

# Additions and Corrections

1964, Volume 3

**R. J. Woodruff, James L. Marini, and J. P. Fackler, Jr.:** The Reaction Product of Bis(2,4-pentanediono)titanium(IV) Dichloride with Anhydrous Iron(III) Chloride.

Page 688. In Figure 1, the labeling of curves A and B should be interchanged.—JOHN P. FACKLER, JR.

1965, Volume 4

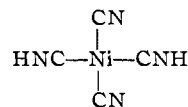
**John L. Burmeister and Monther Y. Al-Janabi:** Selenocyanate Complexes of Cobalt(III), Palladium(II), and Platinum(II).

Page 963. The compounds  $[\text{Pd}(\text{P}(\text{C}_6\text{H}_5)_3)_4]\text{Cl}_2$  and  $[\text{Pt}(\text{P}(\text{C}_6\text{H}_5)_3)_4]\text{Cl}_2$ , reported, on the basis of C, H, and N analyses, to result from the reaction of a warm ethanolic solution of triphenylphosphine with ethanolic solutions of, respectively,  $[\text{Pd}(\text{SeCN})_4]^{2-}$  and  $[\text{Pt}(\text{SeCN})_4]^{2-}$  are, in reality, triphenylphosphine-adulterated samples of triphenylphosphine selenide,  $(\text{C}_6\text{H}_5)_3\text{PSe}$ . The same result is obtained when the reaction between  $[\text{Pt}(\text{SeCN})_4]^{2-}$  and triphenylphosphine is carried out in dimethylformamide, followed by the addition of ice water. Nicpon and Meek have subsequently reported [*Inorg. Chem.*, 5, 1297 (1966)] a general synthesis for tertiary phosphine selenides, wherein the tertiary phosphine is allowed to react with potassium selenocyanate in warm acetonitrile. It should also be noted that when the reaction between  $[\text{Pd}(\text{SeCN})_4]^{2-}$  and triphenylphosphine is carried out in methanol at  $-78^\circ$ , the disubstituted  $[\text{Pd}(\text{P}(\text{C}_6\text{H}_5)_3)_2(\text{SeCN})_2]$  complex is the only product [J. L. Burmeister and H. J. Gysling, *Inorg. Chim. Acta*, 1, 100 (1967)].—JOHN L. BURMEISTER

1968, Volume 7

**Gerald B. Kolski and Dale W. Margerum:** Kinetics of Formation and Dissociation of Tetracyanonickelate(II) Ion.

Page 2243. Formula B should be



and not the isocyanide structure shown.—DALE W. MARGERUM

**R. C. Elder:** Conformations and Crystal Packing. The Crystal and Molecular Structure of *trans*-Bis(2,4-pentanedionato)dipyridinenickel(II),  $\text{Ni}(\text{AA})_2(\text{py})_2$ .

Page 2322. In Table VIII the set of entries in the lower right hand corner should be headed "Ni in Ni stereo" rather than "Ni in Co stereo."—R. C. ELDER

1969, Volume 8

**Gerard R. Dobson:** Octahedral Metal Carbonyls. XI. Reactions of Group VIb Metal Carbonyl Derivatives of 2,2,7,7-Tetramethyl-3,6-dithiaoctane.

Page 91. In column 2, line 2,  $\gtrsim 0.5 M$  should read,  $\lesssim 0.5 M$ .

Page 92. In column 2, in the line of text under eq 7,  $[\text{L}] \gtrsim 0.5 M$  should read  $[\text{L}] \lesssim 0.5 M$ .—GERARD R. DOBSON

**Michael F. Prysak and Melvin D. Joesten:** Metal Complexes of *sym*-Diethyl Tetramethyldiamidopyrophosphate.

Page 1455. The following sentences should be added to the Experimental Section. *Caution!* Both *sym*-diethyl tetramethyldiamidopyrophosphate (SYM) and octamethylpyrophosphoramide (OMPA) are toxic anti-cholinesterase agents. Atropine sulfate is a specific antidote. A detailed description of the toxic properties of these compounds is given by Holmstedt.<sup>6</sup>—MELVIN D. JOESTEN