Inorganic Chemistry

Electrical Properties of Single-Crystalline CaAl₂Si₂

Motoharu Imai,* Hideki Abe, and Kei Yamada

National Institute for Materials Science, 1-2-1 Sengen, Tsukuba, Ibaraki 305-0047, Japan

Received March 4, 2004

Electrical resistivity and Hall coefficient measurements of singlecrystalline $CaAl_2Si_2$ revealed that $CaAl_2Si_2$ is a metal in which both electrons and holes contribute to the transport properties; its dominant carriers are holes at temperature below 150 K but electrons above that temperature.

Binary Ca-Si and ternary Ca-Al-Si systems have compounds with characteristic layered structures, which can be related to a crystalline form of elementary Si, the diamond-type structure. In the diamond-type structure, chairtype corrugated hexagonal layers stack along the [111] direction, as shown in Figure 1a.¹ The binary Ca–Si system has six kinds of compounds.² One of the compounds, CaSi₂, has the CaSi₂-type structure (space group: R3m (No. 166), a = 3.86 Å, c = 30.6 Å, Z = 6), in which hexagonal Ca layers stack alternatively with chair-type corrugated hexagonal Si layers along the [001] direction,³ as shown in Figure 1b. This structure can be made up by intercalating Ca layers to the diamond-type Si and changing its stacking sequence. When some Si atoms are replaced with Al atoms, the corrugated hexagonal layers transform into flat honeycomb layers. Figure 1c shows a crystalline form of $Ca(Al_x, Si_{1-x})_2$ (0.225 $\leq x \leq$ 0.675), the AlB₂-type structure (space group: P6/mmm (No. 191), a = 4.1905(5) Å, c =4.3992(8) Å for x = 0.5, Z = 1), in which Ca hexagonal layers stack alternatively with Al-Si honeycomb layers.^{4,5} The ternary Ca-Al-Si system has another compound with a layered structure, CaAl₂Si₂. CaAl₂Si₂ has the La₂O₃-type structure (space group: P3m1 (No. 164), a = 4.130 Å, c =7.145 Å, Z = 1), where Si and Al atoms are arranged in the chemically ordered double-corrugated hexagonal layers and Ca atoms are intercalated between them,⁴ as shown in Figure 1d. For the characteristic forms of the Al-Si layers, the bonding nature and stability of the layers in CaAl₂Si₂ and

* Corresponding author. E-mail: IMAI.Motoharu@nims.go.jp.

(2) Manfrinetti, P.; Fornasini, M. L.; Palenzona, A. *Intermetallics* 2000, 9, 223.

those in the related compounds have been examined by calculations.^{6–9} The physical properties of these compounds have been studied experimentally. CaSi₂ is a semimetal.^{10–13} Recently, Ca(Al_{0.5}, Si_{0.5})₂, Ca(Al_x, Si_{1-x})₂ for x = 0.5, which is isostructural to a superconducting material MgB₂ with a transition temperature for superconductivity $T_{\rm C}$ of 39 K,¹⁴ was reported to be a superconducting material with a $T_{\rm C}$ of 7.8 K.^{5,15} On the other hand, the physical properties of CaAl₂-Si₂ have not been studied.

In this paper, we report the electrical resistivity and Hall coefficient of single-crystalline $CaAl_2Si_2$ at temperatures ranging from 2 to 300 K. The transport measurements revealed that $CaAl_2Si_2$ is a metal whose dominant carriers are holes at temperatures below 150 K while they are electrons above 150 K.

Single crystals were grown by the floating (FZ) method. The details of the FZ furnace are described elsewhere.¹⁶ The feed rods were prepared by the Ar-arc melting of a mixture of CaSi₂ and Al. Crystal growth was carried out at a growth rate of 10 mm/h in a flowing Ar atmosphere with a flow rate of 2000 cm³/min. During the growth, the seed and feed rods were rotated at 13 rpm in opposite directions to reduce the asymmetry of the temperature distribution and induce forced convection in the molten zone. The grown material had a cylindrical shape with a diameter of about 10 mm and a length of about 60 mm. The surface has a metallic luster. Observation by scanning electron microscope showed that the grown crystal was a single phase.

Chemical compositions determined by inductively coupled plasma mass-spectroscopy are 19.7 at. % Ca, 39.3 at. % Al,

- (6) Zheng, C.; Hoffmann, R.; Nesper, R.; von Schnering, H.-G. J. Am. Chem. Soc. 1986, 108, 1876.
- (7) Ramirez, R.; Nesper, R.; von Schnering, H.-G. Z. Naturforch. 1987, 42a, 670.
- (8) Zheng, C.; Hoffmann, R. J. Solid State Chem. 1988, 72, 58.
- (9) Burdett, J. K.; Miller, G. J. Chem. Mater. **1990**, 2, 12. (10) Fahy, S.; Hamann, D. R. Phys. Rev. B **1990**, 41, 7587.
- (10) Taily, S., Hamann, D. K. Thys. Rev. D 1990, 41, 7307.
- (11) Bisi, O.; Braicovich, L.; Carbone, C.; Lindau, I.; Iandelli, A.; Olsece, G. L.; Palenzona, A. *Phys. Rev. B* **1989**, *40*, 10194.
 (12) Hirano, T. *J. Alloys Compd.* **1991**, *167*, 329.
- (13) Affronte, M.; Laborde, O.; Olcese, G. L.; Palenzona, A. J. Alloys Compd. 1998, 274, 68.
- (14) Nagamatsu, J.; Nakagawa, N.; Muranaka, T.; Zenitani, Y.; Akimitsu, J. *Nature (London)* **2001**, *410*, 63.
- (15) Imai, M.; Sadki, E. S.; Abe, H.; Nishida, K.; Kimura, T.; Sato, T.; Hirata, K.; Kitazawa, H. Phys. Rev. B 2003, 68, 064512.
- (16) Tanabe H.; Hirano, T. Nippon Kinzoku Gakkaishi **1988**, 52, 1154, in Japanese.

10.1021/ic049720n CCC: \$27.50 © 2004 American Chemical Society Published on Web 07/23/2004

⁽¹⁾ For example: Wells, A. F. *Structural Inorganic Chemistry*, 5th ed.; Clarendon Press: Oxford, 1984.

⁽³⁾ Böhm, J.; Hassel, O. Z. Anorg. Allg. Chem. 1927, 160, 152.

⁽⁴⁾ Gladyshevsky, E. I.; Kripyakevich, P. I.; Bodak, O. I. Ukr. Fiz. Zh. 1967, 12, 447. Petzow G.; Effenberg, G. Ternary Alloys; VCH: Weinheim, 1990; Vol. 3.

⁽⁵⁾ Imai, M.; Nishida, K.; Kimura, T.; Abe, H. Appl. Phys. Lett. 2002, 80, 1019.

COMMUNICATION



Figure 1. Crystal structures of (a) Si, (b) CaSi₂, (c) Ca(Al_x,Si_{1-x})₂ (0.225 $\le x \le 0.675$), and (d) CaAl₂Si₂.



Figure 2. Electrical resistivity along the [100] direction ($\rho_{[100]}$) and that along the [001] direction ($\rho_{[001]}$) as a function of temperature. The lower panel shows the anisotropy ratio of the electrical resistivity $\rho_{[001]}/\rho_{[100]}$ as a function of temperature.

and 41.0 at. % Si, which is consistent with an ideal chemical compositions of CaAl₂Si₂. An analysis of the powder X-ray diffraction pattern showed that CaAl₂Si₂ has a hexagonal unit cell with lattice constants of a = 4.1410(3) Å and c = 7.1329(7) Å, which is consistent with the previous report (a = 4.130(10) Å and c = 7.145(15) Å).⁴ X-ray back Laue photographs of the grown crystal indicated that the grown crystal is a good-quality single crystal.

The top part of Figure 2 shows the electrical resistivity of CaAl₂Si₂ along the [100] direction ($\rho_{[100]}$) and that along the [001] direction ($\rho_{[001]}$) as a function of the temperature. Electrical resistivity was measured by means of the standard four-probe method. The resistivity of CaAl₂Si₂, which has a finite value at 2 K, increases with increasing temperature, and saturates around 200 K. The resistivity $\rho_{[001]}$ is larger than $\rho_{[100]}$ throughout the temperature range. The bottom part of Figure 2 shows the anisotropy ratio of the resistivity $\rho_{[001]}$ / $\rho_{[100]}$, which increases from 1.4 to 2.5 with increasing temperature.

Figure 3 shows the Hall coefficient $R_{\rm H}$ as a function of the temperature. The Hall coefficient was measured by means of the standard four-probe method. The current direction is



Figure 3. Hall coefficient $R_{\rm H}$ as a function of temperature. The current direction is the [100] direction, and the direction of the applied magnetic field is the [001] direction.

the [100] direction, and the direction of applied magnetic field is the [001] direction. The Hall coefficient $R_{\rm H}$ was evaluated from the antisymmetric part of transverse resistivity ρ_{xy} measured at a magnetic field of -1 T and +1 T, $R_{\rm H} =$ $[\rho_{xy}(+1T) - \rho_{xy}(-1T)]/2$, to exclude the longitudinal contribution due to a possible misalignment of the Hall contacts. The value of $R_{\rm H}$, which is $\pm 130 \times 10^{-9} \text{ m}^3/\text{C}$ at 5 K, decreases with increasing temperature and reaches $-5.1 \times$ 10^{-9} m³/C at 300 K through zero at about 150 K. The sign reversal in $R_{\rm H}$ is observed also in CaSi₂, a semimetal in which both electrons and holes contribute to the transport properties, although the sign of $R_{\rm H}$ in CaSi₂ changes from negative to positive with increasing temperature.13 In the framework of a two-band model in which the electrons and holes contribute to the transport properties, the electrical conductivity σ (the inverse of electrical resistivity ρ) and $R_{\rm H}$ can be written as follows:

$$\frac{1}{\rho} = \sigma = n_{\rm e} e \mu_{\rm e} + n_{\rm h} e \mu_{\rm h} \tag{1}$$

$$R_{\rm H} = \frac{n_{\rm h} \mu_{\rm h}^2 - n_{\rm e} \mu_{\rm e}^2}{e(n_{\rm h} \mu_{\rm h} + n_{\rm e} \mu_{\rm e})^2} = \frac{n_{\rm h} - n_{\rm e} b^2}{e(n_{\rm h} + n_{\rm e} b)^2}$$
(2)

$$b = \mu_{\rm e}/\mu_{\rm h} \tag{3}$$

Inorganic Chemistry, Vol. 43, No. 17, 2004 5187

COMMUNICATION

Table	1.	Tranport	Parameters	of	CaSi ₂ ,	Ca(Al	0.5,Si0.5)2,	and	CaAl ₂ Si ₂
-------	----	----------	------------	----	---------------------	-------	--------------	-----	-----------------------------------

	$\rho_{4.2K}$ ($\mu\Omega$ cm)	ρ_{273K} ($\mu\Omega$ cm)	ρ _{273K} /ρ _{4.2K}	$\rho_{\rm llc}/\rho_{\rm llab}$ at 273 K	$\frac{R_{\rm H}}{(10^{-10} { m m}^3/{ m C})}$	type of dominant carriers	remarks
CaSi ₂	10.1^{a}	33.2 ^{<i>a</i>}	3.29 ^a	1.4^{b}	-13 at 10 K ^a +3 at 280 K ^a	electrons ^a holes ^a	semimetal
$[100]^{c}$	19.5^{b}	63 ^b	3.2^{b}				
$[001]^{c}$	25.5^{b}	56^{b}	2.2^{b}				
Ca(Al _{0.5} ,Si _{0.5}) ₂				12^{d}	-7.3 at 290 K ^e	electrons ^e	superconductor with a $T_{\rm C}$ of 7.8 K
ab	17^{e}	78.5^{e}	4.62^{e}				
c	539 ^d	949^{d}	1.76^{d}				
CaAl ₂ Si ₂				2.4	+1300 at 5 K -51 at 300 K	holes electrons	
[100] [001]	335 479	548 1327	1.64 2.77				

^{*a*} Reference 13. ^{*b*} Reference 12. ^{*c*} The rhombohedral lattice is used in ref 12. The lattice is transformed into the hexagonal lattice here. ^{*d*} Reference 18. ^{*e*} Reference 15.

where *e*, n_i , and μ_i (i = e or h) are an elemental charge, the carrier density of electrons or holes, and the mobility of electrons or holes, respectively.¹⁷ The positive value of $R_{\rm H}$ means that $n_{\rm h}\mu_{\rm h}^2$ is larger than $n_{\rm e}\mu_{\rm e}^2$, suggesting that the dominant carriers are holes. On the other hand, the negative value of $R_{\rm H}$ means that $n_{\rm e}\mu_{\rm e}^2$ is larger than $n_{\rm h}\mu_{\rm h}^2$, suggesting that dominant carriers are electrons. Therefore, the observed temperature dependence of $R_{\rm H}$ in CaAl₂Si₂ suggests that both electrons and holes contribute to the transport properties, and that the dominant carriers are holes at the temperatures below 150 K but electrons at the temperatures above 150 K.

The transport parameters of CaSi₂, Ca(Al_{0.5}, Si_{0.5})₂, and CaAl₂Si₂ are listed in Table 1.^{12,13,15,18} Both ρ and R_H of CaAl₂Si₂ are one order larger than those of CaSi₂ and Ca-(Al_{0.5}, Si_{0.5})₂, suggesting that the number of carriers in CaAl₂-Si₂ is less than those in CaSi₂ and Ca(Al_{0.5}, Si_{0.5})₂. The anisotropy ratio of CaAl₂Si₂ is the same order as that of CaSi₂. CaAl₂Si₂ is a metal in which both electrons and holes contribute to the transport properties, as is CaSi₂.

The electrical resistivity in most elemental metals can be described by Bloch–Grüneisen theory or Wilson's modification of it:

$$\rho = \rho_0 + A(T/\theta_D)^n \int_0^{\theta_D/T} \frac{z^n \, \mathrm{d}z}{(e^z - 1)(1 - e^{-z})},\tag{4}$$

where ρ_0 , θ_D , and *A* are the residual resistivity, the Debye temperature, and a constant, respectively. The constant *n* is equal to 3 when the "s-d electron-phonon" scattering dominates the resistivity, and the constant *n* is equal to 5 when the "s-s electron-phonon" scattering dominates.^{17,19-21} At temperatures from 2 to 40 K, the resistivity of CaAl₂Si₂ is fitted well with eq 4 with *n* = 3. Above that temperature,

(18) Imai, M. Unpublished.

the resistivity further increases with temperature, but it saturates around 200 K, which is in contrast to temperature dependence described by eq 4, according to which the resistivity just keeps increasing with temperature. In these theories, the main effect of the temperature is on the relaxation time, which in turn affects mobility; however, it has no effect on the number of carriers. The temperature dependence of $R_{\rm H}$ in CaAl₂Si₂ shows that the sign of the numarator of eq 2 ($n_{\rm h} - n_{\rm e}b^2$) changes from negative to positive with increasing temperature, which suggests the possibility that $n_{\rm e}$ and/or b^2 become larger with increasing temperature. When n_i changes with the temperature, the temperature dependence of ρ does not obey eq 4. This may be the reason that the temperature dependence of ρ in CaAl₂Si₂ is different from that in typical elemental metals.

The band structure of CaAl₂Si₂ has not been calculated yet although that of ionized double-hexagonal corrugated Al–Si layers, the Al₂Si₂^{2–} layers, has been calculated.⁷ The band calculations of CaSi₂ and Ca(Al_{0.5}, Si_{0.5})₂ showed that Ca d states locate around E_F and play an important role for transport properties.^{10,22} Also in CaAl₂Si₂, the states originating from Ca are expected to locate around E_F and to affect the transport properties, as do those of CaSi₂ and Ca(Al_{0.5}, Si_{0.5})₂.

In summary, the electrical resistivity and Hall coefficient measurements of single-crystalline CaAl₂Si₂ revealed that CaAl₂Si₂ is a metal in which both electrons and holes contribute to the transport properties; the dominant carriers are holes at temperature below 150 K but electrons above that temperature.

Acknowledgment. The authors thank A. P. Tsai, T. Hirano, I. Nakatani, and H. Mamiya of NIMS for allowing us the use of their arc furnace, FZ furnace, and Physical Properties Measurement System (Quantum Design), respectively.

IC049720N

⁽¹⁷⁾ For example: Ziman, J. M. Principles of the Theory of Solids, 2nd ed.; Cambridge University Press: Cambridge, 1972.

⁽¹⁹⁾ Wilson, A. H. Proc. R. Soc. London 1938, A167, 580.

⁽²⁰⁾ Webb, G. W. Phys. Rev. 1969, 181, 1127.

⁽²¹⁾ Pinski, F. J.; Allen, P. B.; Butler, W. H. Phys. Rev. B 1981, 23, 5080.

⁽²²⁾ Shein, I. R.; Ivanovskaya, V. V.; Medvedeva, N. I.; Ivanovskii, A. L. JETP Lett. 2002, 76, 223.