Orbital Hybridization and Magnetic Coupling of the A-Site Cu Spins in $CaCu_3B_4O_{12}$ (B = Ti, Ge, and Sn) Perovskites

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X-ray absorption spectroscopy (XAS) spectra near the O K-edge of A-site-ordered perovskite with A-site Cu^{2+} ($S = 1/_2$) spins were measured. The spectra of ferromagnetic $CaCu_3Ge_4O_{12}$ and $CaCu_3Sn_4O_{12}$ showed hybridization between Cu 3d and O 2p orbitals, but magnetic circular dichroism measurement revealed that the O 2p orbital played a less important role in magnetic interaction. The XAS spectra of antiferromagnetic $CaCu_3Ti_4O_{12}$, on the other hand, showed strong hybridization of the Cu 3d, Ti 3d, and O 2p orbitals. These results demonstrated that direct exchange interaction of the Cu^{2+} spins primarily determined the ferromagnetic ordering of $CaCu_3Ge_4O_{12}$ and $CaCu_3Sn_4O_{12}$, whereas the involvement of Ti 3d orbitals induced the antiferromagnetic property in $CaCu_3Ti_4O_{12}$.

A-site-ordered perovskites with the general formula $AA'_{3}B_{4}O_{12}$ show diverse and intriguing physical properties.^{1,2} Figure 1 shows the crystal structure of these compounds. Distinctive features of these compounds are not only that they have a very unusual ordered arrangement of A-site ions but also that they contain magnetic Jahn–Teller ions such as Cu^{2+} and Mn^{3+} at the originally 12-fold-coordinated A' site. A compound thus could be a rare A-site magnetic material due to interaction between the magnetic ions at the A site.

CaCu₃Ge₄O₁₂, CaCu₃Ti₄O₁₂, and CaCu₃Sn₄O₁₂ are examples of such A-site magnetic materials; all of them contain magnetic Cu²⁺ ions at the A' site and nonmagnetic ions at the B site. Only Cu²⁺ ($S = \frac{1}{2}$) spins at the A' site thus contribute to the magnetic properties of the oxides. Interestingly, CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂ show ferromagnetic properties with Curie temperatures of 13 and 10 K, respectively, in contrast to an antiferromagnetic one for CaCu₃Ti₄O₁₂ ($T_N = 25$ K).^{3,4} There-

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Figure 1. Crystal structure of $ACu_3B_4O_{12}$ (B = Ti, Ge, and Sn). A (large sphere) and Cu (middle sphere) ions are ordered, and the B-site ions form BO_6 octahedra. The two-headed arrow shows direct interaction between Cu^{2+} spins, and the thick lines show the Cu–O–Ti–O–Cu path.

fore, the A-site Cu^{2+} spins can align either ferromagnetically or antiferromagnetically within the same structural framework, depending on the nonmagnetic B-site ion.

In the special alignment of the CuO₄ units in the ordered structure, normal superexchange interaction through O ions is not induced. Thus, the observed ferromagnetic and antiferromagetic behaviors are completely different from a bond-angle switching in the SeCuO₃-TeCuO₃ perovskites from ferromagnetic to antiferromagetic regions according to the Kanamori-Goodenough rules.⁵ Instead, direct exchange interaction between Cu^{2+} ions at the A' site in CaCu₃B₄O₁₂ appears to primarily determine the magnetic properties, which explains the ferromagnetic behavior in CaCu3Ge4O12 and CaCu3Sn4O12. In CaCu₃Ti₄O₁₂, on the other hand, antiferromagnetic superexchange interaction should overcome ferromagnetic direct interaction. Because the empty Ti 3d band can hybridize with the O 2p and Cu 3d orbitals near the Fermi energy $(E_{\rm F})$, the contribution of the Ti 3d band should play an important role in the magnetic property.

In this study, we investigated the electronic structures of $CaCu_3Ge_4O_{12}$, $CaCu_3Ti_4O_{12}$, and $CaCu_3Sn_4O_{12}$ by using X-ray absorption spectroscopy (XAS). We can clearly see the difference in O K-edge spectra between the ferromagnetic ($CaCu_3Ge_4O_{12}$ and $CaCu_3Sn_4O_{12}$) and the antiferromagnetic

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Figure 2. Temperature dependence of the magnetic susceptibility of $CaCu_3B_4O_{12}$ (B = Ti, Ge, and Sn). These data are from ref 3.

 $(CaCu_3Ti_4O_{12})$ compounds. The observed spectral weight in $CaCu_3Ti_4O_{12}$ is the result of strong hybridization between Ti 3d and O 2p orbitals, and we believe this is clear evidence of the involvement of Ti 3d orbitals in the antiferromagnetic property.

Powder samples of CaCu₃Ge₄O₁₂, CaCu₃Ti₄O₁₂, and CaCu₃-Sn₄O₁₂ were prepared by solid-state reaction processes in the same manner as that described in refs 3 and 6. CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂ were synthesized under high-pressure and high-temperature conditions. All of the samples obtained were confirmed to be single phases by X-ray diffraction. As shown in Figure 2 (data from ref 3), magnetic measurements confirm the ferromagnetic behavior for CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂ and the antiferromagnetic one for CaCu₃Ti₄O₁₂. Effective magnetic moments calculated from the Curie constants in the Curie–Weiss susceptibility also confirm the Cu²⁺ (S = 1/2; μ_{eff} = 1.73) localized spins.

The O K-edge XAS spectra of the samples were measured at 9 and 300 K by a total electron yield method for the synthesized samples at BL25SU and BL27SU of SPring-8 in Japan. The energy resolution $E/\Delta E$ at the O K-edge was greater than 5000. The incident photon energy was calibrated by measuring the energies of the Ti L_{2,3}-edges of TiO₂ and the Ni L_{2,3}-edges of NiO. Powder samples were pasted uniformly on a sample holder by using carbon tape. In order to investigate magnetic interaction, the O K-edges magnetic circular dichroism (MCD) spectra of CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂ were also measured at BL25SU at temperatures below the Curie temperatures. The MCD intensity was defined as the difference between two absorption spectra, which are obtained using parallel and antiparallel photon spins along the magnetization direction.

Figure 3 shows the O K-edge XAS spectra of CaCu₃Ge₄O₁₂, CaCu₃Ti₄O₁₂, and CaCu₃Sn₄O₁₂. The arbitrary unit of intensity in each spectrum was set to unity at E = 550 eV. The observed spectrum is the result of an excitation of electrons from the O 1s states to unoccupied O 2p states in the conduction band. Because the empty O 2p states hybridize with states of cations coordinated by O ions, the observed spectra show mostly the transition-metal character.⁷ The spectral weight from 530 to 532 eV corresponds to the Cu 3d and Ti 3d bands, and that around



Figure 3. O K-edge XAS spectra for $CaCu_3Ge_4O_{12}$ (solid squares), $CaCu_3Ti_4O_{12}$ (solid triangles), and $CaCu_3Sn_4O_{12}$ (solid circles).



Figure 4. O K-edge XAS (closed symbols) and MCD (open symbols) spectra at T = 9 K for CaCu₃Ge₄O₁₂ (squares) and CaCu₃Sn₄O₁₂ (circles).

535 eV corresponds to the Ca 3d, Ge s, and Sn s bands. The absorption spectra around 540-545 eV reflect the character of the Cu 4sp and Ti 4sp bands. These assignments are consistent with previously analyzed CaMnO₃ and germanium oxides.⁷⁻⁹

Figure 4 shows magnified XAS spectra near the Cu 3d–O 2p hybridized region and the XAS-MCD spectra at the O K-edge for the ferromagnetic CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂. The MCD signal for CaCu₃Ge₄O₁₂ is not evident, and that for CaCu₃Sn₄O₁₂ is very weak, less than 2% around the Cu 3d band region. Compared with the O K-edge MCD intensities reported for other ferromagnetic transition-metal oxides, such as 5% for La_{1-x}Sr_xMnO₃¹⁰ 5% for CaCu₃Fe₄O₁₂¹¹ and 4% for BiCu₃-Mn₄O₁₂,¹¹ the observed MCD intensities of the present compounds are apparently weak. Thus, the results clearly show that the magnetic polarization of O 2p electrons in CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂ is very low, suggesting that the Cu–O–Cu path plays a less important role in magnetic interaction. This implies that superexchange interaction is not induced between the Cu spins in the A-site-ordered perovskite compounds and also gives evidence that ferromagnetic direct exchange interaction primarily determines the magnetic properties. This is also consistent with the experimental results on magnetism and our discussion from a structural point of view.³

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Figure 5. Calculated DOS for $CaCu_3Ti_4O_{12}$ (upper panel), $CaCu_3Ge_4O_{12}$ (middle panel), and $CaCu_3Sn_4O_{12}$ (lower panel). The total DOS (shaded region) and partial DOS of Cu (thick lines), B (thin lines), and O (dashed lines) ions are shown in each panel.

For antiferromagnetic CaCu₃Ti₄O₁₂, as shown in Figure 3, it is noted that the XAS intensity at 530–532 eV shows a significant difference from those of ferromagnetic CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂. Because the absorption spectra in the region mainly reflect the hybridization of orbitals between Cu 3d, Ti 3d, and O 2p, this result strongly indicates the contribution of Ti 3d orbitals in the hybridized states in antiferromagnetic CaCu₃Ti₄O₁₂. Thus, hybridization of Ti 3d orbitals should induce antiferromagnetic interaction between the Cu²⁺ spins.

In order to investigate oxygen—metal hybridization in detail, the density of states (DOS) near $E_{\rm F}$ for CaCu₃Ge₄O₁₂, CaCu₃Ti₄O₁₂, and CaCu₃Sn₄O₁₂ were calculated by full potential linearized augmented plane-wave (FLAPW) first-principle calculations within the generalized gradient approximation exchange-correlation function. The antiferromagnetic state for CaCu₃Ti₄O₁₂ was calculated with a *Pm*3 structure model, in which the Cu²⁺ spins are ordered along the [111] direction in an antiparallel manner, whereas ferromagnetic states were investigated for CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂. The FLAPW sphere radii used for the calculations were 2.0, 2.0, 1.9, 1.9, 2.2, and 1.6 au for Ca, Cu, Ge, Ti, Sn, and O, respectively. Self-consistent calculation was carried out on 45 *k*-point meshes in the irreducible Brillouin zone (1000 *k* points in the whole Brillouin zone).

Figure 5 shows the calculated DOS for the three compounds. In order to compare the results with the experimentally observed XAS spectra, we especially pay attention to the unoccupied region above $E_{\rm F}$. In Figure 6, the unoccupied O 2p partial DOS of each compound is plotted in such a manner that the energy position with the maximum partial DOS is shifted to the origin. The widths of the unoccupied O 2p bands are narrow (~0.3 eV) for the ferromagnetic CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂. These bands mainly hybridize with the Cu 3d orbitals. For antiferromagnetic CaCu₃Ti₄O₁₂, on the other hand, the corresponding band is apparently broad (~0.5 eV) and much of the spectral weight is seen at a lower energy region than the O 2p top position. This band hybridizes with not only Cu 3d but also Ti 3d orbitals. Although the O K-edge XAS intensity is not proportional to the O 2p partial DOS, the observed strong XAS



Figure 6. O 2p DOS for CaCu₃Ti₄O₁₂ (thick line), CaCu₃Ge₄O₁₂ (thin line), and CaCu₃Sn₄O₁₂ (dashed line). Each energy position with the maximum partial DOS is shifted to the origin.

intensity at 530-532 eV in antiferromagnetic CaCu₃Ti₄O₁₂ should be the result of strong hybridization of Ti 3d orbitals with O 2p.

The present results of our XAS-MCD experiments and the electronic structure calculation demonstrate that the O 2p orbitals in ferromagnetic CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂ mainly hybridize with Cu 3d but that they do not play an important role in magnetic interaction. On the other hand, the empty Ti 3d orbitals strongly hybridize with O 2p and Cu 3d near E_F . Therefore, we can conclude that direct exchange interaction between the Cu²⁺ spins, as shown in Figure 1, primarily determines the ferromagnetic superexchange interaction through the Cu-O-Ti-O-Cu path, also as shown in Figure 1, overcomes ferromagnetic direct interaction, leading to anti-ferromagnetism in CaCu₃Ti₄O₁₂.

In summary, we investigated the electronic structures and magnetic interaction of A-site magnetic perovskites by XAS and MCD measurements. The O K-edge XAS spectra of ferromagnetic CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂ showed hybridization between Cu 3d and O 2p orbitals, but the MCD spectra showed that magnetic polarization of O 2p electrons was very low. The results implied that ferromagnetic direct exchange interaction primarily determined the magnetic ordering of CaCu₃Ge₄O₁₂ and CaCu₃Sn₄O₁₂. For antiferromagnetic CaCu₃-Ti₄O₁₂, on the other hand, the spectrum showed strong hybridization of Cu 3d, Ti 3d, and O 2p orbitals, suggesting the involvement of Ti 3d orbitals with the antiferromagnetic property. We thus concluded that the antiferromagnetism of CaCu₃Ti₄O₁₂ originated from superexchange interaction through the Cu-O-Ti-O-Cu path, which overcame ferromagnetic direct exchange interaction.

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