

Preparation and Suggested Structure of a Binuclear Palladium Complex containing Two Cyclopentadienyl Groups

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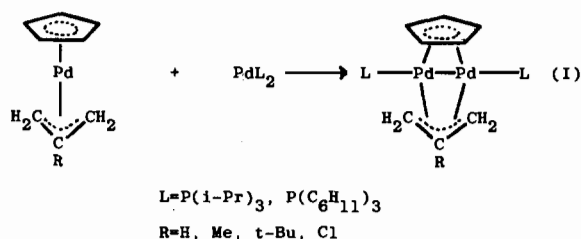
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A cyclopentadienyl group is one of the most important ligands in organometallic chemistry. Recently Werner *et al.* [1] and Turner *et al.* [2] have synthesized new dimeric palladium(I) complexes which contain a cyclopentadienyl group coordinated to the Pd(I)–Pd(I) bond. Further, Werner *et al.* [3] presented an interesting synthetic method which led to formation of complexes of type (I).

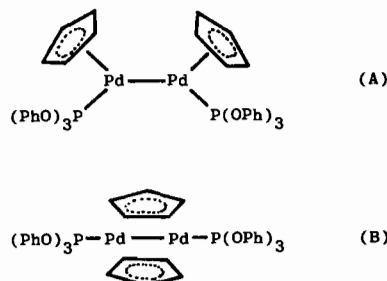


However this method is not suitable for the synthesis of a complex containing two bridging cyclopentadienyl groups, because starting material palladocene has not been obtained yet. Turner *et al.* [4] also tried the reaction of $\mu-(\eta^3-C_5H_5)-\mu-BrPd_2L_2$ ($L = PPh_3, P(i-Pr)_3$) with TiC_5H_5 in an attempt to replace the bridging bromine atom with a cyclopentadienyl group without success. In this paper we describe the preparation of $[\mu-(C_5H_5)PdP(OPh)_3]_2$ (II).

To a solution of 1.7 g of NaC_5H_5 in 200 ml of tetrahydrofuran cooled to 0 °C under nitrogen was added with stirring 4.3 g of $Pd_2Cl_4 \{P(OPh)_3\}_2$ in 200 ml of tetrahydrofuran. After a short time the solvent was distilled off under vacuum and the residue was washed with ether and water and recrystallised from $CHCl_3$ -*n*-pentane to give a yellow crystalline compound (II). *Anal.* Found: C, 57.4; H, 4.7; P, 6.5. Required for $C_{46}H_{40}O_6Pd_2P_2$: C, 57.3; H, 4.2; P, 6.4.

Mol. wt. Found: 1060. Required for $C_{46}H_{40}O_6Pd_2P_2$: 963.

The analytical data and molecular weight of the complex (II) strongly indicate that (II) is a dimeric palladium complex $[\mu-(C_5H_5)PdP(OPh)_3]_2$ for which two structures (A) and (B) are possible.



We can distinguish between (A) and (B) by the 1H nmr spectrum of the cyclopentadienyl protons which couple with the two phosphorus atoms. Thus we can expect a doublet of doublets for (A) and a triplet for (B). As Fig. 1 shows, the cyclopentadienyl and the phenyl protons appear as a triplet (τ 4.93) and a singlet (τ 2.7) respectively. Their intensity ratio 1:3:2 ($C_5H_5:P(OPh)_3$) is in agreement with the formulation of (II). These results allowed us to assign the structure (B) – the structure containing two bridging C_5H_5 groups – to the complex (II). The coordination mode of the C_5H_5 groups is not clear at present: recently, Turner *et al.* [2] showed that the C_5H_5

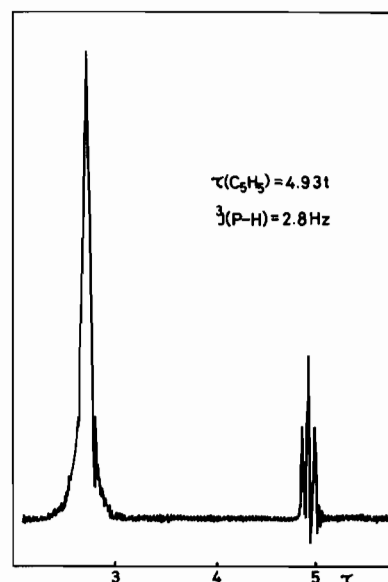


Fig. 1. 1H nmr spectrum of $[\mu-(C_5H_5)PdP(OPh)_3]_2$.

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bridge in $\mu-(\eta^3\text{-C}_5\text{H}_5)\text{-}\mu\text{-BrPd}_2\{\text{P}(\text{i-Pr})_3\}_2$ can be regarded as an allyl plus alkene group. Further, a complex incorporating two bridging allyl groups, $[\mu-(\eta^3\text{-C}_3\text{H}_4)\text{PdP}(\text{i-Pr})_3]_2$, has been synthesised [4]. Therefore, it might be reasonable to assume that the C_5H_5 groups in (II) behave as bridging allyl groups. The reactions of (II) with other compounds are under investigation.

References

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