

# Raising the Spin of Fe<sup>III</sup><sub>7</sub> Disklike Clusters: The Power of Molecular Spin Frustration

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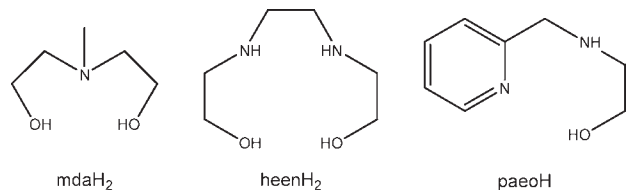
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**S** Supporting Information

**ABSTRACT:** Two clusters with a new type of Fe<sup>III</sup><sub>7</sub> disklike structure have been prepared; in contrast to other Fe<sup>III</sup><sub>7</sub> disks, they possess high ground-state spins ( $S = 15/2$  and  $21/2$ ), which have been rationalized by analysis of the spin-frustration patterns.

Molecules possessing large numbers of unpaired electrons (i.e., large ground-state spin,  $S$ ) represent a fascinating subarea of metal cluster chemistry of importance to many fields.<sup>1,2</sup> For example, when coupled with significant easy-axis magnetoanisotropy, such molecules function as single-molecule magnets (SMMs), providing a molecular approach to nanoscale magnetism.<sup>3,4</sup> In contrast, when they possess little or no anisotropy, they are of interest as components for molecule-based magnetic refrigeration based on the magnetocaloric effect.<sup>5</sup> Also, of course, there is the fundamental desire to understand how the signs and relative magnitudes of the many exchange interactions within a cluster yield its high  $S$  value.

We recently studied the origin of the  $S = 11$  ground state of Mn<sub>7</sub> (4Mn<sup>II</sup>, 3Mn<sup>III</sup>) complexes with a disklike structure<sup>6,7</sup> and showed it to be due to spin-frustration effects from competing exchange interactions of both ferromagnetic (F) and antiferromagnetic (AF) nature. The relative magnitude of the various exchange parameters ( $J$ ) suggested that complexes with the maximum  $S = 16$  might be attainable, and they were successfully prepared from ligand-induced structural perturbations that altered the relative magnitude of the competing interactions. We then wondered whether spin modification might also be possible in the Fe<sup>III</sup><sub>7</sub> disklike complexes that we and others had studied,<sup>8–10</sup> such as [Fe<sub>7</sub>O<sub>3</sub>(O<sub>2</sub>CR)<sub>9</sub>(mda)<sub>3</sub>(H<sub>2</sub>O)<sub>3</sub>] (**1**; mdaH<sub>2</sub> = *N*-methyl-diethanolamine), which possesses a buckled Fe<sub>6</sub> loop around a central Fe atom and an  $S = 5/2$  ground state.



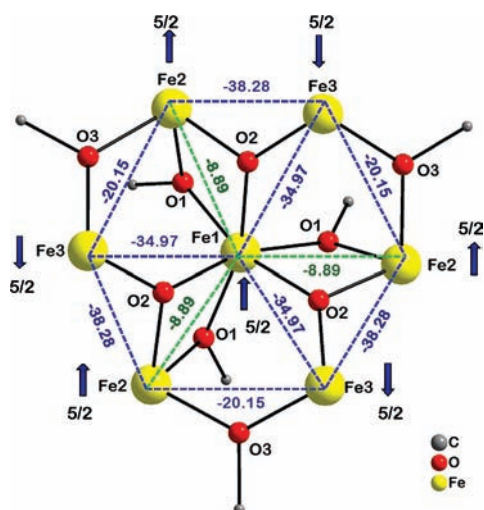
As in our Mn<sub>7</sub> work, the first priority was to identify the origin of the  $S = 5/2$  ground state of **1** because we expected all interactions now to be AF. We did this by determining the  $J$  values using a published magnetostructural correlation originally developed for dimers that employs both the Fe–O distances and

Fe–O–Fe angles.<sup>11</sup> The  $J$  values (Figure 1) are indeed all AF but are of two types: relatively strong ( $-20$  to  $-39$  cm<sup>-1</sup>) and weak ( $-8.89$  cm<sup>-1</sup>). The ground state can thus be rationalized (Figure 1) as comprising an antiparallel alignment of spins controlled by strong interactions and a parallel alignment of spins controlled by weak ones (Fe1Fe2 and its symmetry partners); i.e., these AF interactions are completely frustrated. This does not offer hope for experimentally changing the  $S = 5/2$  ground state of **1** via small ligand-induced perturbations, in contrast to Mn<sub>7</sub>.<sup>6,7</sup> With strong AF interactions both between Fe atoms of the outer ring ( $J_{oo}$ ) and between them and the inner Fe ( $J_{io}$ ), where o = outer and i = inner, it would clearly take a major modification to affect the ground state, and it is not obvious how to target this. However, this was achieved through happenstance when we recently made a new type of Fe<sub>7</sub> disk while exploring Fe<sup>III</sup> chemistry with *N,N'*-bis(2-hydroxyethyl)ethylenediamine (heenH<sub>2</sub>) and 2-(2-pyridylmethyl)aminoethanol (paeoH). This gives much higher ground-state spins.

We have previously used heenH<sub>2</sub> in Fe chemistry but not paeoH.<sup>12</sup> In the present work, the reaction of FeCl<sub>2</sub> and heenH<sub>2</sub> (1:1) in refluxing MeOH gave upon cooling [Fe<sub>7</sub>O<sub>3</sub>(OMe)<sub>3</sub>(heen)<sub>3</sub>Cl<sub>4.5</sub>(MeOH)(H<sub>2</sub>O)<sub>1.5</sub>]Cl<sub>1.25</sub>[FeCl<sub>4</sub>]<sub>1/4</sub> (**2**), isolated as orange needles of 2·2MeOH·<sup>1</sup>/<sub>2</sub>H<sub>2</sub>O in 10% nonoptimized yield after 7 days. Similarly, the reaction of Fe(ClO<sub>4</sub>)<sub>3</sub>, paeoH, and NEt<sub>3</sub> (1:3:1) in MeOH gave [Fe<sub>7</sub>O<sub>3</sub>(OH)<sub>3</sub>Cl(paeo)<sub>6</sub>](Cl)(ClO<sub>4</sub>)<sub>4</sub> (**3**) as orange crystals of 3·2Me<sub>2</sub>CO·<sup>1</sup>/<sub>2</sub>Et<sub>2</sub>O in 14% nonoptimized yield. The cations of **2** and **3** have almost identical Fe<sub>7</sub> cores (Figure 2)<sup>13</sup> consisting of a near-planar Fe<sup>III</sup><sub>6</sub> hexagon linked to a central Fe<sup>III</sup> ion by three μ<sub>3</sub>-O<sup>2-</sup> ions and lying 1.437 Å (**2**) or 1.484 Å (**3**) above the Fe<sub>6</sub> plane. In **2**, each heen<sup>2-</sup> is η<sup>2</sup>:η<sup>1</sup>:η<sup>1</sup>:η<sup>2</sup>:μ<sub>3</sub>, chelating to an outer Fe and bridging to neighboring Fe atoms on either side. In **3**, each of the now six paeo<sup>-</sup> groups is η<sup>2</sup>:η<sup>1</sup>:η<sup>1</sup>:μ<sub>2</sub>, chelating to one Fe and bridging to only one neighbor. Additional bridges between outer Fe atoms are by three μ<sub>2</sub>-OMe<sup>-</sup> (**2**) or μ<sub>2</sub>-OH<sup>-</sup> (**3**) groups, and terminal ligation at three outer Fe atoms in **2** is by a Cl<sup>-</sup> and either H<sub>2</sub>O or MeOH. Ligation at the central Fe is completed by a terminal Cl<sup>-</sup> ion. The main difference between **2/3** and prior Fe<sub>7</sub> disks is the tetrahedral geometry of the central Fe, which also rationalizes the near-planar Fe<sub>6</sub> hexagon. In both **2** and **3**, the cations are surrounded by two types of anions and by solvate molecules; the [FeCl<sub>4</sub>]<sup>-</sup> anion in **2** forms no interactions with the cation, directly or via solvent molecules, and thus is at best only very weakly exchange-coupled to the cation.

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**Figure 1.** Core of **1** ( $R = \text{Bu}^t$ ) showing the calculated  $J$  values ( $\text{cm}^{-1}$ )<sup>11</sup> and resulting spin alignments giving the  $S = 5/2$  ground state. The frustrated interactions are the green dashed lines.

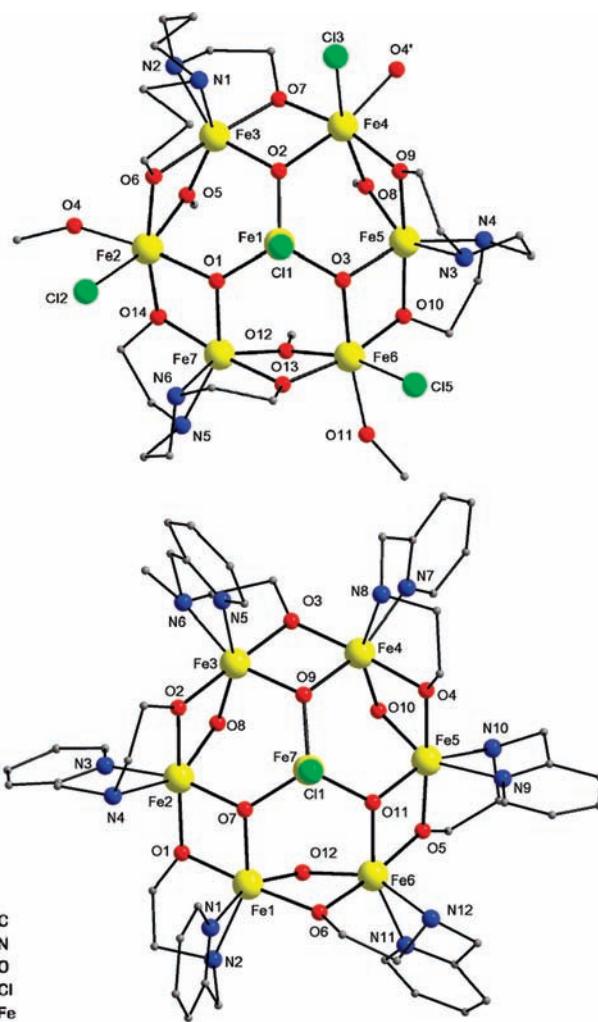
Solid-state direct-current magnetic susceptibility ( $\chi_M$ ) data on dried **2** and **3** were collected on microcrystallites restrained in eicosane in a 0.1 T field in the 5.0–300 K range (Figure 3). For **2**,  $\chi_M T$  (with the contribution of  $[\text{FeCl}_4]^-$  subtracted) increases from 29.92  $\text{cm}^3 \text{K mol}^{-1}$  at 300 K to a maximum of 48.16  $\text{cm}^3 \text{K mol}^{-1}$  at 50 K and then decreases to 32.83  $\text{cm}^3 \text{K mol}^{-1}$  at 5.0 K. For **3**,  $\chi_M T$  increases from 28.42  $\text{cm}^3 \text{K mol}^{-1}$  at 300 K to a maximum of 56.71  $\text{cm}^3 \text{K mol}^{-1}$  at 50 K and then decreases slightly to 53.10  $\text{cm}^3 \text{K mol}^{-1}$  at 5.0 K. Both plots indicate ground-state  $S$  values significantly greater than  $S = 5/2$  for **1**: the 5.0 K value for **2** suggests  $S = 15/2$  (the spin-only value is 31.88  $\text{cm}^3 \text{K mol}^{-1}$ ), with an increase up to 50 K suggesting population of the excited states with  $S > 15/2$ . The low  $T$  data for **3** suggests an  $S = 19/2$  or  $21/2$  ground state (spin-only values of 49.88 and 60.38  $\text{cm}^3 \text{K mol}^{-1}$ , respectively).

Confirmation of the ground states was obtained from fits of magnetization ( $M$ ) data collected in the 1.8–10 K range in fields ( $H$ ) up to 7 T. The data were fit, using the program *MAGNET*,<sup>14</sup> by matrix diagonalization assuming only the ground state is populated, incorporating axial anisotropy ( $D\hat{S}_z^2$ ) and Zeeman terms, and employing a full powder average. The spin Hamiltonian is given by eq 1, where  $\hat{S}_z$  is the  $z$ -axis spin projection,  $g$  is the Landé  $g$  factor,  $\mu_B$  is the Bohr magneton, and  $\mu_0$  is the vacuum permeability.

$$\mathcal{H} = D\hat{S}_z^2 + g\mu_B\mu_0\hat{S}\cdot H \quad (1)$$

For **2**, we could not get an acceptable fit using data collected up to 7 T, a problem often caused by low-lying excited states, especially if some have an  $S$  value larger than the ground state. Using data collected at  $\leq 2$  T gave an acceptable fit with  $S = 15/2$ ,  $D = -0.13 \text{ cm}^{-1}$ , and  $g = 2.12$ . For **3**, a satisfactory fit with all data was obtained with  $S = 21/2$ ,  $D = -0.08 \text{ cm}^{-1}$ , and  $g = 1.97$  (Figures S1–S4 in the Supporting Information, SI). These  $S$  values were supported by alternating-current data that gave extrapolated values at 0 K of just under 30 and 60  $\text{cm}^3 \text{K mol}^{-1}$ , respectively (Figures S5 and S6 in the SI).

Complexes **2** and **3** thus possess much higher  $S$  values than prior  $\text{Fe}^{\text{III}}_7$  disks such as **1**. A high  $S$  has been seen previously for  $\text{Fe}_7$  disks only when they are  $\text{Fe}^{\text{II}}$  ( $S = 10$ )<sup>15</sup> or mixed-valent  $\text{Fe}^{\text{II/III}}$  ( $S = 29/2$ ),<sup>16</sup> leading to some couplings being F. To see a high  $S$  in  $\text{Fe}^{\text{III}}_7$  disks where all couplings are AF was surprising, especially because all magnetically characterized  $\text{Fe}^{\text{III}}_7$  clusters



**Figure 2.** Structures of the cations of **2** (top) and **3** (bottom). H atoms have been removed for clarity.

of any structure in the literature have  $S = 5/2$  or less.<sup>8,9,17,18</sup> We thus sought its rationalization from the  $J$  values of **2** and **3**, obtained as for **1** (Figure 4). The main differences between **2/3** and **1** (Figure 1) are that all  $J_{i0}$  values are now strong in the former, and all of the  $J_{00}$  values are much weaker. Thus, the  $J_{00}$  values are now frustrated, and the outer Fe spins are aligning more or less parallel to each other and antiparallel to the central Fe spin. The limiting situation is  $S = 25/2$  when the outer spins are perfectly parallel, and antiparallel to the central spin; **3** almost reaches this value, but the outer  $J_{00}$  value must not be completely frustrated, and there must be some intermediate-spin alignments from the competition between  $J_{i0}$  and  $J_{00}$ . A tentative rationalization of the  $S$  values is provided in Figure S7 in the SI. **2** has a slightly stronger average  $J_{i0}$  value than **3** but also a slightly stronger  $J_{00}$  value, so we refrain from attempting a more detailed rationalization of their  $S$  values until more accurate  $J$  values from density functional theory calculations are available. The main conclusion for now is that these new  $\text{Fe}_7$  disks have distinct structural differences from previous ones, primarily arising from the tetrahedral central Fe, leading to a major switch in the relative magnitude of the  $J_{i0}$  vs  $J_{00}$  plot and resulting in high ground-state  $S$  values from the resulting spin-frustration pattern. High spins in several other  $\text{Fe}^{\text{III}}_x$  clusters are, of course, known, but it is

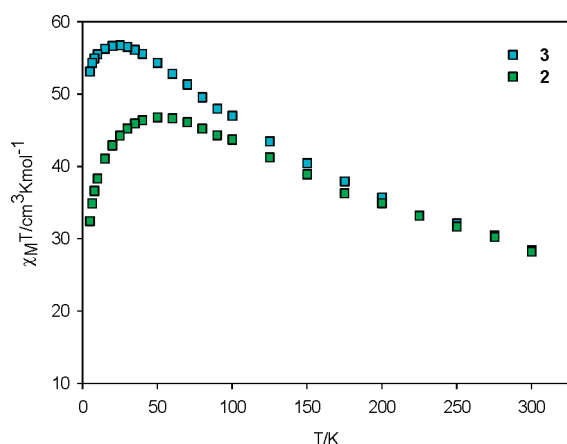


Figure 3.  $\chi_{MT}$  vs  $T$  plots for **2** (with the  $[\text{FeCl}_4]^-$  contribution subtracted) and **3** in a 0.1 T field.

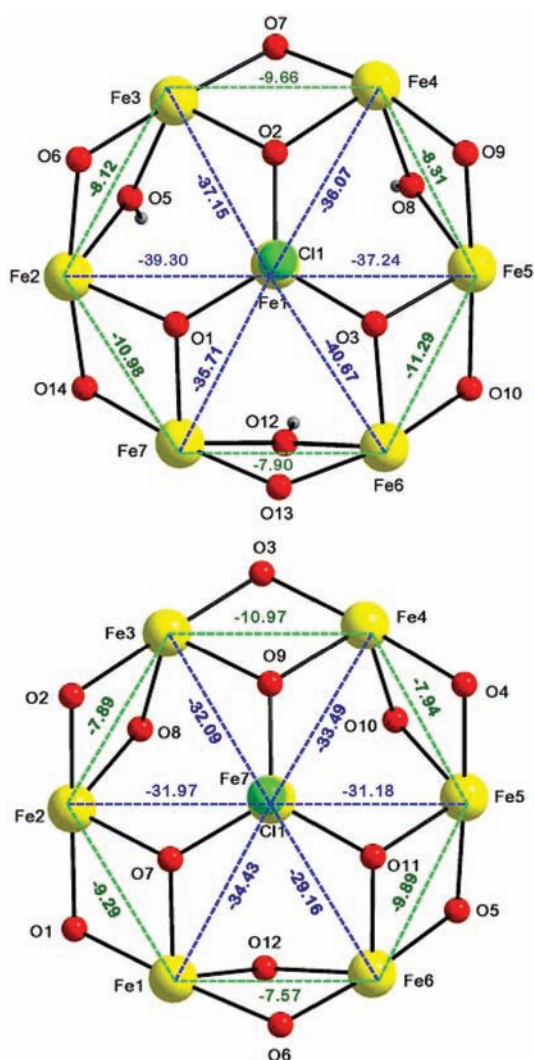


Figure 4. Cores of **2** (top) and **3** (bottom) showing the calculated  $J$  values ( $\text{cm}^{-1}$ ) for each  $\text{Fe}_2$  pair.

interesting that the present  $\text{Fe}_7$  disks also appear to demonstrate the ability to give variable ground states,<sup>19</sup> which has not been seen for these other examples.<sup>20</sup>

Finally, preliminary ZILSH calculations on **1** have given  $J$  values comparable with those in Figure 1, supporting the validity of determining the  $J$  values of **1** via the magnetostructural correlation,<sup>11</sup> and by extension of those of **2** and **3**.

## ASSOCIATED CONTENT

**S** Supporting Information. X-ray crystallographic data (CIF) and magnetism figures. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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