

New Photo-products in the High-intensity Solution Photolysis of Octacyanotungstate(V) Ion: Definition of a High-energy Photochemical Channel

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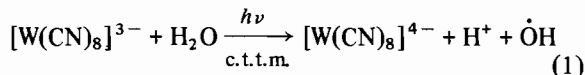
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Photolysis of $[\text{W}(\text{CN})_8]^{3-}$ ion in water [1–3] and non-aqueous solvents [4] normally produces $[\text{W}(\text{CN})_8]^{4-}$ in high yield ($\phi = c. 0.8$ in water [3]) via the photoredox process (in water)



One of us (B.S.) has reported previously [5] that under conditions of *high* light intensity, a new pathway emerges, yielding a red–purple solution (λ_{max} 537 nm). Photolysis in water–methanol (1:1 vol./vol.) led to conversion of the 537 nm product to a deep violet material (λ_{max} 560 nm). Both of these diamagnetic photo-products were attributed [5] to dioxygen complexes of W^{IV} .

We now report that, following precipitation procedures with $\text{PPh}_4^+\text{Cl}^-$, we have isolated from the aqueous photolysate containing the mixture of photo-products, after high-intensity photolysis (focused output of a 200 W Xe/Hg lamp) of a 0.01 mol dm^{-3} solution of $\text{K}_3[\text{W}(\text{CN})_8]$, the almost colourless (palest pink) W^{VI} product $(\text{PPh}_4)_2[\text{WO}(\text{CN})_6] \cdot \text{H}_2\text{O}$ (1), and from methanol–water the analogous almost colourless (palest violet) $(\text{PPh}_4)_2[\text{WO}(\text{CN})_5(\text{OMe})]$ (2). (Both complexes gave analytical and spectral data consistent with their structures. Selected IR data for 1 (KBr): $\nu(\text{CN})$ 2206, 2159, 2143; $\nu(\text{W}=\text{O})$ 963 cm^{-1} and for 2 (KBr): $\nu(\text{CH})$ 2920, 2825; $\nu(\text{CN})$ 2206, 2164; $\nu(\text{W}=\text{O})$ 933 cm^{-1} .)

The structures of both of these complexes (after recrystallisation from methanol) have been obtained by X-ray diffraction, both anions being pentagonal bipyramidal (Fig. 1(a) and (b)) with all equatorial positions occupied by CN^- ligands. (Crystal data for

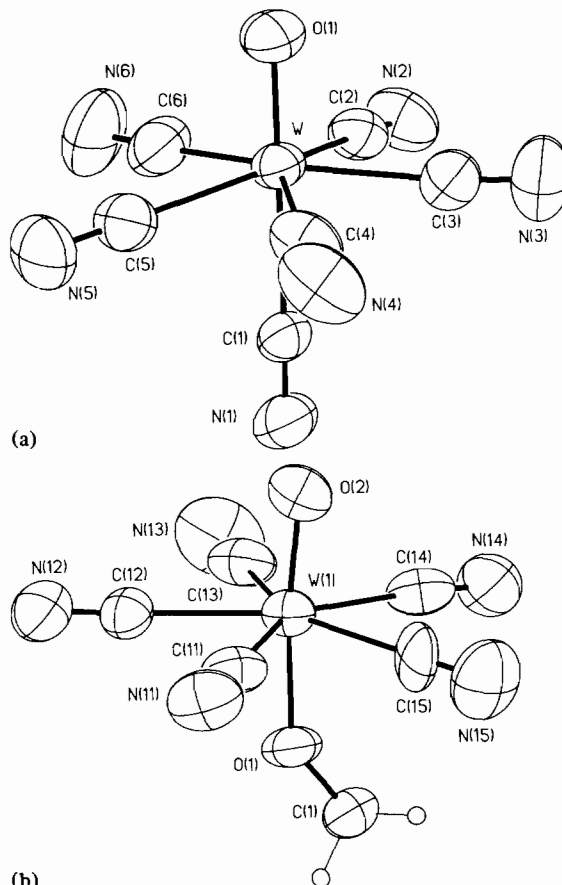


Fig. 1. (a) View of the anion $[\text{WO}(\text{CN})_6]^{2-}$ in crystalline $[\text{PPh}_4]_2[\text{WO}(\text{CN})_6]$. H_2O isolated from the high-intensity photolysate of $[\text{W}(\text{CN})_8]^{3-}$ in water. Selected average distances and angles are: $\text{W}=\text{O}$ 1.691(3), $\text{W}-\text{C}_{\text{ax}}$ 2.277(3), $\text{W}-\text{C}_{\text{eq}}$ 2.150(5) Å (mean), $\text{O}(1)-\text{W}-\text{C}_{\text{ax}}$ 177.1°. (b) View of one of the two independent anions $[\text{WO}(\text{CN})_5(\text{OMe})]^{2-}$ in crystalline $[\text{PPh}_4]_2[\text{WO}(\text{CN})_5(\text{OMe})]$ isolated from the high-intensity photolysate of $[\text{W}(\text{CN})_8]^{3-}$ in water–methanol (1:1). Selected average distances and angles are: $\text{W}=\text{O}(2)$ 1.700(8), $\text{W}-\text{O}(1)$ 1.906(8), $\text{W}-\text{C}_{\text{eq}}$ 2.172(12) Å (mean), $\text{W}-\text{O}(1)-\text{C}(1)$ 143.1(8) and 149.9(8)°.

1: triclinic, $P\bar{1}$, with $a = 10.247(2)$, $b = 14.510(3)$, $c = 16.394(3)$ Å, $\alpha = 103.19(2)$, $\beta = 90.78(1)$, $\gamma = 92.21(2)^\circ$, $U = 2307.8(7)$ Å³, $Z = 2$, $R = 0.028$ for 6785 unique, observed reflections ($I/\sigma(I) > 3.0$). Crystal data for 2: triclinic, $P\bar{1}$, $a = 10.713(3)$, $b = 19.208(5)$, $c = 24.787(6)$ Å, $\alpha = 99.17(2)$, $\beta = 90.64(2)$, $\gamma = 99.40(2)^\circ$, $U = 4964(2)$, $Z = 4$, $R = 0.051$ for 6865 unique observed reflections ($I/\sigma(I) > 2$.) Their geometry conforms to that of such anions as $[\text{V}(\text{CN})_6(\text{NO})]^{4-}$ [6, 7], $[\text{Mo}(\text{CN})_7]^{5-}$ [8], $[\text{Mo}(\text{CN})_7]^{4-}$ [9] and $[\text{Re}(\text{CN})_7]^{4-}$ [10].

The chief mechanistic interest lies in the unexpected production of W^{VI} : this contrasts with the

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normal route, eqn. (1) and implies either disproportionation of a W^V intermediate or a high-energy route. Detailed quantum yield investigation of the formation of the (still uncharacterised) 537 nm photo-product indicates that this is formed in a (probably *sequential*) biphotonic process [11], with a formation rate dependent on the *square* of the incident light intensity, when $360 \text{ nm} > \lambda > 310 \text{ nm}$. (The species absorbing the second photon can be either a long-lived excited state or a reactive intermediate.) Since production of the crystalline W^{VI} species described occurs in parallel with those of the 537 and 560 nm photo-products, we attribute formation of these unusual W^{VI} photo-products also to a biphotonic route.

Such two-quantum processes, while common in organic photochemistry, are rare in solution inorganic photochemistry, recent instances being (i) the photoaquation of $[\text{Cr}(\text{phen})_3]^{3+}$ (phen = 1,10-phenanthroline) [12]; (ii) in luminescence quenching, at high excitation intensities, of a polypyridyl complex of Ru^{II} by a polytungstate anion [13]; (iii) the excited state disproportionation of $[\text{Ru}(\text{bipy})_3]^{2+}$ (bipy = 2,2'-bipyridyl) [14].

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