## X-Ray Crystal Structure of a Platinum-9-Methyladenine Complex

## A. TERZIS, N. HADJILIADIS, R. RIVEST and T. THEOPHANIDES

Department of Chemistry, University of Montreal, Montreal, Quebec, Canada H3C 3V1

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The discovery by Rosenberg and collaborators of the activity of certain platinum compounds as antitumor agents<sup>1</sup> stimulated greatly research in this area.

In our studies<sup>2-4</sup> of the anti-tumor properties of complexes of platinum(II) and their interactions

with purines, nucleosides and nucleotides, we noted that 9-methyladenine reacted with potassium chloroplatinate (K<sub>2</sub>PtCl<sub>4</sub>) in water yielding a deep yellow colored compound. Suitable crystals for X-ray work were obtained by slow evaporation of the acid (3N HCl) solution. We report here the results of the crystal structure analysis of the compound formed having the formula, Pt(9-methyladenine)Cl<sub>3</sub>H<sup>+</sup> (Anal. Found: C, 16.33; H, 1.78; N, 15.96; Pt, 42.76; Cl, 23.01. Calc. C, 15.93; H, 1.77; N, 15.47; Pt, 43.21; Cl, 23.57).

The crystals are triclinic, space group  $P\overline{1}$ ,  $a = 9.495 \pm 0.002$ ,  $b = 9.924 \pm 0.002$ ,  $c = 6.961 \pm 0.001$  Å,  $\alpha = 104.57 \pm 0.03^{\circ}$ ,  $\beta = 93.48 \pm 0.03^{\circ}$ ,  $\gamma = 114.94 \pm 0.03^{\circ}$ , Z = 2, Z = 2.607 g cm<sup>-3</sup>, Z = 2.63 (by flotation).

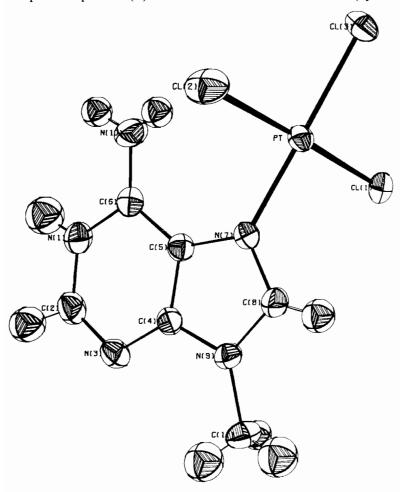


Fig. 1. Molecular structure of Pt(9-methyladenine)Cl<sub>3</sub>H<sup>+</sup>.

The intensities were measured on a Picker automatic diffractometer using graphite-monochromated Mo-radiation ( $\lambda$  = 0.7107 Å) and a  $\theta$ -2 $\theta$  scan mode out to a 2 $\theta$  value of 55°. The position of the platinum was located from a three-dimensional Patterson synthesis. All atoms, including the hydrogens, were located by two Fourier syntheses. Using block-diagonal least-squares refinements, anisotropic temperature factors for the non-hydrogen atoms and isotropic for the hydrogens the R index converged to a present value of 0.025.

As can be seen in the Figure the Platinum is coordinated to the  $N_7$  site and the  $N_1$  hydrogen is involved in a hydrogen bond with the Cl(1) of a centrosymmetrically related molecule, forming in this way weakly bonded dimers which are held together by Van der Waals forces.

We believe that the platinum—9-methyladenine structure is the first to show coordination of a

platinum ion to the  $N_7$  site<sup>3</sup>. This site is a potential target for interaction with platinum ions in DNA<sup>5</sup> whereas the  $N_1$  site is directly involved in hydrogen bonding.

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## References

- B. Rosenberg, L. Van Camp, J. E. Trosko and V. H. Mansour, *Nature*, 222, 385 (1969).
- N. Hadjiliadis, P. Kourounakis and T. Theophanides, Inorg. Chim. Acta, 11, 179 (1973).
- 3 P. C. Kong and T. Theophanides, *Inorg. Chem.*, 13, 1167 (1974).
- 4 P. C. Kong and T. Theophanides, Inorg. Chem., in press.
- 5 J. P. Macquet and T. Theophanides, to be published.