exhibits solvent-dependent Fe(III)−Fe(II) process. For both complexes a ligand-centered oxidation process is also observed. The exact nature of $1e^-$-oxidized species of **1** is under investigation.

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Supporting Information Available: Listings of crystallographic experimental details, positional and isotropic thermal parameters, bond distances and angles, dihedral angles, anisotropic thermal parameters, least-squares planes, and hydrogen atom coordinates for compounds **1** and **2** (Tables S1−S16), magnetic data for **1** (Table S17), X-ray structure of **2** (Figure S1), UV−vis spectra of **1** and **2** (Figure S2), EPR spectra of **1** (Figures S3 and S4), ^{1}H NMR spectra of H_2L (Figure S5) and **2** (Figure S6), cyclic voltammograms of **1** and **2** (Figures S7 and S8), and least-squares plot of $E_{1/2}$ (Fe(III)-Fe(II)) of **1** vs reciprocal of dielectric constant of the solvents (Figure S9) (36 pages). Ordering information is given on any current masthead page.

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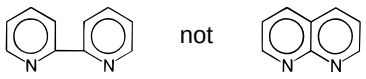
Additions and Corrections

1997, Volume 36

Zhengui Yao and Kenneth J. Klabunde*: The Strong Acidity and the Process for Deprotonation of (η^6-Toluene)-$Fe(H)_2(SiCl_3)_2$.

Pages 6A, 2122, and 2123 (Issue No. 10). There is an error in the drawing of 2,2′-bipyridine in the synopsis, Figure 6, and Scheme 4. A bond is missing, making the structure appear to be a naphthalene derivative rather than 2,2′-bipyridine. The structure should be represented as



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