

absorption band in the visible region for solutions containing the cryptate and the quencher. Spectrophotometric and electrochemical analysis show that 1:1 complexes are formed with stability constants of the order of  $10^2 \text{ l mol}^{-1}$ .

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## B17

### Dependence of the $^5D_0 \rightarrow ^5D_1$ Fluorescence Dynamics Versus the Optical Pumping Frequency in $\text{Sm}^{2+}:\text{BaClF}$

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Very recently, the role of the  $^5D_0 \rightarrow ^5D_1$  upwards radiationless transition in the de-excitation mechanisms of the  $\text{Sm}^{2+}$  ion in  $\text{BaClF}$  at temperatures ranging from 295 to 650 K was clearly established [1]. Using selective pulsed laser excitation into the first metastable excited state  $^5D_0$  ( $14533 \text{ cm}^{-1}$ ), the fluorescence of the  $^5D_1$  multiplet located at  $1339 \text{ cm}^{-1}$  above  $^5D_0$  was detected. Both  $^5D_1$  and  $^5D_0$  fluorescence at long times after the laser pulse were observed to decay exponentially with the same time constant ( $\sim 1.5 \text{ ms}$  at room temperature) characteristic of the  $^5D_0$  lifetime in this material, as shown in Figs. 1a) and b). This experiment confirms that a thermalization process between levels  $^5D_0$  and  $^5D_1$  takes place in this system, as observed 16 years ago using an ultraviolet excitation promoting the  $\text{Sm}^{2+}$  ion up to  $4f^5-5d$  states [2, 3]. Surprisingly enough, the  $^5D_0$  and  $^5D_1$  fluorescence decays under excitation into the  $^5D_1$  level do not bear evidence of such a thermalization process, as shown in Figs. 1c) and d), the  $^5D_1$  fluorescence being observed to decay exponentially with a time constant equal to the intrinsic  $^5D_1$  lifetime ( $\sim 150 \mu\text{s}$  at room temperature). Moreover, all the measurements we performed in previous works pumping the system either in the  $^5D_1$  or  $^5D_2$  level were found to be consistent with the assumption that the  $^5D_0 \rightarrow ^5D_1$  radiationless transition did not contribute signi-

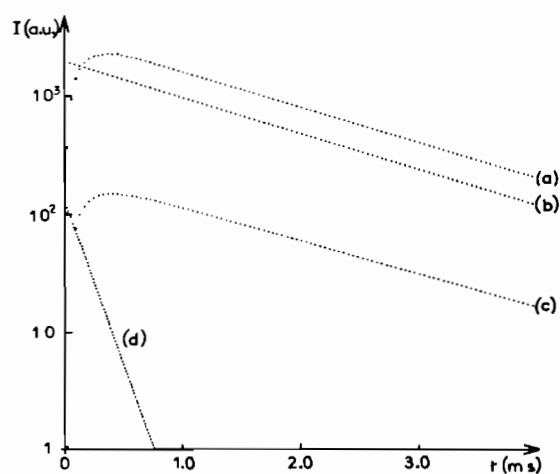


Fig 1. Fitted decay-curves of the  $^5D_0$  (b, c) and  $^5D_1$  (a, d) fluorescences under selective excitation into the  $^5D_0$  (a, b) and  $^5D_1$  (c, d) levels.

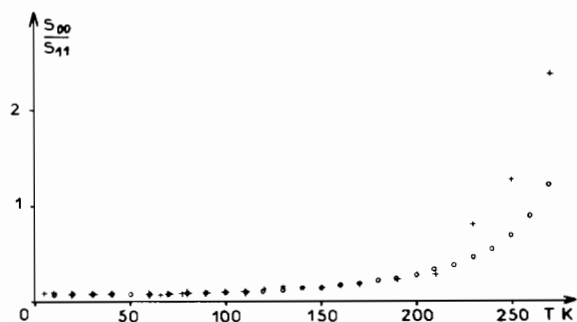


Fig. 2. Comparison between the temperature dependences of the ratio  $S_{00}/S_{11}$  of the integrated intensities of the  ${}^5D_0 \rightarrow {}^7F_0$  emission to the  ${}^5D_1 \rightarrow {}^7F_1$  one in the spectra recorded under pulsed selective excitation into the  ${}^5D_1$  level (o) and continuous ultraviolet excitation (+).

ificantly to the optical properties of the  $\text{Sm}^{2+}:\text{BaClF}$  system at temperatures up to 300 K [4–6]. The physical explanation for these discrepancies lies in the fact that the feeding of levels  ${}^5D_0$  and  ${}^5D_1$  during the excitation pulse depends strongly on the pumping conditions [1]. Obviously, the role of the  ${}^5D_0 \rightarrow {}^5D_1$  radiationless transition will be enhanced if the population of the  ${}^5D_0$  state at the end of the pulse is significant, as is the case for the  ${}^5D_0$  pumping. On the other hand, the contribution of this transition will be reduced if the  ${}^5D_0$  population remains equal to zero during the excitation pulse, as is the case for the  ${}^5D_1$  or  ${}^5D_2$  pumping. Under ultraviolet excitation, the feeding of the  ${}^5D_0$  state during the pulse is ensured by fast non radiative processes connecting the  $4f^5-5d$  states directly to the  ${}^5D_0$  state at temperatures above 200 K as it clearly appears from the comparison between the temperature dependences of the ratio  $S_{00}/S_{11}$  of the integrated intensities of the  ${}^5D_0/{}^7F_0$  emission to the  ${}^5D_1 \rightarrow {}^7F_1$  one in the spectra recorded under selective excitation into the  ${}^5D_1$  level on one hand, and ultraviolet excitation on the other hand (Fig. 2).

Thus, the thermalization near room temperature between two rare-earth levels separated by an energy gap of several times the maximum phonon frequency in the crystal ( $294\text{ cm}^{-1}$  in  $\text{BaClF}$ ) [6] may be observed or not, depending on the optical pumping frequency. Similar investigations in other rare-earth systems such as  $\text{Eu}^{3+}$  doped materials, with larger  ${}^5D_1-{}^5D_0$  gaps and higher  $4f^5-5d$  state energies should be of interest to complement this observation.

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## B18

### Mechanisms for f–f Transition Probabilities in Lanthanide Coordination Compounds

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The Laporte-forbidden transitions of  $\text{Ln(III)}$  coordination compounds acquire a first-order electric-dipole probability from transient dipoles induced in the ligand groups by an allowed even-multipole electric moment of the f–f excitation, and by the mixing of the f–f with f–d and f–g electron promotions under the electrostatic field of the ligands. Applied initially to  $\text{Ln(III)}$  complexes containing monoatomic ligands which have an effective isotropic polarizability the ligand-polarization mechanism is found to depend, on extension to the corresponding polyatomic ligand cases, upon the anisotropy of the ligand polarizability tensor in complexes belonging to the higher non-centric symmetries. The electrostatic field and the ligand polarization mechanisms make complementary intensity contributions to the f–f transitions of a given  $\text{Ln(III)}$  complex type, dependent upon the rank of the leading electric multipole moment. The polarization mechanism contributes principally to the intensities of the ligand-hypersensitive  $2^2$ -pole f–f transitions, whereas the electrostatic mechanism is predominant for the  $2^6$ -pole transition intensities, and makes the more important contribution in the  $2^4$ -pole cases.

## B19

### Absorption and Fluorescence Spectra of Europium(III) Compounds in Non-Aqueous Solutions

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The absorption and fluorescence spectra of  $\text{Eu(III)}$  compounds were investigated at room temperature in different non-aqueous solutions.