H, m); (5.05×10^{-2} mol/L) δ: 2.65 (1H, ddd, *J*=13.9, 6.2, 6.2 Hz), 2.79 (1H, ddd, *J*=13.9, 6.2, 6.2 Hz), 2.96 (1H, m), 3.45 (1H, m), 7.03 (2H, br d, *J*=8.1 Hz), 7.19—7.33 (4H, m). ¹³C-NMR δ: 14.42, 35.79, 45.16, 47.74, 60.76, 68.91, 105.05, 113.65, 116.41, 127.06, 128.46, 128.85, 132.43, 136,40, 143.03, 165.67, 167.84, 172.87, 182.98, 190.99. *Anal.* Calcd for C₂₂H₂₁N₃O₄+H₂O: C, 64.54; H, 5.66; N, 10.26. Found: C, 64.82; H, 5.64; N, 10.00.

Ethyl 2',3'-dihydro-8-methyl-2',4-dioxo-2-(phenethylamino)spiro[2-cyclopentene-1,3'-imidazo-[1,2-*a*]pyridine]-3-carboxylates (7e); From 3b and 6b, 64% (reaction time 8 h), colorless powder, mp 212—215 °C. IR (KBr): v 1620, 1661, 1699, 3231 cm⁻¹. ¹H-NMR (0.47×10^{-2} mol/L) δ : 2.59 (1H, ddd, *J*=13.9, 7.6, 7.6 Hz), 2.78 (1H, ddd, *J*=13.9, 6.4, 6.4 Hz), 2.96 (1H, m), 3.40 (1H, m), 7.02 (2H, br d, *J*=8.1 Hz), 7.19—7.31 (3H, m); (4.00×10^{-2} mol/L) δ : 2.59 (1H, ddd, *J*=13.9, 6.4, 6.4 Hz), 2.78 (1H, ddd, *J*=13.9, 6.2, 6.2 Hz), 2.96 (1H, m), 3.40 (1H, m), 7.02 (2H, br d, *J*=8.1 Hz), 7.19—7.30 (4H, m). ¹³C-NMR δ : 14.44, 17.37, 35.86, 45.16, 47.80, 60.77, 69.53, 104.92, 113.51, 127.05, 127.07, 128.42, 128.79, 129.35, 136.43, 141.27, 165.76, 167.75, 172.99, 183.02, 191.08. *Anal.* Calcd for C₂₃H₂₃N₃O₄+H₂O: C, 65.24; H, 5.95; N, 9.92. Found: C, 65.49; H, 5.93; N, 9.69.

Ethyl 2',3'-dihydro-6,8-dimethyl-2',4-dioxo-2-(phenethylamino)spiro[2-cyclopentene-1,3'-imidazo-[1,2-*a*]pyridine]-3-carboxylates (7f); From 3c and 6b, 85% (reaction time 12 h), colorless powder, mp 243—247 °C. IR (KBr): v 1611, 1665, 1709, 3237 cm⁻¹. ¹H-NMR (0.48×10^{-2} mol/L) δ : 2.59 (1H, ddd, *J*=13.9, 7.6, 7.6 Hz), 2.79 (1H, ddd, *J*=13.9, 6.2, 6.2 Hz), 2.98 (1H, m), 3.42 (1H, m), 7.02 (2H, br d, *J*=8.1 Hz), 7.18—7.30 (3H, m); (7.83×10⁻² mol/L) δ : 2.59 (1H, ddd, *J*=13.9, 6.2, 6.2 Hz), 2.98 (1H, m), 3.44 (1H, m), 7.02 (2H, br d, *J*=8.1 Hz), 7.13—7.29 (4H, m). ¹³C-NMR δ : 14.40, 17.19, 17.56, 35.82, 45.12, 47.82, 60.65, 69.69, 104.77, 123.77, 126.32, 126.93, 127.32, 128.39, 128.68, 136.50, 144.04, 165.74, 166.17, 173.22, 183.05, 191.35. *Anal.* Calcd for C₂₄H₂₅N₃O₄: C, 68.72; H, 6.01; N, 10.02. Found: C, 69.01; H, 6.08; N, 9.75.

Ethyl 2',3'-dihydro-2-(ethoxycarbonylmethyl)amino-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo-[1,2-*a*]pyridine]-3-carboxylates (7g); From 3a, glycine ethyl ester hydrochloride (6c), and triethylamine, 85% (reaction time 12 h), colorless powder, mp 217—221 °C. IR (KBr): v 1606, 1662, 1705, 1742, 3246 cm⁻¹. ¹H-NMR (0.97×10^{-2} mol/L) δ: 1.21 (3H, t, *J*=7.1 Hz), 3.50 (1H, dd, *J*=18.1, 6.6 Hz), 3.99—4.20 (3H, m): (9.28×10^{-2} mol/L) δ: 1.21 (3H, t, *J*=7.1 Hz), 3.47 (1H, dd, *J*=18.0, 6.2 Hz), 3.99—4.20 (3H, m). ¹³C-NMR δ: 14.02, 14.39, 43.88, 47.38, 60.95, 62.44, 69.06, 106.29, 113.91, 116.29, 133.43, 143.31, 165.07, 167.05, 167.80, 172.57, 182.33, 191.27. *Anal.* Calcd for C₁₈H₁₉N₃O₆: C, 57.90; H, 5.13; N, 11.25. Found: C, 57.90; H, 5.40; N, 10.99.

Ethyl 2',3'-dihydro-2-(ethoxycarbonylmethyl)amino-8-methyl-2',4-dioxospiro[2-cyclopentene-1,3'imidazo[1,2-*a*]pyridine]-3-carboxylates (7h); From 3b, 6c, and triethylamine, 99% (reaction time 3 h), colorless powder, mp 230—233 °C. IR (KBr): v 1607, 1662, 1703, 1738, 3244 cm⁻¹. ¹H-NMR $(0.88 \times 10^{-2} \text{ mol/L})$ δ : 1.21 (3H, t, *J*=7.1 Hz), 3.50 (1H, dd, *J*=18.1, 6.6 Hz), 3.97—4.19 (3H, m): $(9.83 \times 10^{-2} \text{ mol/L})$ δ : 1.21 (3H, t, *J*=7.1 Hz), 3.47 (1H, dd, *J*=18.1, 6.4 Hz), 3.95—4.19 (3H, m). ¹³C-NMR δ : 13.98, 14.38,17.32, 43.84, 47.41, 60.67, 62.32, 69.62, 106.13,113.70,126.90, 130.26, 141.47,165.11, 166.98, 167.65, 172.69, 182.32, 191.29. *Anal.* Calcd for C₁₉H₂₁N₃O₆: C, 58.91; H, 5.46; N, 10.85. Found: C, 59.05; H, 5.59; N, 10.58.

Ethyl 2',3'-dihydro-2-(ethoxycarbonylmethyl)amino-6,8-dimethyl-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylates (7i); From 3c, 6c, and triethylamine, 62% (reaction time 12 h), colorless powder, mp 216—220 °C. IR (KBr): v 1640, 1669, 1713, 1746, 3225 cm⁻¹. ¹H-NMR $(1.10\times10^{-2} \text{ mol/L})$ δ: 1.21 (3H, t, *J*=7.1 Hz), 3.47 (1H, dd, *J*=18.1, 6.3 Hz), 3.97—4.20 (3H, m): (5.86×10⁻² mol/L) δ: 1.21 (3H, t, *J*=7.1 Hz), 3.46 (1H, dd, *J*=18.1, 6.3 Hz), 3.97—4.20 (3H, m). ¹³C-NMR δ: 14.02, 14.42, 17.23, 17.55, 43.82, 47.49, 60.88, 62.31, 69.84, 106.10, 123.99, 126.31, 127.94, 144.15, 165.22, 166.16, 167.08, 172.88, 182.36, 191.50. *Anal.* Calcd for C₂₀H₂₃N₃O₆: C, 59.84; H, 5.78; N, 10.47. Found: C, 50.70; H, 5.94; N, 10.44.

Ethyl 2',3'-dihydro-2-dimethylamino-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylates (7j); From 3a and dimethyamine (6d), 78% (reaction time 9 h), colorless powder, mp 270—273 °C. IR (KBr): v 1626, 1659, 1720 cm⁻¹. *Anal.* Calcd for $C_{16}H_{17}N_3O_4$: C, 60.94; H, 5.43; N, 13.33. Found: C, 61.11; H, 5.46; N, 13.14.

Ethyl 2',3'-dihydro-2-dimethylamino-8-methyl-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylates (7k); From 3b and 6d, 79% (reaction time 1 h), colorless powder, mp 238—242 °C. IR (KBr): v 1620, 1678, 1699 cm⁻¹. ¹³C-NMR δ : 14.39, 17.27, 35.07, 48.84, 61.17, 71.77, 110.96, 113.54, 126.90, 129.64, 141.19, 163.68, 167.18, 167.84, 183.43, 192.37. *Anal.* Calcd for C₁₇H₁₉N₃O₄: C, 62.00; H, 5.81; N, 12.76. Found: C, 62.12; H, 5.83; N, 12.62.

Ethyl 2',3'-dihydro-2-dimethylamino-6,8-dimethyl-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo-[1,2 *a*]pyridine]-3-carboxylates (7l); From 3c and 6d, 82% (reaction time 3 h), colorless powder, mp 197—200 °C. IR (KBr): v 1635, 1699, 1732 cm⁻¹. ¹³C-NMR δ : 14.36, 17.12, 17.52, 17.58, 48.91, 61.14, 71.92, 110.88, 123.75, 126.26, 125.30, 143.92, 163.78, 165.58, 167.84, 183.47, 192.57. *Anal.* Calcd for C₁₈H₂₁N₃O₄: C, 62.96; H, 6.16; N, 12.24. Found: C, 62.99; H, 6.13; N, 12.11.

Ethyl 2',3'-dihydro-2',4-dioxo-2-piperidinospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3carboxylates (7m); From 3a and piperidine (6e), 79% (reaction time 7 h), colorless powder, mp 210—214 °C. IR (KBr): v 1626, 1663, 1723 cm⁻¹. ¹H-NMR (0.56×10^{-2} mol/L) δ : 1.50—1.62 (6H, m), 3.11—3.18 (4H, m): (11.14×10⁻² mol/L) δ : 1.47—1.63 (6H, m), 3.07—3.25 (4H, m). ¹³C-NMR δ : 14.43, 17.36, 29.82, 47.52, 60.74, 69.53, 104.94, 113.52, 127.07, 129.57, 141.33, 165.85, 167.93, 174.14, 182.65, 191.04. *Anal.* Calcd for C₁₉H₂₁N₃O₄: C, 64.21; H, 5.96; N, 11.82. Found: C, 64.43; H, 6.04; N,

11.52.

Ethyl 2',3'-dihydro-8-methyl-2',4-dioxo-2-piperidinospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylates (7n); From 3b and 6e, 95% (reaction time 6 h), colorless powder, mp 212—216 °C. IR (KBr): v 1622, 1674, 1709 cm⁻¹. ¹H-NMR ($0.44 \times 10^{-2} \text{ mol/L}$) δ : 1.50—1.59 (6H, m), 3.10—3.19 (4H, m): ($12.03 \times 10^{-2} \text{ mol/L}$) δ : 1.48—1.64 (6H, m), 3.06—3.21 (4H, m). ¹³C-NMR δ : 14.38, 17.27, 23.10, 25.49, 48.82, 60.94, 71.81, 110.30, 113.41, 126.75, 130.11, 141.06, 163.46, 166.89, 167.98, 183.67, 192.49. *Anal.* Calcd for C₂₀H₂₃N₃O₄: C, 65.03; H, 6.28; N, 11.37. Found: C, 65.09; H, 6.57; N, 11.12.

Ethyl 2',3'-dihydro-6,8-dimethyl-2',4-dioxo-2-piperidinospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylates (70); From 3c and 6e, 76% (reaction time 4 h), colorless owder, mp 250-252 °C. IR (KBr): v 1633, 1699, 1726 cm⁻¹. ¹H-NMR (0.68×10⁻² mol/L) δ : 1.50–1.63 (6H, m), 3.09–3.16 (4H, m): (7.81×10⁻² mol/L) δ : 1.50–1.63 (6H, m), 3.06–3.24 (4H, m). ¹³C-NMR δ : 14.37, 17.17, 17.51, 29.79, 47.49, 60.65, 69.68, 104.75, 123.86, 126.28, 127.42, 144.14, 165.80, 166.24, 174.29, 182.62, 191.33. *Anal.* Calcd for C₂₁H₂₅N₃O₄+H₂O: C, 62.83; H, 6.78; N, 10.47. Found: C, 62.95; H, 6.91; N, 10.22.

Crystallography of ethyl 2',3'-dihydro-8-methyl-2-methylthio-2',4-dioxospiro[2-cyclopentene-1,3'-imidazo[1,2-*a*]pyridine]-3-carboxylates (3b): CHCl₃. A single crystal ($0.06 \times 0.32 \times 1.00$ mm) grown from CHCl₃-ether was used for the unit-cell determinations and the data collection by a Rigaku AFC5S four-circle diffractometer with graphite-monochromated MoK_a radiation (l=0.71069 Å). Crystal data of **3b**-CHCl₃: C₁₇H₁₇N₂O₄SCl₃; *M*=451.75; monoclinic, space group *P*2₁/n (#14), *Z*=4 with *a*=18.98 (2) Å, *b*=8.90 (3) Å, *c*=17.003 (14) Å, b=107.83° (9); *V*=2013 (7) Å³, and *D*_{calc.}=1.490 g/cm³. All calculations were performed using CrystalStructure.¹³ The structure was solved by a direct method (SIR).¹⁴ The non-hydrogen atoms were refined anisotropically, and the hydrogen atoms were attached at the idealized position and not refined. The final *R*- and *R*_w-factors after full-matrix least-squares refinements were 0.064 and 0.063 for 1915 (*I*>2.00s(*I*)) observed reflections, respectively.

REFERENCES AND NOTES

- Preparation of new nitrogen-bridged heterocycles. 70. For 69 of this series, see A. Kakehi, H. Suga,
 Y. Okumura, K. Itoh, H. Hatayama, D. Kubo, and K. Kobayashi, *Heterocycl. Commun.*, in press.
- 2. C. Enguehard and A. Gueiffier, Mini-Rev. Med. Chem., 2007, 7, 888.
- 3. Y. Yamaguchi, H. Miyashita, H. Tsunekawa, A. Nouri, H.-C. Kim, K. Saito, T. Matsuno, S. Kawashima, and T. Nabeshima, *J. Pharm. & Exp. Therap.*, 2006, **317**, 1079.

- U. Ito, K. Takuma, H. Mizoguchi, T. Nagai, and K. Yamada, J. Pharm. & Exp. Therap., 2007, 320, 819.
- 5. F. Han, N. Shioda, S. Moriguchi, Y. Yamamoto, A. Y. A. Raie, Y. Yamaguchi, M. Hino, and K. Fukunaga, *J. Pharm. Exp. Therap.*, 2008, **326**, 127.
- 6. W. Jiang, J. J. Fiordeliso, and Z. Sui, PCT Int. Appl. 2006, WO2006049890 A1 20060511.
- Y. Yamaguchi, R. Yui, T. Matsuno, K. Saitoh, H. Miyashita, and T. Nagata, *PCT Int. Appl.*, 2008, WO2008047952 A2 20080424.
- 8. T. Abe, Y. Okumura, H. Suga, and A. Kakehi, *Heterocycles*, 2010, 80, 439.
- L. M. Jackman, 'Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry, Pergamon Press, Inc., New York, 1959.
- C. K. Johnson, "ORTEO II, Report ORNL-5138," Oak Ridge National Laboratory, Oak Ridge, Tennessee, 1976.
- 11. The 2(3*H*)-imidazo[1,2-*a*]pyridinone ring is very rigid and almost planar but the cyclopentenone ring is comparatively movable.
- 12. The addition of ethanol (17 mg) in the CDCl₃ solution of **3a** (0.54×10^{-2} mol/L) did not also provide the ¹H-NMR spectral change.
- CrystalStructure 3.8: Crystal Structure Analysis Package, Rigaku and Rigaku/MSC (2000—2006).
 9009 New Trails Dr. The Woodlands TX 77381 USA.
- 14. SIR92: A. Altomore, M. Cascarano, C. Giacovazzo, and A. Guagliardi, J. Appl. Cryst., 1994, 26, 343.