A Chiral Self-Catenated Dual-Ligand Coordination Polymer Constructed from Three Distinct Interwoven Helical Motifs Interconnected by One-Dimensional Chains

David P. Martin, Richard J. Staples, and Robert L. LaDuca*

Lyman Briggs College and Department of Chemistry, Michigan State University, East Lansing, Michigan 48825

Received August 5, 2008

Hydrothermal synthesis has afforded $[Co_3(oba)_3(bpmp)_2]_n$ [oba = oxybis(benzoate), bpmp = bis(4-pyridylmethyl)piperazine], a chiral coordination polymer possessing an aesthetic self-catenated threedimensional structure with an unprecedented 8-connected uninodal $4^{4}5^{17}6^{7}$ topology. The network is formed by the junction of $\{Co_3O_2\}$ clusters into three different homochiral interlocked helical motifs, linked by one type of "infinite" chain pattern. Antiferromagnetic exchange is observed within the $\{Co_3O_2\}$ clusters.

One of the most commonly encountered features of coordination polymer systems is the mutual interpenetration of networks, with a variety of interesting connectivity patterns both analogous to those in naturally occurring minerals and completely unprecedented.^{1,2} In comparison to interpenetrated coordination polymers, those exhibiting self-catenation, in which connecting rods of the network penetrate through the smallest circuits of the same network, are much rarer.³ Self-catenated structures can range from the relatively simple uniform 3-connected (12,3)-**twt** lattice⁴ and the more intricate 5-connected uniform 6¹⁰-**rld-z** topology⁵ to extremely complicated examples.⁶ While the deliberate construction of highly connected self-catenated coordination polymer networks remains a synthetic challenge, we and other researchers have found that the kinked dicarboxylate

- (a) Batten, S. R. CrystEngComm 2001, 3, 67–73.
 (b) Carlucci, L.; Ciani, G.; Proserpio, D. M. Coord. Chem. Rev. 2003, 246, 247–289.
- (2) Blatov, V. A.; Carlucci, L.; Ciani, G.; Proserpio, D. M. CrystEngComm 2004, 6, 377–395.
- (3) (a) Wang, X.-L.; Qin, C.; Wang, E.-B.; Su, Z.-M. Chem.—Eur. J. 2006, 12, 2680–2691. (b) Bi, M.; Li, G.; Zou, Y.; Shi, Z.; Feng, S. Inorg. Chem. 2007, 46, 604–606. (c) Niel, V.; Thompson, A. L.; Goeta, A. E.; Enachescu, C.; Hauser, A.; Galet, A.; Munoz, M. C.; Real, J. A. Chem.—Eur. J. 2005, 11, 2047–2060. (d) Lloyd, G. O.; Atwood, J. L.; Barbour, L. J. Chem. Commun. 2005, 1845–1847.
- (4) Abraham, B. F.; Batten, S. R.; Grannas, M. J.; Hamit, H.; Hoskins, B. F.; Robson, R. Angew. Chem., Int. Ed. 1999, 38, 1475–1477.
- (5) Montney, M. R.; Mallika Krishnan, S.; Patel, N. M.; Supkowski, R. M.; LaDuca, R. L. *Cryst. Growth Des.* **2007**, *7*, 1145–1153.
 (6) Wang, X. L.; Qin, C.; Wang, E.-B.; Li, Y.-G.; Su, Z.-M.; Xu, L.;
- (6) Wang, X. L.; Qin, C.; Wang, E.-B.; Li, Y.-G.; Su, Z.-M.; Xu, L.; Carlucci, L. Angew. Chem., Int. Ed. 2005, 44, 5824–5827.

9754 Inorganic Chemistry, Vol. 47, No. 21, 2008

Chart 1. Structures of oba and bpmp



Inorg. Chem. 2008, 47, 9754-9756

oxybisbenzoate (oba) ligand can promote their formation. Wang has reported a (4.8^2) $(4.6^4.8^4.10)$ three-dimensional self-catenated topology within the structure of {[Ni(oba)(4,4'-bipyridine)]•H₂O}.⁶ Recently, we were able to prepare {[Ni(oba)(4,4'-dipyridylamine)]•H₂O}, a self-catenated layered phase with a unique nondiamondoid 6⁶ topology.⁷ Inspired by these previous successes in preparing oba/ organodiimine self-catenated coordination polymers, we undertook synthetic explorations toward metal—organic materials incorporating both oba and bis(4-pyridylmeth-yl)piperazine (bpmp)⁸ tethering ligands (Chart 1).

Hydrothermal reaction of cobalt nitrate, H₂oba, and bpmp resulted in a good yield⁹ of crystalline $[Co_3(oba)_3(bpmp)_2]_n$ (1), a chiral coordination polymer with a complex and unprecedented 8-connected self-catenated three-dimensional structure. According to single-crystal X-ray diffraction,¹⁰ compound 1 crystallized in the chiral hexagonal space group $P6_122$. The Flack parameter¹¹ of 0.00(3) illustrates the enantiomeric purity of the single crystal. Another single

- (8) Pocic, D.; Planeix, J.-M.; Kyritsakas, N.; Jouaiti, A.; Abdelaziz, H.; Wais, M. CrystEngComm 2005, 7, 624–628.
- (9) A mixture of Co(NO₃)₂•6H₂O (55 mg, 0.19 mmol), H₂oba (48 mg, 0.19 mmol), and bpmp (100 mg, 0.37 mmol) in 10 mL of H₂O was heated at 120 °C under autogenous pressure in an acid digestion bomb for 72 h. Small orange crystals of compound 1 (68 mg, 0.045 mmol, 71% yield) were isolated after washing with distilled water, ethanol, and acetone. IR (cm⁻¹, powder): 2810(w), 1610(m), 1594(m), 1499(w), 1387(s), 1301(w), 1240(s, sh), 1232(s), 1160(m), 1129(w), 1067(w), 1011(m), 875(w), 853(w), 842(w), 794(w), 779(s), 766(w), 731(w), 694(w), 654(m). Anal. Calcd for C₇₄H₆₄Co₃N₈O₁₅ (1): C, 59.97; H, 4.35; N, 7.56. Found: C, 59.39; H, 4.12; N, 7.62.
- (10) Crystal data for 1: $C_{74}H_{64}Co_3N_8O_{15}$, fw = 1482.12 g mol⁻¹, hexagonal, space group $P6_{1}22$, a = 13.6840(4) Å, c = 61.780(4) Å, V = 10018.5(8) Å³, Z = 6, $\rho_{calcd} = 1.474$ g cm⁻³, μ (Mo K α) = 0.812 mm⁻¹, F(000) = 4590, R [$I > 2\sigma(I)$] = 0.0510, wR [$I > 2\sigma(I)$] = 0.1005, GOF = 0.856, T = 173(2) K, CCDC 689544.
- (11) Flack, H. D. Acta Crystallogr., Sect. A 1983, 29, 876–881.

10.1021/ic801471e CCC: \$40.75 © 2008 American Chemical Society Published on Web 10/09/2008

 $[\]ast$ To whom correspondence should be addressed. E-mail: laduca@msu.edu.

⁽⁷⁾ Martin, D. P.; Supkowski, R. M.; LaDuca, R. L. Inorg. Chem. 2007, 46, 7917–7922.



Figure 1. Coordination environment and trinuclear unit within $[Co_3(oba)_3(bpmp)_2]_n$. Thermal ellipsoids are shown at 50% probability.

crystal from the same synthetic experiment crystallized in the enantiomorphic space group $P6_522$ with a Flack parameter of 0.01(5), showing that a racemic mixture of the two enantiomeric forms likely exists in the bulk crystalline sample of **1**.¹²

The asymmetric unit of **1** consists of two Co atoms, one of which (Co1) lies on a crystallographic 2-fold axis, one half and one full oba units (oba-A and oba-B), and halves of two crystallographically distinct bpmp ligands (bpmp-A and bpmp-B). The bpmp-B ligand is disordered equally over two sets of positions. The operation of the symmetry at Co1 generates a bent {Co₃O₂} trinuclear kernel (Figure 1), where two Co2 atoms are linked to the central Co1 atom through μ_2 -O atoms belonging to carboxylate units of two oba-B ligands. The Co1–Co2 distance within this unit is 3.565 Å.

The central Co1 atom possesses a distorted $\{CoO_6\}$ coordination octahedron, with two symmetry-related cis O donors belonging to two oba-A ligands, two cis O donors belonging to two oba-B ligands, and two trans-disposed O donors from two different oba-B ligands. The Co2 atoms on the periphery of the trinuclear units have a $\{CoN_2O_4\}$ -distorted octahedral coordination sphere, where the two cis N donors belong to bpmp-A and bpmp-B ligands. Three O donors from two oba-B ligands are bound to Co2 in a *fac* pattern, with a pair of these being donated by a chelating carboxylate terminus of a single oba-B ligand. A single O atom from an oba-A carboxylate rounds out the coordination sphere. Bond lengths and angles (Table S1 in the Supporting Information) are standard.

Adjacent {Co₃O₂} units are aggregated by exotetradentate oba-A ligands with a tetra-monodentate binding mode into a {Co₃O₂(oba-A)}_n⁴ⁿ⁺ 6-fold right-handed helix (helix-1), which follows the *c* crystal direction with a long pitch of 61.78 Å (Figures 2a and S1 in the Supporting Information). The Co1•••Co1 distance between neighboring trinuclear units within helix-1 measures 11.493 Å; the related Co2•••Co2 distance is 15.379 Å. The distance across the cylinder inscribed by the rotation of helix-1, as defined by the path of the outermost Co atoms, is ~11 Å.

Two pairs of oba-B ligands also covalently link neighboring trinuclear units together into a 6-fold double helix, $\{Co_3(oba-B)_2\}$, denoted as helix-2 (Figures 2b and S2 in the



Figure 2. Four distinct submotifs within the structure of **1**, all viewed down *c*: (a) $\{Co_3O_2(oba-A)\}_n$ 6-fold right-handed single helix (helix-1); (b) $\{Co_3(oba-B)_2\}_n$ 6-fold right-handed double helix (helix-2); (c) $\{Co_3O_2(bpmp-A)\}_n$ 6-fold right-handed single helix (helix-3); (d) $\{Co_3O_2(bpmp-B)\}_n$ chain.

Supporting Information). While the oba-B ligands adopt an exotetradentate binding mode, one carboxylate terminus serves as a bismonodentate donating group to two Co atoms (Co1 and Co2), while the other chelates to another Co2 atom, with one O atom (O5) also bridging to another Co1 atom. The Co1…Co1 distance within helix-2 measures 13.402 Å, approximately 1.5 Å longer than that within helix-1, although both helix-1 and helix-2 have the same 61.78 Å pitch and right-handed helicity. The inner diameter across the cylinder inscribed by the rotation of helix-2, as defined by the path of the Co1 atoms, is ~17 Å. A view down the *c* crystal axis of a single helix-2 presents an incipient channel-like void; however, this is filled with a helix-1 motif.

A third right-handed 6-fold helical subunit is present within the structure of 1, formed by the linkage of $\{Co_3O_2\}$ trinuclear units at their terminal Co2 atoms through tethering bpmp-A ligands. This {Co₃O₂(bpmp-A)} single helix, helix-3, is coaxial with both helix-1 and helix-2, with identical pitch and handedness (Figures 2c and S3 in the Supporting Information). The longer extent of the bpmp-A ligands relative to the dicarboxylate tethers results in a Co2····Co2 distance of 15.009 Å, within a Co1···Co1 intertrimer distance of 21.245 Å. The inner diameter of the "star"-shaped channel within helix-3, measured in a fashion similar to that of helix-2, is ~ 37.8 Å. This very large internal space accommodates one helix-2 and seven helix-1 subunits inside it. The bpmp-A ligands within helix-3 adopt a bent cis conformation of their methylpyridyl arms across their central piperazinyl rings, with a 51.7° N····N····N torsion angle.

The fourth distinct linkage motif between $\{Co_3O_2\}$ trinuclear units occurs at their Co2 terminal atoms through tethering bpmp-B ligands. These tethers adopt a more linear conformation than bpmp-A, with a N···N···N torsion angle of 171°, providing a longer Co2···Co2 distance of

^{(12) (}a) Li, F.; Li, T.; Li, X.; Li, X.; Wang, Y.; Cao, R. Cryst. Growth Des. 2006, 6, 1458–1462. (b) Sasa, M.; Tanaka, K.; Bu, X.-H.; Shiro, M.; Shionoya, M. J. Am. Chem. Soc. 2001, 123, 10750–10751.



Figure 3. Framework perspective of the $4^{4}5^{17}6^{7}$ self-catenated 8-connected uninodal topology in **1**. Helix-1, helix-2, helix-3, and chain-4 submotifs are represented in red, orange, blue, and green, respectively.

16.979 Å. Because of the nearly linear nature of the bpmp-B tethers, a { $Co_3O_2(bpmp-B)$ } one-dimensional chain motif (chain-4) is formed (Figure 2d), in contrast to the homochiral helices resulting from the other three types of connecting ligands in compound **1**. The Co1···Co1 distance between { Co_3O_2 } trinuclear units within chain-4 is 23.701 Å.

By means of the four types of tethering ligands, producing four coexisting structural patterns (helix-1, helix-2, helix-3, and chain-4), each trinuclear $\{Co_3O_2\}$ unit is directly covalently linked to eight others through 10 connecting ligands (two oba-A, four oba-B, two bpmp-A, and two bpmp-B). Considering each pair of oba-B ligands as a single connection, an 8-connected three-dimensional coordination polymer network can be invoked for 1 (Figure 3). Each helix-2 unit covalently connects the inner perimeters of six individual helix-1 units and surrounds, but does not covalently link, a seventh helix-1 motif. Each helix-1, in turn, connects covalently to 18 other helix-1 units (all neighbors and next-nearest neighbors) through six helix-2 units. Treating the Co1 atoms as 4-connected nodes with single oba-A ligands and oba-B pairs as linkers results in a 6⁴8² chiral quartz (qtz) lattice (Figure S4 in the Supporting Information), according to TOPOS.13

In turn, each helix-3 also covalently connects six individual helix-1 units via their outer perimeters and encapsulates a seventh helix-1 motif within its interior. In a connectivity pattern similar to that promoted by helix-2, six helix-3 units covalently conjoin 18 neighboring and next-nearest-neighboring individual helix-1 units. Analysis of the coordination polymer connectivity provided by helix-1 and helix-3 also produces a $6^{4}8^{2}$ chiral **qtz** lattice (Figure S5 in the Supporting Information). On the other hand, a system including only helix-2 and helix-3 results in a 4-fold interpenetrated chiral qtz lattice (Figure S6 in the Supporting Information). Inclusion of helix-1 into the latter results in interconnectivity between the individual networks within 4-fold interpenetrated lattices of helix-2 and helix-3; thus, the overall network produced by the interweaving of all of the helical motifs in 1 can be considered to be self-catenated. This self-catenated 6-connected net possesses an unprecedented $4^46^{9}8^2$ topology. This network is further linked by the chain-4 motifs that cut through all three helical motifs via Co2 atoms on the periphery of the $\{Co_3O_2\}$ trimeric subunits. If only helix-2, helix-3, and chain-4 are treated as connecting linkers with Co1 atoms as connecting nodes, a 6-connected threedimensional self-catenated network with a remarkably regular 6¹⁵ topology can be invoked. Topological analysis of the full covalent net of 1, including all four submotifs and treating only Co1 atoms as connecting nodes, reveals the presence of a previously unknown self-catenated 8-connected uninodal lattice with 4⁴5¹⁷6⁷ topology,¹⁴ shown in Figure 3. Bent ligands are crucial for the generation of this framework in order to avoid edge crossings.¹⁵ A more complete list of network topologies generated from partial connectivity of the full network is given in Table S2 in the Supporting Information.

A variable-temperature magnetic susceptibility was carried out to investigate spin communication within the {Co₃O₂} trimers within the structure of **1**. The higher temperature data could be fit to the Curie–Weiss law, with C = 2.74 cm³ K mol⁻¹ and $\Theta = -8.1$ K, plausibly indicative of antiferromagnetic coupling within the {Co₃O₂} trimers, although zerofield-splitting effects cannot be ruled out (Figure S7 in the Supporting Information). Thermogravimetric analysis (TGA) of **1** (Figure S8 in the Supporting Information) revealed that no mass loss occurred until ~250 °C, whereupon expulsion of the organic components commenced. The mass remnant at ~875 °C of 19.9% is roughly consistent with the deposition of CoO (calcd 18.4%).

In conclusion, the hydrothermal self-assembly of the selfcatenated 8-connected three-dimensional network of $[Co_3(oba)_3(bpmp)_2]_n$ reveals the feasibility of accessing highly connected coordination polymer networks with unprecedented modes of entanglement through the employment of very long kinked and/or conformationally flexible tethering ligands. Extension of the tether length and variances in conformational flexibility, donor disposition, and hydrogenbonding capability within dicarboxylate or organodiimine ligands are certain to expand the scope of highly connected self-catenated coordination polymer networks.

Acknowledgment. This work was funded by Michigan State University. We thank Dr. Ronald Supkowski of King's College, Wilkes-Barre, PA, for performing the TGA.

Supporting Information Available: X-ray crystallographic data in Cif format, supplementary figures, Curie–Weiss plot, TGA trace, table of relevant bond distances and angles, and a table of network topologies present within **1**. This material is available free of charge via the Internet at http://pubs.acs.org.

IC801471E

⁽¹³⁾ Blatov, V. A.; Shevchenko, A. P.; Serezhkin, V. N. J. Appl. Crystallogr. 2000, 33, 1193.

⁽¹⁵⁾ Monge, A.; Snejko, N.; Gutierrez-Puebla, E.; Medina, M.; Cascales, C.; Ruiz-Valero, C.; Iglesias, M.; Gamez-Lor, B. *Chem. Commun.* 2005, 1291–1293.