Inorg. Chem. 2004, 43, 1808–1809

Inorganic Chemistry

Amino Acid Templating of Inorganic Networks: Synthesis and Structure of L-Asparagine Zinc Phosphite, C₄N₂O₃H₈·ZnHPO₃

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Received June 3, 2003

C₄N₂O₃H₈•ZnHPO₃ is the first zincophosphite framework to be templated by an amino acid (L-asparagine), which bonds to Zn via a carboxyl O atom. It contains infinite, homochiral, helical 4-ring chains of ZnO₄ and HPO₃ groups, stabilized by intra- and interchain N–H···O hydrogen bonds. Crystal data: C₄N₂O₃H₈•ZnHPO₃, M_r = 277.49, orthorhombic, $P2_12_12_1$ (No. 19), a = 5.0349(2) Å, b = 9.4539(4) Å, c = 18.6092(8) Å, V = 885.79 (6) Å³, Z = 4.

Many organically templated zincophosphates (ZnPOs) have now been reported.¹ In these and related materials,² the organic entity is regarded as the key factor in promoting the formation of new types of low-density networks. Most templates are cationic, protonated, polyfunctional amine cations, and multipoint $N-H\cdots O_f$ (f = framework) hydrogen bonds play a key role in stabilizing the resulting ZnO₄/PO₄ inorganic network.³ We have recently extended these studies to organically templated zincophosphites (ZnHPOs) containing the (hydrogen) phosphite [HPO₃]²⁻ moiety.⁴⁻⁸ Depending on the template used, some ZnHPOs are surprisingly similar to corresponding ZnPOs,⁶ whereas others adopt novel structures.⁷ In this Communication, we report the synthesis, crystal structure, and some properties of L-asparagine zinc phosphite, C₄N₂O₃H₈•ZnHPO₃. This new material is the first ZnHPO or ZnPO phase to be templated by a chiral amino acid, and as a result, it contains homochiral polyhedral chains.



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Figure 1. Fragment of C₄N₂O₃H₈·ZnHPO₃ (50% thermal ellipsoids, arbitrary spheres for the H atoms). Symmetry codes: (a) $x - \frac{1}{2}, \frac{1}{2} - y$, 1 - z, (b) x - 1, y, z. Selected bond distances (Å): Zn1-O1 1.9494(13), Zn1-O2 1.9837(13), Zn1-O3 1.9368(14), Zn1-O4 1.9468(13), P1-O1 1.5269(15), P1-O2a 1.5295(13), P1-O3b 1.5174(13), O4-C1 1.278(2), O5-C1 1.225(2), O6-C4 1.235(3), N1-C2 1.487(2), N2-C4 1.326(3), C1-C2 1.529(2), C2-C3 1.519(3), C3-C4 1.522(2).

C₄N₂O₃H₈·ZnHPO₃ was prepared from a mixture of 0.814 g (10 mmol) of ZnO, 0.820 g (10 mmol) of H₃PO₃, 1.501 g (11.4 mmol) of L-asparagine (**I**), and 20 mL of water. These components were shaken together in a polypropylene bottle to result in a white slurry and heated to 80 °C for 5 days. Product recovery by vacuum filtration and rinsing with water resulted in 1.472 g of transparent blades and plates of the title compound (yield based on Zn = 53%). Elemental analysis data were satisfactory (C obsd (%), calcd (%), 17.31, 16.81; H, 2.90, 3.27; N, 9.39, 10.10). Thermogravimetric analysis showed weight loss over the broad range 200–700 °C.



The crystal structure⁹ of $C_4N_2O_3H_8$ ·ZnHPO₃ contains 14 non-H atoms in the asymmetric unit (Figure 1). The zinc atom is tetrahedrally coordinated to three O atoms bridging

10.1021/ic0346127 CCC: \$27.50 © 2004 American Chemical Society Published on Web 02/20/2004

⁽⁹⁾ Bruker SMART 1000 CCD diffractometer, Mo Kα radiation, $\lambda = 0.71073$ Å, T = 20 °C. The H atoms were located in difference maps (N–H) or geometrically placed (C–H) and were refined by riding in idealized positions. Crystal data: C₄N₂O₃H₈·ZnHPO₃, $M_r = 277.47$, orthorhombic, $P2_12_12_1$ (No. 19), a = 5.0349(2) Å, b = 9.4539(4) Å, c = 18.6092(8) Å, V = 885.79(6) Å³, Z = 4, R(F) = 0.024, $R_w(F^2) = 0.055$.



Figure 2. Detail of $C_4N_2O_3H_8$ ·ZnHPO₃ showing a [100] chain of ZnO₄ and HPO₃ units (violet and blue tetrahedra, respectively) and pendant asparagine molecules. Color key: C blue, N green, O red, H gray (all radii arbitrary). The intrachain N2–H6···O6a (a = 1 + x, y, z) hydrogen bond is shown in yellow.

to P and a carboxyl O atom of the asparagine molecule $[d_{av}$ (Zn–O = 1.954 (2) Å]. The P atom bonds to three O atoms $[d_{av}(P-O) = 1.525$ (2) Å], with the fourth tetrahedral vertex occupied by an H atom, which is always observed for a phosphite grouping.¹⁰ The three bridging O atoms have an average Zn–O–P bond angle of 124.1°.

The L-asparagine molecule has typical geometrical parameters.¹¹ The crystal structure study clearly shows the N1 species to be triply protonated; thus, this species is neutral, and (nominally) zwitterionic, which is in accordance with the charge-balancing criterion, assuming the usual valencies for Zn^{2+} and $[HPO_3]^{2-}$. C2 has its expected S chirality, based on the L-asparagine starting material, and refinement of the Flack absolute structure parameter to 0.019(9) indicated that the absolute structure is well defined (i.e., there is no suggestion of racemic or merohedral twinning).

The linkage of the ZnO₄ (i.e., ZnO₃O_A A = asparagine) and HPO₃ tetrahedra in C₄N₂O₃H₈·ZnHPO₃ results in infinite, edge-sharing, ladderlike chains of 4-rings propagating along [100] (Figure 2). The chain motif is chiral (the Zn1-O2-P1-O1 backbone of every chain propagates in a counter-clockwise sense along [100]) and lies on a 2₁ symmetry axis.

Hydrogen bonding appears to be a key factor in defining the molecular packing, and various intra- (via H6) and interchain N–H···O links are present. These interactions help to define a crystal structure (Figure 3) in which the neutral [100] chains are interleaved by means of the N1–H7···O2 and N1–H9···O1 bonds in the *b* direction and N1–H8··· O5 and N2–H5···O4 bonds in the *c* direction (see Table 1



Figure 3. Unit cell packing in $C_4N_2O_3H_8$ ·ZnHPO₃ viewed down [100]. Color key as in Figure 2.

Table	1.	Hydrogen	Bonding	Interactions ^a	(Å,	deg)
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N1-H7•••O2a	2.02	2.782(2)	142
N1-H8····O5b	1.96	2.763(2)	150
N1-H9···O1c	2.15	3.016(2)	165
N2-H5····O4d	2.26	2.955(2)	138
N2-H6···O6e	2.31	3.136(2)	160

^{*a*} The three values refer to $d(H^{\dots}O)$, $d(N^{\dots}O)$, and $\theta(N-H^{\dots}O)$, with d(N1-H) = 0.89 Å and d(N2-H) = 0.86 Å. Symmetry codes: (a) $\frac{1}{2} + x$, $\frac{3}{2} - y$, 1 - z; (b) $x - \frac{1}{2}, \frac{3}{2} - y$, 1 - z; (c) x, 1 + y, z; (d) 1 - x, $\frac{1}{2} + y$, $\frac{1}{2} - z$; (e) 1 + x, y, z.

for acceptor-atom symmetry codes). The chains arrange themselves into pseudo-(001) sheets in which the component tetrahedra all point in the same direction; the sheets configure themselves in *ABAB* all-up/all-down in the [001] direction.

The existence of a template-O-atom-to-Zn bond is a novel feature for ZnPOs and ZnHPOs,¹ although template-N-atom-to-Zn bonds have been seen for both families.^{1,4} However, both Zn–O and Zn–N bonds are commonplace in molecular systems.¹² The possible biological significance of Zn–O and Zn–N bonds has recently been discussed.¹³ We are now exploring the chemistry of ZnHPOs templated by other amino acids with a major aim being the preparation of a chiral, three-dimensional framework.

Acknowledgment. We thank Jim Marr and Brian Paterson for assistance.

Supporting Information Available: CIF file for the title compound. This material is available free of charge via the Internet at http://pubs.acs.org.

IC0346127

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