into the factors controlling the relative product yields observed, and that will be the subject of a future paper.

Acknowledgment is made to the National Institutes of Health under Grant No. 5-RO1-GM32296-05 for support of this research. **Registry No.** $[Ru^{II}(Hedta)(H_2O)]^-$, 117687-36-2; $[Fe^{II}(Hedta)(H_2O)]^-$, 117687-37-3; $[Fe(H_2O)_6]^{2+}$, 15365-81-8; NO₂⁻⁻, 14797-65-0; NH₂OH, 7803-49-8; NH₃, 7664-41-7; N₂O, 10024-97-2; N₂, 7727-37-9; $[Ru^{II}(Hedta)NO]$, 117687-38-4; $[Fe^{II}(Hedta)(NO^*)]^-$, 117687-39-5; $[Fe(TPPS)(H_2O)]^{3-}$, 53194-20-0.

Additions and Corrections

1988, Volume 27

Robert A. Scott, Cheryl E. Coté, and David M. Dooley*: Copper X-ray Absorption Spectroscopic Studies of the Bovine Plasma Amine Oxidase–Sulfide Complex.

Page 3859. The symbols in the caption to Figure 1 have been reversed. The correct caption should read as follows: Copper K absorption edge (a) and EXAFS (b) spectra for resting (oxidized) bovine plasma amine oxidase (---) and its complex with sulfide (--). In the Experimental Section, third sentence, the definition of enzyme activity should read as follows: One enzyme unit catalyzes the oxidation of 1 μ mol min⁻¹ of benzylamine to benzaldehyde.—David M. Dooley