

Zinc, Cobalt and Cadmium Thiolate Complexes: Models for the Zn(S-cys)₃(his) Center in the Gene 32 Protein

DOUGLAS T. CORWIN, JR., ERIC S. GRUFF
and STEPHEN A. KOCH*

Department of Chemistry, State University of New York at
Stony Brook, Stony Brook, N.Y. 11794, U.S.A.

(Received June 12, 1987)

In the past two years, a new class of zinc metallo-proteins has been established [1-6]. A number of DNA and RNA binding proteins have been shown to contain zinc and to be characterized by Zn(S-cys)_x coordination. In particular, it has recently been shown that the gene 32 protein (g32P) isolated from bacteriophage T4 contains a mole of tightly bound zinc. Inspection of the amino-acid sequence of the protein and the spectroscopic properties of the Co(II) and Cd(II) substituted protein led Coleman to propose a [Zn(S-cys)₃(his)] coordination unit for this protein [4, 5]. The coordination of three cysteines to a monomeric zinc center has also been suggested to occur in the enzyme aminolevulinate dehydratase [7]. We wish to report the synthesis, structure, and spectroscopic properties of a series of Zn, Co, and Cd complexes that reproduce this proposed coordination unit.

The reactions of MCl₂ (M = Zn, Co, Cd) with 3 equivalents of LiS-2,3,5,6-Me₄C₆H₃, 1 equivalent of 1-methylimidazole and 1 equivalent of [(*n*-Pr)₄N]Br gives [(*n*-Pr)₄N][M(S-2,3,5,6-Me₄C₆H₃)(1-Me-imid)] in 60-70% yields. The complexes are recrystallized from CH₃CN to give large blue green crystals for Co (1) and colorless crystals for Zn (2) and Cd (3). X-ray crystallographic studies have shown that the [(*n*-Pr)₄N][M(S-2,3,5,6-Me₄C₆H₃)(1-Me-imid)] complexes of cobalt and cadmium are isomorphous. The X-ray structure of the Co complex was determined (Fig. 1). Crystal data: triclinic, space group *P* $\bar{1}$, *a* = 12.321(5), *b* = 18.038(4), *c* = 11.852(3) Å, α = 97.95(2), β = 113.36(2), γ = 76.44(2)°, *V* = 2348(2) Å³, *Z* = 2. *R*(*R*_w) = 0.051(0.063) for 2599 unit reflections with *I* > 3σ*I*. The cobalt is coordinated by three thiolate ligands and the nitrogen of the 1-Me-imid. The average Co-S distance of 2.300(6) Å is similar to that in the related [Co(S-2,3,5,6-Me₄C₆H₃)(CH₃CN)]⁻ anion [8]; the Co-N distance is 2.051(5) Å. The [CoS₃N] core shows small distortions for tetrahedral coordination with the L-M-L angles lying in the range of 105 to 114°.

* Author to whom correspondence should be addressed.

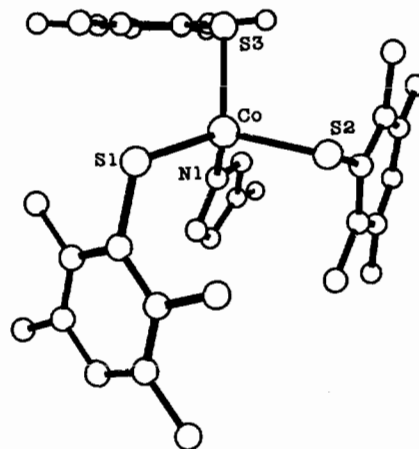


Fig. 1. A ROTOCHEM diagram of the anion of [(*n*-Pr)₄N][Co(S-2,3,5,6-Me₄C₆H₃)(1-Me-imid)]. Bond distances (Å): Co-S1 2.299(2); Co-S2 2.307(2); Co-S3 2.294(2); Co-N1 2.051(5). Bond angles (deg): S1-Co-S2 112.75(7); S1-Co-S3 114.51(7); S1-Co-N1 110.5(2); S2-Co-S3 107.41(8); S2-Co-N1 105.4(1); S3-Co-N1 105.6(2).

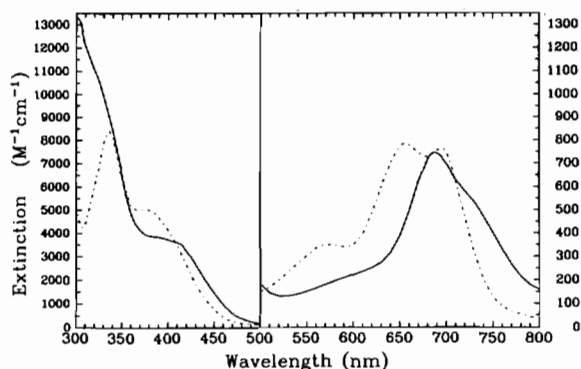


Fig. 2. The electronic spectra (CH₃CN) of [(*n*-Pr)₄N][Co(S-2,3,5,6-Me₄C₆H₃)(1-Me-imid)] (—) and [Co(S-2,4,6-*i*-Pr₃C₆H₂)₂(1-Me-imid)₂] (---).

The electronic spectrum of [Co(S-2,3,5,6-Me₄C₆H₃)(1-Me-imid)]⁻ (Fig. 2) is consistent with a tetrahedral Co(II) center with thiolate coordination [9, 10]. The intense transitions at 390(sh) nm [ϵ = 3830 M⁻¹ cm⁻¹]; 322(sh), [10900] and 301, [13400] are characteristic of RS to Co(II) charge transfer transitions. The ligand field transitions occur at 730(sh) nm, [ϵ = 535]; 687, [745]; and 590(sh), [210]. Although the spectrum of 1 is quite similar to that of Co substituted g32P, this does not definitively imply a congruence in the coordination units of model and the protein. The visible spectrum of 1, although differing in detail, is similar to the spectrum of the corresponding [Co(SR)₂(1-Me-imid)₂] complexes [11]. The ¹H NMR spectra of the Zn and the

Cd complexes are consistent with their formulation. The ^1H NMR spectrum (CD_3CN) of 1 shows paramagnetically shifted resonances in the region of +40 to -40 ppm; all of the individual resonances can be assigned by comparison to related compounds. The resonances of the 1-methylimidazole ligand occur at 38.4 δ (1H), 5-H; 4.02 (3H), 1- CH_3 ; and two broad resonances at 36.7 and 34.2 which are assigned to the 2 and 4 protons. The chemical shifts of the thiolate ligands occur at 28.9 (18H), 2- and 6- CH_3 ; -9.9 (18H), 3- and 5- CH_3 ; -30.7 (3H), 4-H. The ^{113}Cd NMR resonance of the Cd complex occurs at 533 ppm; a chemical shift which is consistent with the $\text{Cd}(\text{S}_3\text{N})$ coordination. The ^{113}Cd spectra of the gene 32 protein has not been reported.

Acknowledgements

This research was supported by the National Institute of Health Grant GM-31849.

References

- 1 J. M. Berg, *Science*, **232**, 485 (1986).
- 2 G. P. Diakun, L. Fairall and A. Klug, *Nature (London)*, **324**, 698 (1986).
- 3 L. Fairall, D. Rhodes and A. Klug, *J. Mol. Biol.*, **192**, 577 (1986).
- 4 D. P. Giedroc, K. M. Keating, K. R. Williams, W. H. Konigsberg and J. E. Coleman, *Proc. Natl. Acad. Sci. U.S.A.*, **83**, 8452 (1986).
- 5 D. P. Giedroc, K. M. Keating, C. T. Martin, K. R. Williams and J. E. Coleman, *J. Inorg. Biochem.*, **28**, 155 (1986).
- 6 E. Wingender and K. H. Seifart, *Angew. Chem., Int. Ed. Engl.*, **26**, 218 (1987).
- 7 S. S. Hasnain, E. M. Wardell, C. D. Garner, M. Schlösser and D. Beyersmann, *Biochem. J.*, **230**, 625 (1985).
- 8 S. A. Koch, R. Fikar, M. Millar and T. O'Sullivan, *Inorg. Chem.*, **23**, 121 (1984).
- 9 A. B. P. Lever, 'Inorganic Electronic Spectroscopy', Elsevier, Amsterdam, 1984, pp. 480-505.
- 10 L. Banci, A. Bencini, C. Benelli, D. Gatteschi and C. Zanchini, *Struct. Bonding (Berlin)*, **52**, 37 (1982).
- 11 D. T. Corwin, Jr., E. S. Gruff and S. A. Koch, *J. Chem. Soc., Chem. Commun.*, 966 (1987).