

## Coordination-Driven Self-Assembly of Three-Dimensional Supramolecular Dendrimers

Yao-Rong Zheng,<sup>†</sup> Koushik Ghosh,<sup>†</sup> Hai-Bo Yang,\*<sup>,‡</sup> and Peter J. Stang\*<sup>,†</sup>

† Department of Chemistry, University of Utah, 315 South 1400 East, Room 2020, Salt Lake City, Utah 84112, and ‡ Shanghai Key Laboratory of Green Chemistry and Chemical Processes, Department of Chemistry, East China Normal University, 3663 N. Zhongshan Road, Shanghai, China 200062

Received February 18, 2010

record and the energy on the Chemical Society Chemical Society Chemical Society Published on The European Chemical Society Published on The European Chemical Society Published on The European Chemical Society Published on The design and synthesis of novel three-dimensional (3-D) supramolecular dendrimers is described. The coordination-driven selfassembly of a 120° diplatinum acceptor and tritopic pyridyl donors bearing [G-0]-[G-3] Fréchet-type dendrons results in a series of supramolecular dendrimers under mild conditions, possessing a robust adamantanoid core of well-defined shape and size. The assemblies were identified using multinuclear (<sup>31</sup>P and <sup>1</sup>H) NMR spectroscopy and electrospray ionization mass spectrometry, as well as pulsed-field-gradient spin-echo (PGSE) NMR measurement together with computational simulations. Isotopically resolved mass spectral data support the existence of the  $[6 + 4]$  assembly of adamantanoid dendrimers, and the NMR results are consistent with the formation of these symmetrical assemblies. PGSE NMR measurements together with MMFF force-field modeling clearly reveal the structural feature of the 3-D supramolecular dendrimers with varying sizes.

Dendrimers of high structural complexity have emerged as one of the most important structures of significant interest in chemistry, biology, and medical research.<sup>1,2</sup> While merging with self-assembly, complicated dendritic structures can be

synthesized in a simple manner. One-dimensional (1-D) helical supramolecular dendrimers, for example, that vividly mimic natural pore-forming proteins can be efficiently selfassembled by a library of amphiphilic dendritic peptides.<sup>2a-c</sup> Recently, via coordination-driven self-assembly, a variety of two-dimensional (2-D), such as rhomboid, square, and hexagon, supramolecular dendrimers have been prepared.<sup>3,4</sup> Most reported systems are limited to utilizing helical and 2-D structures. As compared with 1-D and 2-D structures, threedimensional (3-D) assemblies exhibit a finite cavity of welldefined shape and size and can be valuable in applications such as guest encapsulation, gas storage, catalysis, and drug delivery.<sup>3</sup> However, 3-D supramolecular dendrimers are still rare, $^{2d}$  mainly because of the design constraint of rigid welldefined 3-D structures.

Coordination-driven self-assembly represents a successful methodology for constructing 3-D supramolecules, as indicated by a variety of elaborate supramolecular cages, prisms, and polydedra of well-defined shape and size. $6$  The selfassembled 3-D supramolecules have proven useful for capturing guest molecules, in supramolecular catalysis, and opening access to unusual reactions by virtue of 3-D nanocavities.7 Covalent modifications of molecular subunits

P. J. J. Org. Chem. 2009, 74, 7067.<br>(4) Baytekin, H. T.; Sahre, M.; Rang, A.; Engeser, M.; Schulz, A.; Schalley, C. A. Small 2008, 4, 1823.

(5) (a) Rosi, N. L.; Eckert, J.; Eddaoudi, M.; Vodak, D. T.; Kim, J.; O'Keeffe, M.; Yaghi, O. M. Science 2003, 300, 1127. (b) Trembleau, L.; Rebek, J., Jr. Science 2003, 301, 1219. (c) Yoshizawa, M.; Tamura, M.; Fujita, M. Science 2006, 312, 251. (d) Pluth, M. D.; Bergman, R. G.; Raymond, K. N. Science 2007, 316, 85.

(6) (a) Stang, P. J.; Olenyuk, B. Acc. Chem. Res. 1997, 30, 502. (b) Leininger, S.; Olenyuk, B.; Stang, P. J. Chem. Rev. 2000, 100, 853. (c) Seidel, S. R.; Stang, P. J. Acc. Chem. Res. 2002, 35, 972. (d) Fiedler, D.; Leung, D. H.; Bergman, R. G.; Raymond, K. N. Acc. Chem. Res. 2005, 38, 351. (e) Fujita, M.; Tominaga, M.; Hori, A.; Therrien, B. Acc. Chem. Res. 2005, 38, 369.<br>(7) (a) Pluth, M. D.; Bergman, R. G.; Raymond, K. N. Acc. Chem. Res.

2009, 42, 1650. (b) Yoshizawa, M.; Klosterman, J. K.; Fujita, M. Angew. Chem., Int. Ed. 2009, 48, 3418.

<sup>\*</sup>To whom correspondence should be addressed. E-mail: Stang@ chem.utah.edu.

<sup>(1) (</sup>a) Hawker, C. J.; Fréchet, J. M. J. J. Am. Chem. Soc. 1990, 112, 7638. (b) Fréchet, J. M. J.; Tomalia, D. A. Dendrimers and Other Dendritic Polymers; VCH-Wiley: New York, 2000. (c) Grayson, S. M.; Frechet, J. M. J. Chem. Rev. 2001, 101, 3819. (d) Hecht, S.; Fréchet, J. M. J. Angew. Chem., Int. Ed. 2001, 40, 74. (e) Hecht, S.; Vladimirov, N.; Frechet, J. M. J. J. Am. Chem. Soc. 2001, 123, 18. (f) Fréchet, J. M. J. Proc. Natl. Acad. Sci. U.S.A. 2002, 99, 4782. (g) Landskron, K.; Ozin, G. A. Science 2004, 306, 1529. (h) Lee, C. C.; MacKay, J. A.; Fréchet, J. M. J.; Szoka, F. C. Nat. Biotechnol. 2005, 23, 1517. (i) Medina, S. H.; El-Sayed, M. E. H. Chem. Rev. 2009, 109, 3141.

<sup>(2) (</sup>a) Percec, V.; Dulcey, A. E.; Balagurusamy, V. S. K.; Miura, Y.; Smidrkal, J.; Peterca, M.; Nummelin, S.; Edlund, U.; Hudson, S. D.; Heiney, P. A.; Duan, H.; Magonov, S. N.; Vinogradov, S. A. Nature 2004, 430, 764. (b) Percec, V.; Dulcey, A. E.; Peterca, M.; Ilies, M.; Sienkowska, M. J.; Heiney, P. A. J. Am. Chem. Soc. 2005, 127, 17902. (c) Percec, V.; Dulcey, A. E.; Peterca, M.; Adelman, P.; Samant, R.; Balagurusamy, V. S. K.; Heiney, P. A. J. Am. Chem. Soc. 2007, 129, 5992. (d) Percec, V.; Peterca, M.; Dulcey, A. E.; Imam, M. R.; Hudson, S. D.; Nummelin, S.; Adelman, P.; Heiney, P. A. J. Am. Chem. Soc. 2008, 130, 13079. (e) Rudick, J. G.; Percec, V. Acc. Chem. Res. 2008, 41, 1641. (f) Rosen, B. M.; Wilson, C. J.; Wilson, D. A.; Peterca, M.; Imam, M. R.; Percec, V. Chem. Rev. 2009, 109, 6275.

<sup>(3) (</sup>a) Yang, H.-B.; Das, N.; Huang, F.; Hawkridge, A. M.; Muddiman, D. C.; Stang, P. J. J. Am. Chem. Soc. 2006, 128, 10014. (b) Yang, H.-B.; Hawkridge, A. M.; Huang, S. D.; Das, N.; Bunge, S. D.; Muddiman, D. C.; Stang, P. J. J. Am. Chem. Soc. 2007, 129, 2120. (c) Yang, H.-B.; Northrop, B. H.; Zheng, Y.-R.; Ghosh, K.; Lyndon, M. M.; Muddiman, D. C.; Stang, P. J. J. Org. Chem. 2009, 74, 3524. (d) Yang, H.-B.; Northrop, B. H.; Zheng, Y.-R.; Ghosh, K.; Stang,

**Scheme 1.** Representation of the  $[6 + 4]$  Self-Assembly of Adamantanoid Supramolecular [G-0]-[G-3] Dendrimers 3a-3d



can endow assemblies with a wide variety of fascinating structural and functional properties.<sup>8</sup> Recent examples reported by Fujita et al. demonstrated that functionalization of the interior and/or exterior of a  $M_{12}L_{24}$  palladium-based supramolecular sphere can result in saccharide clusters, nanoscale fluorodroplets, and confined polymerization.<sup>9</sup> Because of the robust structure, unique confined space, and designable framework, the coordination-driven self-assembled 3-D metallosupramolecule is a promising candidate for constructing 3-D supramolecular dendrimers.

We have recently prepared 2-D hexagonal and 3-D cuboctahedral multifunctional scaffolds with a precise number, position, and orientation of functional groups by chemical tailoring of molecular subunits.10 Encouraged by the power and versatility of this methodology, we extend it to the construction of 3-D supramolecular dendrimers and hereby present the self-assembly of 120 diplatinum acceptors 1 with tritopic donors 2a-2d, substituted with Fréchet-type dendrons from  $[G-0]$  to  $[G-3]$ .<sup>10</sup> As shown in Scheme 1, coordination-driven self-assembly allows for the quantitative generation of a series of supramolecular dendrimers 3a-3d possessing a rigid 3-D adamantanoid core. These structures have been identified by multinuclear  $(^{31}P$  and  $^1H)$  NMR

(8) Northrop, B. H.; Yang, H.-B.; Stang, P. J. Chem. Commun. 2008, 5896.



**Figure 1.**  ${}^{31}P_1{}^{1}H$ } NMR (300 MHz, 1/1 acetone- $d_6$ /CD<sub>2</sub>Cl<sub>2</sub>) spectra of (a) 120° organoplatinum acceptor 1 and (b) [G-3] supramolecular of (a)  $120^\circ$  organoplatinum acceptor 1 and (b) [G-3] supramolecular dendrimer 3d.



Figure 2. Calculated (blue, top) and experimental (red, bottom) ESI mass spectra of [G-2] supramolecular dendrimer 3c.

spectroscopy and electrospray ionization (ESI) mass spectrometry, as well as pulsed-field-gradient spin-echo (PGSE) NMR measurement together with computational simulations.

 $[G-0]$ - $[G-3]$  tripyridyl donors  $2a-2d$  were prepared by substituting tripyridin-4-ylmethanol with Fréchet-type dendrons. The  $CD_2Cl_2$  solution of tripyridyl donors 2 was slowly added into an acetone- $d_6$  solution of 120 $\degree$  organoplatinum acceptor 1 in a 2:3 ratio, and after stirring for 15 min at room temperature,  $[6 + 4]$  self-assembly of adamantanoid dendrimers 3 of different sizes was quantitatively obtained. In the mers 3 of different sizes was quantitatively obtained. In the 3<sup>1</sup>P{<sup>1</sup>H} NMR spectra (Figure 1a,b and the Supporting Information) of these mixtures, only a sharp singlet (3a, 14.06 ppm; 3b, 14.06 ppm; 3c, 14.05 ppm; 3d, 14.01 ppm) with concomitant 195Pt satellites can be found. Because of the formation of the Pt-N coordination bond, the signals are shifted upfield from that of the organoplatinum acceptor 1 by approximately 9.0 ppm. Likewise, the <sup>1</sup>H NMR spectra (see the Supporting Information) exhibit sharp signals for the pyridyl protons with small shifts downfield ( $\Delta \delta_{Pv\alpha-H}$ , 0.30 ppm;  $\Delta \delta_{Py\beta-H}$ , 0.75 ppm) due to the loss of electron density that occurs upon coordination. The well-assigned sharp NMR signals in both the  ${}^{31}P$  and  ${}^{1}H$  NMR spectra agree with the high symmetry of 3 and rule out the formation of oligomers.

ESI mass spectrometry further confirmed the  $[6 + 4]$ structures (Supporting Information). In the ESI mass spectrum of the [G-0] adamantanoid dendrimer 3a, peaks corresponding to  $[M - 40Tf]^{4+}$  and  $[M - 50Tf]^{5+}$  can be found at

<sup>(9) (</sup>a) Sato, S.; Iida, J.; Suzuki, K.; Kawano, M.; Ozeki, T.; Fujita, M. Science 2006, 313, 1273. (b) Kamiya, N.; Tominaga, M.; Sato, S.; Fujita, M. J. Am. Chem. Soc. 2007, 129, 3816. (c) Murase, T.; Sato, S.; Fujita, M. Angew. Chem., Int. Ed. 2007, 46, 5133. (d) Suzuki, K.; Kawano, M.; Sato, S.; Fujita, M. J. Am. Chem. Soc. 2007, 129, 10652. (e) Suzuki, K.; Sato, S.; Fujita, M. Nat. Chem. 2010, 2, 25.

<sup>(10) (</sup>a) Yang, H.-B.; Ghosh, K.; Northrop, B. H.; Zheng, Y.-R.; Lyndon, M. M.; Muddiman, D. C.; Stang, P. J. J. Am. Chem. Soc. 2007, 129, 14187. (b) Yang, H.-B.; Ghosh, K.; Zhao, Y.; Northrop, B. H.; Lyndon, M. M.; Muddiman, D. C.; White, H. S.; Stang, P. J. J. Am. Chem. Soc. 2008, 130, 839. (c) Ghosh, K.; Hu, J.; White, H. S.; Stang, P. J. J. Am. Chem. Soc. 2009, 131, 6695.



Figure 3. Computational simulations (MMFF) of 3-D dendrimers 3a-3d (protons and triethylphosphrous moieties of the adamantanoid core omitted for clarity) and their calculated (blue) sizes as well as experimental (red) values from PGSE NMR measurements.

 $m/z$  1742.4 and 1427.3. The ESI mass peak for loss of triflate anions from [G-1] adamantanoid dendrimer 3b at  $m/z$  1912.2  $([M - 5<sup>5+</sup>)$  is observed, likewise for the [G-2] adamantanoid dendrimer 3c at  $m/z$  2251.6 ([M - 5OTf]<sup>5+</sup>; Figure 2). All ESI mass peaks are isotopically resolved and agree with their theoretical distributions. However, for the [G-3] adamantanoid dendrimer 3d, because of its high molecular weight, an isotopically resolved ESI mass signal was not obtained.

Because suitable X-ray-quality crystals could not be obtained, a computational study was carried out to gain insight into the structural characteristics of these assemblies. A 1.0 ns molecular dynamics simulation [molecular mechanics force field (MMFF), 300 K, gas phase] was used to equilibrate each supramolecule, and the output of the simulation was then minimized to full convergence. As shown in Figure 3, each assembly has a 3.0 nm adamantanoid core with  $T_d$  symmetry, and the size of the overall structure is increased from 3.0 nm (for  $[G-0]$  3a) to 4.6 nm (for  $[G-3]$  3d). In 3d, the adamantanoid core is partially covered by a dendritic shell.

PGSE NMR measurements were used to characterize these structures by determination of the translational selfdiffusion coefficient, and in conjunction with the Stokes-Einstein equation, the "effective size" of the overall assembly in solution was obtained.<sup>11</sup> The <sup>1</sup>H NMR PGSE measurements for all assemblies in acetone- $d_6$  (see the Supporting Information) were determined: **3a**,  $3.0 \pm 0.1$  nm; **3b**,  $4.0 \pm$ 0.12 nm;  $3c$ ,  $4.5 \pm 0.1$  nm;  $3d$ ,  $4.5 \pm 0.15$  nm. As indicated in

Figure 3, they are in good agreement with the computational values (3a, 3.0 nm; 3b, 3.8 nm; 3c, 4.3 nm; 3d, 4.6 nm) from the MMFF modeling. Because the dendrons on the exterior are spreading around the adamantanoid core instead of spreading out, the size increase from 3c to 3d is minor, resulting in a similar size in the PGSE NMR measurement.

In conclusion, we have developed a convergent synthesis of 3-D metallosupramolecular dendrimers via coordinationdriven self-assembly, where the exterior dendrons were introduced by premodification of the molecular subunits. These supramolecular dendrimers show a unique 3-D adamantanoid core, which is partially covered by the dendritic exterior, and exploration of the interior environment would be interesting and valuable. Furthermore, functionalization of the dendrons and/or the supramolecular core should allow access to 3-D hollow supramolecular functional systems with potential applications such as guest encapsulation, nanoreactor, and drug delivery.<sup>1</sup>

Acknowledgment. P.J.S. thanks the NIH (Grant GM-057052) for financial support. H.-B.Y. thanks the NSFC (Grant 20902027), Shanghai Pujiang Program (09PJ-1404100), Shanghai Shuguang Program (09SG25), and Innovation Program of SMEC (10ZZ32) for financial support.

Supporting Information Available: General detailed experimental procedures for the synthesis and characterization of dendritic tripyridyl donors 2a-2d and supramolecular dendrimers 3a-3d (http://pubs.acs.org/paragonplus/submission/ jacsat/). This material is available free of charge via the Internet at http://pubs.acs.org.

<sup>(11)</sup> Caskey, D. C.; Yamamoto, T.; Addicott, C.; Shoemaker, R. K.; Vacek, J.; Hawkridge, A. M.; Muddiman, D. C.; Kottas, G. S.; Michl, J.; Stang, P. J. J. Am. Chem. Soc. 2008, 130, 7620 and references cited therein.