

Spin Reorientation in the Square-Lattice Antiferromagnets RMnAsO (R = Ce, Nd): Density Functional Analysis of the Spin-Exchange Interactions between the Rare-Earth and Transition-Metal Ions

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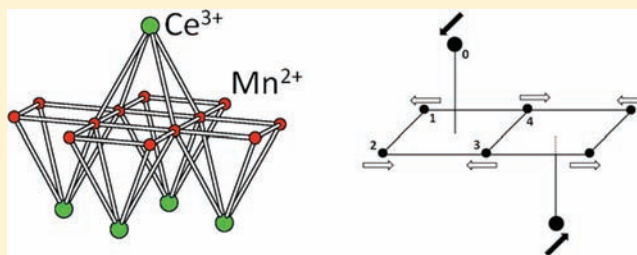
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ABSTRACT: The spin reorientation (SR) phenomenon of the square-lattice antiferromagnets RMnAsO (R = Ce, Nd) was investigated by analyzing the spin-exchange interactions between the rare-earth and the transition-metal ions (R³⁺ and Mn²⁺, respectively) on the basis of density functional calculations. It is found that the symmetry and strength of the Dzyaloshinskii–Moriya (DM) interaction are determined primarily by the partially filled 4f states of the R³⁺ ions and that the DM and biquadratic (BQ) exchanges between the R³⁺ and the Mn²⁺ ions are unusually strong and control the observed spin reorientation phenomenon. Below their SR temperature, the Mn²⁺ and Ce³⁺ moments are orthogonal in CeMnAsO but are collinear in NdMnAsO, because the DM interaction dominates over the BQ interaction for CeMnAsO, while the opposite is the case for NdMnAsO. Experiments designed to test the implications of our findings are proposed.



1. INTRODUCTION

The energy spectrum of a magnetic solid consisting of ions with unpaired spins is commonly described by considering the isotropic bilinear (Heisenberg), the antisymmetric (Dzyaloshinskii–Moriya), and anisotropic exchange between various pairs of its spin sites,^{1,2} although some inadequacy of this approach has been pointed out in the studies of La₂CuO₄.³ For magnetic ions with spin $S > 1/2$, it may become necessary to rectify the inadequacy of using the Heisenberg exchange $\sum_{i<j} J_{ij} \vec{S}_i \cdot \vec{S}_j$ by including the biquadratic (BQ) exchange $\sum_{i<j} K_{ij} (\vec{S}_i \cdot \vec{S}_j)^2$.^{4–6}

The Dzyaloshinskii–Moriya (DM) exchange $\sum_{i<j} \vec{D}_{ij} \cdot (\vec{S}_i \times \vec{S}_j)$ is responsible for a variety of interesting magnetic phenomena such as spin canting in antiferromagnets,^{2,7} ferroelectric polarization induced by chiral magnetic order,^{8,9} and magnon quantum Hall effect,¹⁰ to name a few. In general, the strength of the DM exchange is often estimated by $|D_{ij}/J_{ij}| \approx \Delta g/g$, where Δg is the contribution of the orbital moment to the g factor^{11,12} and is commonly believed to be at most 1/10 of the Heisenberg exchange. Similarly, the strength of the anisotropic exchange is estimated by $(\Delta g/g)^2$ and is at most 1/10 of the DM exchange.^{11,12}

Exchange interactions between transition-metal (TM) and rare-earth (RE) ions have been less extensively studied than

those between TM ions.¹³ Thus far, studies on the DM exchange have been mostly concerned with interactions among identical TM ions. Consequently, the DM exchange between TM and RE ions is not well understood. The anisotropic exchange^{1,11} between RE and TM ions has been examined in connection with the spin reorientation (SR) phenomenon, for example, involving the Nd³⁺ (f^7) and Cu²⁺ (d^9) moments in the square-lattice antiferromagnet Nd₂CuO₄,^{14,15} in which two Nd³⁺ ions are located above and below every square of Cu²⁺ ions forming a Cu₄Nd₂ square bipyramid (Figure 1a). In the ordered magnetic states of Nd₂CuO₄ (Figure 1b),¹⁴ in which the Cu²⁺ ions form a square antiferromagnetic lattice, the net Heisenberg exchange between the Nd³⁺ and the Cu²⁺ ions vanishes. The DM exchange between them also vanishes because the DM interaction of one Nd³⁺ ion lying above each Cu₄ square is canceled by that of the other Nd³⁺ ion located below the Cu₄ square. A nonzero DM exchange would have resulted if the two Nd³⁺ ions lying above and below each Cu₄ square had opposite spin directions. Thus, the anisotropic exchange^{1,11} is left as the only remaining exchange, provided that the BQ exchange is negligible.

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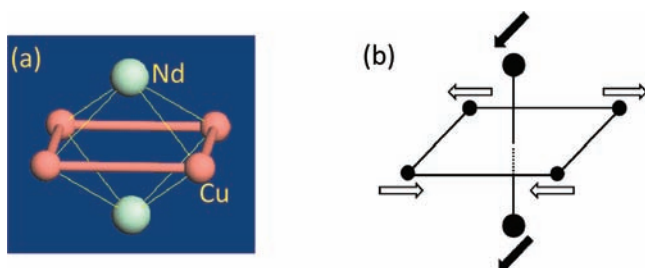


Figure 1. (a) Arrangement of the Nd^{3+} ions above and below the square lattice of Cu^{2+} ions in Nd_2CuO_4 , where the Nd^{3+} and Cu^{2+} ions are represented by white and pink spheres, respectively. Two Nd atoms located above and below each Cu_4 square form a Cu_4Nd_2 square bipyramid. (b) Orthogonal arrangement between the Nd^{3+} and the Cu^{2+} moments in Nd_2CuO_4 , where the larger and smaller circles represent the Nd^{3+} and Cu^{2+} ions, respectively.

A nonzero DM interaction between RE and TM ions can be realized in the square-lattice antiferromagnets NdMnAsO ^{16,17} and PrMnSbO ,¹⁸ which are analogous in their crystal structure to the iron arsenide superconductors.^{19,20} In RMnAsO ($R = \text{Nd, Pr}$), each square sheet of high-spin Mn^{2+} (d^5 , $S = 5/2$) ions (parallel to the ab plane) is sandwiched between two square sheets of As^{3-} ions to form an As-Mn-As slab (Figure 2a) and

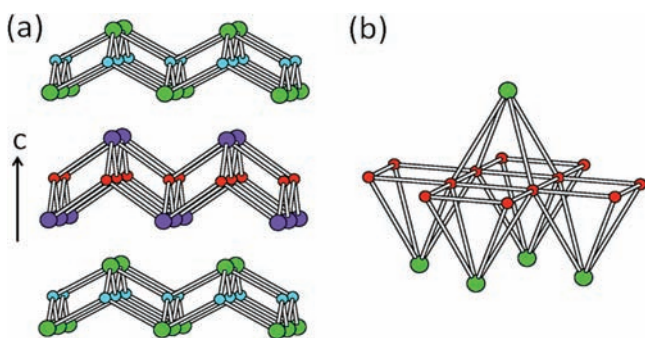


Figure 2. (a) Perspective view of the crystal structure of RMnAsO ($R = \text{Ce, Pr, Nd}$) in which each square sheet of high-spin Mn^{2+} (d^5 , $S = 5/2$) ions parallel to the ab plane is sandwiched between two square sheets of As^{3-} ions to form an As-Mn-As slab ($\text{Mn} = \text{red circle}$, $\text{As} = \text{purple circle}$). Similarly, each square sheet of O^{2-} ions parallel to the ab plane is sandwiched between two square sheets of R^{3+} ions to form a R-O-R slab ($\text{O} = \text{cyan circle}$, $\text{R} = \text{green circle}$). In RMnAsO the As-Mn-As slabs alternate with the R-O-R slabs along the c direction such that each R is located above the center of an As_4 square and each As above the center of a R_4 square (Figure 2a). Consequently, the RE ions R^{3+} ions lying above and below the square sheet of the Mn^{2+} ions occupy the positions of neighboring Mn_4 squares, each R atom forming a Mn_4R square pyramid.

each square sheet of O^{2-} ions (parallel to the ab plane) between two square sheets of R^{3+} ions to form a R-O-R slab (Figure 2a). The As-Mn-As slabs alternate with the R-O-R slabs along the c direction such that each R is located above the center of an As_4 square and each As above the center of a R_4 square (Figure 2a). Consequently, the RE ions R^{3+} ions lying above and below the square sheet of the Mn^{2+} ions occupy the positions of neighboring Mn_4 squares, with each R atom forming Mn_4R square pyramid (Figure 2b). The Mn^{2+} (d^5) ions in the square lattice of RMnAsO ($R = \text{Nd, Pr}$) undergo an antiferromagnetic ordering at high temperatures ($T_N = 359$ and

230 K for $R = \text{Nd}$ and Pr , respectively) with the Mn moments oriented along the c axis.¹⁶ NdMnAsO and PrMnSbO undergo a SR transition at low temperatures ($T_{\text{SR}} = 23$ and 35 K, respectively), below which the Mn moments rotate into the ab plane. Simultaneously, the RE moments occur in the ab plane with ferromagnetic coupling in each sheet of RE atoms parallel to the ab plane, and two such sheets straddling each sheet of the Mn atoms have their moments antiferromagnetically coupled (Figure 3). CeMnAsO consists of Ce^{3+} (f^1) and Mn^{2+} (d^5) ions, is isostructural with NdMnAsO , and undergoes an analogous SR at $T_{\text{SR}} = 34$ K.^{21,22} LaMnAsO exhibits an antiferromagnetic ordering of the Mn moments similar to that found in RMnAsO ($R = \text{Ce, Nd}$). However, with no moment on La^{3+} , LaMnAsO does not undergo a SR on cooling and keeps the Mn moments aligned along the c axis.¹⁷ This indicates that the SR in NdMnAsO and CeMnAsO is caused by the interaction between the R^{3+} and the Mn^{2+} ions. There is an important difference between the magnetic structures of CeMnAsO and NdMnAsO below T_{SR} , namely, the R^{3+} and Mn^{2+} moments are orthogonal in CeMnAsO ^{21,22} but are collinear in NdMnAsO .¹⁷

In the present work we probe the cause for the SR of RMnAsO ($R = \text{Ce, Nd}$) on the basis of density functional theory (DFT) calculations. We show that the DM and BQ exchange interactions between the R^{3+} and the Mn^{2+} ions are strong and govern the nature of their SR phenomenon and that the DM interaction dominates over the BQ exchange in CeMnAsO , but the opposite is the case for NdMnAsO . In addition, we explore the implications of our findings that can be tested by further experiments and theoretical analyses.

2. INTERACTIONS BETWEEN RARE-EARTH AND TRANSITION-METAL IONS

A. Nature of the DM Exchange. Let us examine under what condition the DM exchange between the R^{3+} and the Mn^{2+} ions of RMnAsO (located, respectively, at sites 0 and i of the Mn_4R square pyramid, Figure 3a and 3c) can become strong. The DM exchange between R^{3+} and Mn^{2+} in a Mn_4R square pyramid is written as $\vec{D}_{0i} \cdot (\vec{S}_0 \times \vec{S}_i)$ ($i = 1-4$) with the DM vector expressed as $\vec{D}_{0i} = \lambda J_{0i} (\delta\vec{L}_0 - \delta\vec{L}_i)$,^{2,23} where λ is the spin-orbit coupling (SOC) constant of the magnetic ions while $\delta\vec{L}_0$ and $\delta\vec{L}_i$ are essentially the unquenched orbital momenta of the magnetic ions at 0 and i , respectively.²³ Thus, the magnitude of \vec{D}_{0i} can be increased by increasing the $(\delta\vec{L}_0 - \delta\vec{L}_i)$ value. For a $S = 5/2$ TM ion (e.g., Mn^{2+}), $L = 0$ to a first approximation so that $\delta\vec{L} \approx 0$. For a RE ion R^{3+} , the quenching of the orbital angular momentum is generally weak. The f orbitals of a R^{3+} ion overlap poorly with the orbitals of its surrounding ligands, so that its spherical electron density distribution is weakly distorted by the surrounding ligands. Therefore, the unquenched orbital momentum $\delta\vec{L}$ of a RE ion R^{3+} can be large. Therefore, the DM vector for the exchange between such TM and RE ions can be approximated by

$$\vec{D}_{0i} \approx \sqrt{\lambda_0 \lambda_i} J_{0i} (\delta\vec{L}_0 - \delta\vec{L}_i) \approx \sqrt{\lambda_0 \lambda_i} J_{0i} \delta\vec{L}_0 \quad (1)$$

Since the SOC constant of $4f$ ions is about three times larger than that of $3d$ ions,²⁴ such combinations of RE and TM ions would increase the length of the associated DM vector. Use of a $S = 5/2$ TM ion is also favorable in strengthening the DM exchange, because it can increase the length of the vector $(\vec{S}_0 \times \vec{S}_i)$.

When the spins of the RE and TM ions are both large, the BQ exchange can become strongly enhanced because it is proportional to $(\vec{S}_0 \cdot \vec{S}_i)^2$. Therefore, it is an important issue

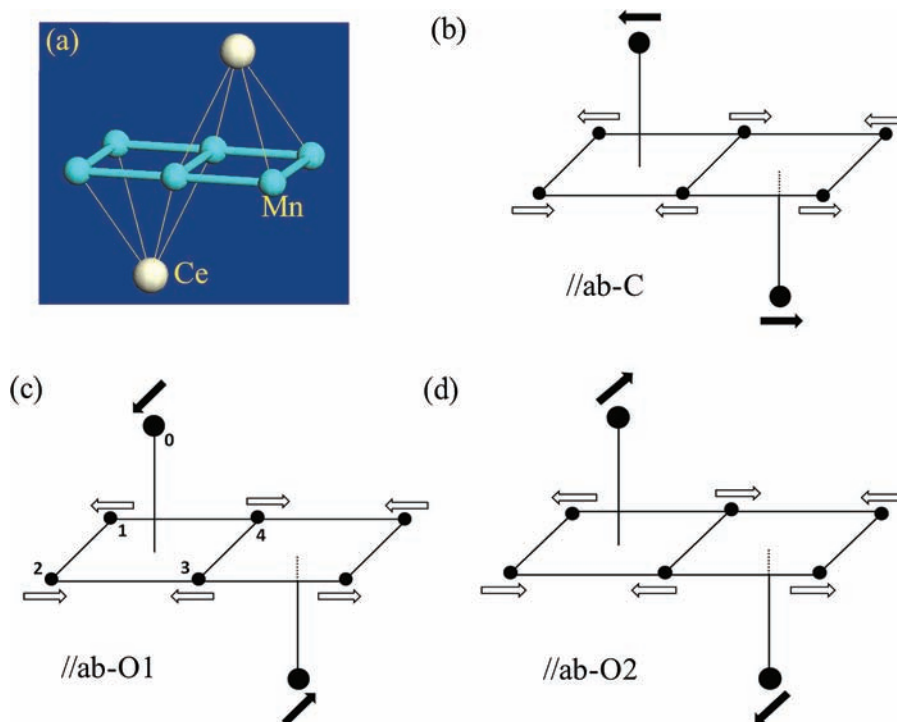


Figure 3. (a) Arrangement of the R^{3+} ions above and below the square lattice of Mn^{2+} ions in $RMnAsO$ ($R = Ce, Nd$), where the R^{3+} and Mn^{2+} ions are represented by white and cyan spheres, respectively. Each R atom located above or below every Mn_4 square forms a Mn_4R square pyramid. (b–d) Orthogonal and collinear arrangements between the R^{3+} and the Mn^{2+} moments in $RMnAsO$ ($R = Ce, Nd$), where the larger and smaller circles represent the R^{3+} and Mn^{2+} ions, respectively. The $\parallel ab-C$, $\parallel ab-O1$, and $\parallel ab-O2$ arrangements of the R^{3+} and Mn^{2+} moments are presented in b, c, and d, respectively. In the square lattice of the Mn^{2+} ions, the Mn^{2+} moments are antiferromagnetically coupled. R^{3+} and Mn^{2+} moments lie in the ab plane (i.e., the plane of the square lattice). R^{3+} moments located above the square lattice of the Mn^{2+} ions are antiparallel to those located below the square lattice of the Mn^{2+} ions.

whether the DM or the BQ exchange dominates, because the DM exchange favors an orthogonal arrangement between the RE and TM moments whereas the BQ exchange favors a collinear arrangement.

B. DM and BQ Exchange Energies. In their powder neutron diffraction study of $NdMnAsO$,¹⁶ Marcinkova et al. were unable to resolve whether the ordered Nd and Mn moments below $T_{SR} = 23$ K, which lie in the ab plane, are collinear (Figure 3b) or orthogonal (Figure 3c or 3d). However, Emery et al.¹⁷ recently reported that the ordered Mn and Nd moments below T_{SR} have the collinear arrangement depicted in Figure 3b. For the convenience of our discussion, the ab plane collinear arrangement in Figure 3b will be referred to as the $\parallel ab-C$ arrangement, the ab -plane orthogonal arrangement in Figure 3c as the $\parallel ab-O1$ arrangement, and the alternative ab plane orthogonal arrangement in Figure 3d as the $\parallel ab-O2$ arrangement. The latter arrangement results from the $\parallel ab-O1$ arrangement by reversing the directions of the RE moments while keeping the Mn moments unchanged. The recent neutron powder diffraction study¹⁷ showed that the Mn moments of $LaMnAsO$ at room temperature are parallel to the c axis, as found for those of $NdMnAsO$ between T_{SR} and T_N .

We now consider the energies of the $\parallel ab-C$, $\parallel ab-O1$, and $\parallel ab-O2$ spin arrangements of $RMnAsO$ ($R = Ce, Nd$), which we denote as $E(\parallel ab-C)$, $E(\parallel ab-O1)$, and $E(\parallel ab-O2)$, respectively, in terms of the spin Hamiltonian defined by the Heisenberg, BQ, and DM spin-exchange interactions

$$H_{\text{spin}} = \sum_{i<j} J_{ij} \vec{S}_i \cdot \vec{S}_j + \sum_{i<j} K_{ij} (\vec{S}_i \cdot \vec{S}_j)^2 + \sum_{i<j} \vec{D}_{ij} \cdot (\vec{S}_i \times \vec{S}_j) \quad (2)$$

(With the above definition, ferromagnetic and antiferromagnetic Heisenberg exchanges are given by $J_{ij} < 0$ and $J_{ij} > 0$, respectively.) Then, the net Heisenberg exchange between the R^{3+} and the Mn^{2+} ions is zero for the $\parallel ab-C$ arrangement and also for the $\parallel ab-O1$ and $\parallel ab-O2$ arrangements due to their orthogonal spin arrangement. For the $\parallel ab-C$ arrangement (Figure 3b), the DM exchange energy (E_{DM}) is zero but the BQ exchange energy (E_{BQ}) is not, because the R^{3+} and Mn^{2+} moments are collinear. For the $\parallel ab-O1$ and $\parallel ab-O2$ arrangements, $E_{BQ} = 0$ because the R^{3+} and Mn^{2+} ions moments are orthogonal and their E_{DM} energies are nonzero and opposite in sign. Therefore

$$2|E_{DM}| = |E(\parallel ab-O1) - E(\parallel ab-O2)| \quad (3)$$

Let us use the notation $\parallel ab-O_i$ to indicate the lower energy one of the $\parallel ab-O1$ and $\parallel ab-O2$ arrangements. Then the energy difference between the $\parallel ab-C$ and the $\parallel ab-O_i$ states has both the BQ and the DM contributions. Thus

$$E_{BQ} = E(\parallel ab-C) - E(\parallel ab-O_i) - |E_{DM}| \quad (4)$$

Consequently, one can evaluate E_{DM} and E_{BQ} by determining the relative energies of the $\parallel ab-C$, $\parallel ab-O1$, and $\parallel ab-O2$ arrangements on the basis of DFT+U+SOC calculations.

3. ANALYSIS OF THE ELECTRONIC AND MAGNETIC STRUCTURES

A. Computational Details. We examine the electronic structures of $LaMnAsO$, $CeMnAsO$, and $NdMnAsO$ on the basis of DFT electronic structure calculations. For $LaMnAsO$,

we employed the frozen-core projector-augmented wave method encoded in the Vienna Ab initio Simulation Package²⁵ with the generalized-gradient approximation²⁶ for the exchange-correlation functional, the plane-wave cutoff energy of 400 eV, and 27 *k* points for the irreducible Brillouin zone. To describe the electron correlation in the Mn 3d states, the DFT plus on-site repulsion *U* (DFT+*U*) method²⁷ was used with effective on-site Coulomb repulsion *U* = 4.5 eV on the Mn atom, a representative value for Mn.^{28,29} The threshold for the self-consistent-field convergence of the total electronic energy was 10⁻⁶ eV.

To evaluate the Heisenberg exchange interactions between the R³⁺ and the Mn²⁺ ions in RMnAsO (R = Ce, Nd), it is necessary to treat the *f* electron of the R³⁺ ions explicitly. Thus, for RMnAsO (R = Ce, Nd), we carried out DFT+*U* calculations (with *U* = 4.5 and 5.4 eV on the Mn and R atoms, respectively³⁰) using the full-potential linearized augmented plane-wave method encoded in the WIEN2k package³¹ with 64 *k* points for the irreducible Brillouin zone, the threshold of 10⁻⁶ Ry for the energy convergence, the cutoff energy parameters of R *K*_{max} = 7.0 and *G*_{max} = 12, and the energy threshold of -6.0 Ry for the separation of the core and valence states. We use the following sets of basis orbitals: [Ar] 3d⁵ 4s² for Mn, [Xe] 4f² 6s² for Ce, [Ar] 3d¹⁰ 4s² 4p³ for As, and [He] 2s² 2p⁴ for O. The muffin-tin sphere radii used are 2.5, 2.34, 2.25, and 2.08 for Mn, Ce, As, and O, respectively. To determine the spin orientations of the R³⁺ and Mn²⁺ ions and evaluate the DM interaction energies of RMnAsO (R = Ce, Nd), we perform DFT+*U* plus SOC³² (DFT+*U*+SOC) calculations.

B. Electronic Structures. The total density of states (DOS) plots obtained from the DFT+*U*+SOC calculations for the $\parallel ab$ -C spin arrangement of CeMnAsO and NdMnAsO are presented in Figures 4a and 5a, respectively. The partial DOS (PDOS) plots obtained for the Mn 3d and Ce 4f states are given Figure 4b and 4c, respectively, and those obtained for the Mn 3d and Nd 4f states are given in Figure 5b and 5c, respectively. The upper and lower panels in each diagram refer to the up-spin and down-spin states, respectively. Both CeMnAsO and NdMnAsO exhibit a band gap and hence are magnetic insulators. The PDOS plots for the Mn 3d states show that the up-spin states are all filled but the down-spin states are all empty, consistent with the presence of high-spin Mn²⁺ ions (i.e., *S* = 5/2) in RMnAsO (R = Ce, Nd). The PDOS plots for the Ce 4f and Nd 4f states are consistent with the Ce³⁺(*f*¹) and Nd³⁺(*f*³) electron counts, respectively (for further discussion, see below).

C. Magnetic Anisotropy of Individual Magnetic Ions. On the basis of DFT+*U*+SOC calculations we first estimate the magnetic anisotropy of the individual Mn²⁺, Ce³⁺, and Nd³⁺ ions in RMnAsO (R = Ce, Nd) in the absence of interactions between R³⁺ and Mn²⁺ ions. Our DFT+*U*+SOC calculations for the magnetic ground state of LaMnAsO, in which the Mn moments in the square lattices are antiferromagnetically ordered, show that the Mn²⁺ moment parallel to the *c* axis ($\parallel c$) is more stable than that perpendicular to the *c* axis ($\perp c$) by 0.20 meV per Mn, which is consistent with the experimental observations for LaMnAsO at room temperature and for NdMnAsO and PrMnSbO between *T*_{SR} and *T*_N.^{16–18} Our DFT+*U*+SOC calculations for the structure RZnAsO, resulting from RMnAsO when the magnetic Mn²⁺ ions are replaced with the nonmagnetic Zn²⁺ ions, show that the Ce³⁺ spin moment favors the $\parallel c$ orientation over the $\perp c$ orientation by 3.5 meV per Ce. In contrast, the Nd³⁺ spin moment prefers the $\perp c$ orientation to

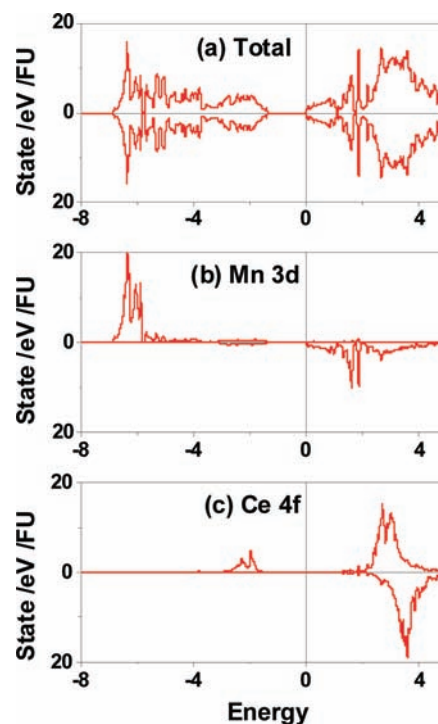


Figure 4. Total DOS and PDOS plots of CeMnAsO obtained from the DFT+*U*+SOC calculations for the $\parallel ab$ -C spin arrangement. Bottom of the empty bands is taken as the Fermi level.

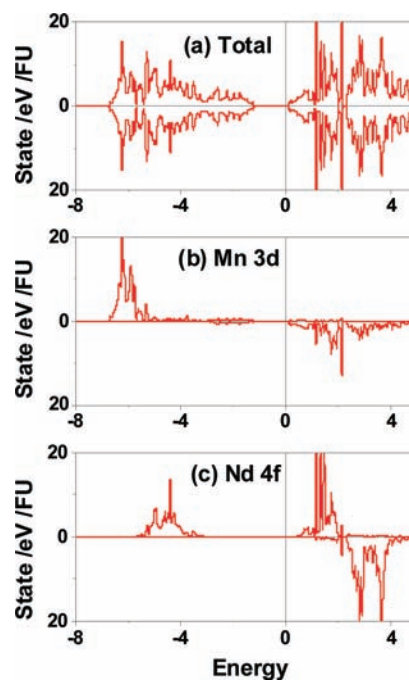


Figure 5. Total DOS and PDOS plots of NdMnAsO obtained from the DFT+*U*+SOC calculations for the $\parallel ab$ -C spin arrangement. Bottom of the empty bands is taken as the Fermi level.

the $\parallel c$ orientation by 20.7 meV per Nd. Thus, the Mn²⁺ ions of RMnAsO have weak easy-axis anisotropy as may be expected for the *L* = 0 ion Mn²⁺. The Ce³⁺ ions of CeMnAsO have substantial easy-axis anisotropy. In contrast, the Nd³⁺ ions of NdMnAsO exhibit strong easy-plane anisotropy. For CeMnAsO

to undergo a SR with a rotation of the Mn moments into the *ab* plane, the interaction between the Ce³⁺ and the Mn²⁺ ions needs to overcome the easy-axis anisotropy of the individual Ce and Mn moments.

The single-ion magnetic anisotropy of a TM ion is a weak effect arising from SOC. As found for numerous magnetic systems such as Ag₂MnO₂,³³ TbMnO₃,³⁴ SrFeO₂,³⁵ and Sr₃Fe₂O₅,³⁶ not to mention LaMnAsO examined in the present work, this anisotropy is well reproduced by DFT+U+SOC calculations despite the inherent limitations of the exchange-correlation functionals. This indicates that the deficiency of the functional cancel out when the energy difference between different spin orientations is calculated.

D. Energy Mapping Analysis for the DM and BQ Exchange Energies. The relative energies calculated for the $\parallel ab$ -C, $\parallel ab$ -O1, and $\parallel ab$ -O2 arrangements of RMnAsO (R = Ce, Nd) are summarized in Table 1, and so are the values of E_{DM}

Table 1. Relative Energies E (in meV per FU) of the $\parallel ab$ -C, $\parallel ab$ -O1, and $\parallel ab$ -O2 Spin Arrangements of RMnAsO (R = Ce, Nd) Obtained from DFT+U+SOC Calculations and the Associated DM and BQ Exchange Energies (in meV per FU) between the R³⁺ and Mn²⁺ Ions

	CeMnAsO	NdMnAsO
$E(\parallel ab\text{-C})$	0	0
$E(\parallel ab\text{-O1})$	-4.81	+18.39
$E(\parallel ab\text{-O2})$	+4.96	+17.70
E_{DM}	-4.88	-0.35
E_{BQ}	-0.07	-18.05

and E_{BQ} of RMnAsO (R = Ce, Nd) calculated from the relative energies by using eqs 3 and 4. For CeMnAsO, the orthogonal arrangement $\parallel ab$ -O1 is the ground state, and the DM exchange dominates over the BQ exchange, with $E_{DM} = -4.88$ meV per formula unit (FU) and $E_{BQ} = -0.07$ meV/FU. The DM exchange favoring the $\parallel ab$ -O1 arrangement overrides not only the BQ exchange favoring the $\parallel ab$ -C arrangement but also the easy-axis anisotropy of the individual Ce³⁺ and Mn²⁺ ions (i.e., 3.7 meV/FU). Therefore, the SR of CeMnAsO is caused by the DM exchange between the Ce³⁺ and the Mn²⁺ ions. For NdMnAsO, the collinear arrangement $\parallel ab$ -C is the ground state and the BQ exchange dominates over the DM exchange with $E_{BQ} = -18.05$ meV/FU and $E_{DM} = -0.35$ meV/FU. Thus, the SR of NdMnAsO is induced by the strong easy-plane anisotropy of the Nd³⁺ ion plus the strong BQ exchange between the Nd³⁺ and the Mn²⁺ ions.

E. Heisenberg Spin Exchange between the R³⁺ and Mn²⁺ Ions. For our discussion in the next section it is necessary to evaluate the Heisenberg spin-exchange constant J_{0i} between the R³⁺ and the Mn²⁺ ions (eq 1). As depicted in Figure 6, one may consider the five Heisenberg exchange paths J_1 – J_5 in RMnAsO (R = Ce, Nd), where J_{0i} is defined as J_5 . To extract the values of J_1 – J_5 , we determine the relative energies of the six ordered spin states FM and AF1–AF5 depicted in Figure 7 by DFT+U calculations. Given the spin Hamiltonian

$$H^c = \sum_{i<j} J_{ij} \hat{S}_i \cdot \hat{S}_j \quad (5)$$

defined in terms of the spin exchanges $J_{ij} = J_1$ – J_5 , the total spin-exchange interaction energies of the FM and AF1–AF5 states

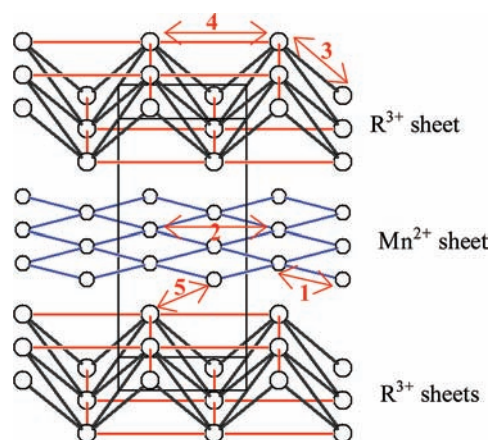


Figure 6. Five spin-exchange paths J_1 – J_5 of RMnAsO. For simplicity, only the Mn²⁺ and R³⁺ ions are shown as small and large circles. Numbers 1–5 refer to the spin-exchange paths J_1 – J_5 , respectively. J_5 path is the Heisenberg exchange J_{0i} between the R³⁺ and the Mn²⁺ ions.

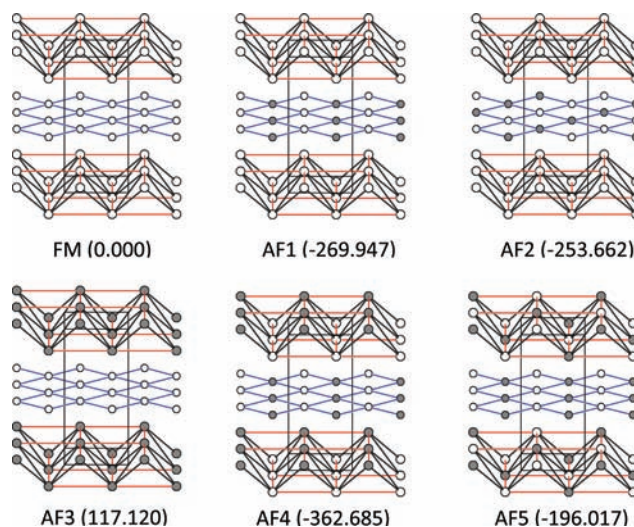


Figure 7. Six ordered spin states of RMnAsO (R = Ce, Nd), where the shaded circles represent the up-spin R and Mn sites and the unshaded circles the down-spin R and Mn sites. Numbers in parentheses refer to the relative energies (in meV per FU) determined for CeMnAsO from the DFT+U calculations.

(per two FUs) are obtained as summarized below by applying the energy expressions obtained for spin dimers³⁷

$$\begin{aligned} E_{FM} &= (+4J_1 + 4J_2)N^2/4 + (+4J_3 + 4J_4)M^2/4 \\ &\quad + (+8J_5)NM/4 \\ E_{AF1} &= (-4J_1 + 4J_2)N^2/4 + (+4J_3 + 4J_4)M^2/4 \\ E_{AF2} &= (-4J_2)N^2/4 + (+4J_3 + 4J_4)M^2/4 \\ E_{AF3} &= (+4J_1 + 4J_2)N^2/4 + (+4J_3 + 4J_4)M^2/4 \\ &\quad + (-8J_5)NM/4 \\ E_{AF4} &= (-4J_1 + 4J_2)N^2/4 + (-4J_3 + 4J_4)M^2/4 \\ E_{AF5} &= (-4J_1 + 4J_2)N^2/4 + (-4J_4)M^2/4 \end{aligned} \quad (6)$$

where N and M are the numbers of the unpaired spins at the Mn^{2+} and R^{3+} sites of RMnAsO ($\text{R} = \text{Ce}, \text{Nd}$), namely, $N = 5$ and $M = 1$ for CeMnAsO and $N = 5$ and $M = 3$ for NdMnAsO . Thus, by mapping the relative energies of the six ordered spin states determined from the DFT+ U calculations onto the corresponding energies obtained from the total Heisenberg exchange energies,³⁸ we obtain the J_1 – J_5 values of CeMnAsO summarized in Table 2, indicating that the Heisenberg exchange between the

Table 2. Distances between the Magnetic Ions in the Heisenberg Exchange Paths J_1 – J_5 of CeMnAsO and Values of J_1 – J_5 Obtained from the Mapping Analysis Based on the DFT+ U Calculations

exchange paths	distance (Å)	J_i (meV)
J_1 (within each Mn layer)	Mn–Mn = 2.884	+13.1
J_2 (within each Mn layer)	Mn–Mn = 4.078	+5.9
J_3 (between adjacent Ce layers)	Ce–Ce = 3.722	+92.8
J_4 (within each Ce layer)	Ce–Ce = 4.078	–120.3
J_5 (between adjacent Ce and Mn layers)	Ce–Mn = 3.871	–11.2

R^{3+} and the Mn^{2+} ions in CeMnAsO is ferromagnetic (–11.2 meV). According to eq 5, J_5 can be determined by considering only the FM and AF3 states. This is how we determine J_5 is antiferromagnetic (+38.4 meV) for NdMnAsO . Therefore, J_5 is larger in magnitude for NdMnAsO than for CeMnAsO . We note that DFT+ U calculations for magnetic compounds made up of TM ions are known to overestimate Heisenberg spin-exchange constants typically by a factor up to 4–5 depending upon what U values are used, but the trends in their relative values remain nearly independent of the U values used in the DFT+ U calculations.^{39–42}

4. DISCUSSION

A. Nature of the DM Exchange Energy. To see why the DM exchange is strong in CeMnAsO but weak in NdMnAsO , we examine the nature of the DM exchange on the basis of eq 1. Consider the DM interaction around one R^{3+} ion of RMnAsO in the $\parallel ab$ -O1 arrangement (Figure 3c), where site 0 is used for the R^{3+} ion and sites 1–4 for the Mn^{2+} ions. Then the E_{DM} is expressed as

$$\begin{aligned}
 E_{\text{DM}} &= \vec{D}_{01} \cdot (\vec{S}_0 \times \vec{S}_1) + \vec{D}_{02} \cdot (\vec{S}_0 \times \vec{S}_2) + \vec{D}_{03} \cdot (\vec{S}_0 \times \vec{S}_3) \\
 &\quad + \vec{D}_{04} \cdot (\vec{S}_0 \times \vec{S}_4) \\
 &= (\vec{D}_{01} - \vec{D}_{02} + \vec{D}_{03} - \vec{D}_{04}) \cdot (\vec{S}_0 \times \vec{S}_1)
 \end{aligned} \tag{8}$$

Equation 1 shows that the symmetry of the DM vectors \vec{D}_{0i} ($i = 1$ –4) is determined largely by that of the unquenched orbital momentum $\delta\vec{L}_0$ of the R^{3+} ion, namely, by the symmetry of the occupied 4f orbitals of the R^{3+} ion. Our DFT+ U +SOC calculations for CeMnAsO show that the occupied 4f states of the Ce^{3+} ion have only the $l_z = \pm 2$ orbitals (Figure 8a). In Cartesian coordinates, the latter correspond to the f_{xy} and $f_{z(x^2-y^2)}$ orbitals, which switch their sign when rotated around the z axis by 90° . This means that $\vec{D}_{01} = \vec{D}_{03} = -\vec{D}_{02} = -\vec{D}_{04}$, so that

$$E_{\text{DM}} = 4\vec{D}_{01} \cdot (\vec{S}_0 \times \vec{S}_1) \tag{9}$$

Thus, for the $\parallel ab$ -O2 arrangement (Figure 3d), $E_{\text{DM}} = -4\vec{D}_{01} \cdot (\vec{S}_0 \times \vec{S}_i)$.

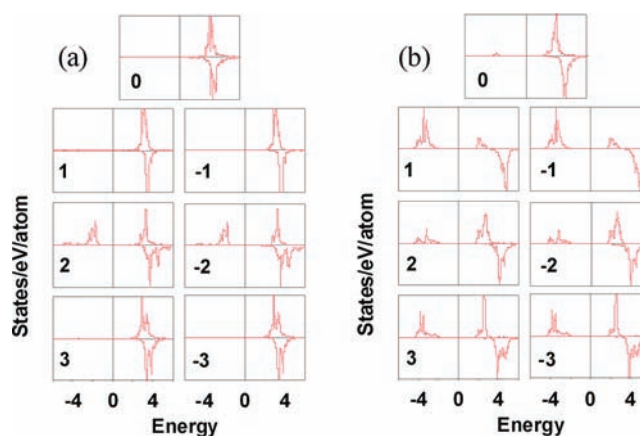


Figure 8. PDOS plots for (a) the Ce 4f states of CeMnAsO and (b) the Nd 4f states of NdMnAsO obtained from the DFT+ U +SOC calculations for the $\parallel ab$ -C spin arrangement, where the numbers 0, ± 1 , ± 2 , and ± 3 refer to the l_z values of the Ce^{3+} and Nd^{3+} ions. Top of the filled bands is taken as the Fermi level. In each diagram, the upper and lower panels represent the up-spin and down-spin states, respectively. In each panel, the vertical PDOS axis covers from 0 to 1.5 states/eV/atom and the numbers on the horizontal energy axis are in eV.

Our calculations for NdMnAsO show that the occupied 4f states of the Nd^{3+} ion have a strong contribution from the $l_z = \pm 1$ orbitals but weak contributions from the $l_z = \pm 2$ and $l_z = \pm 3$ orbitals (Figure 8b). The $l_z = \pm 1$ orbitals, corresponding to the $f_{x(4z^2-x^2-y^2)}$ and $f_{y(4z^2-x^2-y^2)}$ orbitals in Cartesian coordinates, switch their sign when rotated around the z axis by 180° , so that $\vec{D}_{01} = -\vec{D}_{03}$ and $\vec{D}_{02} = -\vec{D}_{04}$. Consequently, the DM exchange associated with these orbitals is zero. The $l_z = \pm 2$ orbitals will give rise to $E_{\text{DM}} = 3n_{\pm 2}(-4\vec{D}_{01} \cdot \vec{S}_c)$, where $3n_{\pm 2}$ represents the number of 4f electrons in the $l_z = \pm 2$ orbitals. The $l_z = \pm 3$ orbitals, which correspond to the $f_{x(x^2-3y^2)}$ and $f_{y(3x^2-y^2)}$ orbitals, switch their sign when rotated around the z axis by 120° . If it is assumed that the DM vectors associated with the $l_z = \pm 3$ orbitals change linearly as a function of the rotation angle around the z axis, one might expect that $\vec{D}_{02} = -\vec{D}_{01}/2$, $\vec{D}_{03} = 0$, and $\vec{D}_{04} = \vec{D}_{01}/2$. Thus, the DM exchange energy associated with the $l_z = \pm 3$ orbitals would be $3n_{\pm 3}(-\vec{D}_{01} \cdot \vec{S}_c)$, where $3n_{\pm 3}$ represents the number of 4f electrons in the $l_z = \pm 3$ orbitals. The DM exchange of NdMnAsO is small due to the low occupations of the $l_z = \pm 2$ and ± 3 orbitals. This explains why the DM exchange is weak in NdMnAsO .

Equation 1 shows that the sign of the DM vector \vec{D}_{0i} between the R^{3+} and Mn^{2+} ions depends on the sign of their Heisenberg exchange constant J_{0i} . As noted above, J_{0i} is ferromagnetic ($J_{0i} < 0$) in CeMnAsO but antiferromagnetic ($J_{0i} > 0$) in NdMnAsO . This accounts for why the $\parallel ab$ -O1 spin arrangement is more stable than the $\parallel ab$ -O2 spin arrangement in CeMnAsO , while the opposite is the case for NdMnAsO (Table 1).

B. Experimental Implications. There are two important implications of our work that can be tested experimentally. One deals with the local nature of the DM and BQ exchanges. The antiferromagnetic coupling in the Mn lattice of LaMnAsO is strong since the Mn moments order well above room temperature.¹⁷ Thus, when the moments of a few Mn^{2+} ions are forced to change their directions, the moments of the remaining Mn^{2+} ions will follow them in order to maintain a collinear antiferromagnetic arrangement. For the solid solution $\text{La}_{1-x}\text{Ce}_x\text{MnAsO}$, in which the magnetic ions Ce^{3+} are diluted with nonmagnetic ions

La³⁺, the SR is expected to occur as long as the DM exchange of the remaining Ce³⁺ ions with their surrounding Mn²⁺ ions is stronger than the easy-axis anisotropy of the Mn²⁺ moment (i.e., $4.88x > 0.2$). This prediction has recently been confirmed by Tsukamoto et al.,²¹ who showed that the SR in La_{1-x}Ce_xMnAsO persists for x down to 0.10. Similarly, the solid solution La_{1-x}Nd_xMnAsO is predicted to exhibit the SR for $x > 0.01$. In addition, the solid solution Ce_{1-x}Nd_xMnAsO is predicted to adopt the $\parallel ab$ -C arrangement for $x > 0.22$ according to the E_{DM} and E_{BQ} values of CeMnAsO and NdMnAsO in Table 1. The other implication is concerned with how to tilt the balance between the DM and the BQ exchanges. For the DM exchange to dominate over the BQ exchange, it would be necessary to reduce the $(\bar{S}_0 \cdot \bar{S}_C)^2$ value, and hence combination of the $S = 1/2$ ion Ce³⁺ with TM ions with $S < 5/2$ (e.g., $S = 2$ ion such as Cr²⁺ and Fe²⁺) might be interesting.

5. CONCLUDING REMARKS

In summary, our study shows that the SR phenomena of the square-lattice antiferromagnets RMnAsO (R = Ce, Nd) are well explained in terms of the DM and BQ exchanges between R³⁺ and Mn²⁺ ions determined in terms of DFT calculations. The symmetry of the DM vector in such systems follows that of the partially filled 4f states of the R³⁺ ions (eq 1), and consideration of the DM and BQ exchanges between RE and TM ions is essential in describing magnetic solids containing both types of ions. Below their SR temperature, the Mn²⁺ and Ce³⁺ moments are orthogonal in CeMnAsO but collinear in NdMnAsO because the DM interaction dominates over the BQ interaction in CeMnAsO while the opposite is the case for NdMnAsO. The implications of our findings discussed above need to be tested by further experiments and theoretical analyses.

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) Erdős, P. *J. Phys. Chem. Solids* **1966**, *27*, 1705.
- (2) Moriya, T. *Phys. Rev.* **1960**, *120*, 91.
- (3) (a) Shekhtman, L.; Entin-Wohlman, O.; Aharony, A. *Phys. Rev. Lett.* **1992**, *69*, 836. (b) Yildirim, T.; Harris, A. B.; Aharony, A.; Entin-Wohlman, O. *Phys. Rev. B* **1992**, *52*, 10239.
- (4) Sólyom, J. *Phys. Rev. B* **1987**, *36*, 8642.
- (5) Novák, P.; Chaplygin, I.; Seifert, G.; Gemming, S.; Laskowski, R. *Comput. Mater. Sci.* **2008**, *79*, 44.

- (6) Furrer, A.; Juranyi, F.; Krämer, K. W.; Strässle, Th. *Phys. Rev. B* **2008**, *77*, 174410.
- (7) Dzyaloshinskii, I. *J. Phys. Chem. Solids* **1958**, *4*, 241.
- (8) Katsura, H.; Nagaosa, N.; Balatsky, A. V. *Phys. Rev. Lett.* **2005**, *95*, 057205.
- (9) Sergienko, I. A.; Dagotto, E. *Phys. Rev. B* **2006**, *73*, 094434.
- (10) Onose, Y.; Ideue, T.; Katsura, H.; Shiomi, Y.; Nagaosa, N.; Tokura, Y. *Science* **2010**, *329*, 297.
- (11) Ginsberg, A. P. *Inorg. Chim. Acta Rev.* **1971**, *5*, 45.
- (12) Bencini, A.; Gatteschi, D. *EPR of Exchange Coupled Systems*; Springer: New York, 1989.
- (13) Benelli, C.; Gatteschi, D. *Chem. Rev.* **2002**, *102*, 2369.
- (14) Skanthakumar, S.; Zhang, H.; Clinton, T. W.; Li, W.-H.; Lynn, J. W.; Fisk, Z.; Cheong, S.-W. *Physica C* **1989**, *160*, 124.
- (15) Sachidanandam, R.; Yildirim, T.; Harris, A. B.; Aharony, A.; Entin-Wohlman, O. *Phys. Rev. B* **1997**, *56*, 260.
- (16) Marcinkova, A.; Hansen, T. C.; Curfs, C.; Margadonna, S.; Bos, J.-W. G. *Phys. Rev. B* **2010**, *82*, 174438.
- (17) Emery, N.; Wildman, E. J.; Skakle, J. M. S.; Mclaughlin, A. C.; Smith, R. I.; Fitch, A. N. *Phys. Rev. B* **2011**, *83*, 144429.
- (18) Kimber, S. A. J.; Hill, A. H.; Zhang, Y.-Z.; Jeschke, H. O.; Valenti, R.; Ritter, C.; Schellenberg, I.; Hermes, W.; Pöttgen, R.; Argyriou, D. N. *Phys. Rev. B* **2010**, *82*, 100412.
- (19) Kamihara, Y.; Watanabe, T.; Hirano, M.; Hosono, H. *J. Am. Chem. Soc.* **2008**, *130*, 3296.
- (20) Ishida, K.; Nakai, Y.; Hosono, H. *J. Phys. Soc. Jpn.* **2009**, *78*, 062001.
- (21) Tsukamoto, Y.; Okamoto, Y.; Matsuhira, K.; Whangbo, M.-H.; Hiroi, Z. *J. Phys. Soc. Jpn.* **2011**, *80*, 094708.
- (22) Lee, J.-S.; Ji, S.; Lee, S.-H. Private communication.
- (23) Dai, D.; Xiang, H.-J.; Whangbo, M.-H. *J. Comput. Chem.* **2008**, *29*, 2187.
- (24) In *Concise Encyclopedia of Magnetic and Superconducting Materials*, 2nd ed.; Buschow, K. H. J., Ed.; Elsevier: Amsterdam, 2006; p 150.
- (25) (a) Kresse, G.; Hafner, J., *Phys. Rev. B* **1993**, *62*, 558 (1993). (b) Kresse, G.; Furthmüller, J. *Comput. Mater. Sci.* **1996**, *6*, 15. (c) Kresse, G.; Furthmüller, J. *Phys. Rev. B* **1996**, *54*, 11169.
- (26) Perdew, J. P.; Burke, K.; Ernzerhof, M. *Phys. Rev. Lett.* **1996**, *77*, 3865.
- (27) Dudarev, S. L.; Botton, G. A.; Savrasov, S. Y.; Humphreys, C. J.; Sutton, A. P. *Phys. Rev. B* **1998**, *57*, 1505.
- (28) Ji, S.; Kan, E. J.; Whangbo, M.-H.; Kim, J.-H.; Qiu, Y.; Matsuda, M.; Yoshida, H.; Hiroi, Z.; Green, M. A.; Ziman, T.; Lee, S.-H. *Phys. Rev. B* **2010**, *81*, 094421.
- (29) Ben Yahia, H.; Gaudin, E.; Boulahya, K.; Darriet, J.; Son, W.-J.; Whangbo, M.-H. *Inorg. Chem.* **2010**, *49*, 8578.
- (30) Blaha, P.; Schwarz, K.; Madsen, G. K. H.; Kvasnicka, D.; Luitz, J. *WIEN2k*; Vienna University of Technology: Vienna, 2001 (ISBN 3-9501031-1-2).
- (31) The f electrons of the RE ions are more localized than the d electrons of the TM ions, so the use of a larger U for the RE ion is justified.
- (32) Kuneš, K.; Novák, P.; Diviš, M.; Oppeneer, P. M. *Phys. Rev. B* **2001**, *63*, 205111.
- (33) Ji, S.; Kan, E. J.; Whangbo, M.-H.; Kim, J.-H.; Qiu, Y.; Matsuda, M.; Yoshida, H.; Hiroi, Z.; Green, M. A.; Ziman, T.; Lee, S.-H. *Phys. Rev. B* **2010**, *81*, 094421.
- (34) Xiang, H. J.; Wei, S.-H.; Whangbo, M.-H.; Da Silva, J. L. F. *Phys. Rev. Lett.* **2008**, *101*, 037209.
- (35) Xiang, H. J.; Wei, S.-H.; Whangbo, M.-H. *Phys. Rev. Lett.* **2008**, *100*, 167207.
- (36) Koo, H.-J.; Xiang, H. J.; Lee, C.; Whangbo, M.-H. *Inorg. Chem.* **2009**, *48*, 9051.
- (37) (a) Dai, D.; Whangbo, M.-H. *J. Chem. Phys.* **2001**, *114*, 2887. (b) Dai, D.; Whangbo, M.-H. *J. Chem. Phys.* **2003**, *118*, 29.
- (38) Whangbo, M.-H.; Koo, H.-J.; Dai, D. *J. Solid State Chem.* **2003**, *176*, 417.

- (39) Xiang, H. J.; Lee, C.; Whangbo, M.-H. *Phys. Rev. B: Rapid Commun.* **2007**, *76*, 220411(R).
- (40) Koo, H.-J.; Whangbo, M.-H. *Inorg. Chem.* **2008**, *47*, 128.
- (41) Koo, H.-J.; Whangbo, M.-H. *Inorg. Chem.* **2008**, *47*, 4779.
- (42) Koo, H.-J.; Lee, C.; Whangbo, M.-H.; McIntyre, G. J.; Kremer, R. K. *Inorg. Chem.* **2011**, *50*, 3582.