# Highly Selective Mitochondria-Targeting Amphiphilic Silicon(IV) Phthalocyanines with Axially Ligated Rhodamine B for Photodynamic Therapy

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## **S** Supporting Information

ABSTRACT: Two axially ligated rhodamine−Si(IV)−phthalocyanine (Rh-SiPc) conjugates, bearing one and two rhodamine B, were synthesized and their linear and two-photon photophysical, subcellular localization and photocytotoxic properties were studied. These Rh-SiPc conjugates exhibited an almost exclusive mitochondrial localizing property in human nasopharyngeal carcinoma (HK-1) cells and human cervical carcinoma (HeLa) cells. Strong photocytotoxic but low dark cytotoxic properties were also observed for the two Rh-SiPc conjugates toward the HK-1 cells. Using nuclei staining method and flow cytometric DNA content analysis, apoptotic cell death was induced by these conjugates upon photoactivation. This observation is consistent with their mitochondrial localization property. The observed properties of these conjugates qualify them as promising PDT agents.



Rh-SiPc-Rh

# **NO INTRODUCTION**

Photodynamic therapy (PDT) has attracted increasing attention as a highly selective anticancer treatment modality where a nontoxic photosensitizer showing preferential accumulation in tumor tissue or its vasculature when administered inside the body is activated by the tumor-targeted photoirradiation at a specific wavelength to produce cytotoxic reactive oxygen species (ROS), principally singlet oxygen  $(^1O_2)$ , which then kills the tumor cells.<sup>1,2</sup> An ideal anticancer PDT agent should possess the following properties: (1) minimal dark cytotoxicity, (2) relatively high  $^{1}O_{2}$  $^{1}O_{2}$  $^{1}O_{2}$  $^{1}O_{2}$  $^{1}O_{2}$  quantum yield, (3) preferential uptake by target tissues, (4) strong absorption at longer wavelengths, that is,  $\lambda > 650$  nm, where absorption by biomolecules, such as hemoglobin, is minimal, (5) high chemical stability, and (6) rapid clearance from normal tissues.<sup>3</sup> As singlet oxygen is believed to be the principal cytotoxic agent, its short reactive range  $(0.01-0.02 \mu m)$  in biological sys[te](#page-9-0)ms<sup>4</sup> is an important factor to consider in designing an efficacious PDT agent. To overcome the short reactive ranges of  ${}^{1}O_{2}$  and other ROS, photosensitizer needs to closely localize to the targets that can effectively trigger cell death. Among the intracellular targets that have been shown to cause cell death, mitochondrion is considered quite appealing as it can cause cell death by apoptosis, which is a programmed shutdown of the cell machinery with all its components recycled without eliciting any undesirable

inflammatory responses often seen in conventional chemotherapy.<sup>5</sup> In this work, we report the synthesis of a mitochondria-targeting PDT agent by covalently connecting a well-known mitochondri[a](#page-9-0)targeting agent, namely, rhodamine B (Rh B), to a tumorlocalizing photosensitizer, namely, phthalocyanine (Pc), and evaluated its subcellular localization, one- and two-photon PDT properties in several cancer and noncancer cell models.

This design is based on the fact that positively charged rhodamine, such as Rh B and rhodamine  $123<sup>6</sup>$  accumulates specifically in the mitochondria of living cells and has been used as probes to measure and monitor mitoc[h](#page-9-0)ondrial membrane potential in bioassays.<sup>7</sup> On the other hand,  $Pc$  is a class of photosensitizers that absorbs strongly in the tissue-penetrating spectral window of 6[5](#page-9-0)0−800 nm and has been extensively studied as a potential PDT agent due to its tumor-localizing properties.<sup>3,8</sup> Since most phthalocyanines are insoluble and tend to aggregate in aqueous media, they have to be encapsulated in [lip](#page-9-0)osomes,<sup>9</sup> polymeric micelles<sup>10</sup> or nanoparticles<sup>11</sup> before their administration into test animals or cells. The aqueous solubility of Pc can be improve[d](#page-9-0) by introduction [of](#page-9-0) hydrophilic substituents, such as sulfonate, $12$  carboxylate, $13$ N-methylpyridinium,<sup>14</sup> and polyethylene glycol (PEG),<sup>15</sup> in the

Received: May 31, 2[01](#page-9-0)1 Published: December 22, 2011 macrocyclic ring. Axial ligation of hydrophilic or amphiphilic groups in metallophthalocyanines, including silicon(IV) phthalocyanine (SiPc), which is currently under clinical trial, $\delta$ is another approach to increase water solubility and reduce ag-gregation in aqueous media.<sup>16</sup> Recently, the synthesis and PD[T](#page-9-0) activity of the hydrophobic monohydroxy trisphenylporphyrin (TPP-OH) tethered to Rh [B v](#page-9-0)ia a saturated hydrocarbon linker has been reported.<sup>17</sup> Enhanced (>5-fold) photocytotoxicity, relative to its components (i.e., TPP-OH and Rh B), and preferential accumulati[on](#page-9-0) at mitochondria were observed for this conjugate. Similar conjugates of sulfono- and carboxy-substituted phthalocyanines with rhodamines have also been synthesized<sup>18</sup> and their efficient intramolecular energy transfer (from Rh to Pc) studied as well.<sup>19</sup> In our design, Rh B, instead of connecting [to](#page-9-0) the macrocyclic ring as peripheral substituents as in previous works,<sup>17-19</sup> is [axi](#page-9-0)ally connected to SiPc via a polyether linkage to form two SiPc-Rh B conjugates, 3 and 2, possessing one and two R[h](#page-9-0) [B m](#page-9-0)oieties, respectively. This axial ligation design can reduce Pc aggregation, thereby increasing its excited state lifetimes<sup>20</sup> and improving its cellular uptake and imaging properties. The synthesis, linear and two-photon photophysical, subcellu[lar](#page-9-0) localization, and in vitro PDT activity of these conjugates are reported herein.

# **EXPERIMENTAL SECTION**

General Remarks. All reactions were carried out in a dry nitrogen atmosphere. Solvents were dried by standard procedures, distilled, and deaerated prior to use. All chemicals were obtained from Aldrich Chemical Co. and, where appropriate, degassed before use. NMR spectra were recorded with a Varian INOVA 400 NMR spectrometer. High-resolution matrix assisted laser desorption/ionization time-offlight (MALDI-TOF) mass spectra were recorded with a Bruker Autoflex MALDI-TOF mass spectrometer. Electronic absorption spectra in the UV/vis region were recorded with a Varian Cary 100 UV/vis spectrophotometer. Steady-state visible fluorescence and photoluminescence excitation spectra were recorded with a Photon Technology International (PTI) Alphascan spectrofluorimeter and singlet-oxygen near-IR (NIR) emission spectra were recorded directly by its phosphorescence emission at 1270 nm with an InGaAs detector, using  $\hat{H}_2$ TPP ( $\Phi$ <sub>Δ</sub> = 0.55  $\pm$  0.11)<sup>21</sup> as the reference compound. The filter was LG-697-F from Corion Company. Infrared spectra (KBr pellets) were recorded with a Nic[ole](#page-9-0)t Magna 550 FTIR spectrometer. All measurements were performed at ambient temperature  $(20 \pm 2 \degree C)$ and pressure. The UV/vis absorption and NIR emission spectra of all solution samples were measured in a 10 mm quartz cell.

Cell Culture. Human nasopharyngeal carcinoma (HK-1) cells were cultured in RPMI-1640 medium (Gibco) supplemented with 10% fetal bovine serum (Gibco) and antibiotics (penicillin 50 U/mL; streptomycin 50 μg/mL). Human cervical carcinoma (HeLa) cells were maintained in an RMPI 1640 medium supplemented with 10% fetal bovine serum (FBS) and 1% penicillin and streptomycin. Cells were incubated at 37 °C in a humidified incubator with 5%  $CO_2$ . Culture medium in each dish was changed prior to exposure to the test compounds. Stock solutions of the test compounds (1 mM) were prepared in DMSO and stored in dark at room temperature. These compounds, when used in the imaging and bioassay experiments involving cultured cells, were diluted with the corresponding culture media to appropriate concentrations.

Confocal Microscopic Imaging of Rh-SiPc Conjugates. HK-1 cells  $(1 \times 10^5)$  were seeded onto coverslip in 35-mm culture dishes for overnight. The cells were initially incubated with Rh-SiPc-Rh  $(2, 50 \text{ nM})$ , Rh-SiPc  $(3, 50 \text{ nM})$ , Rh B  $(1 \mu\text{M})$  or C-SiPc-C<sub>7</sub>  $(4, 1 \mu\text{M})$ for 30 min in dark. The cells were then washed and stained with 100 nM mitochondria-specific probe, Mito Tracker Green FM dye M7514, lysosomes-specific probe Lyso Tracker Green DND-26 L7526, Golgispecific probe BODIPY FL C5-ceramide complexed to BSA B-22656 and the endoplasmic reticulum probe ER Tracker Blue−White DPX dye E12353 (Invitrogen) for 30 min. The emitted fluorescent signals of 2, 3, Rh B, 4 and the organelle-specific probes were examined using the (Olympus) FV1000 confocal microscope equipped with a spectral detection system. A diode laser line at 405 nm was used for excitation of the ER Tracker. An argon-ion laser line at 488 nm was used for the excitation of Mito Tracker, Lyso Tracker and Golgi Tracker. A helium−neon laser line at 543 nm was used for the excitation of 2, 3, Rh B and 4. Emission signals at 425−475 nm (ER Tracker), 500− 530 nm (Mito, Lyso and Golgi Trackers), 620−720 nm (2 and 3), 560−660 nm (Rh B) and 630−710 nm (4) were collected. A 60x oil immersion objective and pinhole size of 110  $\mu$ m was used for image capturing. Images were processed and analyzed using the FV10-ASW software (Olympus).

**Photocytotoxicity Assay.** HK-1 cells  $(2 \times 10^4/\text{well})$  were incubated in wells of 96-well plate for overnight. The cells were treated with 2 (100 nM), 3 (100 nM), Rh B (100 nM), or 4 (100 nM) for 6 h in dark. The culture medium was then replaced with fresh medium and the cells were exposed to yellow light  $(1-4$  J/cm<sup>2</sup>) produced from a 400 W tungsten lamp fitted with a heat-isolation filter and a 500 nm long-pass filter. The fluence rate was 4 mW/cm<sup>2</sup>. Cells viability was determined by the MTT reduction assay at 24 h post-PDT.<sup>22</sup> The cell monolayers were rinsed twice with phosphatebuffered saline (PBS) and then incubated with 250  $\mu$ g/mL MTT soluti[on](#page-9-0) at 37 °C for 3 h. The formazan crystal formed was dissolved in DMSO and the absorbance of dissolved formazan crystal at 540 and 690 nm was measured using a 96-well plate reader (ELx800 Absorbance Microplate Reader).

[(absorbance of cell control - absorbance of blank)

− (absorbance of treatment − absorbance of blank)]

− /(absorbance of cell control absorbance of blank)

 $\times$  100%

**DNA Content Analysis.** HK-1 cells  $(3 \times 10^5/\text{well})$  were seeded onto 35 mm Petri dish for overnight. The cells were treated with 2 (100 nM) for 6 h in dark. The cells were then irradiated  $(1-4$  J/cm<sup>2</sup>) as described above. At 24 h post-PDT, both floating and adherent cells were collected, washed with PBS twice and then fixed in 70% ethanol for at least 1 h at 4 °C. After fixation, the cells were stained with DNAbinding buffer (40  $\mu$ g/mL propidium iodide, 1 mg/mL RNase, 0.1% Triton X-100 in PBS) for 30 min. The fluorescence profiles of the stained cells were analyzed using the FACSCalibur Flow Cytometer (Becton−Dickinson).<sup>23</sup> Laser with a wavelength of 488 nm was used for excitation, and the fluorescence signal was detected using the FL-2 channel ( $\lambda_{\text{em}}$  = 564–[60](#page-9-0)6 nm). At least 10 000 events were counted. DNA content was analyzed using the Cell Quest and the Modfit LT Version 3.0 Software.

Staining of Apoptotic Nuclei. Hoechst 33258 was used in the staining of apoptotic nuclei.<sup>24</sup> Overnight cultured HK-1 cells were treated with 2 (100 nM) for 6 h in dark and then irradiated  $(1-4$  J/cm<sup>2</sup>) as described previously. [At](#page-9-0) 3 and 16 h after PDT, detached and adherent cells were collected and fixed with 4% paraformaldehyde in PBS for 15 min. Cell membrane was permeabilized with absolute methanol for 20 min, followed by staining with Hoechst 33258  $(0.12 \mu g/mL)$  in PBS for 15 min. Fluorescence images of apoptotic nuclei were visualized and captured under a fluorescence microscope (Zeiss, Axioskop2).

Two-Photon-Induced Confocal Microscopic Imaging of Rh-SiPc Conjugates. HeLa cells  $(1 \times 10^5)$  were seeded onto coverslip in 35-mm culture dishes for overnight. The cells were initially incubated with Rh-SiPc-Rh  $(2, 5 \mu M)$ , Rh-SiPc  $(3, 5 \mu M)$ , or  $C_7$ -SiPc- $C_7$  (4, 5  $\mu$ M) for 6 h in dark. The two-photon-induced fluorescent signals of 2, 3, and 4 were captured using the Leica SP5 (upright configuration) confocal microscope equipped with a femtosecondpulsed Ti:Sapphire laser (Libra II, Coherent) inside the tissue culture chamber (5%  $CO<sub>2</sub>$ , 37 °C). The excitation beam produced by the femtosecond laser, which was tunable from 680 to 1050 nm, was focused on the adherent cells through a 40× oil immersion objective.

The time-lapse images (0−30 min, one laser shot per min) were obtained with femtosecond laser excitation at 850 nm (laser power ∼8 mW).

Two-Photon Absorption Measurement. Two-photon absorption spectra of 2, 3, and 4 were measured at 850 nm by the openaperture  $Z$ -scan method<sup>25</sup> using 100-fs laser pulses with a peak power of 276 GW cm<sup>−</sup><sup>2</sup> from an optical parametric amplifier operating at a 1-kHz repetition rate [ge](#page-9-0)nerated from a Ti:Sapphire regenerative amplifier system. The laser beam was split into two parts by a beam splitter. One was monitored by a photodiode (D1) as the incident intensity reference,  $I_0$ , and the other beam was detected by the photodiode (D2) as the transmitted intensity. After passing through a lens with  $f = 20$  cm, the laser beam was focused and passed through a quartz cell. The position of the sample cell, z, was moved along the laser-beam direction  $(z \text{ axis})$  by a computer-controlled translatable table so that the local power density within the sample cell could be changed under the constant incident intensity laser power level. Finally, the transmitted intensity from the sample cell was detected by the photodiode D2 interfaced to a computer for signal acquisition and averaging. Each transmitted intensity data represent an average of over 100 measurements. Assuming a Gaussian beam profile, the nonlinear absorption coefficient  $\beta$  can be obtained by curve fitting to the observed open-aperture traces,  $T(z)$ , with eq 1,

$$
T(z) = 1 - \frac{\beta I_0 (1 - e^{-\alpha_0 t})}{2a_0 [1 + (z/z_0)]^2}
$$
 (1)

where  $a_0$  is the linear absorption coefficient, l is the sample length (1 mm quartz cell) and  $z_0$  is the diffraction length of the incident beam.

After the nonlinear absorption coefficient  $\beta$  was obtained, the twophoton absorption cross-section  $\sigma_2$  of the sample molecule (in units of GM, where  $1 \text{ GM} = 10^{-50} \text{ cm}^4 \text{ s} \text{ photon}^{-1}$  can be calculated using eq 2

$$
\sigma_2 = \frac{1000\beta h\nu}{N_{\rm A}d} \tag{2}
$$

where  $N_A$  is the Avogadro constant,  $d$  is the concentration of the sample compound in solution,  $h$  is the Planck constant, and  $v$  is the frequency of the incident laser beam.

Singlet Oxygen Quantum Yield Measurement. Singlet oxygen was detected directly by its phosphorescence emission at 1270 nm using an InGaAs detector on a PTI QM4 luminescence spectrometer. The singlet oxygen quantum yields  $(\Phi_{\Delta})$  of the test compounds were determined in  $CHCl<sub>3</sub>$  by comparing the singlet oxygen emission intensity of the sample solution to that of a reference compound  $(H_2 TPP, \Phi_\Delta = 0.55 \text{ in CHCl}_3)^{26}$  according to eq 3<sup>27</sup>

$$
\Phi_{\Delta}^{\rm S} = \Phi_{\Delta}^{\rm REF} \times \left(\frac{n_{\rm S}}{n_{\rm REF}}\right)^2 \frac{G_{\Delta}^{\rm S}}{G_{\Delta}^{\rm REF}} \times \frac{A_{\rm REF}}{A_{\rm S}} \tag{3}
$$

where  $\Phi_{\Delta}$  is the singlet oxygen quantum yield,  $G_{\Delta}$  is the integrated emission intensity, A is the absorbance at the excitation wavelength,  $n$ is the refractive index of the solvent. Superscripts REF and S correspond to the reference and the sample, respectively. In all measurements, the  ${}^{1}O_{2}$  emission spectra were obtained using an excitation with the absorbance set at 0.05 in order to minimize reabsorption of the emitted light.

Synthesis of Tetraethylene Glycol Rhodamine B, 1. A mixture of tetraethylene glycol monoiodide (912 mg, 3 mmol), rhodamine B (1.437 g, 3 mmol) and  $K_2CO_3$  (415 mg, 3 mmol) in DMF (50 mL) was heated at 70 °C for three days. The product mixture was filtered and the filtrate was evaporated under vacuum. The solid obtained was dissolved in a minimum amount of  $CH_2Cl_2$  and then loaded onto a silica gel column. The first purple-red band, which corresponded to the unreacted rhodamine B, was eluted with EtOAc/MeOH/ Et<sub>3</sub>N (85:10:5). The second purple-red band, which corresponded to product 1, was eluted with  $CH_2Cl_2/MeOH$  (75/25). Removal of solvent by evaporation afforded the purple-brown color sticky product 1 (987 mg, 53.1%): UV/vis  $\left[\text{CH}_2\text{Cl}_2, \lambda_{\text{max}}/ \text{nm} \right]$  (log  $\varepsilon$ )] 521  $(3.94)$ , 557  $(4.66)$ ; <sup>1</sup>H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>):  $\delta = 8.28$ 

 $(d, J = 7.6 \text{ Hz}, 1\text{H}; H_{ar}), 7.77 \text{ (m, 2H; H}_{ar}), 7.24 \text{ (d, } J = 7.6 \text{ Hz}, 1\text{H};$  $H_{\text{ar}}$ ), 7.02 (d, J = 9.6 Hz, 2H;  $H_{\text{ar}}$ ), 6.78 (d, J = 9.6 Hz, 2H;  $H_{\text{ar}}$ ), 6.72 (s, 2H;  $H_{ar}$ ), 4.08 (t, J = 4.7 Hz, 2H; OCH<sub>2</sub>), 3.52 (m, 8H; NCH<sub>2</sub>CH<sub>3</sub>), 3.45 (m, 14H; OCH<sub>2</sub>), 2.56 (broad s, 1H; OH), 1.25 (t,  $J = 7.1$  Hz, 12H; NCH<sub>2</sub>CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (100 MHz, CD<sub>2</sub>Cl<sub>2</sub>)  $\delta =$ 165.3, 159.5, 158.2, 155.9, 134.0, 133.4, 131.8, 131.7, 130.8, 130.5, 130.2, 114.5, 113.9, 96.5, 72.9, 70.9,70.8, 70.7, 70.6, 69.0, 65.1, 61.8, 46.5, 12.8 ppm; HRMS (MALDI)  $m/z$  calcd for  $[C_{36}H_{47}N_2O_7]^+$ 619.3378 [M – Cl] <sup>+</sup>; found 619.3374;  $\Delta_{\rm m}$  = -0.65 ppm.

Synthesis of Rh-SiPc-Rh (2) and Rh-SiPc (3). A mixture of tetraethylene glycol rhodamine B  $(1, 123.8 \text{ mg } 0.2 \text{ mmol})$ , SiPcCl<sub>2</sub> (61.2 mg, 0.1 mmol) and 0.5 mL dried pyridine in dried toluene (20 mL) was refluxed for 24 h. After cooling to room temperature, the solvent, which contained the unreacted 1 and the byproduct SiPctetraethylene glycol without the Rh B, was poured out. The sticky blue liquid mixture adhered to the bottom of the reaction flask was dissolved in a minimum volume of  $CH_2Cl_2$  and loaded onto a basic  $\text{Al}_2\text{O}_3$  column. Product 3 was eluted with  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  (80:20) as a blue band and product 2 was eluted with  $CH_2Cl_2/MeOH$  (75:25) as a purple-blue band. Removal of solvent by passage of air stream afforded the purplish blue solid product 3 (7.76 mg, 4.2%) and the deep bluepale purple colored solid product 2 (19.4 mg, 10.5%). Both products were purified by recrystallization from  $CH_2Cl_2/Et_2O$ . 2: UV/vis  $[CH_2Cl_2, \lambda_{\text{max}}/n$ m  $(\log \varepsilon)]$  353 (4.63), 558 (4.89), 607 (4.13), 645 (4.09), 675 (4.93); <sup>1</sup>H NMR (400 MHz,  $CD_2Cl_2$ )  $\delta = 9.54$  (d, J = 2.7) Hz, 8H; Pc-1,4-H<sub>ar</sub>), 8.29 (d, J = 2.7 Hz, 8H; Pc-2,3-H<sub>ar</sub>), 8.04 (d, J = 7.6 Hz, 2H; Rh- $H_{ar}$ ), 7.63 (t, J = 7.6 Hz, 2H; Rh- $H_{ar}$ ), 7.49 (t, J = 7.8 Hz, 2H; Rh- $H_{ar}$ ), 7.20 (d, J = 7.8 Hz, 2H; Rh- $H_{ar}$ ), 6.86 (d, J = 9.4 Hz, 4H; Rh- $H_{ar}$ ), 6.62 (d, J = 9.4 Hz, 4H; Rh- $H_{ar}$ ), 6.54 (s, 4H; Rh- $H_{ar}$ ), 3.81 (t, J = 4.5 Hz, 4H; OCH<sub>2</sub>), 3.45 (m, 16H; NCH<sub>2</sub>CH<sub>3</sub>), 3.06 (t,  $J = 4.5$  Hz, 4H; OCH<sub>2</sub>), 2.91 (t,  $J = 4.7$  Hz, 4H; OCH<sub>2</sub>), 2.75 (t,  $J =$ 4.7 Hz, 4H; OCH<sub>2</sub>), 2.30 (t, J = 4.7 Hz, 4H; OCH<sub>2</sub>), 1.58 (t, J = 4.7 Hz, 4H; OCH<sub>2</sub>), 1.12 (m, 24H; NCH<sub>2</sub>CH<sub>3</sub>), 0.27 (t, J = 5.6 Hz, 4H; Si-OCH<sub>2</sub>CH<sub>2</sub>O),  $-2.05$  ppm (t, J = 5.6 Hz, 4H; Si-OCH<sub>2</sub>CH<sub>2</sub>O); HRMS (MALDI)  $m/z$  calcd for  $[C_{104}H_{108}N_{12}O_{12}Si]^{2+}$  1777.7901 [M – 2Cl]<sup>+</sup>; found 1777.7910;  $\Delta_m = 0.46$  ppm. 3: UV/vis [CH<sub>2</sub>Cl<sub>2</sub>,  $\lambda_{\text{max}}/\text{nm}$  (log  $\varepsilon$ )] 343 (4.57), 559 (4.73), 611 (4.24), 649 (4.16), 679 (4.99); <sup>1</sup>H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>)  $\delta$  = 9.49 (m, 8H; Pc-1,4-H<sub>ar</sub>), 8.23 (d, J = 7.6 Hz, 1H; Rh- $H_{ar}$ ), 8.18 (m, 8H; Pc-2,3- $H_{ar}$ ), 7.67 (m, 2H; Rh- $H_{ar}$ ), 7.30 (d, J = 7.6 Hz, 1H; Rh- $H_{ar}$ ), 6.92 (d, J = 9.4 Hz, 2H; Rh- $H_{ar}$ ), 6.52 (d, J = 9.4 Hz, 2H; Rh- $H_{ar}$ ), 6.34 (s, 2H; Rh- $H_{ar}$ ), 4.01  $(t, J = 4.6 \text{ Hz}, 2\text{H}; \text{OCH}_2)$ , 3.20 (m, 10H; OCH<sub>2</sub> + NCH<sub>2</sub>CH<sub>3</sub>), 2.95  $(t, J = 4.6 \text{ Hz}, 2H; OCH<sub>2</sub>)$ , 2.69  $(t, J = 4.6 \text{ Hz}, 2H; OCH<sub>2</sub>)$ , 2.25  $(t, J =$ 4.8 Hz, 2H; OCH<sub>2</sub>), 1.56 (t, J = 4.8 Hz, 2H; OCH<sub>2</sub>), 1.00 (t, J = 7.2 Hz, 12H; NCH<sub>2</sub>CH<sub>3</sub>), 0.29 (t, J = 5.4 Hz, 2H; SiOCH<sub>2</sub>CH<sub>2</sub>O), -2.00  $(t, J = 5.40 \text{ Hz}, 2\text{H};$  SiOCH<sub>2</sub>CH<sub>2</sub>O) ppm; HRMS (MALDI)  $m/z$  calcd for  $[C_{68}H_{62}N_{10}O_8Si]^+$  1175.4594  $[M - Cl]^+$ ; found 1175.4604;  $\Delta_m =$ 0.85 ppm.

**Synthesis of C<sub>7</sub>-SiPc-C<sub>7</sub>** (4). A mixture of tri(ethyl glycol)monomethylether (79  $\mu$ L, 0.5 mmol), SiPcCl<sub>2</sub> (30.5 mg, 0.05 mmol), and anhydrous  $K_2CO_3$  (27.6 mg, 0.2 mmol) in dried toluene (10 mL) was refluxed for 14 h. After it was cooled to room temperature, the mixture was directly loaded onto a basic  $Al_2O_3$  column. Product 4 was eluted with  $CH_2Cl_2/MeOH$  (80:5) as a green band. Removal of solvent by evaporation afforded a green solid product 4 (6.7 mg, 15.6%). The crude product was purified by recrystallization from  $\text{CH}_2\text{Cl}_2/\text{Et}_2\text{O}$ : UV/vis $\text{[CH}_2\text{Cl}_2$   $\lambda_{\text{max}}/\text{nm}$  (log  $\varepsilon)$  ] 354 (4.01), 605  $(3.80)$ , 643  $(3.72)$ , 673  $(4.62)$ ; <sup>1</sup>H NMR  $(400 \text{ MHz}, \text{CD}_2\text{Cl}_2)$   $\delta = 9.63$  $(m, 8H; Pc-1,4-H<sub>ar</sub>), 8.34 (m, 8H; Pc-2,3-H<sub>ar</sub>), 3.14 (s, 6H; OCH<sub>3</sub>),$ 3.12 (t, J = 4.6 Hz, 4H; OCH<sub>2</sub>), 2.95 (t, J = 4.8 Hz, 4H; OCH<sub>2</sub>), 2.45  $(t, J = 4.9 \text{ Hz}, 4\text{H}; \text{OCH}_2)$ , 1.67  $(t, J = 4.9 \text{ Hz}, 4\text{H}; \text{OCH}_2)$ , 0.38  $(t, J = 1.9 \text{ Hz}, 4\text{H}; \text{OCH}_2)$ 5.7 Hz, 4H; SiOCH<sub>2</sub>CH<sub>2</sub>O),  $-1.92$  (t, J = 5.7 Hz, 4H; SiOCH<sub>2</sub>CH<sub>2</sub>O) ppm; <sup>13</sup>C NMR (100 MHz, CD<sub>2</sub>Cl<sub>2</sub>)  $\delta$  = 29.70, 54.68, 58.79, 68.56, 69.30, 69.70, 71.46, 123.64, 130.83, 135.98, 149.21 ppm; HRMS (MALDI)  $m/z$  calcd for  $[C_{46}H_{46}N_8O_8SiNa]^+$  889.3100  $[M + Na]^+$ ; found 889.3149;  $\Delta_{\rm m}$  = 5.51 ppm.

X-ray Crystallography. X-ray diffraction data were collected at 173 K using graphite-monochromated Mo K $\alpha$  radiation ( $\lambda = 0.71073$ ) Å) on a Bruker Axs SMART 1000 CCD diffractometer. The collected frames were processed with the software SAINT+<sup>28</sup> and an absorption

Scheme 1. Synthesis Route to the  $Si(IV)Pc-Rhodamine B$  Conjugates, 2 and 3, and the  $Si(IV)Pc$  with Axially Ligated Polyethylene Glycol, 4



correction (SADABS)<sup>29</sup> was applied to the collected reflections. The structure was solved by the Direct method  $(SHELXTL)^{30}$  in conjunction with s[tan](#page-9-0)dard difference Fourier techniques and subseque[n](#page-9-0)tly refined by full-matrix least-squares analyses on  $F^2$ . . Hydrogen atoms were generated in their idealized positions and all non-hydrogen atoms were refined anisotropically. Crystal data for 1:  $C_{46}H_{46}N_8O_8Si$ ; *M* = 867.00; triclinic; space group  $\bar{P}1$ ; *a* = 7.9377(8), b = 10.210(1), c = 13.216(1) Å;  $\alpha$  = 94.452(2),  $\beta$  = 94.984(2),  $\gamma$  = 105.604(2)°;  $V = 1022.1(2)$  Å<sup>3</sup>;  $T = 173(2)$  K;  $Z = 1$ ; reflections collected/unique, 6493/4666,  $R_{int} = 0.0184$ ; final R indices  $[I > 2\sigma(I)],$  $R_1 = 0.0366$ ,  $R_2 = 0.0969$  for 4162 observed reflections; R indices (all data),  $R_1 = 0.0415$ ,  $R_2 = 0.1013$ ; GOF = 1.040; CCDC 779323.

#### ■ RESULTS AND DISCUSSION

Synthesis of Silicon(IV) Phthalocyanine-Rhodamine B **Conjugates.** The synthetic route for the two silicon $(IV)$  phthalocyanine-rhodamine B conjugates, 2 and 3, is shown in Scheme 1. Briefly, tetraethylene glycol monoiodide was prepared according to the literature procedure.<sup>31</sup> Treatment of the commercially available Rh B with tetraethylene glycol monoiodide in the presence of  $K_2CO_3$  in DMF g[ave](#page-9-0) the tetraethylene glycol-rhodamine B, 1, in 53% yield after its separation from the unreacted Rh B by silica gel column chromatography. Further washing of the silica gel column with the eluting solvent allowed a second harvest of 1, resulting in a combined yield of 70%.

Products 2 and 3 were synthesized according to the literature procedure.<sup>32</sup> Bases, such as  $K_2CO_3$ , NaH, quinoline, DMF, pyridine, have been used to render silicon(IV) phthalocyanine dichloride  $(SiPcCl<sub>2</sub>)$  reactive to hydroxyl derivatives. In our preparation of 2 and 3,  $K_2CO_3$  and pyridine were tried. Both bases gave a number of byproduct of SiPc containing one and



Figure 1. Perspective drawing of 4, with the thermal ellipsoids drawn at the 25% probability level. Selected bond lengths (Å) and angles (deg): Si−N(1) 1.920(1), Si−N(3) 1.9333(9), Si−O(1) 1.6971(8); N(1)−Si−N(3) 89.88(4), N(1)−Si−O(1) 86.61(4).

two polyethers without the Rh B. These byproduct were identified after TLC separation by MALDI-TOF-MS and NMR. The yields of 2 and 3 were found to be similar for both bases. However, with pyridine as base, products 2 and 3 were found in the sticky liquid adhered to the bottom of the reaction flask, while the byproduct and the starting material, 1, were present in the toluene solvent bulk and were therefore removed readily. Products 2 and 3 were then isolated and purified by basic alumina column chromatography. Dichloromethane/methanol was used as the eluting solvent to elute products 2 (yield 10.5%) and 3 (yield 4.2%) from the basic alumina column. Since methanol has been reported to replace the axial ligand of  $SiPc<sub>2</sub><sup>32,33</sup>$  other mixed solvents without methanol, such as  $CH_2Cl_2/a$ cetonitrile,  $CH_2Cl_2/$ tetrahydrofuran (THF), and ethy[l ace](#page-9-0)tate/THF, were tried as eluent but to no avail. This observation suggests that the cations present in 2 and 3 might have increased their adsorption to the basic alumina column, thus reducing the recoverable yields. Another reason for the low observed yield could be the result of the displacement of the axially ligated Rh B by the methanol used in elution. In contrast, product 4 is more stable and no obvious decomposition was observed during purification. Both 2 and 3 are soluble in most organic solvents, such as  $CH_2Cl_2$ , CHCl<sub>3</sub>, methanol, acetonitrile, THF, dimethylformamide, and dimethylsulfoxide, and they even showed slight solubility in water, giving purple red solutions.

All products were characterized by NMR and high-resolution mass spectrometry. Compared to the <sup>1</sup>H NMR spectrum of Rh B, product 1 showed two additional peaks at  $\delta$  3.56 (14 H) and 4.08 ppm (2H), corresponding to the ethylene protons of the tetra(ethylene glycol) that are distal and proximal, respectively, to the Rh B (Supporting Information Figure S1). The eight new carbon peaks found at  $\delta$  62−73 ppm in the <sup>13</sup>C NMR spectrum of 1 [are due to the eight](#page-8-0) carbon atoms of the tetra(ethylene glycol). MALDI-TOF mass spectrum of 1 gave a peak at  $m/z$  of 619.3374, which compared favorably ( $\Delta_m$  =  $-0.65$  ppm) to the calculated *m/z* of 619.3378 for the [M –  $Cl$ <sup>+</sup> ion of 1 (Supporting Information Figure S2)

The <sup>1</sup>H NMR spectra of 2, 3, and 4 showed two downfield groups of peaks ( $\delta = 8-10$  ppm) typical of the  $\alpha$ - and  $\beta$ protons of the Pc [ring](#page-8-0) [\(Supporting](#page-8-0) [Info](#page-8-0)rmation Figures S3, S5, and S7). Because of the strong ring current effect of the Pc, the signals of the ethylene [glycol protons of](#page-8-0) 2, 3, and 4 are upfield shifted when compared to 1. These ethylene glycol signals appeared as eight four-proton triplets from  $\delta$  –2.05 to 3.81 ppm for 2; seven two-proton triplets from  $\delta$  –2.00 to 4.01 ppm and another two-proton signal overlapped with Rh B methylene group at  $\delta$  3.20 ppm for 3; and six four-proton triplets from

 $\delta$  –1.91 to 3.12 for 4. The methoxy protons of 4 appeared as a sharp singlet at  $\delta$  3.14 ppm. High resolution mass spectra were also measured to confirm the identity of these compounds. Molecular ion peak at  $m/z$  1777.7910 observed for 2 is in good agreement with the calculated  $m/z$  value of 1777.7901 for the [M−2Cl]+ ion of 2 (Supporting Information Figure S4). For 3, two major molecular ion peaks were seen at  $m/z$  1175.4604 and  $m/z$  1158.411[3, corresponding to its](#page-8-0)  $[M - Cl]^+$  and  $[M - Cl - OH]$ <sup>+</sup> ions, respectively (Supporting Information Figure S6). Three major molecular ion peaks were observed for 4, that is, at  $m/z$  889.3149,  $m/z$  728.3082 and  $m/z$  703.2029, which corresponded to its [M+Na]<sup>+</sup> ion, [M−[tri\(ethyl](#page-8-0) [glycol\)](#page-8-0) monomethylether+Na]<sup>+</sup> ion and [M−tri(ethyl glycol) monomethyl ether]+ ion, respectively (Supporting Information Figure S8). The molecular structure of 4 was also confirmed by X-ray crystallography (see Figure 1). [The asymmetric unit](#page-8-0) consists of half of a molecule and the two halves are related by a center of inversion at the Si atom. The Si center is six-coordinate with four Pc nitrogen donor atoms and two axial oxygen donors from the polyethylene glycol units arranged in an octahedral environment.

Photophysical Properties of 2 and 3. The UV/vis spectra of 2 and 3, together with those of 1 (Rh) and 4



Figure 2. UV/vis spectra of the 1 (Rh,  $\varepsilon_{520nm} = 46600$ ;  $\varepsilon_{557nm} =$ 145000), 2 (Rh-SiPc-Rh,  $\varepsilon_{520nm} = 26400$ ;  $\varepsilon_{557nm} = 5990$ ;  $\varepsilon_{607nm} =$ 46600;  $\varepsilon_{673\text{nm}}$  = 50000), 3 (SiPc-Rh ( $\varepsilon_{558\text{nm}}$  = 28700;  $\varepsilon_{607\text{nm}}$  = 16490;  $\varepsilon_{639nm} = 21000$ ;  $\varepsilon_{672nm} = 74000$ ), and  $4 (C_7\text{-}SiPc-C_7, \varepsilon_{672nm} = 16400)$ at concentration of  $~\sim$ 1 µM.

 $(C_7$ -SiPc- $C_7$ ) in CH<sub>2</sub>Cl<sub>2</sub> are shown in Figure 2. The simultaneous appearance of the rhodamine and SiPc characteristic absorptions in the spectra of 2 and 3 confirms their rhodamine-SiPc conjugate structure. Furthermore, the absorption intensity of 2 at 557 nm (rhodamine absorption peak) is ∼2-fold to that of 3. This observation is consistent with the ligation of two and one Rh B



Figure 3. (a) UV/vis spectra of different concentrations of 2 in  $CH<sub>2</sub>Cl<sub>2</sub>$ . The inset shows a plot of the absorbance of 2 at 557 nm (square), due to the Rh B moiety, and at 672 nm (triangle), due to the SiPc moiety, as a function of its concentration. (b) Emission spectra of different concentrations of 2 in  $CH_2Cl_2$ . The inset shows a plot of the emission of 2 at 560 nm (square), due to the Rh B moiety, at 681 nm (circle) and at 753 nm (triangle), due to the SiPc moiety, as a function of its concentration.

moiety in 2 and 3, respectively. Figure 3 shows the absorption and emission spectra of 2 as a function of its concentration in  $CH<sub>2</sub>Cl<sub>2</sub>$ . The spectra exhibit typical features of nonaggregated phthalocyanines. The inset in Figure 3a shows a plot of the absorption intensities of the SiPc Q-band and rhodamine versus the concentration of 2. The inset in Figure 3b shows a plot of the emissions of 2 due to its Rh B (560 nm) and SiPc (681 nm) moieties versus its concentration. These linear plots observed in both the absorption and emission spectra of 2 shows that this compound is essentially free from aggregation in the indicated concentration ranges in  $CH<sub>2</sub>Cl<sub>2</sub>$ .

The fluorescence spectra of 2 and 3 are shown in Figure 4. For both conjugates, excitation of the rhodamine band at 515 nm led to emissions from both the Rh and the Pc moieties while excitation of the Pc chromophore at 640 nm resulted in emission exclusively from the Pc moiety. Furthermore, the emission from the Pc moiety excited at 515 nm coincided with the emission obtained from excitation at 640 nm. These observations indicate an intramolecular energy transfer from the excited Rh B to the Pc moiety in 2 ( $\lambda_{\text{em}} = 681 \text{ nm}$ ) and 3  $(\lambda_{em} = 688 \text{ nm})$ , very similar to that observed previously for  $Zn(II)Pc$  with peripherally substituted rhodamines.<sup>19</sup> These results clearly indicate that such intramolecular energy transfer from Rh to Pc can occur readily with either axially [lig](#page-9-0)ated or peripherally substituted rhodamines.

Recently, phthalocyanines  $(H_2Pc$  and  $ZnPc)$  peripherally substituted by four tetraphenylporphyrins  $(H_2TPP$  and ZnTPP) via alkyne linkages in a windmill fashion was synthesized and the intramolecular porphyrin-to-Pc energy transfer studied.<sup>36</sup> Compared to the almost quantitative TPPto-Pc energy transfer, the efficiency of Rh-to-Pc energy transfer in 2 and 3 is si[gn](#page-9-0)ificantly lower. The difference is presumably due to the long and nonconjugative alkoxy linkage between the donor and acceptor in 2 and 3 as compared to the short and conjugative alkyne linkage between TPP and Pc.

Production of Singlet Oxygen from 2 and 3. To assess their potential as effective PDT agents, the  ${}^{1}O_{2}$  production capabilities of these Rh-SiPc conjugates were measured based on the phosphorescence intensities of the  ${}^{1}O_{2}$  produced upon photoirradiation of these compounds. Figure 5a shows the  $^{1}O_{2}$ phosphorescence spectra of 2, 3, and  $H_2TPP$  in CHCl<sub>3</sub>. Using the <sup>1</sup>O<sub>2</sub> quantum yield of H<sub>2</sub>TPP ( $\Phi$ <sub>Δ</sub> = 0.[55](#page-6-0) ± 0.11) as the reference standard,<sup>21,26</sup> the relative  $\Phi_{\Delta}$  of Rh-SiPc-Rh (2) and Rh-SiPc (3) were estimated to be  $0.48 \pm 0.11$  and  $0.25 \pm 0.11$ ,



Figure 4. Normalized excitation spectrum (black, monitored at 750 nm) and fluorescence spectra (blue,  $\lambda_{\text{exc}} = 515$  nm and red,  $\lambda_{\text{exc}} = 640$  nm) of 2 (a) and 3 (b) in CH<sub>2</sub>Cl<sub>2</sub> (∼1.0 × 10<sup>-6</sup> M). Note that the emissions from the Pc moiety of 2 and 3 excited at 515 and 640 nm coincide almost completely.

<span id="page-6-0"></span>

**Figure 5.** Near-IR phosphorescence spectra of the  $^1O_2$  generated from the following photosensitizers: (a)  $\rm H_2TPP$  (black), Rh-SiPc-Rh (2, blue), and Rh-SiPc (3, red) excited at 549 nm in CHCl<sub>3</sub>; (b) ZnPc (black), Rh-SiPc-Rh (2, red), C<sub>7</sub>-SiPc-C<sub>7</sub> (4, blue), and Rh-SiPc (3, green) excited at 666 nm in toluene. Note that the concentrations of the photosensitizers used are different between a and b.

respectively, when excited at the Rh B absorption maximum of 549 nm. Since rhodamine has very low  ${}^{1}O_{2}$  production capability,<sup>37</sup> the substantial  ${}^{1}O_{2}$  generation observed must be the result of an energy transfer from the photoexcited Rh B to the Pc m[oie](#page-9-0)ty which produced the  ${}^{1}O_{2}$  observed in 2 and 3.

To evaluate the  ${}^{1}O_{2}$  production capability of the SiPc moiety alone, the  ${}^{1}O_{2}$  phosphorescence spectrum of C<sub>7</sub>-SiPc-C<sub>7</sub> (4), soluble in toluene, upon photoexcitation at 549 nm was measured and is shown in Figure 5b, together with those spectra obtained from ZnPc, 2 and 3, measured under identical conditions. The relative  $\Phi_{\Delta}$  estimated for ZnPc, Rh-SiPc-Rh (2), Rh-SiPc (3), and C<sub>7</sub>-SiPc-C<sub>7</sub> (4) in toluene were  $0.58 \pm 0.11$ ,  $0.50 \pm 0.10$ ,  $0.23 \pm 0.07$ , and  $0.38 \pm 0.08$ , respectively. These measurements show that (1) SiPc is quite capable of producing singlet oxygen even when excited at a nonoptimal wavelength (e.g., 4 with  $\Phi_{\Delta}$  = 0.38 at 549 nm), (2) the ligation of two Rh B to SiPc increased its <sup>1</sup>O<sub>2</sub> quantum yield (2 with  $\Phi_{\Delta} = 0.50$ ), and (3) the presence of an additional Rh B chromophore in 2 resulted in a 2-fold increase in the  $^{1}O_{2}$  quantum yield of 3 ( $\Phi_{\Delta}$  = 0.23). Thus, while the introduction of the Rh moiety to SiPc is mainly for its mitochondria-targeting function in cells (vide infra), its presence also allows for  ${}^{1}O_{2}$  generation from Pc's at shorter wavelengths that support the Rh-to-Pc energy transfer.

Two-Photon Absorption Properties of 2, 3, and 4. Recent studies showed that  ${}^{1}O_{2}$  can also be generated via twophoton excitation of a sensitizer.<sup>38</sup> In this approach, a chromophore such as Rh B which absorbs at  $\lambda$  < 650 nm, that is, in the tissue nontransparent spectral r[eg](#page-9-0)ion, by one-photon absorption, can be excited by two-photon absorption (2PA) at  $\lambda > 650$ nm, which allows much better depth penetration through tissues. Other advantages of 2PA-PDT include (1) high-resolution (femtoliter scale) targeting with pinpoint accuracy, thus minimizing photodamage to tissues along the beam path of a one-photon PDT, (2) negligible photobleaching of the sensitizing chromophore, and (3) minimal endogenous autofluorescence.39,40

To assess the potential of 2 and 3 as 2PA-PDT agents, we measur[ed th](#page-9-0)eir two-photon absorption cross sections,  $\sigma_2$ , by the open aperture Z-scan method.<sup>24</sup> The results are shown in Figure 6, from which the 2PA cross sections of 2 and 3 were derived to be 1494 G[M](#page-9-0) and 706 GM (where GM =  $10^{-50}$  cm<sup>4</sup> s photon<sup>-1</sup>), respectively, at 850 nm. Compared to the  $\sigma_2$  of Rh B, that is, ~120 GM at 850 nm,<sup>42</sup> an apparent 6-fold enhan-



Figure 6. Open-aperture Z-scan traces of 0.1 mM of Rh-SiPc-Rh, (2, top panel) and Rh-SiPc, (3, bottom panel) excited at 850 nm. The 2PA cross sections determined for 2 and 3 were 1494 GM and 706 GM, respectively.

cement in  $\sigma_2$  for each Rh B chromophore can be seen in these Rh-SiPc conjugates. To ascertain that this enhancement of  $\sigma_2$  in 2 and 3 is derived principally from Rh B being ligated to SiPc and not from the SiPc moiety itself, we compared these  $\sigma_2$ values to that of  $C_7$ -SiPc- $C_7$ , 4, which was measured to be 318 GM (data shown in Supporting Information Figure S11). This comparison clearly shows that the enhancement of  $\sigma_2$  in 2 and 3 is due to the [ligation of Rh B to SiP](#page-8-0)c. This notion is corroborated by the 2-fold higher  $\sigma_2$  value in 2, with two ligated Rh B, as compared to that in 1, with only one ligated Rh B. Such enhancement in two-photon absorption efficiency in 2 and 3 might be due to a possible intramolecular electron transfer

between Rh B and SiPc, particularly considering the short polyethylene glycol linkage between them. A more detailed photophysical study using transient absorption spectroscopy should shed more light on this issue. Nonetheless, the substantial  $\sigma_2$ measured for 2 and 3 suggests that these conjugates possess potent 2PA-PDT activity.

Subcellular Localization of 2 and 3. To ascertain the mitochondria-localizing a[bili](#page-9-0)ty of Rh-SiPc-Rh (2) and Rh-SiPc (3), the human nasopharyngeal carcinoma cells (HK-1) were costained with mitochondria-specific probe. Judging from the overlapping of the fluorescence signals (yellow dots) emitted by these Rh-SiPc conjugates (2 and 3, red dots) and the mitochondria-specific probe (Mito Tracker Green, green dots), both 2 (Supporting Information Figure S11c, upper panel) and 3 (Supporting Information Figure S11d, upper panel) were found t[o localize in the mito](#page-8-0)chondria of HK-1 cells. The localization of 2 and 3 in mitochondria was further confirmed by [analyzing](#page-8-0) [the](#page-8-0) [fluorescen](#page-8-0)ce-intensity profile drawn across the cells. The fluorescence profiles of these Rh-SiPc conjugates along the arrow (Supporting Information Figures S11c and S11d, lower panels) are similar to the fluorescence profile of the mitochondria-specif[ic probe. Furthermore, th](#page-8-0)e patterns of colocalization of 2, 3, and Rh B (Supporting Information Figure S11a, upper panel) with the mitochondria-specific probe are very similar. In contrast, mitochondrial localization of  $C_7$ -SiPc- $C_7$  (Supporting Information Figure S11b, upper panel), which possesses no Rh moiety, was not observed. These observations indicat[e that Rh B](#page-8-0) [functioned a](#page-8-0)s a carrier for phthalocyanine to gain entry into the mitochondria. The mitochondria-localizing capability of 2 and 3 was also demonstrated in a noncancerous cell line, namely, human umbilical vein endothelial cells (HUVEC), which is shown in Supporting Information Figure S12.

The localizing properties of 2 and 4 in other subcellular [organelles, namely, lyso](#page-8-0)somes, Golgi bodies and endoplasmic reticulum (ER), were also studied using the corresponding organelle-specific probes. The confocal microscopic images obtained from these colocalization experiments are shown in Supporting Information Figures S13, S14, and S15. No significant localization of 2 in lysosomes, Golgi bodies and [endoplasmic reticulum in](#page-8-0) the HK-1 cells was seen (Supporting Information Figure S13). As for the  $C_7$ -SiPc- $C_7$  control (4), no significant organelle-specific localization in lysoso[mes, Golgi](#page-8-0) [bodies and](#page-8-0) endoplasmic reticulum was detected (Supporting Information Figures S14 and S15). These observations, taken together, indicate an almost exclusive subcellular lo[calization of](#page-8-0) [the Rh-SiPc](#page-8-0) conjugate, 2, in the mitochondria of these cells.

Photocytotoxicity of 2, 3 and 4 Toward HK-1 cells. The dark and photo-cytototoxicity of 2 and 3 were measured, together with Rh B and  $C_7$ -SiPc- $C_7$  (4), which served as the reference. The results are shown in Figure 7. With light doses from 1 to 4  $J/cm^2$ , 2, 3, and 4 exhibited a dose-dependent cytotoxicity to the HK-1 cells. However, photocytotoxicity was not observed in the Rh B-treated HK-1 cells. In addition, no significant cytotoxicity (i.e., <10%) due to 2 and 3 was observed when the HK-1 cells were incubated with these Rh-SiPc conjugates without photoirradiation. Furthermore, the observed photocytotoxicity of <sup>2</sup>, <sup>3</sup>, and <sup>4</sup> does not correlate with the measured <sup>1</sup>  ${}^{1}O_{2}$  quantum yields (2 > 4 > 3) of these compounds, suggesting that their subcellular localization might exert a crucial influence on their photocytotoxicity. However, recently Saha et al showed that the amount of complexes taken up by the cells is also a key factor in determining their in vitro toxicities



Figure 7. Photocytotoxicity of Rh B and Rh-SiPc conjugates in HK-1 cells. HK-1 cells  $(2 \times 10^4 \text{ cells/well})$  were incubated in wells of 96-well plate for overnight. The cells were then treated with 100 nM of Rh B,  $C_7$ -SiPc- $C_7$  (4), Rh-SiPc (3), or Rh-SiPc-Rh (2) for 6 h in dark, followed by photoirradiation (1−4 J/cm<sup>2</sup> ) and MTT cytotoxicity assay. Results were expressed as the mean  $\pm$  SD of three independent experiments.

(dark and light). $43$  Further studies on the cellular uptake of these conjugates will be conducted to address this issue.

With the est[ab](#page-9-0)lishment of mitochondria-localizing and photocytotoxic properties of these Rh-SiPc conjugates, the next logical question is the mode of cell death (i.e., apoptosis vs. necrosis) induced by these conjugates. To determine the purported apoptotic cell death associated with the mitochondria-localizing property of Rh-SiPc conjugates, both nuclei staining method and flow cytometric DNA content analysis were performed on the HK-1 cells after PDT with 2. No apoptotic nuclei formation was observed under the light dose 1  $I/cm<sup>2</sup>$ . . However, apoptotic nuclei were observed for 3 h after PDT with light doses of 2 and 4  $J/cm<sup>2</sup>$  (Supporting Information Figure S16).

The extent of apoptosis was furt[her evaluated using flow](#page-8-0) cytometry and the results are shown in Supporting Information Figure S17. 2-PDT increased the percentage of sub-G1 cells from 2% (control) to 10.9% (2  $\bar{J}/\text{cm}^2$ ) to 20% (4  $\bar{J}/\text{cm}^2$ ), indicating that 2-PDT dose-dependently induced tumor cell apoptosis. Under the light dose of  $1$  J/cm<sup>2</sup>, 2-PDT was found to increase the percentage of G1 cells from 50.8% in the cell control group to 81.8% after PDT. This observation suggests that a low dose of 2-PDT would induce G1 cell cycle arrest.

Two-Photon-Induced Cytotoxicity of 2, 3, and 4 Toward Human Cervical Carcinoma (HeLa) and Human Nasopharyngeal Carcinoma (HK-1) Cells. HK-1 cells were incubated with 0.5  $\mu$ M of Rh-SiPc-Rh (2), Rh-SiPc (3), or C<sub>7</sub>-SiPc-C<sub>7</sub> (4) containing no Rh B moiety, for 6 h. The cells were then excited at 850 nm, where Rh B and its SiPc conjugates, 2 and 3, showed a two-photon absorption cross-section of ∼120 GM,<sup>42</sup> 1494 GM and 706 GM, respectively. The confocal images were captured at one laser shot per minute for a total of 3[0 m](#page-9-0)in. Figure 8 shows the confocal images of the cells at two-photon laser irradiation time  $t = 0$ , 5, 25, and 30 min. Bright fluorescent [im](#page-8-0)ages (i.e., images captured at  $t = 0$  min) were seen in cells treated with Rh-SiPc-Rh (2). The fluorescent intensity of the treated cells was time-dependently decreased. Similar changes in fluorescent intensity were also seen in cells treated with Rh-SiPc (3). There are two possible explanations for the reduced fluorescent intensity in cells after two-photon irradiation, namely, the photobleaching and the leakage of the Rh-SiPc conjugates from the damaged cell membrane after PDT. Photobleaching of the Rh-SiPc

<span id="page-8-0"></span>

Figure 8. Confocal microscopic images of HK-1 cells treated with 0.5  $\mu$ M of Rh-SiPc-Rh, (2, top), Rh-SiPc (3, middle), and C<sub>7</sub>-SiPc-C<sub>7</sub> (4, bottom) obtained at various time points (0, 5, 25, and 30 min) under continuous laser flash excitation at 850 nm (one laser shot per min). The cells were incubated with 2, 3, and 4 for 6 h before the laser flash excitation.

conjugates is unlikely an explanation as photobleaching of chromophores resulting from two-photon excitation is considered negligible.39−<sup>41</sup> Leakage of the Rh-SiPc conjugates through the damaged membrane might be the possible explanation for the [r](#page-9-0)e[du](#page-9-0)ced fluorescent intensity of the stained cell. This explanation is supported by the fact that 2 or 3-PDT treated cells tend to round up, an early sign of cell death, after 2PA-PDT treatment. In contrast to Rh-SiPc and Rh-SiPc-Rh, the morphology of the 4-PDT treated HeLa cells is similar to the cell image captured at before laser irradiation (i.e.,  $t = 0$  min). These observations clearly showed that 2 and 3 possess 2PA-PDT activities when excited at 850 nm whereas 4, which bear no Rh B chromophore, displayed no significant 2PA-PDT activity. This experiment, however, precludes a quantitative comparison of the PDT activities of 2 and 3 resulted from one- versus two-photon excitation because the experimental conditions used (e.g., concentrations of the conjugates, cell type, etc.) were quite different. Nonetheless, the result does demonstrate substantial PDT activities of 2 and 3 induced via two-photon excitation of the Rh B chromophore potentially. In addition, Figure 8 also shows that 2 gave a stronger in vitro emission than 3 under two-photon excitation at 850 nm. This observation is consistent with the 2-fold higher  $\sigma_2$  value measured for 2 than 3 (Figure 6).

# **EN CONCLUDING REMARKS**

Two axially ligated rhodamine-Si(IV)-phthalocyanine conjugates, bearing one and two rhodamine B, were synthesized and their subcellular localizing and photocytotoxic properties induced by one- and two-photon excitation studied. These Rh-SiPc conjugates, 2, exhibited an almost exclusive mitochondrial localizing property, together with strong photocytotoxic but low dark cytotoxic properties. Evidence of apoptotic cell death, which is consistent with their mitochondrial localization property, was also seen in the photocytotoxicity induced by these conjugates. These properties qualify them as highly promising PDT agents.

## ■ ASSOCIATED CONTENT

#### **8** Supporting Information

NMR spectra, high-resolution MALDI-TOF mass spectra, in vitro imaging, and cytotoxicity results of the compounds synthesized. This material is available free of charge via the Internet at http://pubs.acs.org.

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#### ■ REFERENCES

(1) (a) Hasan, T.; Ortel, B.; Moor, A. C. E.; Pogue, B. W. In Holland-Frei Cancer Medicine 6; Kufe, D. W., Pollock, R. E., Weichselbaum, R. R., Bast, R. C., Jr., Gansler, T. S., Holland, J. F., Frei, E., III, Eds.; Decker, Hamilton, 2003, pp 605−622. (b) Triesscheijn, M.; Bass, P.; Schellens, J. H. M.; Stewart, F. A. Oncologist 2006, 11, 1034−1044. (c) Brown, S. B.; Brown, E. A.; Walker, I. Lancet Oncol. 2004, 5, 497− 508. (d) Dolmans, D. E. J. G. J.; Fukumura, D.; Jain, R. K. Nature Rev. Cancer 2003, 3, 380−387.

(2) (a) Dougherty, T. J.; Gomer, C. J.; Henderson, B. W.; Jori, G.; Kessel, D.; Korbelik, M.; Moan, J.; Peng, Q. J. Natl. Cancer Inst. 1998, 90, 889−905. (b) Juzeniene, A.; Peng, Q.; Moan, J. Photochem. Photobiol. Sci. 2007, 6, 1234−1245. (c) Castano, A. P.; Demidova, T. N.; Hamblin, M. R. Photodiagn. Photodyn. Ther. 2004, 1, 279−293. (d) Castano, A. P.; Demidova, T. N.; Hamblin, M. R. Photodiagn. Photodyn. Ther. 2005, 2, 91−106.

<span id="page-9-0"></span>(3) Plaetzer, K.; Krammer, B.; Berlanda, J.; Berr, F.; Klesslich, T. Lasers Med. Sci. 2009, 24, 259−268. M Detty, M. R.; Gibson, S. L.; Wagner, S. J. J. Med. Chem. 2004, 47, 3897−3915. Moan, J. J. Photochem. Photobiol. B: Biol. 1990, 5, 521−524.

(4) Moan, J. J. Photochem. Photobiol. B: Biol. 1990, 6, 343−344.

(5) (a) Oleinick, N. L.; Morris, R. L.; Belichenko, I. Photochem. Photobiol. Sci. 2002, 1, 1−21. (b) Fulda, S.; Galluzzi, L.; Kroemer, G. Nature Rev. Drug Discovery 2010, 9, 447−464. (c) Sasnauskiene, A.; Kadziauskas, J.; Veselyte, N.; Jonusiene, V.; Kirveliene, V. Apoptosis 2009, 14, 276−286. (d) Kessel, D.; Luo, Y. Cell Death Differ. 1999, 6, 28−35. (e) Green, D. R.; Reed, J. C. Science 1998, 281, 1309−1312. (6) Johnson, L. V.; Walsh, M. L.; Chen, L. B. Proc. Natl. Acad. Sci. U.S.A. 1980, 77, 990−994.

(7) (a) Lee, D. R.; Helps, S. C.; Macardle, P. J.; Nilsson, M.; Sims, N. R. Neurochem. Res. 2009, 34, 1857−1866. (b) Ferlini, C.; Scambia, G. Nat. Protocols 2007, 2, 3111−3114. (c) Baracca, A.; Sgarbi, G.; Solaini, G.; Lenaz, G. Biochim. Biophys. Acta-Bioenergy 2003, 1606, 137-146.

(8) (a) Moreira, L. M.; dos Santos, F. V.; Lyon, J. P.; Maftoum-Costa, M.; Pacheco-Soares, C.; de Silva, N. S. Aust. J. Chem. 2008, 61, 741− 754. (b) Miller, J. D.; Baron, E. D.; Scull, H.; Hsia, A.; Berlin, J. C.; McCormick, T.; Colussi, V.; Kenney, M. E.; Cooper, K. D.; Oleinick, N. L. Toxicol. Appl. Pharmacol. 2007, 224, 290−299. (c) Ogura, S.; Tabata, K.; Fukushima, K.; Kamachi, T.; Okura, I. J. Porphyrins Phthalocyanines 2006, 10, 1116−1124. (d) Lukyanets, E. A. J. Porphyrins Phthalocyanines 1999, 3, 424−432.

(9) (a) Longo, J. P. F.; Lozzi, S. P.; Simioni, A. R.; Morais, P. C.; Tedesco, A. C.; Azevedo, R. B. J. Photochem. Photobiol. B: Biol. 2009, 94, 143−146. (b) Vittar, N. B. R.; Prucca, C. G.; Strassert, C.; Awruch, J.; Rivarola, V. A. Int. J. Biochem. Cell Biol. 2008, 40, 2192−2205. (c) Derycke, A. S. L.; Kamuhabwa, A.; Gijsens, A.; Roskams, T.; de Vos, D.; Kasran, A.; Huwyler, J.; Missiaen, L.; de Witte, P. A. M. J Natl. Cancer Inst. 2004, 96, 1620−1630. (d) Qualls, M. M.; Thompson, D. H. Int. J. Cancer 2001, 93, 384–392. (e) Ginevra, F.; Biffanti, S.; Pagnan, A.; Biolo, R.; Reddi, E.; Jori, G. Cancer Lett. 1990, 49, 59−65. (10) (a) Rijcken, C. J. F.; Hofman, J.-W.; van Zeeland, F.; Hennink,

W. E.; van Nostrum, C. F. J. Controlled Release 2007, 124, 144−153. (b) Tailler, J.; Jones, M. C.; Brasseur, N.; van Lier, J. E.; Leroux, J. C. J. Pharm. Sci. 2000, 89, 52−62. (c) Foley, M. S. C.; Beeby, A.; Parker, A. W.; Bishop, S. M.; Phillips, D. J. Photochem. Photobiol. B: Biol. 1997, 38, 18−24.

(11) (a) Weider, M. E.; Hone, D. C.; Cook, M. J.; Handsley, M. M.; Gavrilovic, J.; Russell, D. A. Photochem. Photobiol. Sci. 2006, 5, 727− 734. (b) Ricci-Junior, E.; Marchetti, J. M. Int. J. Pharm. 2006, 310, 187−195. (c) Ricci-Junior, E.; Marchetti, J. M. J. Microencapsul. 2006, 23, 523−538.

(12) (a) Kuznetsova, N. A.; Gretsova, N. S.; Derkacheva, V. M.; Kaliya, O. L.; Lukyanets, E. A. J. Porphyrins Phthalocyanines 2003, 7, 147−154. (b) Cauchon, N.; Tian, H.; Langlois, R.; La Madeleine, C.; Martin, S.; Ali, H.; Hunting, D.; van Lier, J. E. Bioconjug. Chem. 2005, 16, 80−89. (c) Kluson, P.; Drobek, M.; Kalaji, A.; Karaskova, M. Res. Chem. Intermed. 2009, 35, 103−116.

(13) (a) Kaliya, O. L.; Lukyanets, E. A.; Vorozhtsov, G. N. J. Porphyrins Phthalocyanines 1999, 3, 592−610. (b) Liu, W.; Jensen, T. J.; Fronczek, F. R.; Hammer, R. P.; Smith, K. M.; Vicente, M. G. H. J. Med. Chem. 2005, 48, 1033−1041. (c) Verdree, V. T.; Pakhomov, S.; Su, G.; Allen, M. W.; Countryman, A. C.; Hammer, R. P.; Soper, S. A. J. Fluores. 2007, 17, 547−563. (d) Ng, A. C. H.; Li, X.-Y.; Ng, D. K. P. Macromolecules 1999, 32, 5292−5298.

(14) Kuznetsova, N.; Makarov, D.; Yuzhakova, O.; Strizhakov, A.; Roumbal, Y.; Ulanova, L.; Krasnovsky, A.; Kaliya, O. Photochem. Photobiol. Sci. 2009, 8, 1724−1733.

(15) (a) Huang, J. D.; Wang, S. Q.; Lo, P. C.; Fong, W. P.; Ko, W. H.; Ng, D. K. P. New J. Chem. 2004, 28, 348−354. (b) Gijsens, A.; Derycke, A.; Missiaen, L.; de Vos, D.; Huwyler, J.; Eberle, A.; de Witte, P. Int. J. Cancer 2002, 101, 78−85. (c) Wohrle, D.; Muller, S.; Shopova, M.; Mantareva, V.; Spassova, G.; Vietri, F.; Ricchelli, F.; Jori, G. J. Photochem. Photobiol. B: Biol. 1999, 50, 124−128. (d) Egorin, M. J.; Zuhowski, E. G.; Sentz, D. L.; Dobson, J. M.; Callery, P. S.; Eiseman, J. L. Cancer Chemother. Pharmacol. 1999, 44, 283−294.

(e) Allemann, E.; Rousseau, J.; Brasseur, N.; Kudrevich, S. V.; Lewis, K.; van Lier, J. E. Int. J. Cancer 1996, 66, 821−824.

(16) (a) Lo, P.-C.; Chan, C. M. H.; Liu, J.-Y.; Fong, W.-P.; Ng, D. K. P. J. Med. Chem. 2007, 50, 2100−2107. (b) Lo, P.-C.; Wang, S.; Zeug, A.;

Meyer, M.; Roder, B.; Ng, D. K. P. Tetrahed. Lett. 2003, 44, 1967−1970. (17) Ngen, E. J.; Rajaputra, P.; You, Y. Bioorg. Med. Chem. 2009, 17, 6631−6640.

(18) Derkacheva, V. M.; Mikhalenko, S. A.; Solov'eva, L. I.; Alekseeva, V. I.; Marinina, L. E.; Savina, L. P.; Butenin, A. V.; Luk'yanets, E. A. Russ. J. Gen. Chem. 2007, 77, 1117−1125.

(19) Kuznetsova, N.; Makarov, D.; Derkacheva, V.; Savvina, L.; Alekseeva, V.; Marinina, L.; Slivka, L.; Kaliya, O.; Lukyanets, E. J. Photochem. Photobiol. A Chem. 2008, 200, 161−168.

(20) Howe, L.; Zhang, J. Z. J. Phys. Chem. A 1997, 101, 3207−3213.

(21) Schmidt, R.; Afshari, E. J. Phys. Chem. 1990, 94, 4377−4378.

(22) Mak, N. K.; Li, K. M.; Leung, W. N.; Wong, R. N.; Huang, D. P.; Lung, M. L.; Lau, Y. K.; Chang, C. K. Biochem. Pharmacol. 2004, 68, 2387−2396.

(23) Ting, C. M.; Lee, Y. M.; Wong, C. K.; Wong, A. S.; Lung, H. L.; Lung, M. L.; Lo, K. W.; Wong, R. N.; Mak, N. K. Biochem. Pharmacol. 2010, 79, 825−841.

(24) Chan, P. S.; Koon, H. K.; Wu, Z. G.; Wong, R. N.; Lung, M. L.; Chang, C. K.; Mak, N. K. Photochem. Photobiol. 2009, 85, 1207−1217.

(25) Sheik-Bahae, M.; Said, A. A.; Wei, T.-H.; Hagan, D. J.; Van Stryland, E. W. IEEE J. Quantum Electron. 1990, 26, 760−769.

(26) Zhang, J; Wong, K.-L.; Wong, W.-K.; Mak, N.-K.; Kwong, W. J.

D.; Tam, H. L. Org. Biomol. Chem. 2011, DOI: 10.1039/c1ob05415e. (27) Li, Y.; Pritchett, T. M.; Huang, J.; Ke, M.; Shao, P.; Sun, W.

J. Phys. Chem. A 2008, 112, 7200−7207.

(28) SAINT+, version 6.02a; Bruker Analytical X-ray System, Inc., Madison, WI, 1998.

(29) Sheldrick, G. M. SADABS, Empirical Absorption Correction Program; University of Göttingen, Germany, 1997.

(30) Sheldrick, G. M. SHELXTLTM, Reference Manual, version 5.1; Bruker AXS, Inc.: Madison, WI, 1997.

(31) Bauer, H.; Stier, F.; Petry, C.; Knorr, A.; Stadler, C.; Staab, H. A. Eur. J. Org. Chem. 2001, 3255−3278.

(32) Huang, J.-D.; Lo, P.-C.; Chen, Y.-M.; Lai, J. C.; Fong, W.-P.; Ng, D, K. P. J. Inorg. Biochem. 2006, 100, 946−951.

(33) Cammidge, A. N.; Nekelson, F.; Helliwell, M.; Heeney, M. J.; Cook, M. J. J. Am. Chem. Soc. 2005, 127, 16382−16383.

(34) Lo, P.-C.; Huang, J.-D.; Cheng, D. Y. Y.; Chan, E. Y. M.; Fong,

W.-P.; Ko, W.-H.; Ng, D. K. P. Chem.-Eur. J. 2004, 10, 4831-4838. (35) Zhao, Z.; Poon, C.-T.; Wong, W.-K.; Wong, W. Y.; Tam, H. L.; Cheah, K.-W.; Xie, T.; Wang, D. Eur. J. Inorg. Chem. 2008, 119−128.

(36) Mak, N. K.; Li, K. M.; Leung, W. N.; Wong, R. N.; Huang, D. P.; Lung, M. L.; Lau, Y. K.; Chang, C. K. Biochem. Pharmacol. 2004, 68, 2387−2396.

(37) (a) Fredericksen, P. K.; Jorgensen, M.; Ogilby, P. R. J. Am. Chem. Soc. 2001, 123, 1215−1221. (b) Poulsen, T. D.; Fredericksen, P. K.; Jorgensen, M.; Mikkelsen, K. V.; Ogilby, P. R. J. Phys. Chem. A 2001, 105, 11488−11495. (c) Karotki, A.; Dorbizhev, M.; Kruk, M.; Spangler, C.; Nickel, E.; Mamar-Dashvili, N.; Rebane, A. J. Opt. Soc. Am. B 2003, 20, 321−332.

(38) Williams, R. M.; Piston, D. W.; Webb, W. W. FASEB J. 1994, 8, 804−813.

(39) Denk, W.; Strickler, J. H.; Webb, W. W. Science 1990, 248, 73−76. (40) Ogawa, K.; Kobuke, Y. Anti-Cancer Agents Med. Chem. 2008, 8, 269−279.

(41) Xu, C.; Webb, W. W. J. Opt. Soc. Am. B 1996, 13, 481−491.

(42) Poon, C.-T.; Chan, P.-S.; Man, C.; Jiang, F.-L.; Wong, R. N. S.; Mak, N. K.; Kwong, D. W. J.; Tsao, S.-W.; Wong, W.-K. J. Inorg. Biochem. 2010, 104, 62−70.

(43) Saha, S.; Mallick, D; Majumdar, R.; Roy, M.; Dighe, R.; Jemmis, E.; Akhil, R.; Chakravarty, A. Inorg. Chem. 2011, 50, 2975−2987.