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Cobalt-Centered Ten-Vertex Germanium Clusters: The Pentagonal Prism as an Alternative to Polyhedra Predicted by the Wade–Mingos Rules

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Supporting Information

ABSTRACT: One of the most exciting recent (2009) discoveries in metal cluster chemistry is the pentagonal prismatic $Co@Ge_{10}^{3-}$ ion, found in $[K(2,2,2\text{-crypt})]_4[Co@Ge_{10}][Co(1,5-C_8H_{12})_2]$ ·toluene and characterized structurally by X-ray diffraction. The complete absence of triangular faces in the pentagonal prismatic structure of $Co@Ge_{10}^{3-}$ contradicts expectations from the well-established Wade–Mingos rules, which predict polyhedral structures having mainly or entirely triangular faces. A theoretical study on $Co@Ge_{10}^z$ systems (z = -5 to +1) predicts a singlet D_{5h} pentagonal prismatic global minimum for the trianion $Co@Ge_{10}^{3-}$ in accord with this experimental result. Redox reactions on this pentagonal prismatic Co@ Ge_{10}^{3-} trianion generate low-energy pentagonal prismatic structures for $Co@Ge_{10}^z$ where z =



0, -1, -2, -4, and -5 having quartet, triplet, doublet, doublet, and triplet spin states, respectively. Similar theoretical methods predict a singlet $C_{3\nu}$ polyhedral structure for the monoanion $Co@Ge_{10}^-$, similar to previous theoretical predictions on the isoelectronic neutral Ni@Ge_{10} and the structure realized experimentally in the isoelectronic Ni@In_{10}^{10-} found in the K_{10}In_{10}Ni intermetallic. Redox reactions on this $C_{3\nu}$ polyhedral $Co@Ge_{10}^-$ monoanion generate low energy $C_{3\nu}$ polyhedral structures for $Co@Ge_{10}^z$ where z = 0, -2, -3, and -4 having doublet, doublet, triplet, and quartet spin states, respectively.

1. INTRODUCTION

Modern work on bare post-transition element clusters dates back to the 1960s after Corbett and co-workers found ways to obtain crystalline derivatives by use of suitable counterions.^{1,2} This provided for the first time the possibility of definitive elucidation of the structures of such clusters using X-ray crystallography. Such crystalline derivatives of the cluster anions had cryptate or polyamine complexed alkali metals as countercations.¹ Similarly, crystalline derivatives of the cluster cations had counteranions, such as AlCl₄⁻, derived from metal halide strong Lewis acids.²

The information arising from the early structural studies of such clusters suggested an analogy with the polyhedral boranes, the chemistry of which was developing concurrently. The borane dianions $B_n H_n^{2-}$ and isoelectronic carboranes $C_2 B_{n-2} H_n$ having 2n + 2 skeletal electrons ($6 \le n \le 12$) were found to exhibit structures based on deltahedra, that is, polyhedra with all triangular faces. The rules correlating shape of polyhedral boranes with the skeletal electron count were first observed by Williams³ and developed further by Wade^{4,5} and then Mingos.^{6,7} They have subsequently become known as the Wade–Mingos rules.

The Wade–Mingos rules^{4–7} have been applied not only to polyhedral boranes and carboranes but also to metal clusters of various types, including bare post-transition metal clusters. However, limitations in the applications of the Wade–Mingos rules to rationalize the structures of such clusters have become increasingly apparent as these rules were applied to an increasing variety of systems. For example, the icosahedral borane $B_{12}H_{12}^{2-}$ as well as the isoelectronic carboranes $CB_{11}H_{12}^{-}$ and $C_2B_{10}H_{12}$ are all very stable species as predicted by the Wade–Mingos rules for *n*-vertex clusters having 2n + 2 skeletal electrons. However, the isoelectronic icosahedral bare group 14 element clusters E_{12}^{2-} (E = Si, Ge, and Sn) remain essentially unknown. In this case the Nucleus Independent Chemical Shifts (NICS) test for aromaticity developed by Schleyer and collaborators⁸ provides an explanation for this anomaly. Thus using the NICS test, $B_{12}H_{12}^{2-}$ is found to be aromatic but the isoelectronic and still unknown Si_{12}^{2-} is found to be antiaromatic.⁹ Furthermore, for the bare icosahedral clusters E_{12}^{2-} (E = Si, Ge, Sn, Pb), descending the group 14 column of the Periodic Table increases the aromaticity so that the icosahedral lead cluster Pb_{12}^{2-} is found to be aromatic.¹⁰

sahedral lead cluster Pb_{12}^{2-} is found to be aromatic.¹⁰ The Wade–Mingos rules^{4–7} predict cluster structures based on deltahedra with only triangular faces for systems having 2n + 2 skeletal electrons. The predicted polyhedra for electron-rich clusters with *n* vertices having more than 2n + 2 skeletal electrons may have one or more nontriangular faces but most of the faces are still triangles. Thus it was a real surprise when the Ge₁₀ polyhedra in the ten-vertex clusters M@Ge₁₀^{3–} (M = Co,¹¹ Fe¹²) were recently (2009) found to be pentagonal prisms with no triangular faces at all!

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To understand the nature of these very unusual transition metal-centered pentagonal prismatic 10-vertex clusters, the properties of both the interstitial transition metal atom and the surrounding 10-vertex polyhedron need to be considered. Metal clusters containing interstitial transition metal atoms predate the discovery of bare post-transition element clusters with such interstitial atoms by several years. A particularly informative cluster is the icosahedral nickel carbonyl cluster Ni@Ni₁₀[Sb \rightarrow Ni(CO)₃]₂(CO)₁₈⁴⁻, first synthesized by Longoni and co-workers in 1990.¹³ Electron bookkeeping in this cluster counting Ni(CO)₂ vertices as donors of two skeletal electrons like BH vertices in carboranes and Sb→Ni(CO)₃ vertices as donors of three skeletal electrons like CH vertices in carboranes suggests that the interstitial nickel atom is a donor of zero skeletal electrons. This is not unreasonable because a neutral nickel atom has a filled d^{10} shell.¹⁴ By analogy the interstitial neutral d^9 cobalt atom in Co@Ge₁₀³⁻ requires one electron to attain a similar filled d^{10} shell and thus can be considered to be a one-electron acceptor or, alternatively stated, a -1 electron donor. Thus $Co@Ge_{10}^{3-}$ becomes isoelectronic with the 2n + 2 skeletal electron ten-vertex system Ge_{10}^{2-} . Density functional theory¹⁵ as well as the Wade–Mingos rules⁴⁻⁷ predicts a D_{4d} bicapped square antiprism structure for Ge_{10}^{2-} and thus, by isoelectronic analogy, also for Co@Ge₁₀³⁻. Thus the experimental observation of a pentagonal prism rather than a bicapped square antiprism structure for $Co@Ge_{10}^{3-}$ is another example of the limitations of the Wade-Mingos rules. The observation of a pentagonal prismatic structure for Co@ Ge10³⁻ rather than the bicapped square antiprismatic structure predicted by the Wade-Mingos rules may relate to the larger volume of the pentagonal prism, which thus can more readily accommodate the interstitial cobalt atom.

The experimental report of a pentagonal prismatic rather than a bicapped square antiprismatic structure for $\text{Co}@\text{Ge}_{10}^{3-}$ was accompanied by some theoretical studies on $\text{Co}@\text{Ge}_{10}^{3-}$ as well as the isoelectronic $\text{Ni}@\text{Ge}_{10}^{2-}$ and the empty Ge_{10}^{2-} cluster.¹¹ These studies indicate that the D_{5h} pentagonal prism is the ground state structure for $\text{Co}@\text{Ge}_{10}^{3-}$ in accord with the experimental work. However, both these theoretical studies and other concurrently published theoretical studies¹⁶ indicate the D_{4d} bicapped square antiprism to be the ground state for both $\text{Ni}@\text{Ge}_{10}^{2-}$ and the empty Ge_{10}^{2-} . The D_{5h} pentagonal prism structure for $\text{Ni}@\text{Ge}_{10}^{2-}$ is found to lie only ~6 kcal/mol above the global minimum.

Binary Co/Ge clusters have also been observed in the gas phase in laser vaporization experiments.^{17,18} The monoanion $CoGe_{10}^-$ exhibited a particularly strong peak in the mass spectra from such experiments. An endohedral $Co@Ge_{10}^-$ structure was suggested for this species.

In addition to the unexpected observation of pentagonal prismatic structures for the $M@Ge_{10}^{3-}$ clusters (M = Fe,¹² Co¹¹), the ten-vertex post-transition element clusters are of interest because of the variety of polyhedra of distinctive symmetries that have been found experimentally by X-ray crystallography in cluster structures containing interstitial atoms (Figure 1). The D_{4d} bicapped square antiprism is found to encapsulate a post-transition metal atom in the anionic indium cluster $Zn@In_{10}^{8-}$ found in the intermetallic¹⁹ K₈In₁₀Zn as well as the lead clusters $M@Pb_{10}^{2-}$ in $[K(2,2,2-crypt)]_2[M@Pb_{10}]$ (M = Ni, Pd, Pt).^{20,21} A C_{3v} tetracapped trigonal prism is found in the $M@In_{10}^{10-}$ clusters found in the K₁₀In₁₀M intermetallics (M = Ni, Pd, Pt), isoelectronic with $Zn@In_{10}^{8-.22}$ A pentagonal antiprism of bismuth atoms is the host polyhedron for an interstitial



Figure 1. Four ten-vertex polyhedra found as host polyhedra for interstitial metal atoms. For clarity the edges of capping vertices are shown in green.

palladium atom in the cationic bismuth cluster $Pd@Bi_{10}^{4+}$ in $Bi_{14}PdBr_{16}$ (= $[Pd@Bi_{10}][BiBr_4]_4$).²³

The surprising pentagonal prismatic geometry realized experimentally in $Co@Ge_{10}^{3-}$ makes of interest the theoretical study of a wide range of cobalt-centered 10-vertex germanium clusters having various numbers of skeletal electrons. This paper reports a density functional theory study of the cobaltcentered ten-vertex germanium clusters $Co@Ge_{10}^{z}$ with charges ranging from z = -5 to +1. If the interstitial cobalt atom is considered to be a -1 electron donor (i.e., an acceptor of a single electron), then these systems have from 24 to 18 skeletal electrons. These Co@Ge10^z systems includes not only the trianion Co@Ge $_{10}^{3-}$, shown experimentally¹¹ to have pentagonal prismatic geometry, but also the monoanion $\text{Co}@\text{Ge}_{10}^{-}$ observed in the gas phase in laser vaporization experiments^{17,18} and isoelectronic with Ni@In₁₀¹⁰⁻ and Zn@In₁₀⁸⁻ found in the intermetallics K₁₀In₁₀Ni²² and K₈In₁₀Zn,¹⁹ respectively. The current paper thus extends significantly the theoretical studies limited to the D_{5h} , D_{5d} , and D_{4d} structures of Co@Ge₁₀³⁻ and the essentially isoelectronic Ni $@Ge_{10}^{2-}$ and Ge_{10}^{2-} included in the original report by Fässler and collaborators¹¹ on the synthesis and structural characterization of Co@Ge103-. In particular, the earlier authors did not consider $Co@Ge_{10}^{z}$ species having charges other than -3 nor any possible structures with 3-fold symmetry, such as $C_{3\nu}$. We find $C_{3\nu}$ structures to be low energy structures for many of the Co@Ge₁₀^z species, including particularly singlet Co@Ge₁₀⁻.

2. THEORETICAL METHODS

Geometry optimizations were carried out at the hybrid DFT B3LYP level $^{24-27}$ with the 6-31G(d) (valence) double- ζ quality basis functions extended by adding one set of polarization (d) functions for both the interstitial and germanium atoms. The Gaussian 09 package of programs²⁸ was used in which the fine grid (75,302) is the default for numerically evaluating the integrals and the tight (10^{-8}) hartree stands as default for the self-consistent field convergence. Computations were carried out using six initial geometries including ten-vertex polyhedra with 3-fold, 4-fold, and 5-fold symmetry including the four polyhedra in Figure 1 as well as a prolate $C_{3\nu}$ structure and the D_{4h} bicapped cube. The symmetries were maintained during the initial geometry optimization processes. Symmetry breaking using modes defined by imaginary vibrational frequencies was then used to determine optimized structures with minimum energies. Spin states from singlets to quintets were considered. Vibrational analyses show that all of the final optimized structures discussed in this paper are genuine minima at the B3LYP/6-31G(d) level without any significant imaginary frequencies

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 $(N_{imag} = 0)$. In a few cases the calculations ended with acceptable small imaginary frequencies²⁹ and these values are indicated in the corresponding figures.

The optimized structures found for the $Co@Ge_{10}^{z}$ derivatives are labeled by the number of skeletal electrons and their relative energies. In determining the numbers of skeletal electrons the interstitial cobalt atom is assumed to attain the d¹⁰ closed-shell configuration by being a one-electron acceptor, that is, a -1 skeletal electron donor. The germanium vertices are assumed to be donors of two skeletal electrons each as suggested by the Wade-Mingos rules.4-7 Spin states are indicated by S, D, T, Q, and P for singlet, doublet, triplet, quartet, and quintet, respectively. Thus the lowest energy structure of the singlet trianion $Co@Ge_{10}^{3-}$ is labeled 22–18. Additional details of all of the optimized structures, including all interatomic distances, the initial geometries leading to a given optimized structure, and structures with energies too high to be of possible chemical relevance are provided in the Supporting Information. Only structures within 25 kcal/mol of the global minima are considered. In assigning polyhedra to the optimized structures, the Ge-Ge distances less than ~3.2 Å were normally considered as polyhedral edges; significant exceptions are noted in the text. Similarly Co-Ge distances less than ~2.8 Å are considered bonding distances; most such CoGe bonding distances were less than \sim 2.5 Å except for those some of the less regular polyhedral structures.

3. RESULTS

3.1. Neutral Co@Ge₁₀. Three structures were found for the neutral Co@Ge₁₀ within 25 kcal/mol of the global minimum (Figure 2). The global minimum is a doublet $C_{3\nu}$ structure



Figure 2. Three structures for the neutral $Co@Ge_{10}$ within 25 kcal/ mol of the global minimum.

19–1D. A quartet pentagonal prismatic structure 19–2Q lies 5.6 kcal/mol above this global minimum. A doublet distorted pentagonal prismatic structure 19–3D is also found at 10.3 kcal/mol above 19–1D.

3.2. Anionic Co@Ge₁₀^{*z*} Derivatives (z = -1 to -5). Four structures were found for the monoanion Co@Ge₁₀⁻ within 25 kcal/mol of the global minimum (Figure 3). The lowest energy



Figure 3. Four structures for the monoanion $Co@Ge_{10}^{-}$ within 25 kcal/mol of the global minimum.

structure is the singlet $C_{3\nu}$ structure **20–1S** based on a polyhedron similar to the isoelectronic In_{10}^{10-} polyhedron found experimentally²² in the intermetallic $K_{10}In_{10}Ni$ and also predicted theoretically¹⁶ for the likewise isoelectronic neutral Ni@Ge₁₀. The next highest energy Co@Ge₁₀⁻ structure is the triplet D_{5h} pentagonal prismatic structure **20–2T**, already lying 16.5 kcal/mol above **20–1S**. The large energy difference between **20–1S** and **20–2T** suggests that **20–1S** is a highly

favorable structure. Two higher energy distorted $\text{Co}@\text{Ge}_{10}^{-1}$ structures are also found, namely, singlet 20–38, lying 19.2 kcal/mol above 20–18 and derived from a pentagonal prism, and triplet 20–4T, lying 21.8 kcal/mol above 20–18.

Three structures were found for the dianion $Co@Ge_{10}^{2-}$ within 25 kcal/mol of the global minimum (Figure 4). The



Figure 4. Three structures for the dianion $Co@Ge_{10}^{2-}$ within 25 kcal/ mol of the global minimum.

lowest energy structure by only 0.2 kcal/mol is the doublet C_s structure **21–1D** with a polyhedron similar to the **20–1S** global minimum of the monoanion Co@Ge₁₀⁻ (Figure 3). The doublet dianion structure **21–2D** is a slightly distorted pentagonal prism lying only 0.2 kcal/mol above **21–1D** (Figure 4). A $C_{3\nu}$ quartet structure **21–3Q** for the dianion Co@Ge₁₀^{2–} lies at a considerably higher energy, namely, 24.9 kcal/mol above **21–1D**.

Three structures were found for the trianion $\text{Co}@\text{Ge}_{10}^{3-}$ within 25 kcal/mol of the global minimum 22–18 (Figure 5).



Figure 5. Three structures for the trianion $\text{Co}@\text{Ge}_{10}^{3-}$ within 25 kcal/ mol of the global minimum.

The singlet spin state pentagonal prism $\text{Co}@\text{Ge}_{10}^{3-}$ structure **22–1S** is found experimentally in $[K(2,2,2\text{-crypt})]_4[\text{Co}@\text{Ge}_{10}][\text{Co}(1,5\text{-}\text{C}_8\text{H}_{12})_2]$ ·toluene¹¹ and lies ~17 kcal/mol in energy below any other of the $\text{Co}@\text{Ge}_{10}^{3-}$ structures. The predicted length of the "vertical" edges of the pentagonal prism in **22–1S** is 2.59 Å as compared with the experimental value of 2.66 Å. Similarly, the predicted length of the "horizontal" edges of the pentagonal prism in **22–1S** (i.e., the edges of the pentagonal faces) is 2.50 Å as compared with the experimental value of 2.51 Å.

The next $Co@Ge_{10}^{3-}$ structure in terms of energy, namely, the triplet structure **22–2T**, lies ~17 kcal/mol above the pentagonal prism global minimum **22–1S** suggesting that the experimentally observed **22–1S** is a highly favored structure. Structure **22–2T** (Figure 5) is a triplet spin state $C_{3\nu}$ polyhedron similar to the polyhedra in the singlet **20–1S** monoanion global minimum (Figure 3) and the doublet Co@ Ge_{10}^{2-} dianion global minimum structure **21–1D** (Figure 4). A distorted $C_{3\nu}$ singlet structure for the Co@Ge₁₀³⁻ trianion **22–3S** is also found at 18.0 kcal/mol above the global minimum **22–1S**, similar to the triplet monoanion structure **20–4T** (Figure 3).

Four structures were found for the tetraanion $\text{Co}@\text{Ge}_{10}^{4-}$ within 25 kcal/mol of the global minimum (Figure 6). The



Figure 6. Four structures for the tetraanion $Co@Ge_{10}^{4-}$ within 25 kcal/mol of the global minimum.

lowest energy of these structures, namely, doublet **23–1D**, is a slightly distorted pentagonal prism. This structure is reasonably favorable since the next $Co@Ge_{10}^{4-}$ structure, namely, doublet **23–2D**, lies 11 kcal/mol above **23–1D**. Structure **23–2D** is a slightly distorted $C_{3\nu}$ polyhedron similar to the $C_{3\nu}$ polyhedron found in singlet **22–3S**.

The next $\text{Co}@\text{Ge}_{10}^{4-}$ tetraanion structure in terms of energy, namely 23–3D at 15.1 kcal/mol above 23–1D (Figure 6), is derived from the D_{4d} bicapped square antiprism but is so highly distorted that its origin is hardly recognizable. This suggests that the cavity in a bicapped square antiprism, or actually that in the underlying square antiprism, is not large enough to accommodate an interstitial cobalt atom.

The lowest lying quartet structure for the $Co@Ge_{10}^{4-}$ tetraanion, namely, **23–4Q** at 16.2 kcal/mol in energy above **23–1D** (Figure 6), is an undistorted $C_{3\nu}$ polyhedron similar to the $C_{3\nu}$ polyhedron found in the global minima for the Co@ Ge_{10}^{-} monoanion (singlet **20–1S** in Figure 3) and the Co@ Ge_{10}^{2-} dianion (doublet **21–1D** in Figure 4).

The potential energy surface of the $Co@Ge_{10}^{5-}$ pentaanion is more complicated since eight structures are found within 25 kcal/mol of the global minimum **24–1T** (Figure 7). These



Figure 7. Eight structures for the pentaanion $\text{Co}@\text{Ge}_{10}^{5-}$ within 25 kcal/mol of the global minimum.

eight structures include five singlet structures, two triplet structures, and a single quintet structure.

The lowest energy $Co@Ge_{10}^{5-}$ structure is the D_{5h} triplet pentagonal prism **24–1T** (Figure 7). The other triplet Co@ Ge_{10}^{5-} structure is the C_s polyhedral structure **24–6T** lying 12.4 kcal/mol above **24–1T**. The Ge_{10} polyhedron in **24–6T** is a relatively open polyhedron with six quadrilateral faces and four triangular faces corresponding to six degree 4 vertices and four degree 3 vertices. The lone quintet $Co@Ge_{10}^{5-}$ structure 24-7P, lying 15.7 kcal/mol above 24-1T, is a distorted pentagonal prism.

The five singlet Co@Ge₁₀⁵⁻ structures are all distorted from ideal polyhedra (Figure 7). The lowest energy singlet structure 24-2S lies 3.2 kcal/mol above the global minimum 24-1T and is a distorted pentagonal prism. The next singlet Co@Ge₁₀⁵⁻ structure 24-3S lies 4.3 kcal/mol above 24-1T. Structure 24-**3S** is derived from a $C_{3\nu}$ polyhedron distorted to C_s symmetry. In the singlet structure 24-4S, lying 9.1 kcal/mol above 24-1T, the Ge₁₀ polyhedron has split into two Ge₅ units. Thus structure 24-4S can be considered to be two octahedra sharing a vertex. The shared vertex is the cobalt atom. The singlet Co@ Ge_{10}^{5-} structure 24–5S, lying 11.2 kcal/mol above 24–1S, has a $C_{2\nu}$ Ge₁₀ polyhedron very similar to the C_s Ge₁₀ polyhedron in the triplet structure 24-6T discussed above. The highest energy singlet Co@Ge₁₀⁵⁻ structure, namely, **24–8S** lying 17.3 kcal/mol above **24–1T**, has a $C_{2\nu}$ Ge₁₀ polyhedron which is more closed than that in 24-5S. The Ge₁₀ polyhedron in 24-8S has two quadrilateral faces and 12 triangular faces corresponding to four degree 5 vertices and six degree 4 vertices.

3.3. Cation Co@Ge₁₀⁺. Four structures were found for the Co@Ge₁₀⁺ cation within 25 kcal/mol of the global minimum (Figure 8). The lowest energy Co@Ge₁₀⁺ structure is the



Figure 8. Four structures for the cation $Co@Ge_{10}^+$ within 25 kcal/mol of the global minimum.

singlet C_s distorted pentagonal prism 18–1S. The next Co@ Ge_{10}^+ structure is the singlet $C_{3\nu}$ polyhedral structure 18–2S, lying 11.7 kcal/mol above 18–1S. The final singlet Co@Ge_{10}^+ structure is the highly distorted polyhedral structure 18–3S, lying 11.9 kcal/mol above 18–1S. A single triplet structure, namely, the D_{4d} bicapped square antiprismatic structure 18–4T, was found at 23.2 kcal/mol above 18–1S with a small imaginary vibrational frequency at 20*i* cm⁻¹. Structure 18–4T is the only Co@Ge_{10}^{z} (z = -5 to +1) structure found based on this D_{4d} deltahedron within 25 kcal/mol of a global minimum.

4. DISCUSSION

Four 10-vertex polyhedra, namely, the D_{4d} bicapped square antiprism, the $C_{3\nu}$ tetracapped trigonal prism, the D_{5d} pentagonal antiprism, and the D_{5h} pentagonal prism (Figure 1), have been found in centered 10-vertex clusters that have been synthesized and structurally characterized by X-ray crystallography. Among these four polyhedra only the tetracapped trigonal prism and pentagonal prism are found in undistorted form in several of the low-energy Co@Ge₁₀^z structures (z = -5to +1). The bicapped square antiprism is found only in the relatively high energy triplet $Co@Ge_{10}^+$ structure 18-4T (Figure 8). No low energy $Co@Ge_{10}^{z}$ structures with pentagonal antiprismatic geometry were found. In general attempted optimizations of $Co@Ge_{10}^z$ structures of various spin multiplicities starting with polyhedra other than the tetracapped trigonal prism and pentagonal prism lead to structures at very high energies, highly distorted low energy

structures, or structures with pentagonal prismatic or tetracapped trigonal prismatic geometry.

A singlet pentagonal prismatic structure **22–1S** (Figure 5) was found to be the global minimum for the trianion Co@ Ge_{10}^{3-} in accord with the observation of this structure by X-ray crystallography in $[K(2,2,2-crypt)]_4[Co@Ge_{10}][Co(1,5-C_8H_{12})_2]$ ·toluene.¹¹ This structure is obviously a very favorable "closed shell" structure since it lies ~17 kcal/mol in energy below the next lowest energy Co@Ge₁₀³⁻ structure, namely, the triplet **22–2T** (Figure 5). Addition of an electron to the Co@Ge₁₀³⁻ structure **23–1D** for Co@Ge₁₀⁴⁻ (Figure 6). This structure can be considered formally as a radical tetraanion. Further addition of an electron to the tetraanion **23–1D** gives the triplet pentaanion Co@Ge₁₀⁵⁻ structure **24–1T** (Figure 7). Both **23–1D** and **24–1T** are global minima indicating the generally high stability of pentagonal prismatic Co@Ge₁₀^z

Higher spin state pentagonal prismatic $Co@Ge_{10}^{z}$ structures can also be obtained by removal of electrons from the singlet stable trianion Co@Ge $_{10}^{3-}$ structure 22–1S (Figure 5). Thus removal of one electron from 22-1S gives the doublet Co@ $\operatorname{Ge}_{10}^{2-}$ dianion pentagonal prismatic structure 21–2D, which lies within 0.2 kcal/mol of the global minimum. The doublet pentagonal prismatic structure 21-2D for the dianion Co@ $\operatorname{Ge_{10}}^{2-}$ is isoelectronic with the known pentagonal prismatic iron-centered ten-vertex germanium polyhedral anion Fe@ Ge₁₀³⁻, which has been synthesized and characterized by X-ray crystallography.¹² Further successive removal of electrons from the $\text{Co}@\text{Ge}_{10}^{2-}$ dianion structure **21–2D** gives successively low-energy pentagonal prismatic structures for the Co@Ge₁₀ anion (triplet 20-2T in Figure 3) and for neutral Co@Ge₁₀ (quartet 19-2Q in Figure 2). The latter two structures are not global minima but lie relatively close to $C_{3\nu}$ tetracapped trigonal prismatic global minima. The existence of low energy pentagonal prismatic Co@Ge10^z structures for the six charges (z) from 0 to -5 suggests that the redox chemistry of the known Co@Ge₁₀³⁻ is likely to be very rich. However, meaningful electrochemical studies of Co@Ge₁₀³⁻ would probably require generation of this trianion in the absence of other potentially redox active species. Thus, electrochemical studies on the known derivative $[K(2,2,2-crypt)]_4[Co@Ge_{10}][Co(1,5-crypt)]_4]$ C_8H_{12} $)_2$ ·toluene are likely to be complicated by the redox chemistry of the $Co(1,5-C_8H_{12})_2^-$ anion also present in this derivative.

A central question in this work is why the D_{5h} pentagonal prism structure is strongly preferred over the D_{4d} bicapped square antiprism structure for Co@Ge10³⁻. Thus the species Co@Ge10³⁻ has 22 skeletal electrons if the interstitial cobalt atom is considered to be a -1 electron donor (i.e., an oneelectron acceptor). The Wade-Mingos rules⁴⁻⁷ would suggest a D_{4d} bicapped square antiprism structure for Co@Ge₁₀³⁻ similar to the experimental structure for $B_{10}H_{10}^{2-}$ or structures predicted theoretically for Ge_{10}^{2-} and $\text{Ni}(@\text{Ge}_{10}^{2-})$. However, the internal volume of a polyhedron having a given number of vertices is maximized by minimizing the number of polyhedral edges. For example, polyhedra having all degree 3 vertices, such as the prisms, maximize the internal volumes for a given even number of vertices. We suggest that the larger internal volume of a pentagonal prism relative to a bicapped square antiprism makes the pentagonal prism the preferred Ge₁₀ polyhedron to encapsulate a cobalt atom in a 22 skeletal electron system. This suggestion is supported by our previous theoretical studies¹⁶ on

the related 22 skeletal electron systems $M@Ge_{10}^{2-}$ (M = Ni, Pd, Pt). These theoretical studies were completed and submitted for publication before the first experimental publication on $Co@Ge_{10}^{3-}$ in 2009.¹¹ For the nickel derivative the D_{4d} bicapped square antiprism is found to be the global minimum with the D_{5h} Ni@Ge_{10}^{2-} pentagonal prism lying only 5.7 kcal/mol above this global minimum. This result is essentially confirmed by Fässler and collaborators¹¹ in their independent density functional theory work using a different basis set. However, for $M@Ge_{10}^{2-}$ with the larger interstitial palladium and platinum atoms, our work¹⁶ shows that the D_{5h} pentagonal prism is the lowest energy structure.¹⁶ For Pd@Ge_{10}^{2-} the D_{4d} bicapped square antiprism is found to lie ~13 kcal/mol in energy above the D_{5h} pentagonal prism. For Pt@Ge_{10}^{2-} with an even larger interstitial platinum atom, no low-energy D_{4d} bicapped square antiprismatic structure is even found.

The geometry of a regular pentagonal prism can be characterized by the ratio v/h between the lengths of the vertical edges shared by two rectangular faces (v) and the horizontal edges (h) of the pentagonal faces (Table 1). This v/h ratio is

Table 1. Dimensions of Pentagonal Prismatic $Co@Ge_{10}^{z}$ Clusters^{*a*}



seen to be near unity for all of the low energy pentagonal prismatic $Co@Ge_{10}^z$ structures (z = 0 to -5) having from 19 to 24 skeletal electrons. Thus the rectangular faces in all of these pentagonal prisms are squares or almost squares. The higher energy distorted $Co@Ge_{10}^z$ structures 19-3D (Figure 2) and 20-3S (Figure 3) still recognizable as irregular pentagonal prisms are also included in Table 1. These structures represent lower spin state distorted versions of the regular pentagonal prismatic structures 19-2Q and 20-2T, where the distortion appears to be related to the Jahn–Teller effect. To determine effective v/h values for these distorted structures the non-equivalent vertical and horizontal edge lengths were first averaged to find effective values of v and h. These distorted pentagonal prismatic $Co@Ge_{10}^z$ structures also have effective v/h ratios close to unity.

The other type of 10-vertex polyhedra found frequently in low-energy $Co@Ge_{10}^{z}$ structures are polyhedra derived from the tetracapped trigonal prism (Figure 1). The four capping vertices in this polyhedron are of two types, namely, three vertices capping the three rectangular faces of the underlying trigonal prism and a fourth vertex capping one of the triangular

Table 2. Dimensions of Co@Ge₁₀^z Clusters with Ge₁₀ Polyhedra Derived from the Tetracapped Trigonal Prism^a

axial vertex

cluster	ν	h_c	h_u	Δh	\overline{h}	ν/\overline{h}	M@E ₁₀ distances
$Co@Ge_{10}^+$ (18–2S)	2.77	3.53	3.17	0.36	3.35	0.83	2.31(3),2.45(3),2.52(3),2.64(1)
Co@Ge ₁₀ (19–1D)	2.88	3.54	2.87	0.67	3.20	0.90	2.35(3),2.36(3),2.49(3),2.62(1)
$Co@Ge_{10}^{-}$ (20–18)	2.99	3.62	2.67	0.95	3.14	0.95	2.35(3),2.40(3),2.42(3),2.44(1)
Ni@Ge ₁₀ (singlet)	3.00	3.62	2.70	0.92	3.16	0.95	2.36(3),2.41(3),2.43(3),2.46(1)
Ni@In ₁₀ ¹⁰⁻ (expt.)	~3.3	~4.3	~3.2	~1.1	~3.75	0.88	2.8(9),2.7(1)
$Co@Ge_{10}^{-} (20-4T)^{*}$	3.02	3.75	2.65	1.10	3.20	0.94	2.38(3),2.45(3),2.49(3),2.37(1)
$Co@Ge_{10}^{2-} (21-1D)^*$	3.02	3.69	2.73	0.96	3.21	0.94	2.38(3),2.40(3),2.44(3),2.47(1)
$Co@Ge_{10}^{2-}$ (21–3Q)	2.91	3.74	2.87	0.87	3.31	0.79	2.33(3),2.48(3),2.50(3),2.53(1)
$Co@Ge_{10}^{3-}$ (22–2T)	3.08	3.75	2.66	0.89	3.11	0.99	2.41(3),2.44(3),2.45(3),2.43(1)
$Co@Ge_{10}^{4-}$ (23–4Q)	3.31	3.56	2.62	0.94	3.09	1.07	2.39(3),2.44(3),2.47(3),2.83(1)
$Co@Ge_{10}^{5-}$ (24–38)*	3.58	3.18	2.72	0.46	2.95	1.21	2.32(3),2.46(3),2.48(3),3.50(1)
^a Distances in Å. Structures dis	storted signific	cantly from C	symmetry :	are starred.			

faces of the underlying trigonal prism. This capping pattern reduces the D_{3h} symmetry of the underlying trigonal prism to $C_{3\nu}$ symmetry. This symmetry point group partitions the ten vertices of the tetracapped trigonal prism into the following four sets:

- The three vertices of the uncapped triangular face of the underlying trigonal prism;
- (2) The three vertices capping the rectangular faces of the underlying trigonal prism;
- (3) The three vertices of the capped triangular face of the underlying trigonal prism;
- (4) The unique vertex capping a triangular face of the underlying trigonal prism.

The edges of the tetracapped trigonal prism can be partitioned into six types. Three of these types of edges correspond to edges of the underlying trigonal prism whereas the other three types of edges are connected to the capping atoms. The edge lengths of the underlying trigonal prism can be used to characterize the geometries of the tetracapped trigonal prismatic structures encountered in this work. These edge lengths are as follows (Table 2):

- The lengths v of the vertical edges of the trigonal prism, i.e., the edges shared by two rectangular faces of the underlying trigonal prism;
- The lengths *h_c* of the edges of the capped triangular face of the underlying trigonal prism;
- (3) The lengths h_u of the edges of the uncapped triangular face of the underlying trigonal prism.

Table 2 lists these edge lengths of the $Co@Ge_{10}^{z}$ structures of C_{3v} symmetry derived from the tetracapped trigonal prism. In general the capping of only one of the two triangular faces makes the lengths h_c and h_u very different so that their difference $h_c - h_u = \Delta h$ can be used to measure the distortion of the underlying trigonal prism from its original D_{3h} symmetry to the C_{3v} structure with a cap at one end. The unique vertex capping the triangular face of the underlying trigonal prism is conveniently called the axial vertex in these structures since it is the only one of the ten vertices on the C_3 axis. Table 2 also lists the mean v, h_c and h_u edge lengths for $Co@Ge_{10}^{z}$ structures obviously derived from the $C_{3\nu}$ tetracapped trigonal prism but distorted severely from ideal $C_{3\nu}$ symmetry. These distorted structures are starred in Table 2.

An ideal C_{3v} tetracapped trigonal prism, like the D_{4d} bicapped square antiprism, is a deltahedron having 16 triangular faces (Figure 1). However, in ten-vertex germanium clusters two types of distortion of the ideal $C_{3\nu}$ tetracapped trigonal prism are observed without breaking the $C_{3\nu}$ symmetry. In such types of distortion one set of three symmetry-equivalent edges is elongated beyond "bonding distance" so that the resulting polyhedron is best considered having three quadrilateral faces and ten triangular faces. In the oblate or flattened $C_{3\nu}$ polyhedra, which are exhibited by all of the lowest energy Co@ Ge_{10}^{z} structures with C_{3v} symmetry, the axial vertex is pushed toward the center of the polyhedron. This brings all 10 germanium atoms within "bonding distance" of the cobalt atom, that is, less than 3.0 Å (typically less than 2.6 Å in most cases). This lengthens the edges of the capped triangular face of the underlying trigonal prism so that h_c becomes greater than 3.5 Å. The other possible $C_{3\nu}$ polyhedron is the prolate or elongated polyhedron where the axial vertex remains beyond bonding distance to the central cobalt atom. In this case, the vertical edges of the underlying trigonal prism are elongated so that v now becomes greater than 3.5 Å. The only $Co@Ge_{10}^{2}$ structure of this type that was found in this work is the distorted Co@Ge₁₀⁵⁻ structure 24–3S, in which the average length of the vertical edges (v) is 3.58 Å and the interstitial cobalt atom lies 3.50 Å from the axial germanium atom. Both the oblate and prolate $C_{3\nu}$ polyhedral structures were found in a recent theoretical study³⁰ on the beryllium-centered cluster dianion Be@Ge₁₀²⁻.

The monoanion $Co@Ge_{10}^{-}$ appears to be the species where $C_{3\nu}$ geometry is the highly favored structure since the singlet **20–1S** lies more than 16 kcal/mol in energy below any of the other $Co@Ge_{10}^{-}$ structures found in this work (Figure 3). This monoanion is known only in the gas phase and has not been characterized structurally.^{17,18} However, this monoanion could arise from the two-electron oxidation of the known pentagonal prismatic trianion $Co@Ge_{10}^{3-}$. Such an oxidation process would necessarily be a very delicate process since the cobalt

atoms in these clusters have a much lower formal oxidation state than in normal stable inorganic cobalt compounds so that "overoxidation" to cobalt(II) or even cobalt(III) is a distinct problem with most oxidizing agents. The rearrangement of the pentagonal prismatic structure of the trianion $Co@Ge_{10}^{-3}$ upon two-electron oxidation to the monoanion $Co@Ge_{10}^{-}$ is a distinct possibility since for the monoanion the $C_{3\nu}$ polyhedral structure is predicted to lie 16.5 kcal/mol below the D_{5h} pentagonal prismatic structure. However, such a polyhedral rearrangement is predicted to require a change in spin state from the triplet pentagonal prismatic $Co@Ge_{10}^{-}$ (20–2T) to the singlet $C_{3\nu}$ polyhedral $Co@Ge_{10}^{-}$ (20–1S).

The lowest-energy $C_{3\nu}$ structure 20–18 predicted for the monoanion $Co@Ge_{10}^{-}$ can be related to theoretical and experimental studies on isoelectronic species. Thus the monoanion $Co@Ge_{10}^{-}$ is isoelectronic with neutral Ni@Ge_{10}, which is shown in previous theoretical work¹⁶ to have a $C_{3\nu}$ global minimum. The values for v, h_{c} , h_{w} , Δh , \overline{h} , and v/\overline{h} as well as the 10 M-Ge distances for the isoelectronic species Co@Ge10 and Ni@Ge10 are essentially identical (within 0.03 Å and less in most cases). This comparison can be extended to the experimentally realized and structurally characterized species Ni@In10¹⁰⁻ found in the intermetallic K10In10Ni.²² The larger size of the indium atoms in Ni@In₁₀¹⁰⁻¹ relative to the germanium atoms in Co@Ge10⁻ make the In-In and Ni-In distances in the former necessarily larger than the Ge-Ge and Co-Ge distances in the latter so that they are not necessarily directly comparable. However, the v/\overline{h} value of 0.88 found experimentally in Ni@In₁₀¹⁰⁻ is reasonably close to the ν/\bar{h} values of 0.95 predicted for both Co@Ge₁₀⁻ and Ni@Ge₁₀ when the differences in cluster atoms, cluster environments, and cluster charges are considered.

5. SUMMARY

The lowest energy singlet $\text{Co}@\text{Ge}_{10}^z$ structures that are predicted to be particularly favorable energetically are the pentagonal prismatic $\text{Co}@\text{Ge}_{10}^{3-}$ trianion structure (**22–1S** in Figure 5) and the $C_{3\nu}$ oblate polyhedral $\text{Co}@\text{Ge}_{10}^-$ monoanion structure derived from a tetracapped trigonal prism (**20–1S** in Figure 3). The pentagonal prismatic $\text{Co}@\text{Ge}_{10}^{3-}$ structure **22– 1S** has been realized experimentally in $[\text{K}(2,2,2\text{-crypt})]_4[\text{Co}@\text{Ge}_{10}][\text{Co}(1,5\text{-}C_8\text{H}_{12})_2]$ -toluene, characterized structurally by X-ray crystallography.¹¹ The $C_{3\nu}$ polyhedral $\text{Co}@\text{Ge}_{10}^-$ structure has not been realized experimentally. However, analogous isoelectronic structures have been found theoretically¹⁶ for the neutral Ni@Ge_{10} and experimentally²² for the Ni@In₁₀¹⁰⁻ anion in the intermetallic K₁₀In₁₀Ni, characterized by X-ray crystallography.

The lowest energy structures for the remaining $\text{Co}@\text{Ge}_{10}^{z}$ species (z = -5, -4, -2, 0, and +1) are derived by addition or removal of electrons from these favorable $\text{Co}@\text{Ge}_{10}^{3-}$ and $\text{Co}@\text{Ge}_{10}^{-}$ structures. In general, these structures have one or more unpaired electrons. Thus low-energy pentagonal prismatic structures are found for $\text{Co}@\text{Ge}_{10}^{z}$ where z = 0, -1, -2, -4, and -5 with quartet, triplet, doublet, doublet, and triplet spin states, respectively. Similarly, low-energy $C_{3\nu}$ polyhedral structures are found for $\text{Co}@\text{Ge}_{10}^{z}$ where z = 0, -2, -3, and -4 with doublet, doublet, triplet, and quartet spin states, respectively.

ASSOCIATED CONTENT

S Supporting Information

Tables of distances; complete Gaussian reference (reference 28). This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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