

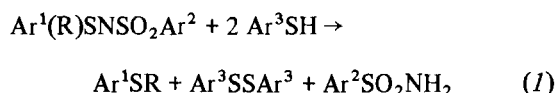
Organic Chemistry

The Solvent Effect in the Reaction of N-Sulphonyl-sulphilimines with Arenethiols in Alcohol

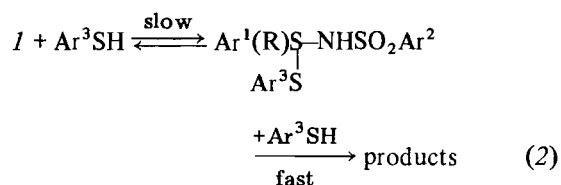
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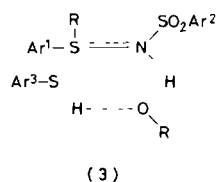
A detailed study of the solvent effect and of activation parameters indicates that most likely the mechanism of sulphilimine reduction by arenethiols in alcohol [1] thus



is as follows



in which the formation of 2 occurs through a cyclic transition state 3 involving the participation of one molecule of sulphilimine, one of thiol and one of alcohol



A quantitative analysis of the solvent effect through multiparametric equations evidences the relative weight of the specific and non specific solute-solvent interactions. For $\text{Ar}^1 = \text{R} = \text{CH}_3$, $\text{Ar}^2 = p\text{-CH}_3\text{C}_6\text{H}_4$, $\text{Ar}^3 = \text{C}_6\text{H}_5$ the second order rate constants satisfy the following equations

$$\log k = -7.11 + 12.99(\pm 2.92)f(\epsilon) -$$

$$-1.216(\pm 0.415)\sigma^* + 0.552(\pm 0.094)E_s^c$$

$$(r = 0.947, s = 0.184) \text{ (Chapman treatment [2])}$$

$$\log k = -4.11 + 1.03(\pm 0.21)\pi^* + 2.88(\pm 0.17)\alpha$$

$$(r = 0.993, s = 0.079) \text{ (Kamlet and Taft treatment [3])}$$

References

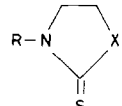
- 1 G Guanti, G Garbarino, C Dell'Erba and G Leandri, *Gazzetta*, 105, 849 (1975)
- 2 N B Chapman, M R J Dack and J Shorter, *J Chem Soc (B)*, 834 (1971)
- 3 M J Kamlet and R W Taft, *J Chem Soc Perkin II*, 337, 349 (1979),
M J Kamlet, M E Jones and R W Taft, *ibid.*, 342 (1979)

1 1 Molecular Adducts between I₂ and Tio- (or Seleno-) amido Group Contained in Some Heterocyclic Rings

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The donor properties of Y (= S or Se) in RN·CY·X group are affected by the nature of X. This fact has been pointed out by the stability constants (K) of the 1 1 molecular adducts between I₂ and the above group contained in the following pentatomic rings [1, 2]

	R	X	R	X
	H	CH ₂	H	NMe
	H	O	H	NEt
	H	S	Me	S
	H	NH	Me	NMe
			Et	NEt