

Energy Levels of Pr^{3+} in the Modulated Incommensurate Structure of ThCl_4

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Abstract

The matrix $\beta\text{-ThCl}_4$ which presents an incommensurate modulated structure at low temperature is doped with the lanthanide ion Pr^{3+} . The strong features in the absorption and the laser-excited fluorescence spectra have been interpreted as arising from levels of Pr^{3+} in the site of D_{2d} symmetry. In a least-squares fit calculation, the best parameters are determined.

Introduction and Review of some Results Obtained on the Matrix ThCl_4

Recently there has been great interest in studying the optical spectra of U^{4+} ($5f^2$) in the ThBr_4 [1–3] and ThCl_4 [4] matrices. $\beta\text{-ThBr}_4$ and $\beta\text{-ThCl}_4$ both undergo a second order structural phase transition at 95 and 70 K respectively. At low temperature, Raman [5], nuclear quadrupolar resonance [6], and neutron diffraction experiments [7, 8] reveal that the tetragonal structure is incommensurate, resulting in a loss of periodicity along the c axis. At room temperature, the matrix $\beta\text{-ThCl}_4$ has a tetragonal structure ($I4_1/amd$) [9] and is isostructural with $\beta\text{-ThBr}_4$. The thorium ion is at a site of relatively high symmetry D_{2d} . At low temperature the site symmetry of the thorium ion or of its substitute is reduced. For $\beta\text{-ThCl}_4$ that is presently being studied by neutron diffraction experiments, those symmetries are C_2 , D_2 and S_4 [8] as a consequence of the continuous modulation of the distance halogen–metal. Then the crystal field perceived by the optically active ion varies continuously. It was shown that in this case the crystal field transitions give rise to absorption and emission lines which can be quite broad and are characterized by edge singularities [3] reflecting the effect of the crystal field changes on the energy levels of U^{4+} . Given those interesting characteristics in the U^{4+} spectra in ThX_4 , it seemed of particular interest to study the spectra of the Pr^{3+} ion ($4f^2$) isoelectronic with the actinide ion U^{4+} ($5f^2$), to see whether the existence of the incommensurate structure can be seen with a lanthanide probe. Indeed the optical spectra of Pr^{3+} in $\beta\text{-ThBr}_4$

[10] were analyzed at a time when the low temperature structure of the $\beta\text{-ThBr}_4$ matrix was not yet known. This article reports a tentative assignment of the energy levels of Pr^{3+} in $\beta\text{-ThCl}_4$.

Symmetry Considerations

The Th^{4+} ions in the crystal are assumed to be substituted for by the Pr^{3+} ions. Pr^{3+} is a tripositive $3+$ impurity in the tetravalent matrix. The missing charge distorts the site and brings up new symmetries due to charge compensation. Thus additional lines appear in the spectra. Following the work on ThBr_4 : Pr^{3+} [10] and as was proved by a careful study on the line and satellites of the $^3\text{H}_4 \rightarrow ^3\text{P}_0$ transition [11], we assumed that the strongest lines in the spectra are the signature of the Pr^{3+} ions in the high temperature D_{2d} site symmetry which transforms into S_4 – C_2 – D_2 symmetries at low temperature.

Because of the weak effect of the modulation which suggests that the environment of the actinide ions in the different symmetries are close together, it was not possible to distinguish between the D_2 , C_2 and S_4 sites inside the bands of absorption and emission spectra. Moreover, a contribution from the S_4 symmetry is expected in the intense lines of the spectra, as will be seen later.

It can be noticed that the selection rules for the electronic dipole transitions in the D_{2d} and S_4 (subgroup of D_{2d}) groups are different only for the transitions (Tables I and II) $\Gamma_3 \rightarrow \Gamma_1$ and $\Gamma_2 \rightarrow \Gamma_4$ that are forbidden in the D_{2d} symmetry and allowed with a π polarization in the S_4 symmetry. The other selection rules are the same.

In the S_4 symmetry, the loss of symmetry elements results in the introduction of imaginary terms in the Hamiltonian: $im B_4^4$ and $im B_4^6$, whereas all the crystal field parameters in the D_{2d} symmetry are real. Interpreting the data in the D_{2d} symmetry instead of the S_4 symmetry amounts to neglecting those imaginary terms in the even rank crystal field components that are used for the calculations of the energy levels of the optically active ions. Nevertheless, the use of the D_{2d} symmetry instead of the S_4 symmetry should give the

TABLE I. Spectroscopic Best Fit Parameters from Fitting Pr^{3+} in $\beta\text{-ThCl}_4$. Set 1 Corresponds to a Γ_5 Ground State and Set 2 to a Γ_4 Ground State

	Set 1	Set 2
F^2	67947(112)	67866(180)
F^4	50576(304)	50219(574)
F^6	33468(161)	33322(311)
	742(3)	742(6)
α^a	21(1)	19(2)
$\beta/12^a$	-39(4)	-43(6)
γ^a	1343	1343
B_0^2	545(37)	20(76)
B_0^4	-657(90)	292(155)
B_4^4	876(58)	-964(95)
B_0^6	1398(115)	-1525(199)
B_4^6	508(97)	52(188)
Number of levels	52	52
r.m.s.: σ^b	34	66

^a γ is fixed at the value for aqueous solution [14]; α and β were varied and then fixed. ^br.m.s.: $\sigma = (\sum (E_{\text{calc}} - E_{\text{obs}})^2 / (n_{\text{obs}} - n_{\text{par}}))^{1/2}$, where E_{calc} and E_{obs} are the calculated and observed energy respectively; n_{obs} and n_{par} are the number of observed energy levels and the number of varying parameters.

main features of the spectra. Esterowitz *et al.* [12] had already noticed with Pr^{3+} doped in LiYF_4 at a site of S_4 symmetry that the D_{2d} selection rules have been useful in identifying the energy levels of Pr^{3+} .

Thus we shall interpret the spectra of Pr^{3+} in $\beta\text{-ThCl}_4$ in the D_{2d} symmetry. The crystal field eigenstates are the Γ_1 , Γ_2 , Γ_3 , Γ_4 and Γ_5 point group representations associated with the D_{2d} symmetry. Only the Γ_5 representation is doubly degenerate, the other ones being non-degenerate.

Analysis of the Spectra

$\beta\text{-ThCl}_4:\text{Pr}^{3+}$ single crystals were grown by the Bridgeman method [13]. The doping material was Pr_6O_{11} or PrCl_3 using approximately 1% by weight. The absorption and fluorescence spectra were recorded with a one-meter monochromator Jobin Yvon 'HR 1000' equipped for the visible region with a 1200 lines/mm grating and a photomultiplier, and with a 600 lines/mm grating and a PbS photocell for the IR region. Fluorescence of the crystal was excited with a Sopra nitrogen pumped dye laser tuned in the 3P_0 region (around $20\,500\text{ cm}^{-1}$) with a minimal bandwidth of 0.75 \AA (3 cm^{-1}).

Because of the charge compensation, many more lines appeared and we could not get polarized spectra.

TABLE II. Energy Levels of Pr^{3+} in ThCl_4 (52 observed energies). Set 1 corresponds to a Γ_5 ground state. Set 2 corresponds to a Γ_4 ground state

L-S state	Observed (cm^{-1})	Set 1		Set 2	
		Calculated (cm^{-1})	Irreducible representation in D_{2d} symmetry	Calculated (cm^{-1})	Irreducible representation in D_{2d} symmetry
3H_4	0	33	Γ_5	43	Γ_4
	172 ^a	198	Γ_1	185	Γ_1
		206	Γ_2	196	Γ_5
	317 ^a	255	Γ_3	244	Γ_2
	318 ^a	357	Γ_1	312	Γ_3
	497 ^a	434	Γ_5	446	Γ_1
	557 ^a	572	Γ_4	486	Γ_5
3H_5	2123 ^a	2159	Γ_2	2225	Γ_3
	2153 ^a	2180	Γ_1	2252	Γ_5
	2288 ^a	2265	Γ_5	2322	Γ_5
	2307 ^a	2332	Γ_5	2343	Γ_2
	2340 ^a	2353	Γ_4	2405	Γ_4
	2423 ^a	2476	Γ_2	2474	Γ_5
	2475 ^a	2514	Γ_5	2476	Γ_1
	2612 ^a	2616	Γ_3	2578	Γ_2
3H_6	4250	4215	Γ_5	4286	Γ_4
	4422	4400	Γ_1	4325	Γ_5
	4445	4459	Γ_3	4412	Γ_3
	4477	4494	Γ_5	4435	Γ_4

(continued on facing page)

TABLE II. (continued)

L-S state	Observed (cm ⁻¹)	Set 1		Set 2	
		Calculated (cm ⁻¹)	Irreducible representation in D _{2d} symmetry	Calculated (cm ⁻¹)	Irreducible representation in D _{2d} symmetry
	4504	4497	Γ ₄	4455	Γ ₁
	4522	4500	Γ ₃	4561	Γ ₃
	4539	4530	Γ ₂	4580	Γ ₂
	4629	4638	Γ ₁	4640	Γ ₅
	4784	4756	Γ ₅	4730	Γ ₁
	4833	4789	Γ ₄	4812	Γ ₅
³ F ₂	4976	5042	Γ ₁	5051	Γ ₄
	5039	5048	Γ ₅	5076	Γ ₅
	5100	5106	Γ ₃	5103	Γ ₁
	5160	5153	Γ ₄	5119	Γ ₃
³ F ₃	6377	6328	Γ ₅	6338	Γ ₂
	6405	6412	Γ ₃	6392	Γ ₃
	6444	6458	Γ ₂	6446	Γ ₅
	6486	6486	Γ ₅	6449	Γ ₄
	6533	6498	Γ ₄	6538	Γ ₅
³ F ₄	6818	6819	Γ ₃	6843	Γ ₁
	6894	6886	Γ ₅	6890	Γ ₅
	6929	6912	Γ ₁	6929	Γ ₄
	6967	6995	Γ ₂	6947	Γ ₂
	7023	7010	Γ ₄	7018	Γ ₁
	7044	7057	Γ ₅	7040	Γ ₅
	7066	7070	Γ ₁	7099	Γ ₃
¹ G ₄		9634	Γ ₃	9654	Γ ₁
		9800	Γ ₄	9721	Γ ₅
		9825	Γ ₁	9947	Γ ₂
		9885	Γ ₅	10001	Γ ₁
		9969	Γ ₂	10012	Γ ₄
		10142	Γ ₅	10035	Γ ₅
		10378	Γ ₁	10372	Γ ₃
¹ D ₂	16710	16679	Γ ₁	16656	Γ ₃
	16734	16781	Γ ₄	16861	Γ ₅
	16921	16899	Γ ₅	16911	Γ ₁
	17122	17120	Γ ₃	17063	Γ ₄
³ P ₀	20517	20459	Γ ₁	20457	Γ ₁
³ P ₁	20994	21026	Γ ₅	21082	Γ ₅
	21235 (Set 1)	21204	Γ ₂	21090	Γ ₂
¹ I ₆	21235 (Set 2)	20934	Γ ₄	21161	Γ ₁
		20935	Γ ₃	21168	Γ ₅
		21348	Γ ₅	21265	Γ ₃
		21371	Γ ₄	21339	Γ ₅
	21406	21392	Γ ₁	21376	Γ ₂
		21457	Γ ₅	21463	Γ ₄
	21522	21535	Γ ₂	21506	Γ ₃
		21787	Γ ₅	21634	Γ ₁

(continued overleaf)

TABLE II. (continued)

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<i>L-S</i> state	Observed (cm ⁻¹)	Set 1		Set 2	
		Calculated (cm ⁻¹)	Irreducible representation in <i>D</i> _{2d} symmetry	Calculated (cm ⁻¹)	Irreducible representation in <i>D</i> _{2d} symmetry
		21799	Γ ₁	21653	Γ ₅
		21800	Γ ₃	21693	Γ ₄
³ P ₂	22179	22226	Γ ₁	22310	Γ ₃
	22427 (Set 2)	22312	Γ ₅	22313	Γ ₅
	22427 (Set 1)	22415	Γ ₄	22333	Γ ₁
	22483	22517	Γ ₃	22402	Γ ₄
¹ S ₀		46660	Γ ₁	46570	Γ ₁

^aCorresponds to the lines observed in fluorescence spectra.

We overcame the difficulty by comparing the spectra of Pr³⁺ in ThCl₄ with those of Pr³⁺ in ThBr₄ that were polarized [10]. The similarity of those two sets of spectra encouraged us to follow their analysis. In ThBr₄:Pr³⁺, the (³P₀) Γ₁ line clearly showed a σ polarization, thus leading to a (³H₄) Γ₅ ground state in the *D*_{2d} symmetry.

The selection rules for the zero phonon electric dipole transitions in the *D*_{2d} symmetry are the following (notation of Koster *et al.* [14]):

Γ₅ → Γ₅ polarization π

Γ₅ → Γ₁, Γ₂, Γ₃ and Γ₄ polarization σ

In the most intense lines, a contribution due to Pr³⁺ ions in the *D*_{2d} symmetry is expected. Indeed, the intensity of electric dipole transition in the same configuration and regardless of the symmetry of the ion site is proportional to a matrix element involving the crystal field parameters of odd rank (*B*^{*k**}_{*q*}). The *D*₂ symmetry is close to the *D*_{2d} symmetry at low temperature. Thus the additional parameter *B*^{*5**}₄ to be added to the odd rank parameters for the *D*₂ symmetry *B*^{*3**}₂ and *B*^{*5**}₂ – common with the *D*_{2d} symmetry – can be supposed to be weak. Therefore the intensity of the lines corresponding to ions in the *D*_{2d} and *D*₂ symmetries will follow the same trend as was observed on ThBr:U⁴⁺ [1] and ThCl₄:U⁴⁺ [4].

Another argument is in favour of this hypothesis: the most intense lines of the spectrum at 4.2 K still exist at 80 K, above the phase transition temperature where the symmetry to be considered is the initial *D*_{2d} symmetry. The weak lines were systematically excluded from our analysis whether they are due to charge compensation and thus correspond to other sites of symmetry or whether they come from transitions with weak intensity in *D*_{2d} (e.g., some transitions towards the ¹I₆ Stark levels) or transitions

allowed in *D*₂ symmetry. So we tried to sort out the lines that correspond to Pr³⁺ in *D*_{2d} symmetry, though the selection rules seem to be somewhat relaxed because more transitions can be seen than are expected.

The absorption and fluorescence data obtained by exciting the ³P₀ were used in the fitting program which adjusted the various spectroscopic parameters of [15]: (i) interelectronic repulsion *F*^{*k*}, with *k* = 0, 2, 4, 6. (ii) spin-orbit coupling ξ. (iii) crystal field characteristic of the *D*_{2d} symmetry: *B*₀², *B*₀⁴, *B*₄⁴, *B*₀⁶, *B*₄⁶. In addition to these major contributions the configuration interaction parameters α, β and γ were used.

The starting values were those of Pr³⁺ in ThBr₄ [10], and in the first step only the absorption data were fitted. Once reasonable values for *B*_{*q*}^{*k*} were obtained, the calculated values were compared to the lines determined from fluorescence experiments. These fluorescence lines were assigned in the order derived from the fit, and the calculation was continued. The final set of best fit parameters values is listed in Table I. In the last step, the configuration interaction parameters α and β were varied, but not γ, whose value depends on the position of both the ³P₀ and ¹S₀ levels. The starting values of α and β and γ were fixed at the values in aqueous solution [16].

The values of the experimental and calculated energies with their irreducible representations and the *L-S* states from which they are derived are given in Table II.

Conclusion

We already stressed the similarity between β-ThCl₄:Pr³⁺ and β-ThBr₄:Pr³⁺ [10]. Assuming that the ground state is Γ₅ as in β-ThBr₄:Pr³⁺, the fitting (set 1, Tables I and II) results in a calculated set of energy and parameters very close to those of ThBr₄:Pr³⁺,

TABLE III. Spectroscopic Parameters of Pr³⁺ in the D_{2d} symmetry and of U⁴⁺ for comparison

	ThCl ₄ :Pr ³⁺		ThBr ₄ :Pr ³⁺ [10]	LuPO ₄ :Pr ³⁺ [17]	YPO ₄ :Pr ³⁺ [17]	ThBr ₄ :U ⁴⁺ [1]	ThCl ₄ :U ⁴⁺ [4]
	Set 1	Set 2					
F ²	67947	67866	68354	67688	677779	42253	42752
F ⁴	50576	50219	50310	48633	49603	40458	39925
F ⁶	33468	33322	33799	32151	32413	25881	24519
ξ	742	742	739	744	739	1783	1808
α	21	19	21	21	21	31	30
β/12	-39	-43	-67	-55	-55	-54	-41
γ	1343	1343	1343	1534	1534	1200	1200
B ₀ ²	545	20	260	21	78	-1096	-1054
B ₀ ⁴	-657	292	-644	280	321	1316	1146
B ₄ ⁴	876	-964	929	-808	-849	-2230	-2767
B ₀ ⁶	1398	-1525	1089	-1658	-1377	-3170	-2135
B ₄ ⁶	508	52	241	291	35	686	-(312)
Number of levels	52	52	42	18	35	26	25
σ (cm ⁻¹)	34	66	61	27	15	36	46

but with crystal field parameters B_q^k different in sign from those found in LuPO₄:Pr³⁺ [17] and ThX₄:U⁴⁺ (X = Br [1]; X = Cl [4]) [Table III]. This difference seemed somewhat surprising in view of the similar coordination polyhedron of U⁴⁺ or Pr³⁺ in the D_{2d} symmetry in all those compounds.

Following the calculation of the energy levels of Pr³⁺ in LuPO₄ and YPO₄ [17] and of U⁴⁺ in ThX₄ [1, 4], we repeated the calculation with a Γ₄ ground state (set 2 in Tables I and III). The starting values for the B_q^k were taken with the same sign as in those compounds. We concluded upon a fitting with a somewhat larger discrepancy between calculated and experimental levels but whose consistency according to the other studies is more satisfactory.

In the crystal that was studied, the effects of the structure modulation of the matrix are not clear in absorption. In fluorescence, the emission spectra showed some differences when the ³P₀ line was scanned. Unfortunately the laser used in our experiments did not have a bandwidth small enough to permit us to clearly relate the effect of the modulation on the emission spectra. Moreover, the crystals were too concentrated for that type of study. But the effect of the modulation with Pr³⁺ in ThCl₄ has been observed when studying the fluorescence of the ³P₀ line in less concentrated crystals (0.01%) and using a laser with a better definition [11].

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