

Peculiar Fluorescence of Eu^{3+} in Oxyapatites*B. PIRIOU[†], D. FAHMI, J. DEXPERT-GHYS*Laboratoire des Eléments de Transition dans les Solides, C.N.R.S., 1 place Aristide Briand, F-92195 Meudon Cédex, France*

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The continuous solid solution $\text{Ca}_{10-x}\text{Eu}_x(\text{PO}_4)_6(\text{O}_{1+x/2}\square_{1-x/2})$, with $0.05 \leq x \leq 2$, was studied by the Eu^{3+} fluorescence at 300 and 77 K. The emission mainly arises from the $^5\text{D}_0$ level. The dependence upon the composition shows that Eu^{3+} ions occupy

preferentially site II. Eu^{3+} ions located in site I are only characterized for $x > 1$. Due to the occurrence of vacancies the lines are broad.

The site selective laser excitation of Eu^{3+} in site II allows us to classify them into two families. A structural model is proposed as explanation.

The spectra are unusual because of the high value of the $^5\text{D}_0$ level ($\approx 17\,500\text{ cm}^{-1}$), the large $^7\text{F}_1$ splitting and the very strong $^5\text{D}_0 \rightarrow ^7\text{F}_0$ emission line. The covalent character of the $\text{Eu}^{3+}-\text{O}^{2-}$ bond correlated to a strong crystal field of nearly $C_{\infty v}$ symmetry allows us to assign the $^5\text{D}_0 \rightarrow ^7\text{F}_{1,2}$ lines. Discussion of this exceptional behavior of Eu^{3+} is given in terms of J -mixing.

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