Additions and Corrections

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John L. Hubbard,* Christopher R. Zoch, and Wendy L. Elcesser: Oxygen Atom Transfer between *Cis*-Coordinated Nitrite and Nitrosyl Ligands: The Case of the $CpCr(NO)_2(NO_2)/CpCr(NO)_2(ONO)$ Linkage Isomers.

Pages 3336–7. Figure 7 should be correctly labeled as an Arrhenius plot. The activation energy (E_a) for the oxygen atom transfer process in MeOH is 11.3(6) kcal/mol. Subsequent Eyring analysis employing the plot of $\ln(k/T)$ versus $1/T$ (×1000) yields $\Delta H^{\ddagger}$ = 10.7(6) kcal/mol and $\Delta S^{\ddagger}$ = –50(2) eu in MeOH. In toluene, E_a is 14(3) kcal/mol and the activation parameters are $\Delta H^{\ddagger}$ = 14.0(4) kcal/mol and $\Delta S^{\ddagger}$ = –40(1) eu. In terms of the mechanism of ^{15}N scrambling, the negative activation entropy is consistent with either the nitro or the nitrito pathway for an intramolecular oxygen atom transfer to the NO ligands (Scheme II, p 3337).

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