

$(\eta^3\text{-P}_3)\text{W}(\text{OCH}_2\text{Bu}^t)_3(\text{HNMe}_2)$ Formed by Cleavage of the $\text{W}\equiv\text{W}$ Bond in the Reaction Between $\text{W}_2(\text{OCH}_2\text{Bu}^t)_6(\text{HNMe}_2)_2$ and P_4

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Recent work has resulted in the synthesis and structural characterization of compounds of empirical formula $(\text{RO})_3\text{M}\equiv\text{X}$ where $\text{M} = \text{Mo}$, $\text{R} = \text{Bu}^t, \text{Pr}^i, \text{CH}_2\text{Bu}^t$ and Et and $\text{X} = \text{N}$ [1], or $\text{M} = \text{W}$, $\text{R} = \text{Bu}^t$, and $\text{X} = \text{N}$ or CR' where $\text{R}' = \text{alkyl}$ or aryl [2]. These findings led us to examine the reactivity of P_4 towards $\text{W}_2(\text{OR})_6$ compounds [3] in an attempt to prepare a related $(\text{RO})_3\text{W}\equiv\text{P}$ compound. We describe here our discovery of the first $\eta^3\text{-P}_3$ metal complex supported by π -donor ligands, namely alkoxides, which is formed in the reaction between $\text{W}_2(\text{OCH}_2\text{Bu}^t)_6(\text{HNMe}_2)_2$ [3] and P_4 .

The reaction between $\text{W}_2(\text{OR})_6$ compounds and P_4 (purified by sublimation) in dry and deoxygenated hydrocarbon solvents (N_2 atmospheres) is sensitive to the steric requirements of R . Reactions employing $\text{W}_2(\text{OCH}_2\text{Bu}^t)_6(\text{HNMe}_2)_2$, which acts as a source of the coordinatively unsaturated compound $\text{W}_2(\text{OCH}_2\text{Bu}^t)_6$, and P_4 gave a mixture of two products. One, a brown compound which crystallizes from CH_2Cl_2 solutions, displays an ambiguously simple ^1H NMR spectrum (two CMe_3 singlets in the integral ratio of 2:1 and an AB quartet and singlet for their respective methylene protons) but contains no detectable phosphorus by either ^{31}P NMR or elemental analysis. Efforts to characterize this product by single crystal X-ray diffraction have as yet been unsuccessful due to problems associated with twinning, high symmetry space groups, and no diffraction beyond $2\theta = 15^\circ$. The other compound, which is more soluble and can subsequently be crystallized from hexanes

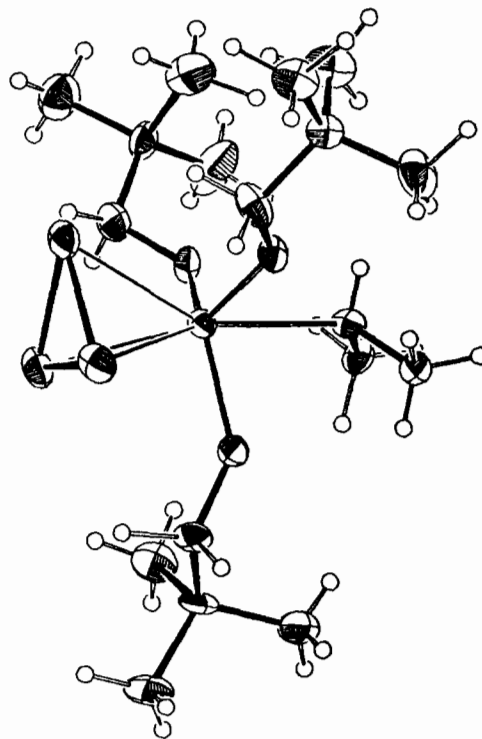


Fig. 1. ORTEP diagram of $(\eta^3\text{-P}_3)\text{W}(\text{OCH}_2\text{Bu}^t)_3(\text{HNMe}_2)$. Bond lengths (Å) and angles ($^\circ$) (averaged where appropriate) are: W–P, 2.4930(25); W–O, 1.908(5); W–N, 2.315(7); P–P, 2.155(4); P–W–P, 51.23(9); P–W–O (proximal), 88.34(17); P–W–O (distal), 134.15(17); W–O–C, 136.6(5); P–W–N, 150.03(18); O–W–O, 114.17(24); O–W–N, 75.80(24); W–P–P, 64.39(9).

as yellow prisms, is reliably formulated as $(\eta^3\text{-P}_3)\text{W}(\text{OCH}_2\text{Bu}^t)_3(\text{HNMe}_2)^{**}$. The yield of this latter compound (as determined by ^1H NMR) is *ca.* 30% based on tungsten.

An ORTEP diagram of the molecule is given in Fig. 1. If the $\eta^3\text{-P}_3$ ligand is counted as occupying a single coordination site X, then the WO_3XN moiety can be viewed as a trigonal bipyramid with X and N occupying the axial sites. The tungsten atom is displaced 0.47 Å from the O_3 plane toward the P_3 ring. The phosphorus atoms of the $\eta^3\text{-P}_3$ ligand are nearly perfectly staggered with respect to the three W–O bonds in the equatorial plane. The P–P distance of 2.155(4) Å is slightly shorter than that found in P_4 but at the upper end of the range seen for $\eta^3\text{-P}_3$ compounds containing only σ -donor ancillary ligands [4].

The ^1H and ^{31}P NMR spectra of the new compound are simple and consistent with expectations based on the solid-state molecular structure. There is

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**Crystal data for $(\eta^3\text{-P}_3)\text{W}(\text{OCH}_2\text{Bu}^t)_3(\text{HNMe}_2)$ at -155°C : $a = 10.444(3)$, $b = 17.171(8)$, $c = 14.619(6)$ Å, $\beta = 105.02(2)^\circ$, $Z = 4$, $D_c = 1.530$ gm cm^{-3} , space group $P2_1/n$. Of the 4422 unique intensities collected using Mo $\text{K}\alpha$ radiation ($6^\circ \leq 2\theta \leq 45^\circ$), the 2716 having $F > 3\sigma(F)$ were used in a full matrix least-squares refinement. The hydrogen atoms were located and refined and no absorption or extinction corrections were used. Final residual values are $R = 0.0335$ and $R_w = 0.0340$.

one type of neopentoxide ligand and one type of amine [^1H NMR, benzene- d_6 , 22° , δ 0.85 (s, 27 H), 2.41 (s, br, 6 H), 3.00 (s, 1 H), 4.52 (s, 6H)] which requires rapid rotation about the W–N σ bond. In the $^{31}\text{P}(^1\text{H})$ spectrum the $\eta^3\text{-P}_3$ ligand resonates at $\delta -205.2$ ppm (relative to 85% H_3PO_4) with a small coupling to ^{183}W ($I = 1/2$, 14.5% natural abundance): $^1J(^{183}\text{W}-^{31}\text{P}) = 16$ Hz. The small value for $^1J(^{183}\text{W}-^{31}\text{P})$ is understandable in terms of the small degree of phosphorus 3s orbital contribution to the $\eta^3\text{-P}_3\text{-to-W}$ bond. A Fenske–Hall molecular orbital calculation indicates that the molecular orbitals of a and e symmetry for the $\eta^3\text{-P}_3\text{-to-W}$ moiety are composed of phosphorus 3p and tungsten $5d_{z^2}$, d_{xz} and d_{yz} orbitals while the phosphorus 3s orbitals are used predominantly to form P–P bonds. Similarly small values of $^1J(\text{M}-^{31}\text{P})$ for $\eta^3\text{-P}_3$ complexes of platinum and rhodium have been noted [4a].

Further studies are in progress.

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