

A Novel Undecametallic Iron(III) Cluster with an $S = 11/2$ Spin Ground State

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The reaction of $[\text{NEt}_4]_2[\text{Fe}_2\text{OCl}_6]$ with sodium benzoate, 4,6-dimethyl-2-hydroxypyrimidine (dmhp), and 1,1,1-tris(hydroxymethyl)ethane (H_3thme) gives the undecametallic compound $[\text{NEt}_4][\text{Fe}_{11}\text{O}_4(\text{O}_2\text{CPh})_{10}(\text{thme})_4(\text{dmhp})_2\text{Cl}_4]$. X-ray crystallography, EPR spectroscopy, bulk magnetic susceptibility studies, and low-temperature single-crystal magnetic measurements were used to characterize the compound. Magnetic measurements indicate an $S = 11/2$ ground state with the parameters $g = 2.03$ and $D = -0.46 \text{ cm}^{-1}$. Single-crystal magnetic studies show hysteresis of molecular origin at $T < 1.2 \text{ K}$ with fast quantum mechanical tunneling at zero field.

Iron–oxo clusters have been investigated for many years for a variety of reasons. These range from biological models for the iron oxo core of the iron storage protein ferritin,¹ the study of biomineralization processes that form iron–oxo minerals,² and more recently for single-molecule magnetism (SMM) behavior. Although a variety of Mn-containing SMMs have been reported in recent years,³ fewer Fe SMMs are known.^{4–8} The requirement for SMM behavior is the

combination of a large spin ground state with a negative D value, but there are relatively few Fe(III) clusters with large ground spin states: Fe_4 , $S = 5$;⁵ Fe_4 , $S = 8$;^{8,9} Fe_8 , $S = 10$;⁴ Fe_{10} , $S = 11$;⁶ Fe_7 , $S = 29/2$;¹⁰ and Fe_{19} , $S = 33/2$;¹¹ Following on from our recent success using the ligand 1,1,1-tris(hydroxymethyl)ethane (H_3thme) in Mn and Fe cluster chemistry,^{12,13} we herein report the synthesis, structure, and magnetic properties of a new undecametallic Fe cluster with an $S = 11/2$ spin ground state.

Reaction of $[\text{NEt}_4]_2[\text{Fe}_2\text{OCl}_6]$ (1 equiv) with NaO_2CPh (2 equiv), dmhp (1 equiv), and H_3thme (1 equiv) in MeCN leads to the formation of $[\text{NEt}_4][\text{Fe}_{11}\text{O}_4(\text{O}_2\text{CPh})_{10}(\text{thme})_4(\text{dmhp})_2\text{Cl}_4]$ **1**¹⁴ in $\leq 20\%$ yield after 1 week. **1** (Figure 1) crystallizes in the monoclinic space group $P2_1/c$.

The core of **1** (Figure 2) consists of four fused butterfly $\{\text{Fe}_4\text{O}_2\}^{8+}$ motifs. The two butterflies in the center of the core (Fe1, Fe2, Fe3, Fe5, and symmetry equivalents) share a body Fe ion (Fe3) forming a planar central $\{\text{Fe}_7\text{O}_4\}^{13+}$ unit (Fe1–Fe3–Fe1A = 180.0°). The two peripheral but-

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- (14) Crystal data for **1**: $\text{C}_{131}\text{H}_{130}\text{Cl}_4\text{Fe}_{11}\text{N}_{14}\text{O}_{38}$, $M = 3264.64$, monoclinic, space group $P2_1/c$, $a = 13.9889(9) \text{ \AA}$, $b = 21.5807(15) \text{ \AA}$, $c = 23.9145(16) \text{ \AA}$, $\beta = 94.753(1)^\circ$, $V = 7194.7(8) \text{ \AA}^3$, $Z = 2$, $D_c = 1.507 \text{ g cm}^{-3}$, $\mu = 1.229 \text{ mm}^{-1}$ (Mo $K\alpha$, $\lambda = 0.71073 \text{ \AA}$), $T = 100(2) \text{ K}$, CCD diffractometer, 41276 measured, 14643 unique data ($R_{\text{int}} = 0.0408$), of which 11583 have $I > 2\sigma(I)$. $R(I > 2\sigma(I)) = 0.0400$, $wR2$ (on F^2 , all data) = 0.1105 with 922 refined parameters, goodness-of-fit = 1.028.

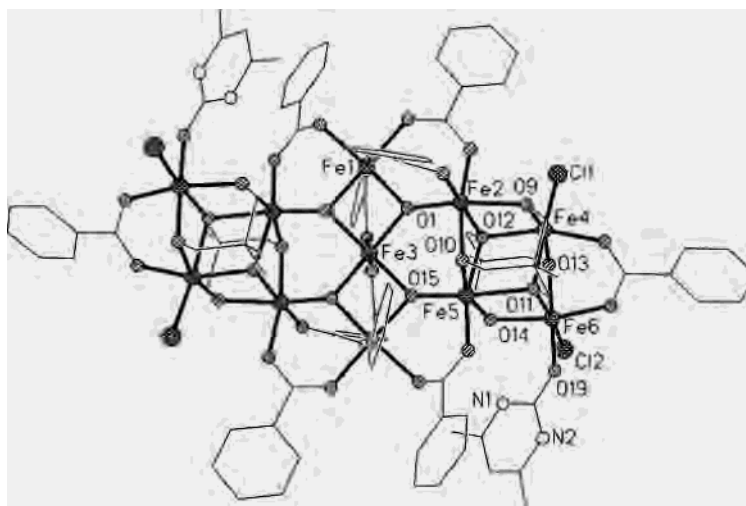


Figure 1. The structure of the complex in **1**. Selected interatomic distances (Å): Fe³⁺–O, 1.827(18)–2.182(18), Fe³⁺–Cl, 2.266(8)–2.301(8).

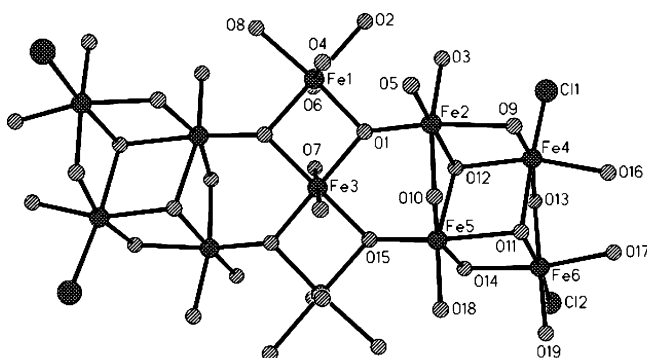


Figure 2. The core of the complex in **1**.

terfly units (Fe2, Fe4, Fe5, Fe6, and symmetry equivalents) are nonplanar and are situated, respectively, above and below the plane of the central {Fe₇O₄}¹³⁺ core. The wing tip Fe atoms (Fe2 and Fe6) lie respectively above and below the plane of the body Fe atoms Fe4 and Fe5. The four thme³⁻ ligands are found in the peripheral {Fe₄O₂} units, with two oxygen atoms acting as μ₂-bridges, linking the Fe atoms within the {Fe₄O₂} unit as well as between the {Fe₄O₂} and {Fe₇O₄}¹³⁺ units. The third oxygen atom acts as a μ₃-bridge within the {Fe₄O₂} unit. The 10 PhCO₂⁻ ligands bridge in their familiar μ₂-mode, while the Cl⁻ ions are all terminal. The neutral dmhp ligands are also terminally bonded, with the nitrogen atom protonated and hydrogen bonding to a μ₂-oxygen of a thme³⁻ ligand (O14–N1, 2.796(8) Å).

{M₄O₂} butterfly units are common building blocks in the structures of Fe(III) and Mn(III) clusters. The {M₇O₄}¹³⁺ subunit is also a common feature in larger clusters, including Fe₁₇/19¹¹ and Fe₁₁.¹⁵ In fact the Fe₁₁ core is similar to a Mn₁₁ cluster reported by Christou and co-workers, but in this case the peripheral units are [Mn₄] cubes not butterflies.¹⁶ **1** is only the second undecametallic Fe complex reported;

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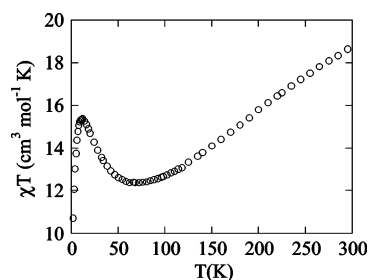


Figure 3. Thermal variation of $\chi_M T$ in the ranges 300–50 K at 0.2 T and 50–2 K at 0.05 T.

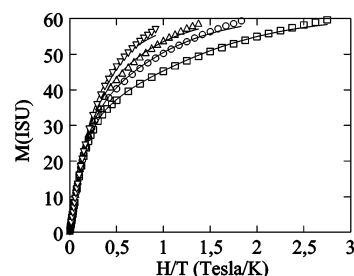


Figure 4. Magnetization vs H/T plot at 2(□), 3(○), 4(Δ), and 6(∇) K.

the original has no structural resemblance to **1**, being based on a pentacapped trigonal prism.

Solid state dc magnetization measurements were performed on **1** in the range 2–300 K in a field of 0.2 T between 300 and 50 K, and 0.05 T between 50 and 2 K. (Figure 3). The room temperature $\chi_M T$ value of approximately 18.7 cm³ mol⁻¹ K decreases to a minimum of 12.4 cm³ mol⁻¹ K at 70 K, and then rises to a maximum of 15.4 cm³ mol⁻¹ K at 12 K and then drops. This indicates that **1** has a nonzero spin ground state and that the ground spin state is higher than $9/2$. In order to obtain the ground state spin, magnetization data were collected in the ranges 2.0–6.0 K and 0.01–5.5 T (Figure 4). The fits were made simultaneously on the four different temperatures (2, 3, 4, and 6 K), assuming that only the ground state is populated, giving a best fit of $S = 11/2$, $g = 2.03$, and $D = -0.46$ cm⁻¹. The maximum value of $\chi_M T$ at low temperature (15.4 cm³ mol⁻¹ K) and the fact that the magnetization data at 2 and 6 K can be fit with one set of parameters strongly indicate that the ground state is

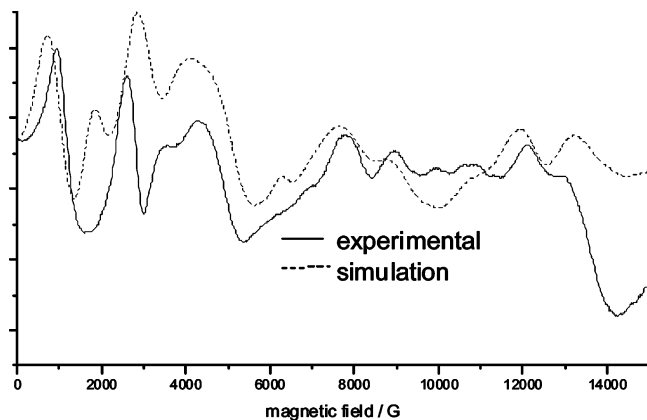


Figure 5. Q-band EPR spectrum of **1**. Experimental: $\nu = 33.9250$ GHz; $T = 5$ K, power = 1.9 mW; modulation amplitude = 10 G; Simulation: $g = 2.03$; $D = -0.38$ cm $^{-1}$; $E = 0.055$ cm $^{-1}$; $\Delta B = 1300$ G; Gaussian line shapes were used.

indeed $S = 11/2$ and that the first excited state is rather high in energy. The previously reported FeII complex has an $S = 1/2$ spin ground state.

The polycrystalline EPR spectrum of **1** at Q-band frequency and 5 K (Figure 5) confirmed both the spin state and magnitude and sign of the zero-field splitting deduced from the magnetization curves. Significantly, it was impossible to produce an adequate simulation of the spectrum without inclusion of rhombicity ($E \neq 0$). Given the nature of S and sign of D , measurements were taken to see if **1** acts as a SMM. Low-temperature (1.2–0.04 K) single-crystal magnetic measurements were performed on **1** using a micro-SQUID instrument equipped with three orthogonal fields, allowing the magnetic field to be scanned in all directions.

Below 1.2 K hysteresis loops are seen in magnetization vs field studies whose coercivities increase with decreasing temperature (Figure 6). A detailed study of the field sweep rate dependence of the hysteresis loops showed that the hysteresis at nonzero fields is due not to a phonon bottleneck but to slow relaxation because of the anisotropy barrier. Slow relaxation is seen at $H = 0$, but the presence of strong tunneling does not allow for a reliable Arrhenius plot. For all SMMs the barrier is reduced at $H = 0$ because of the presence of tunneling. Even half-integer spin systems, which in theory should not tunnel, do so because of coupling with the environment: dipolar coupling between molecules, hyperfine coupling, spin–spin cross relaxation, and other multibody quantum processes. For example for a system with $S = 11/2$, $D = -0.46$ cm $^{-1}$, $E = -0.055$ cm $^{-1}$ and assuming an internal transverse field of approximately 10 mT, one can estimate a tunnel splitting of ca. 2.8×10^{-6} K and thus a tunnel probability of $P = 0.68$ for a sweep rate of 0.1 T s $^{-1}$, in good agreement with our measurements. This has also been observed in a [Mn4] cluster with $S = 9/2$.¹⁷

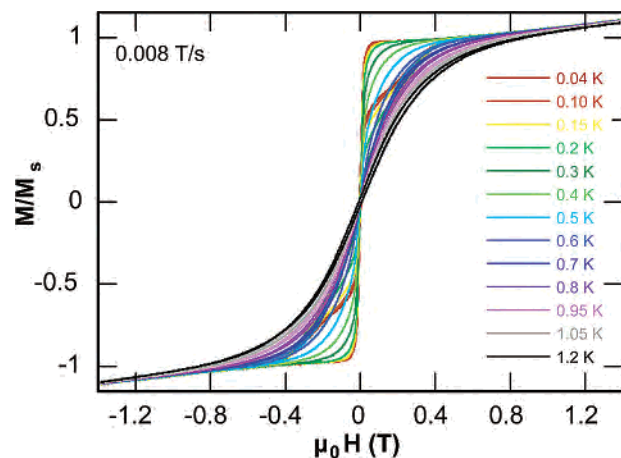


Figure 6. Magnetization of **1** (M) plotted as a fraction of maximum (M_s) vs applied magnetic field ($\mu_0 H$).

Magnetic studies of $\{\text{Fe}_4\text{O}_2\}^{8+}$ butterflies indicate complexes with $S = 0$ spin ground states,¹⁸ in contrast to $\{\text{Mn}_4\text{O}_2\}^{8+}$ complexes which usually exhibit an $S = 3$ spin ground state.¹⁹ The difference has been explained by considering the different competing exchange interactions present: in $\{\text{Mn}_4\text{O}_2\}^{8+}$ complexes the body–body interaction is greater than the wing tip–body interaction, resulting in a “frustrated” spin alignment of the wing tip ions and an $S = 3$ ground state. In $\{\text{Fe}_4\text{O}_2\}^{8+}$ complexes the body–wing tip exchange interaction dominates, resulting in an $S = 0$ spin ground state. **1** contains four such $\{\text{Fe}_4\text{O}_2\}$ units fused together, creating a number of competing exchange interactions leading to the stabilization of an intermediate spin ground state of $S = 11/2$. **1** is only the second FeII cluster reported (although various derivatives of the original FeII have been synthesized)²⁰ and is the first to exhibit an appreciable spin ground state. Combined with a negative D value this leads to **1** being a new example of a SMM.

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Supporting Information Available: Crystallographic data in CIF format. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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