

The Molar Refraction of Liquid Organosilicon Compounds

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Computation of least squares values for 11 species of silicon bonds from literature data on 255 compounds demonstrates that the Eisenlohr molar refraction MR_E provides better correlation of refractive index with molecular structure than is afforded by the classical Lorenz-Lorentz expression.

MODERN ATTEMPTS to correlate refractive index with molecular structure of liquid organosilicon compounds began with a 1946 article of Sauer (39), who used the literature data of 49 silicon compounds to devise a system which involved both group and bond refractions. Later in the same year, Warrick (62), in extending the Denbigh (7) concept of bond refraction to 72 silicon compounds, calculated values for 10 species of silicon bonds, using Denbigh's values for bonds of elements other than silicon. Results differing only slightly from Warrick's were obtained from 139 silicon compounds by Vogel, Cresswell, and Leicester (58), who used values from Vogel's table (57) for elements other than silicon. In all of this prior work, the classical Lorenz-Lorentz equation

$$MR_L = \frac{(n^2 - 1)}{(n^2 + 2)} \cdot \frac{M}{d}$$

was the basis for the calculations, inasmuch as it is the only relation resting on accepted theoretical considerations.

As the concluding investigation of a series begun in 1957 on the molar refraction of liquid organometallic compounds, the present work aimed at providing further support for the strong inference already drawn from the results of calculations on compounds of phosphorus (40), of tin (41), and of boron (42)—namely, that the empirical Eisenlohr "molecular refraction coefficient" (11) $MR_E = Mn_D^{20}$ is more in accord with the facts of nature than is the Lorenz-Lorentz equation. Literature data for 255 representative silicon compounds served for the calculations, in which the values $R_L = 1.424$ and $R_E = 14.26$ were used for the $C_{al}-C_{ar}$ bond (42), supplementing those from Vogel's table (57) for elements other than silicon. Both Lorenz-Lorentz and Eisenlohr values for 11 species of silicon bonds were computed by the method of least squares.

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Table I shows that the new Lorenz-Lorentz values are in fairly close agreement with those of Warrick and of Vogel.

The literature data and observed molar refractions of each compound are listed in Table II, together with the molar refractions calculated from the newly derived parameters of Table I. For comparison of the precision of the two systems under consideration, the coefficient of variation, C_v , was calculated for each—Lorenz-Lorentz = 0.0450, Eisenlohr = 0.0271.

Of the two systems investigated, that of Eisenlohr, aside from its other advantages, is clearly the more precise in its application to silicon compounds, as it has already been shown to be for the other principal classes of liquid organometallics; by a fairly reasonable inference, it seems likely that the same relative superiority obtains for all organic liquids. However, the evidence so far rests on calculations in which a substantial and sometimes preponderant part of the molar values is contributed by various carbon bonds, and of these the only ones derived by the method of least squares are $C-C$ and $C-H$, obtained by Vickery and Denbigh (56) from eight *n*-alkanes and by Vogel from 12 *n*-alkanes and a considerable number of alkyl halides including both straight-chain and branched compounds. Although the 35 values, mostly of carbon bonded with other elements, given in Vogel's table (57) have been eminently useful, it is believed that reliance on molar refraction in research will eventually require that all bond values be computed by the least squares method from thoroughly reliable data, and that finer distinctions than those now in use be drawn between bond species. In particular, the numerous subspecies of $C_{al}-C_{al}$ bonds should be clearly differential rather than lumped together as most of them have been in the past. Hence, the present work merely completes the preliminary portion of a comprehensive study of organic compounds in general, the next phase of which is already in progress on the molar refraction of liquid alkanes.

Table I. Calculated Parameters
Lorenz-Lorentz Refractions

Bond	Present Work			Eisenlohr Refractions		
	Warrick (62)	Vogel (58)	Number of iterations	Values	Number of iterations	Values
Si—C _{al}	2.50	2.52	656	2.47	689	18.57
Si—C _{ar}	2.56	2.93	30	2.93	31	23.68
Si—H	3.20	3.17	96	3.17	96	10.39
Si—F	1.50	1.7	28	1.95	33	32.26
Si—Cl	7.20	7.11	81	7.33	85	62.21
Si—Br	10.20	10.08	39	10.31	39	135.19
Si—I			9	15.92	10	213.77
Si—N	2.00	2.16	74	2.10	83	20.59
Si—O	1.75	1.80	157	1.83	158	21.93
Si—S	6.25	6.14	70	5.94	70	40.96
Si—Si	5.65	5.89	26	5.85	27	30.02

Table II. Molar Values for Organosilicon Compounds

No.	Compound	Lorenz-Lorentz Molar Refractivity ^a						Eisenlohr Molar Refraction Product ^b		
		Ref.	M	t	n' _D	d' _i	n' _D	Obsd.	Calcd.	Obsd.
1	(CH ₃) ₃ Si	(50)	88.230	20	0.6426	1.3582	30.166	29.987	119.83	120.72
2	(CH ₃) ₃ SiC ₂ H ₅	(64)	102.257	20	0.6849	1.3820	34.746	34.645	141.32	141.32
3	(CH ₃) ₃ SiCH ₂ CH=CH ₂	(47)	114.268	20	0.7020	1.4074	39.140	38.815	160.82	160.71
4	(CH ₃) ₃ SiC ₂ H ₇	(64)	116.284	20	0.7181	1.3929	39.525	39.294	161.97	161.92
5	(CH ₃) ₃ SiC ₄ H ₉	(64)	130.311	20	0.7437	1.4030	44.283	43.941	182.83	182.52
6	(C ₂ H ₅) ₃ SiCH ₃	(64)	130.311	20	0.7437	1.4160	43.971	43.941	184.52	182.52
7	(CH ₃) ₂ Si(C ₂ H ₇) ₂	(50)	144.338	20	0.7399	1.4146	48.809	48.589	204.18	203.12
8	(CH ₃) ₂ SiC ₂ H ₁₁	(64)	144.338	20	0.7313	1.4096	48.859	48.589	203.46	203.12
9	(C ₂ H ₅) ₂ Si	(64)	144.338	20	0.7662	1.4268	48.347	48.589	205.94	203.12
10	(CH ₃) ₂ SiC ₆ H ₁₃	(64)	158.366	20	0.7422	1.4154	53.477	53.237	224.15	223.72
11	(C ₂ H ₅) ₂ SiC ₃ H ₇	(64)	158.365	20	0.7724	1.4308	53.050	53.237	226.59	223.72
12	(CH ₃) ₂ SiC ₂ H ₁₅	(64)	172.392	20	0.7506	1.4201	58.134	57.885	244.81	244.32
13	(C ₂ H ₅) ₂ SiC ₄ H ₉	(64)	172.392	20	0.7786	1.4348	57.753	57.885	247.35	244.32
14	(CH ₃) ₂ SiCH ₂ CHClCH ₂ Cl	(47)	185.182	20	1.030	1.4581	49.067	48.961	270.01	267.78
15	(CH ₃) ₂ SiC ₄ H ₁₇	(64)	186.419	20	0.7581	1.4242	62.773	62.533	265.50	264.92
16	(C ₂ H ₅) ₂ SiC ₅ H ₁₁	(64)	186.419	20	0.7835	1.4377	62.422	62.533	268.01	264.92
17	(CH ₃) ₂ SiCH ₂ CH ₂ CH ₂ Br	(47)	195.192	20	1.113	1.4580	47.853	47.007	284.59	282.56
18	(C ₃ H ₇) ₂ Si	(27)	200.446	20	0.7870	1.4378	66.833	67.181	288.20	285.52
19	(C ₂ H ₅) ₃ SiC ₆ H ₁₃	(64)	200.446	20	0.7880	1.4400	67.040	67.181	288.64	285.52
20	(CH ₃) ₂ SiC ₁₀ H ₂₁	(64)	214.473	20	0.7705	1.4310	72.062	71.829	306.91	306.12
21	(C ₂ H ₅) ₂ SiC ₇ H ₁₅	(64)	214.473	20	0.7907	1.4422	71.797	71.829	309.31	306.12
22	(C ₂ H ₅) ₃ SiC ₈ H ₁₇	(64)	228.500	20	0.7971	1.4438	76.117	76.477	329.91	326.72
23	(CH ₃) ₂ SiCH ₂ CH ₂ CH ₂ I	(47)	242.186	20	1.496	1.4400	81.265	81.125	362.31	360.51
24	(CH ₃) ₂ SiC ₁₂ H ₂₅	(64)	242.527	20	0.7800	1.4358	81.393	81.125	348.22	347.32
25	(C ₄ H ₉) ₂ SiC ₃ H ₇	(32)	242.527	20	0.7949	1.4462	85.354	85.773	350.74	347.32
26	(C ₂ H ₅) ₂ SiC ₁₀ H ₂₁	(64)	256.554	20	0.8036	1.4472	90.321	90.421	371.28	367.92
27	(CH ₃) ₂ SiC ₁₄ H ₂₉	(64)	270.581	20	0.7911	1.4410	58.086	57.939	389.91	385.52
28	CH ₃ CH[Si(CH ₃) ₃] ₂	(34)	174.444	20	0.7821	1.4340	81.265	81.125	250.02	246.56
29	CH ₂ CH ₂ CH ₂ CH ₂ Si(CH ₃) ₂	(18)	114.268	20	0.7931	1.4330 ^c	37.445	37.237	163.75	167.04
30	CH ₂ CH ₂ CH ₂ CH ₂ Si(C ₂ H ₅) ₂	(18)	142.322	20	0.8256	1.4471	46.068	46.533	205.95	208.24
31	(CH ₃) ₂ SiOH	(38)	90.203	20	0.8112	1.3880	26.239	25.983	125.20	125.62
32	(CH ₃) ₂ SiOCH ₃	(47)	104.230	20	0.7573	1.3743	35.687	35.459	142.11	141.54
33	(CH ₃) ₂ SiOC ₂ H ₅	(38)	118.257	20	0.7682	1.3838	40.243	40.107	162.52	162.14
34	(CH ₃) ₂ SiOC ₃ H ₇	(21)	132.284	20	0.8647	1.4329	39.752	39.927	183.05	182.74
35	(C ₂ H ₅) ₂ SiOH	(16)	132.284	20	0.7774	1.3925	44.867	44.555	187.42	186.34
36	(CH ₃) ₂ SiOC ₂ H ₉	(38)	146.311	20	0.7849	1.4000	49.522	49.403	223.47	223.94
37	(CH ₃) ₂ SiOC ₃ H ₁₁	(21)	160.338	20	1.0238	1.4607	44.663	44.761	243.55	240.35
38	CH ₃ CHCl(C ₂ H ₅) ₂ SiOH	(45)	166.733	20	0.8169	1.4242	63.255	63.347	288.28	285.74
39	(C ₃ H ₇) ₂ SiOC ₂ H ₅	(18)	202.419	20	0.8068	1.4228	72.713	72.673	327.92	326.94
40	(CH ₃) ₂ SiOC ₁₀ H ₂₁	(21)	230.473	20	0.8322	1.4350	76.665	77.291	350.86	347.54
41	(C ₄ H ₉) ₂ SiOC ₂ H ₅	(18)	244.500	20	0.7619	1.3774	48.647	48.647	223.68	224.94
42	[(CH ₃) ₂ Si] ₂ O	(38)	162.390	20	0.8402	1.4345 ^c	76.535	76.535	353.68	348.54
43	[(C ₂ H ₅) ₂ Si] ₂ O	(49)	246.552	25						

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Table II. Molar Values for Organosilicon Compounds (Continued)

No.	Compound	<i>M</i>	<i>t</i>	<i>n</i> _D '	<i>d</i> ₄ '	Lorenz-Lorentz Molar Refractivity ^a		Eisenlohr Molar Refraction Product ^b	
						<i>n</i> _D ²⁰	Obsd.	Calcd.	Obsd.
44	$(C_3H_7)_3SiO$	(27)	330.714	20	0.8381	1.4412	104.423	104.423	476.62
45	$(CH_3)_2Si(OCH_2H_5)_2$	(18)	148.284	20	0.8410	1.3815	40.985	40.922	204.85
46	$(C_2H_5)_2Si(OCH_2H_5)_2$	(18)	176.338	20	0.8622	1.4022	49.822	50.218	247.26
47	$(C_4H_7)_2Si(OCH_2H_5)_2$	(18)	204.392	20	0.8558	1.4100	59.173	59.514	288.19
48	$(C_4H_9)_2Si(OCH_2H_5)_2$	(18)	232.446	20	0.8510	1.4182	68.863	68.810	285.95
49	$CH_2CH_2CH_2CH_2Si(OCH_3)_2$	(18)	146.268	20	0.9623	1.4269	39.018	38.866	327.15
50	$CH_2CH_2CH_2CH_2Si(OCH_2H_5)_2$	(18)	174.322	20	0.9468	1.4300	47.562	48.162	249.28
51	$CH_3(CH_3)_2SiO[Si(CH_3)_3]$	(31)	236.550	20	0.8200	1.3848	67.572	67.297	329.15
52	$CH_3(CH_3)_2SiO[Si(CH_3)_3]$	(31)	310.710	20	0.8536	1.3895	86.188	85.946	431.73
53	$CH_3(CH_3)_2SiO[Si(CH_3)_3]$	(31)	384.870	20	0.8755	1.3925	104.799	104.596	537.93
54	$CH_3(CH_3)_2SiO[Si(CH_3)_3]$	(31)	459.030	20	0.8910	1.3948	123.456	123.246	641.79
55	$(CH_3)_2SiO_4$ (Cyclic)	(31)	290.640	20	0.9558	1.3968	73.195	74.599	405.97
56	$(CH_3)_2SiO_5$ (Cyclic)	(31)	370.800	20	0.9593	1.3982	93.332	93.248	518.45
57	$(CH_3)_2SiO_6$ (Cyclic)	(31)	444.960	20	0.9672	1.4015	111.897	111.898	623.61
58	$(CH_3)_2SiO_7$ (Cyclic)	(31)	519.120	20	0.9730	1.4040	130.481	130.548	728.84
59	$(C_2H_9)_2Si(NHC_6H_5)_2$	(17)	339.560	20	0.9598	1.4664	98.056	96.706	497.93
60	$(CH_3)_2Si(N(CH_3)_2)_2$	(3)	146.316	22	1.4169	0.809	1.4178	1.4151	521.07
61	$(CH_3)_2Si(NHC_2H_5)_2$	(23)	146.316	20	0.8067	1.4015	45.429	45.201	625.29
62	$(CH_3)_2Si(NH_3)_2$	(46)	174.370	20	1.0090	1.4534	120.973	120.973	729.50
63	$(CH_3)_2Si(NHCH_2C_6H_5)_2$	(23)	270.458	20	0.8297	1.4425	84.217	84.417	494.18
64	$(CH_2)_2Si(NHC_7H_5)_2$	(23)	286.586	20	1.4604	0.9162	91.482	91.681	415.82
65	$[(CH_3)_2C(C_6H_5)Si(NH_3)_2$	(46)	342.694	20	0.8884	1.4427	45.473	45.581	206.02
66	$C_3H_7Si(NHCH_3)_3$	(51)	161.332	20	0.850	1.4338 ^c	53.553	53.373	209.50
67	$CH_3Si[N(CH_3)_2]_2$	(3)	175.359	22	1.4324	0.8545	1.4380	62.590	295.08
68	$C_3H_7Si(NHC_2H_5)_3$	(51)	203.413	20	0.8545	1.4436	76.256	76.043	356.88
69	$C_3H_7Si(NHC_3H_7)_3$	(51)	245.494	20	0.8545	1.4436	85.320	85.909	392.86
70	$C_2H_5Si[N(C_2H_5)_2]$	(53)	273.548	20	0.8773	1.4595	89.978	90.557	413.46
71	$C_3H_7Si[N(C_2H_5)_2]$	(53)	287.575	20	0.8752	1.4599	90.179	89.987	416.24
72	$C_3H_7Si(NHC_3H_7)_3$	(51)	287.575	20	0.8527	1.4474	120.973	120.923	418.68
73	$C_3H_7Si(NHCH_2C_6H_5)_3$	(52)	389.626	20	1.0503	1.5665	120.973	120.923	604.56
74	$[(CH_3)_2SiNH_3]_2$	(38)	161.406	20	0.7742	1.4080	51.432	50.947	229.52
75	$(CH_3)_2SiNH_3$ (Cyclic)	(6)	219.528	20	0.9196	1.4448	63.511	62.848	317.17
76	$(C_2H_5)_2SiNH_3$ (Cyclic)	(6)	303.690	20	0.9287	1.4670	90.735	90.736	445.57
77	$(C_2H_5)_2SiNH_4$ (Cyclic)	(6)	404.920	20	0.9521	1.4769	120.148	120.982	599.98
78	$(CH_3)_3SiCl$	(37)	108.652	20	0.8581	1.3884	29.906	29.825	152.75
79	$CH_2(CH_2H_5)(CH_2 = CH)CH_2SiCl$	(26)	148.717	20	0.9109	1.4380	42.858	43.291	213.34
80	$(C_2H_5)_2SiCl$	(8)	150.733	20	0.8967	1.4314	43.547	43.769	215.76
81	$CH_3(CH_2H_5)(CH_2CH_2CH_2)SiCl$	(26)	185.182	20	1.0408	1.4548	48.255	48.603	267.48
82	$CH_3(CH_2H_5)(CNH_2SiCl)$	(53)	222.846	20	0.9259	1.4419	63.669	63.621	321.32
83	$CH_3(CH_2H_5)_2N[SiCl]$	(53)	222.846	20	0.9343	1.4453	63.518	64.001	322.08
84	$C_2H_5(CH_3)_2CNH_2SiCl$	(53)	236.873	20	0.9280	1.4465	68.133	68.269	344.53
85	$C_2H_5(CH_3)_2N[SiCl]$	(53)	236.873	20	0.9395	1.4517	67.978	68.649	343.87
86	$C_2H_7(CH_3)_2CNH_2SiCl$	(51)	250.900	20	0.9153	1.4468	73.212	72.917	363.00
87	$C_3H_7(CH_2H_5)_2N[SiCl]$	(51)	250.900	20	0.9323	1.4530	72.740	73.297	364.56
88	$C_1H_7(CH_3(C_6H_5)N[SiCl]$	(51)	318.934	20	1.5675				499.93

89	$\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{SiCl}_2$	(18)	155.112	20	1.1505	37.278	36.892	227.25	231.10
90	$\text{CH}_3(\text{CH}_2=\text{CHCH}_2)\text{SiCl}_2$	(26)	155.112	20	1.0686	1.4406	38.301	223.45	224.77
91	$\text{CH}_3(\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2)\text{SiCl}_2$	(26)	169.139	20	1.0521	1.4470	42.954	43.118	244.74
92	$\text{C}_2\text{H}_2(\text{CH}_2=\text{CHCH}_2)\text{SiCl}_2$	(26)	169.139	20	1.0529	1.4483	43.059	43.118	244.96
93	$\text{CH}_3(\text{CH}_2\text{CICH}_2\text{CH}_2)\text{SiCl}_2$	(26)	191.577	20	1.2040	1.4585	43.458	43.782	279.42
94	$\text{C}_2\text{H}_2(\text{CH}_2\text{CICH}_2)\text{SiCl}_2$	(33)	191.577	20	1.2174	1.4678	43.729	43.782	278.91
95	$\text{C}_2\text{H}_2(\text{CH}_3\text{CHCl})\text{SiCl}_2$	(33)	191.577	20	1.2017	1.4578	43.484	43.782	279.28
96	$\text{C}_2\text{H}_2(\text{CH}_2\text{CICH}_2)\text{SiCl}_2$	(26)	205.604	20	1.1810	1.4663	48.244	48.430	301.48
97	$[(\text{CH}_3)_3\text{C}]_2\text{SiCl}_2$	(55)	213.236	20	1.009	1.4561	57.459	57.540	310.49
98	$\text{C}_2\text{H}_2[(i\text{-C}_3\text{H}_7)_2\text{N}]\text{SiCl}_2$	(51)	242.279	20	1.0192	1.4557	64.582	65.332	352.69
99	$\text{C}_2\text{H}_2[(\text{C}_6\text{H}_{11})_2\text{N}]\text{SiCl}_2$	(54)	322.409	20	1.2426	1.4295	33.542	33.649	483.29
100	$\text{CH}_2=\text{CHSiCl}_3$	(34)	161.507	20	1.2224	1.4445	38.181	38.297	230.87
101	$\text{CH}_2=\text{C}(\text{CH}_3)_3\text{SiCl}_3$	(26)	175.534	20	1.2398	1.4412	37.403	38.297	253.36
102	$\text{CH}_2=\text{C}(\text{CH}_3\text{CH}_2)\text{SiCl}_3$	(34)	175.534	20	1.2036	1.4535	42.610	42.945	256.80
103	$\text{CH}_2=\text{C}(\text{CH}_3\text{CH}_2\text{SiCl}_3)$	(26)	189.561	20	1.3590	1.4668	43.269	43.609	277.53
104	$\text{CH}_2\text{CICH}_2\text{CH}_2\text{SiCl}_3$	(26)	211.999	20	1.5069	1.4820	56.222	56.902	310.94
105	$\text{CH}_3\text{CH}(\text{SiCl}_3)_2$	(34)	296.976	20	1.3733	1.4927	65.781	61.550	440.12
106	$(\text{CH}_3)_3\text{C}(\text{SiCl}_3)_2$	(34)	311.003	20	0.832	1.4512	43.487	43.722	464.23
107	$(\text{CH}_3)_3\text{SiSiC}_2\text{H}_5$	(7)	134.323	20	0.844	1.4524	47.453	47.370	194.93
108	$\text{CH}_3\text{SiSC}_2\text{H}_7$	(7)	148.350	20	0.824	1.4497	48.355	47.370	215.46
109	$(\text{CH}_3)_3\text{SiSiC}_2(\text{CH}_3)_2$	(1)	148.350	20	0.854	1.4550	51.588	52.018	217.15
110	$(\text{CH}_3)_3\text{SiSiC}_4\text{H}_9$	(1)	162.377	20	0.834	1.4570	53.025	52.018	236.26
111	$(\text{CH}_3)_3\text{SiSiC}(\text{CH}_3)_3$	(7)	162.377	20	0.834	1.4518	52.018	52.018	237.75
112	$(\text{CH}_3\text{S})_4\text{Si}$	(4)	216.494	35	1.5989	1.1888	62.209	62.305	196.55
113	$(\text{C}_2\text{H}_5)_3\text{Si}$	(4)	272.602	25	1.5638	1.0860	81.619	80.897	217.15
114	$(\text{C}_3\text{H}_7)_3\text{Si}$	(4)	328.710	25	1.5431	1.0328	100.335	99.489	217.15
115	$[(\text{CH}_3)_2\text{CHSi}]_3\text{Si}$	(4)	328.710	35	1.5350	1.0099	101.334	99.489	236.26
116	$(\text{C}_2\text{H}_5)_3\text{SiH}$	(50)	116.284	20	0.7318	1.4119	39.609	39.609	236.58
117	$\text{C}_6\text{H}_5\text{CH}_2(\text{CH}_3)_3\text{SiH}$	(16)	150.301	20	0.9486	1.6057	62.209	62.305	347.62
118	$(\text{C}_3\text{H}_7)_3\text{SiH}$	(63)	158.365	20	0.7897	1.5661	81.619	80.897	426.92
119	$(\text{C}_3\text{H}_7)_3\text{SiH}$	(63)	200.446	20	0.7793	1.4980	53.647	53.553	424.05
120	$\text{CH}_3(\text{C}_6\text{H}_5\text{CH}_2)_3\text{SiH}$	(16)	226.399	20	0.9936	1.5650	67.521	67.497	226.21
121	$(\text{C}_5\text{H}_11)_3\text{SH}$	(63)	242.527	20	0.7894	1.4436	74.219	74.177	224.54
122	$(\text{C}_2\text{H}_5)_2(\text{C}_2\text{H}_5\text{O})\text{SiH}$	(18)	132.284	20	0.8465	1.3869	40.609	40.422	288.24
123	$\text{C}_2\text{H}_5(\text{C}_2\text{H}_5\text{O})_2\text{SiH}$	(18)	148.284	20	0.8701	1.4284	52.289	54.561	286.34
124	$(\text{CH}_2=\text{CH}_2\text{CH}_2\text{O})_2\text{SiH}$	(17)	200.317	25	1.4270	0.9836	111.427	111.771	354.31
125	$[\text{C}_2\text{H}_5\text{C}(\text{CH}_3)\text{O}]_3\text{SiH}$	(17)	248.446	20	0.8661	1.4054	70.369	69.939	348.46
126	$(\text{C}_3\text{H}_7)_3\text{SiH}$	(17)	290.527	27	1.4195	0.8710	1.4210	81.441	348.14
127	$(\text{C}_6\text{H}_5)_3\text{SiH}$	(17)	308.416	26	1.5621	1.1158	84.328	83.883	410.59
128	$(\text{C}_6\text{H}_5\text{O})_3\text{SiH}$	(17)	332.608	25	1.4270	0.8701	1.5636	86.547	266.63
129	$(\text{C}_7\text{H}_5\text{O})_3\text{SiH}$	(17)	374.689	26	1.4315	0.8713	89.146	86.547	468.67
130	$[(\text{C}_6\text{H}_5\text{O})_2\text{SiH}]_2\text{O}$	(17)	254.444	25	1.3850	0.9442	1.3864	63.152	328.43
131	$(\text{CH}_3)_3\text{SiH}$	(66)	170.401	25	1.5671	1.1423	49.895	48.738	266.63
132	$(\text{C}_2\text{H}_5)_3\text{SiH}$	(66)	212.482	25	1.5440	1.0484	63.980	63.839	328.54
133	$(\text{C}_3\text{H}_7)_3\text{SiH}$	(66)	254.563	25	1.5278	0.9991	1.5300	78.432	390.23
134	$[(\text{CH}_3)_2\text{CHSi}]_3\text{SiH}$	(66)	254.563	25	1.5221	0.9864	1.5243	77.783	390.23
135	$(\text{C}_3\text{H}_7)_3\text{SiH}$	(66)	296.644	25	1.5160	0.9819	1.5182	91.251	450.36

(Continued on page 150)

Table II. Molar Values for Organosilicon Compounds (Continued)

No.	Compound	<i>M</i>	<i>t</i>	<i>n</i> _D	<i>d</i> ₄	Eisenlohr Molar Refraction Product ^b		Lorenz-Lorentz Molar Refractivity ^a	
						Obsd.	Calcd.	Obsd.	Calcd.
136	$[(CH_3)_2CHCH_2S]_2SiH$	(66)	296.644	25	1.5160	0.9694	1.5182 ^c	92.427	91.727
137	Cl(CH ₃)(C ₆ H ₅ CH ₃) ₂ SiH	(16)	170.723	27	1.5129 ^d	1.0020	1.5160	51.202	49.748
138	Cl[(CH ₃) ₂ CN]H ₂ SiH	(53)	208.819	20		0.9261	1.4379	59.179	59.288
139	Cl[(C ₂ H ₅) ₂ N]SiH	(53)	208.819	20		0.9354	1.4433	59.218	59.668
140	Cl[(C ₂ H ₅ S) ₂]SiH	(66)	242.919	25	1.5030	1.0358	1.5052 ^c	69.328	69.534
141	Cl[(CH ₃) ₂ CS] ₂ SiH	(66)	242.919	25	1.5040	1.0222	1.5062 ^c	70.368	69.534
142	Cl ₂ (CH ₂) ₂ SiH	(34)	127.058	20		1.1222	1.4160	28.413	29.489
143	Cl ₂ [CH ₂ = C(CH ₃) ₂]SiH	(34)	141.085	20		1.0787	1.4310	33.855	34.137
144	Cl ₂ (CH ₂ CHC(CH ₃))SiH	(33)	163.523	20		1.2614	1.4484	34.731	34.801
145	(C ₂ H ₅) ₂ SiH ₂	(12)	88.230	20	0.6832	1.3918	30.628	30.738	30.628
146	(CH ₂ =CHCH ₂) ₂ SiH ₂	(35)	112.252	20	0.7533	1.4420	39.428	38.968	38.968
147	(C ₃ H ₇) ₂ SiH ₂	(12)	116.284	20	0.7194	1.4112	40.151	39.924	39.924
148	(C ₄ H ₉) ₂ SiH ₂	(63)	144.338	20	0.7458	1.4241	49.395	49.220	49.220
149	(C ₅ H ₁₁) ₂ SiH ₂	(63)	172.392	20	0.7636	1.4324	58.604	58.516	58.516
150	CH ₂ =CHCH ₂ SiH ₃	(35)	72.187	20	0.6764	1.4050	26.157	25.817	25.817
151	C ₃ H ₅ SiH ₃	(12)	74.203	20	0.6434	1.3759	26.457	26.295	26.295
152	CH ₂ ClSiH ₃	(35)	80.598	20	0.9286	1.4157	21.767	21.833	21.833
153	C ₃ H ₅ SiH ₃	(50)	88.230	20	0.6786	1.3922	30.975	30.943	30.943
154	(CH ₃) ₂ CHCH ₂ SiH ₃	(50)	88.230	20	0.6753	1.3905	31.006	30.943	30.943
155	CH ₃ CHClSiH ₃	(33)	94.625	20	0.8846	1.4147	26.770	26.481	26.481
156	C ₃ H ₅ SiH ₃	(63)	102.257	20	0.7019	1.4042	35.645	35.591	35.591
157	CH ₃ CHClICH ₂ SiH ₃	(35)	108.652	20	0.8983	1.4305	31.277	31.129	31.129
158	C ₃ H ₅ SiH ₃	(30)	114.268	25	1.4464	0.7958	1.4486 ^d	38.320	38.027
159	C ₄ H ₉ SiH ₃	(63)	116.284	20	0.7189	1.4131	40.343	40.239	40.239
160	CH ₃ CHSiH(CH ₃) ₂	(34)	146.390	20	0.7597	1.4252	49.292	49.273	49.273
161	(CH ₃) ₂ SiCH ₂ SiH(CH ₃) ₂	(54)	146.390	20	0.7454	1.4158	49.263	48.958	48.958
162	(CH ₃) ₂ C[SH(CH ₃) ₂] ₂	(34)	160.417	20	0.7779	1.4360	53.919	53.921	53.921
163	(CH ₃) ₂ SiCH(CH ₃) ₂ SiH(CH ₃) ₂	(34)	160.417	20	0.7756	1.4288	53.299	53.606	53.606
164	(CH ₃) ₂ SiC(CH ₃) ₂ SiH(CH ₃) ₂	(34)	174.444	20	0.7939	1.4378	57.658	58.254	58.254
165	Cl ₂ SiCH(CH ₃) ₂ SiHCl ₂	(34)	262.527	20	1.4310	1.4740	51.558	52.742	52.742
166	(CH ₃) ₂ SiBr	(24)	153.111	25	1.4211	1.1727	1.4223 ^d	33.116	32.805
167	CH ₃ Si(C ₂ H ₅) ₂ SiBr	(25)	181.165	25	1.4464	1.1515	1.4486 ^d	41.987	42.101
168	(C ₂ H ₅) ₂ SiBr	(61)	195.192	20	1.1454	1.4573	46.438	46.749	46.749
169	H(CH ₃) ₂ SiBr	(67)	203.216	25	1.5660	1.5041	1.5682 ^c	44.072	44.626
170	H(CH ₃) ₂ SiBr	(67)	231.270	25	1.5408	1.3717	1.5430 ^c	53.922	53.922
171	(CH ₃) ₂ SiBr	(67)	249.309	25	1.5978	1.4988	1.6000 ^d	56.736	57.035
172	H[(CH ₃) ₂ CHSi] ₂ SiBr	(67)	259.324	25	1.5195	1.2720	1.5217 ^d	61.929	63.218
173	H[(CH ₃) ₂ CHCH ₂ Si] ₂ SiBr	(67)	287.378	25	1.5159	1.2481	1.5181 ^d	69.535	72.514
174	(C ₂ H ₅) ₂ SiBr	(67)	291.390	25	1.5628	1.3508	1.5650 ^c	70.039	70.979
175	(C ₃ H ₇) ₂ SiBr	(67)	333.471	25	1.5418	1.2444	1.5440 ^d	84.312	84.923
176	[(CH ₃) ₂ CHSi] ₂ SiBr	(67)	333.471	25	1.5410	1.2244	1.5432 ^d	85.584	84.923
177	C ₃ H ₅ CH ₃ C ₆ H ₅ N ₂ SiBr	(53)	369.393	20	1.5282	1.2126	99.312	96.645	96.645
178	[(CH ₃) ₂ CHCH ₂ Si] ₂ SiBr	(67)	375.552	25	1.4696	1.1823	1.5304 ^d	97.842	98.867
179	(CH ₃) ₂ SiBr ₂	(24)	217.992	25	1.4718 ^d	1.6952	1.5852	35.612	32.084
180	(C ₂ H ₅) ₂ SiBr ₂	(24)	246.046	25	1.4850	1.5767	1.4872 ^d	44.725	44.908

^a Lorenz-Lorentz Molar Refractivity^a^b Eisenlohr Molar Refraction Product^b^c Calculated^d Observed

181	$(C_2H_5)_2SiBr_2$	310.178	25	1.5658	1.6541	61.151	61.062	486.36	482.41
182	$[CH_3)_2CHCH_2S][SiBr_2]$	(67)	366.286	25	1.497	1.3566	79.099	79.654	549.14
183	CH_3SiBr_3	(24)	282.873	25	1.5152	2.2130	1.5174 ^d	1.5174 ^d	429.23
184	$C_2H_5SiBr_3$	(24)	296.900	25	1.5155	2.0671	1.5177 ^d	43.347	450.60
185	$C_2H_5(CH_3)_2SiF$	(10)	106.222	25	1.3570	0.8360	1.3900	38.074	144.38
186	$(C_2H_5)_3SiF$	(59)	134.276	20		1.0100	1.4277	43.004	38.558
187	$CH_2CICH_2(C_2H_5)_2SiF$	(45)	168.925	20		1.4208	1.4208	43.228	38.419
188	$CH_3CICH_2(C_2H_5)_2SiF$	(45)	168.925	20	0.9955	1.4117	52.338	52.338	456.35
189	$(C_2H_5)_3SiF$	(59)	176.357	20	0.8384	1.4238	66.297	66.282	456.35
190	$(C_2H_5)_3SiF$	(59)	218.438	20	0.8404	1.4238	29.017	29.017	143.40
191	$CH_2=CH(C_2H_5)SiF_2$	(36)	122.198	20	0.9578	1.3356	26.429	26.429	184.60
192	$(C_2H_5)_2SiF_2$	(36)	124.214	20	0.9357	1.3598	27.811	27.811	237.53
193	$[(CH_3)_2CH]_2SiF_2$	(10)	152.168	22	1.3663	1.3672 ^d	248.96	248.96	237.53
194	$CH_2CICH_2(C_2H_5)SiF_2$	(36)	158.663	20	1.1467	1.3900	32.799	33.033	208.04
195	$CH_3CICH_2(C_2H_5)SiF_2$	(36)	158.663	20	1.1155	1.3790	32.869	33.033	207.28
196	$(C_2H_5)_2SiF_2$	(48)	180.322	25	1.4049	0.3881	50.155	46.791	308.20
197	$CH_2CICH_2SiF_3$	(36)	148.601	20	1.3661	1.3418	22.910	22.838	165.87
198	$CH_3CHCISiF_3$	(36)	148.601	20	1.2606	1.3180	23.251	22.838	166.08
199	$C_6H_{10}SiF_3$	(48)	168.244	25	1.3689	1.1012	1.302 ^d	34.390	207.28
200	$CH_2[Si(CH_3)_2F]_2$	(5)	168.347	20	0.8681	1.5125	37.438	36.940	207.28
201	$H_3SiCH_2H_5$	(12)	108.220	20	1.5251	0.9797	43.252	43.212	207.28
202	$H_3SiC_6H_5OCH_3$	(30)	138.247	25	0.8748	1.4918	49.829	49.938	207.28
203	$(CH_3)_2SiC_6H_5$	(22)	150.301	20	1.6030	1.0054	1.6052 ^d	54.075	219.01
204	$H_3SiC_6H_5(a-)$	(30)	158.280	25		0.8666	1.4910	54.919	54.919
205	$C_6H_5CH_3(p-)$	(15)	164.328	20		0.9905	1.5170	55.079	55.079
206	$HO(C_2H_5)_2SiC_6H_5$	(44)	180.328	20		1.0002	1.5090	55.153	55.153
207	$(CH_3)_2SiC_6H_4Cl(p-)$	(15)	184.750	20	1.5555	1.3632	1.5577 ^d	44.092	207.28
208	$BrH_3SiC_6H_5$	(24)	187.128	25		0.8816	1.5024	64.443	64.443
209	$(C_2H_5)_2SiC_6H_5$	(13)	192.382	20		0.9249	1.5338	64.969	64.969
210	$(CH_3)_2SiC_6H_4N(CH_3)_2(p-)$	(15)	193.371	20		1.5198	91.770	278.79	278.79
211	$[(CH_3)_2C](NH_2)_2SiC_6H_5$	(46)	194.360	20		1.0252	1.5130	59.061	59.061
212	$Cl(C_2H_5)_2SiC_6H_5$	(44)	198.777	20		1.2666	1.5316 ^d	52.429	52.429
213	$Br(CH_3)_2SiC_6H_5$	(25)	215.182	25	1.5294	1.0109	1.5229	68.534	332.60
214	$CH_3CICH_2(C_2H_5)_2SiC_6H_5$	(45)	226.831	20		1.7293	1.5800 ^d	51.039	342.32
215	$BrHSiC_6H_5$	(24)	266.036	25	1.5778	0.8719	1.5802	61.193	417.72
216	$(CH_3)_2SiC_6H_5$	(15)	276.544	20		1.5970	1.5992 ^d	91.770	412.99
217	$Br_2SiC_6H_5$	(22)	344.944	25	1.5958	2.0228	58.102	58.360	542.61
218	$(CH_3)_2SiC_6H_5OCH_3(p-)$	(24)	374.971	25	1.6179	1.9089	1.5980 ^d	66.819	585.27
219	$[(C_2H_5)_2SiC_6H_5]_2O$	(44)	342.134	25		0.9828	1.5214	106.228	106.228
220	$H_3Si(C_6H_5)_2$	(63)	146.390	20		1.0027	1.5802	51.220	51.220
221	$(CH_3)_2Si(C_6H_5)_2$	(22)	212.372	20	0.9880	1.5639	69.903	69.879	291.26
222	$(C_2H_5O)_2Si(C_6H_5)_2$	(18)	272.426	20		1.0334	1.5235	80.595	80.803
223	$Br_2Si(C_6H_5)_2$	(24)	342.134	25		1.5869	1.6201 ^d	75.527	544.47
224	$[(CH_3)_2SiC_6H_5]_2$	(60)	146.390	20		0.7268	1.4229	51.279	52.107
225	$(CH_3)_2Si(C_6H_5)_2F$	(20)	150.355	20	0.8083	1.4031	45.403	45.299	213.18
	$[(CH_3)_2FSi]_2$	(20)	154.320	20	0.9120	1.3837	39.535	39.752	215.26

Table II. Molar Values for Organosilicon Compounds (Continued)

No.	Compound	Ref.	<i>M</i>	<i>t</i>	<i>n</i> _D ⁱ	<i>d</i> _i	Lorenz-Lorentz Molar Refractivity ^a		Eisenlohr Molar Refraction Product ^b	
							<i>n</i> _D ^b	<i>n</i> _D ^c	Obsd.	Calcd.
227	(CH ₃) ₃ Si—Si(CH ₃) ₂ Cl	(43)	166.812	20	1.4430	1.4201	44.920	45.126	240.71	243.14
228	(CH ₃) ₂ Clsi—Si(CH ₃) ₂ F	(20)	170.777	20	0.9623	1.4229	56.192	56.318	242.52	245.20
229	(CH ₃) ₂ Si—Si(OC ₂ H ₅) ₂	(20)	176.417	20	0.7993	1.4576	55.796	55.680	251.02	252.52
230	CH ₃ Cl(CH ₃) ₂ Si—Si(CH ₃) ₂	(19)	180.839	20	0.8837	1.4545	50.234	50.501	263.59	264.04
231	[(CH ₃) ₂ Clsi—Si(CH ₃) ₂] ₂	(20)	187.234	20	1.0103	1.4662	59.993	60.439	272.33	275.17
232	CH ₂ Cl(CH ₃) ₂ Si—Si(C ₂ H ₅) ₂ (CH ₃) ₂	(19)	194.866	20	0.8933	1.4735	55.343	55.507	284.93	285.71
233	CH ₃ Cl(CH ₃) ₂ Si—Si(Cl)(CH ₃) ₂	(19)	201.161	20	1.0206	1.4240	61.924	61.770	296.41	296.07
234	[(CH ₃) ₂ C ₂ H ₅ Os—Si—] ₂	(20)	206.444	20	0.8507	1.0506	70.831	70.787	293.98	293.94
235	(CH ₃) ₂ Si—Si(CH ₃) ₂ C ₂ H ₅	(14)	208.461	20	0.8738	1.4790	78.288	78.734	313.86	317.91
236	[(CH ₃) ₂ Si—Si—] ₂	(60)	230.552	20	0.8351	1.4220	67.724	67.233	334.71	336.26
237	(CH ₃) ₂ (C ₂ H ₅ O) ₂ Si—Si(OC ₂ H ₅) ₂ CH ₃	(20)	236.471	20	0.8873	1.4070	54.491	55.731	335.36	336.00
238	[(CH ₃ O) ₃ Si—Si—] ₂	(28)	242.390	20	1.095	1.4200	72.657	72.695	378.43	376.77
239	[CH ₃ (C ₂ H ₅ O) ₂ Si—Si—] ₂	(20)	266.498	20	0.9282	1.5606	90.467	90.728	422.19	424.84
240	(CH ₃) ₂ Si—Si(CH ₃)(C ₆ H ₅) ₂	(14)	270.532	20	0.9678	1.4721	106.311	106.622	463.29	458.31
241	[(C ₃ H ₇) ₂ Si—Si—] ₂	(60)	314.714	20	0.8291	1.4102	83.205	83.619	460.50	459.60
242	[(C ₂ H ₅ O) ₃ Si—Si—] ₂	(29)	326.552	20	0.9728	1.4694	133.837	134.510	581.91	580.44
243	[(CH ₃) ₂ Si—Si—] ₂	(60)	398.876	20	0.8306	1.4667	162.174	162.398	708.47	705.51
244	[(i-C ₄ H ₉) ₂ Si—Si—] ₂	(60)	483.038	20	0.8260	1.4691	188.848	190.286	833.27	829.11
245	[(C ₆ H ₅) ₂ Si—Si—] ₂	(60)	567.200	20	0.8366	1.4612	72.330	71.695	298.89	301.49
246	(CH ₃) ₂ SiSi(CH ₃) ₂ Si(CH ₃) ₂	(65)	204.550	20	0.7763	1.4877	93.789	92.543	391.88	390.83
247	[(CH ₃) ₃ SiSi(CH ₃) ₂ —Si—] ₂	(65)	262.710	20	0.8066	1.4742	1.5044	1.4949	37.517	38.032
248	(CH ₃) ₂ SiI	(1)	200.105	20	1.903	1.351	1.229	1.229	362.48	368.38
249	Cl ₂ CH ₂ SiI	(2)	240.949	20	1.949	1.4880	66.640	66.301	362.04	366.11
250	(C ₂ H ₅) ₂ SiI	(61)	242.186	20	1.086	1.4842	94.399	94.189	422.99	427.91
251	(C ₃ H ₇) ₂ SiI	(61)	284.267	20	1.600	1.4810	107.573	108.133	607.96	613.31
252	[(CH ₃) ₂ CHCH ₂ CH ₃) ₂ SiI	(61)	368.429	20	2.451	1.5714	56.713	56.126	519.94	534.33
253	(C ₆ H ₅) ₂ SiI	(2)	332.402	20	1.9711	1.5714	529.11	529.11	534.33	534.33
254	ClCH ₂ SiI ₂	(9)	340.034	20						

$$^a MR_i = \frac{(n^2 - 1)}{(n^2 + 2)} \cdot \frac{M}{d} \quad ^b MR_e = Mn_D^{20}$$

^cThe refractive index of Compound 29 is given as 1.5330 on Page 17 and in C.A. (18); this is obviously a typographical error, since Page 13 gives 1.4330, and 37.41 for MR_i . ^d Corrected by use of coefficient 0.0045/^oC.

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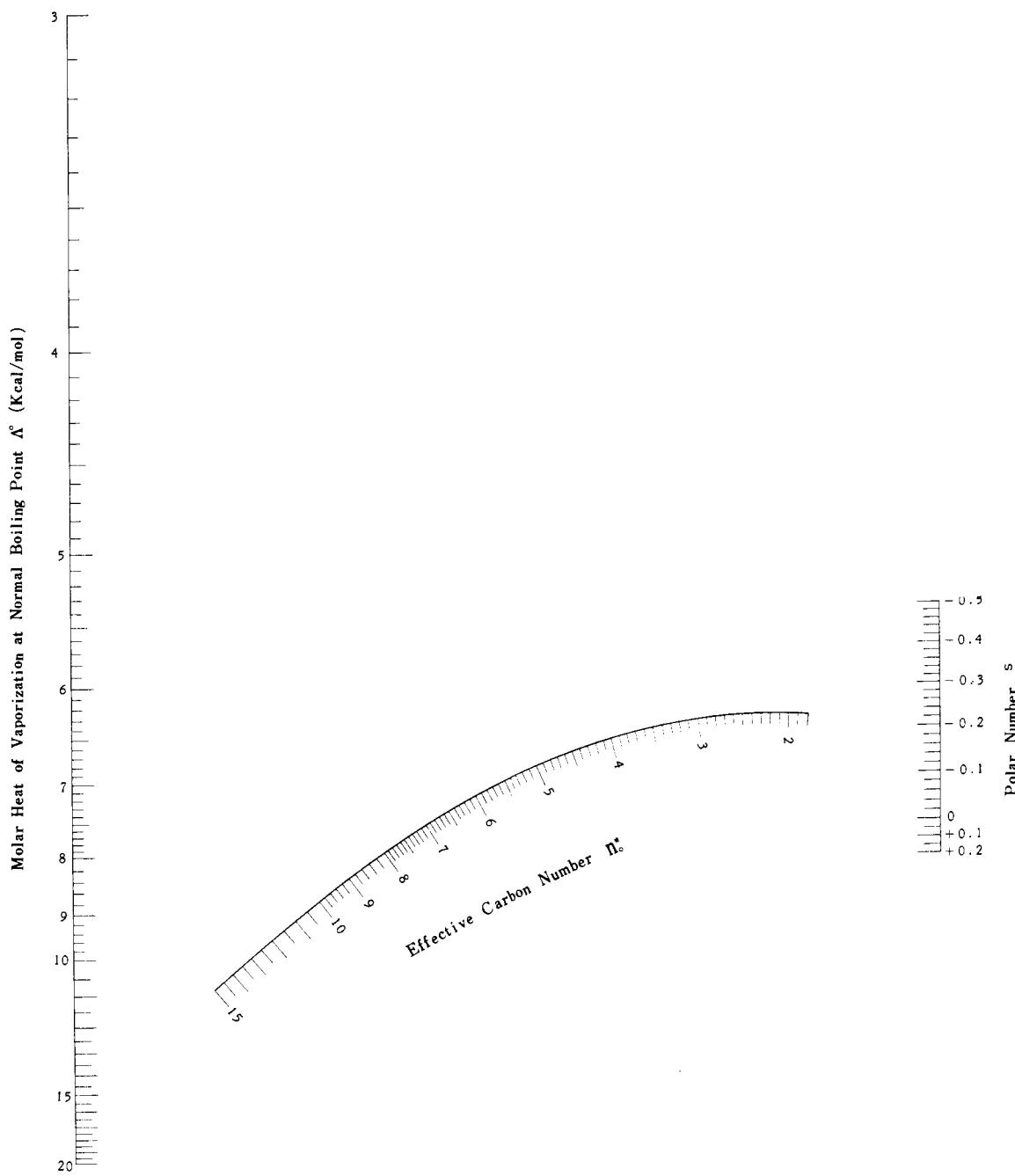


Figure 6. Heat of vaporization at normal boiling point nomogram (nomogram 3)

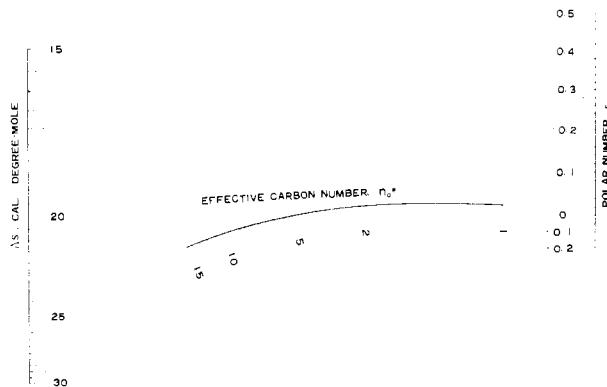


Figure 7. Entropy of vaporization at normal boiling point nomogram (nomogram 4)

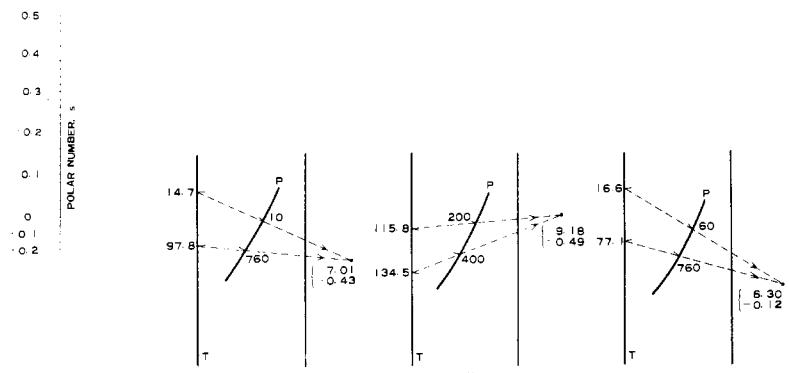


Figure 8. Determination of point representing each compound on nomogram

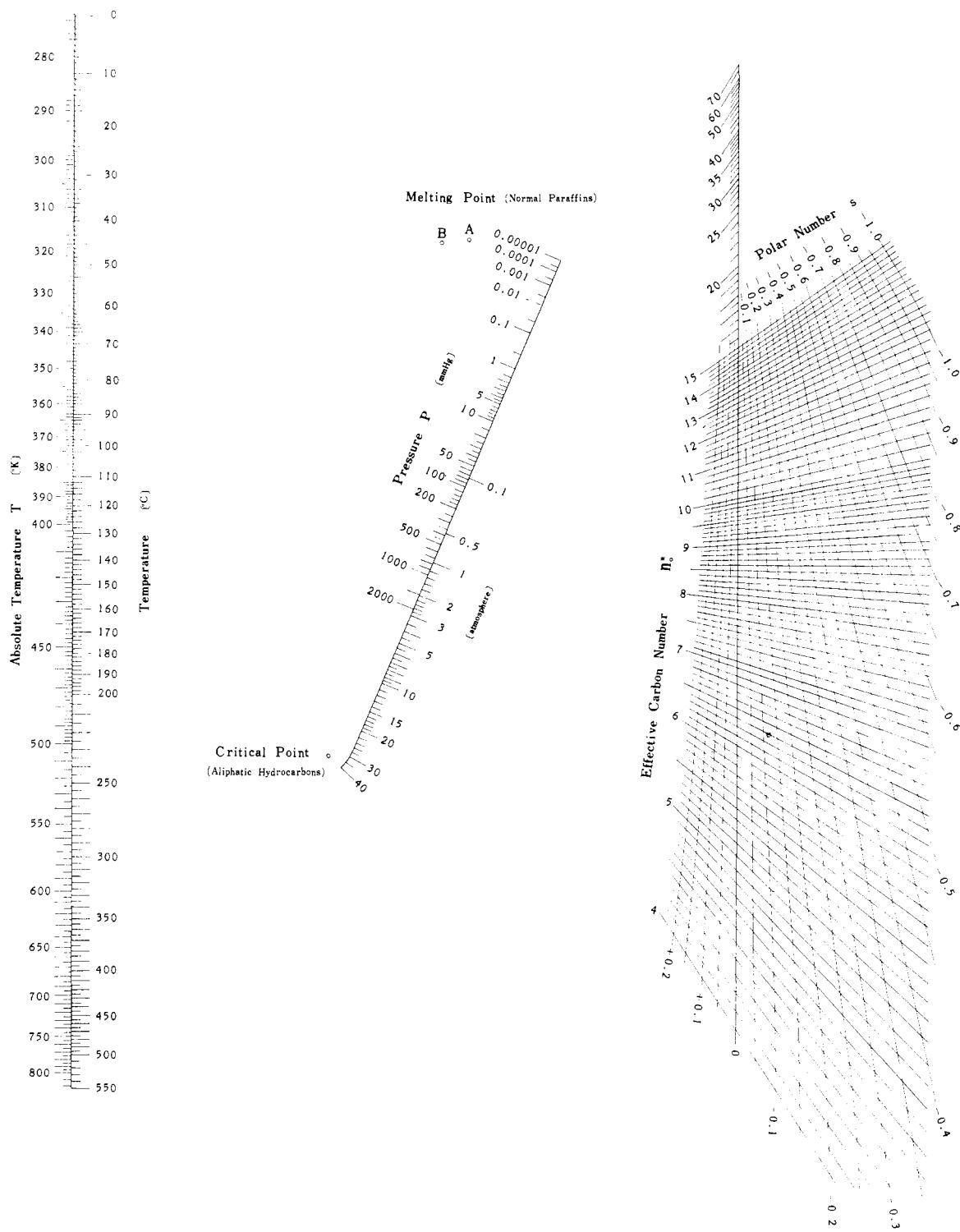


Figure 5. Generalized vapor pressure nomogram II (nomogram 2)

CORRECTION:

In the article, "Generalized Vapor Pressure Nomogram for Organic Liquids," [J. CHEM. ENG. DATA 8, 355 (1963)], the figures and captions were not in proper order. The figures with their correct captions are published from pages 154-157.

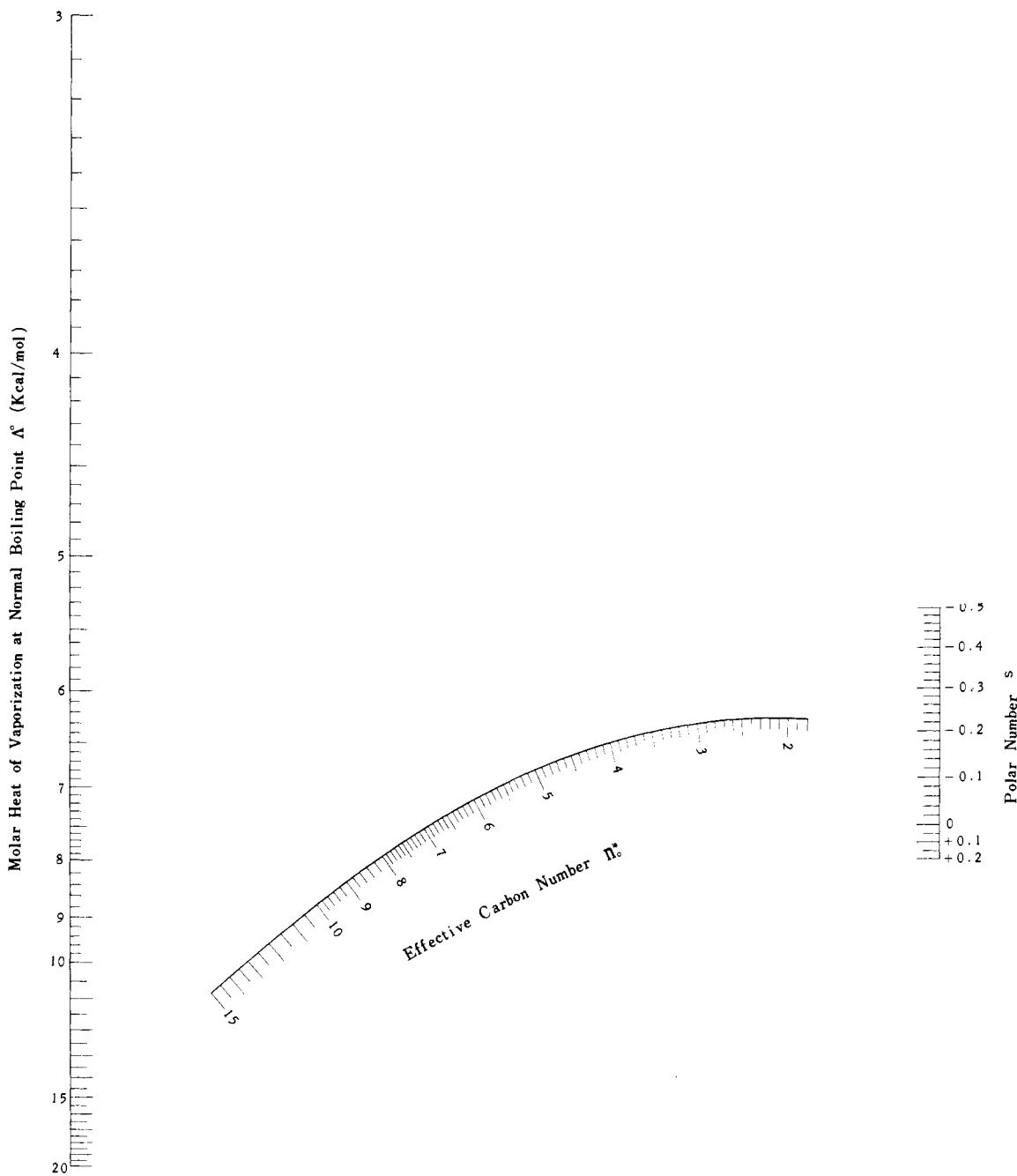


Figure 6. Heat of vaporization at normal boiling point nomogram (nomogram 3)

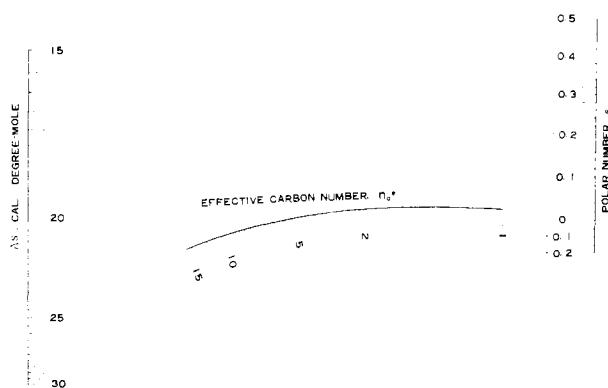


Figure 7. Entropy of vaporization at normal boilint point nomogram (nomogram 4)

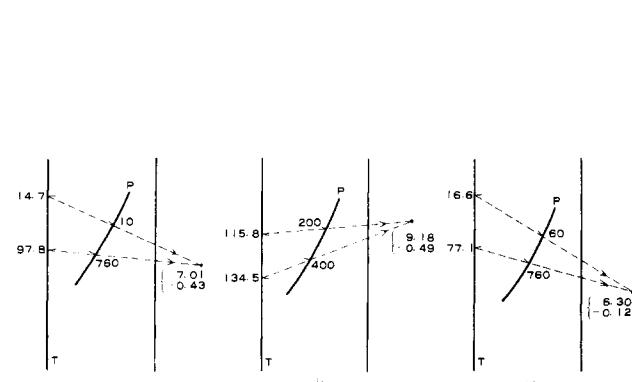


Figure 8. Determination of point representing each compound on nomogram

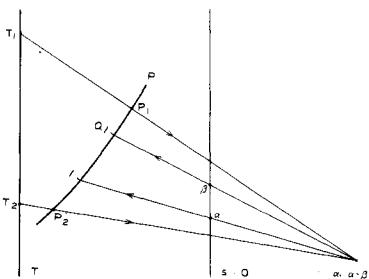


Figure 9. Determination of n_o^* and s on nomogram

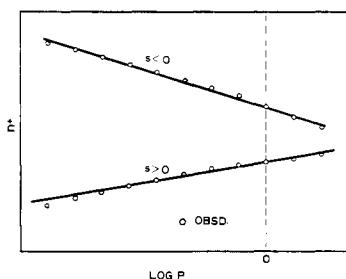


Figure 10. Detailed analysis of n^* vs. $\log P$ data

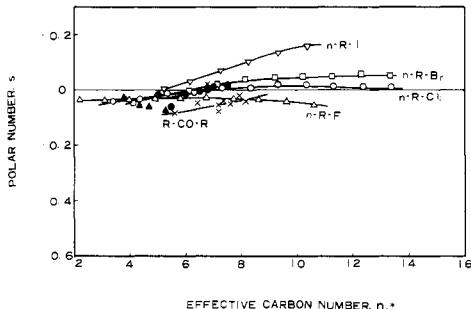


Figure 11. s vs. n_o^* relationship in mono-substituted alkane homologs (1) R- and R'- are alkyl groups and n-R-, normal alkyl group: closed marks are RX

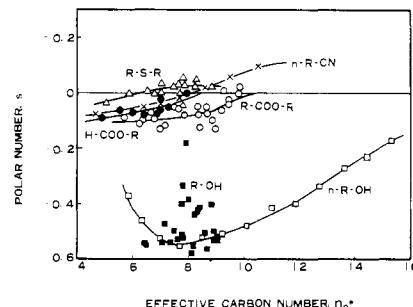


Figure 12. s vs. n_o^* relationship in mono-substituted alkane homologs (2)

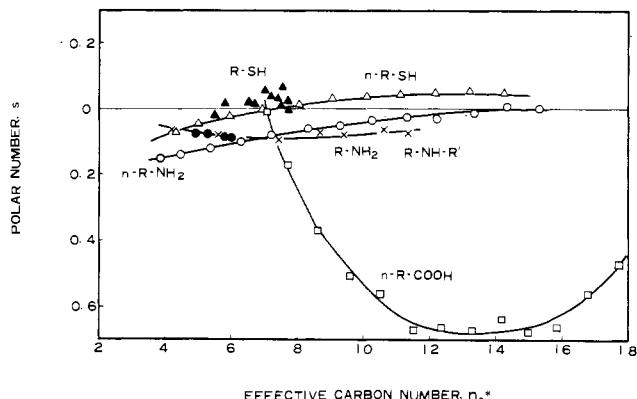


Figure 13. s vs. n_o^* relationship in mono-substituted alkane homologs (3)

CORRECTION:

In the article "Vapor-Liquid Equilibria" [J. CHEM. ENG. DATA 8, 549 (1963)] the name of one of the coauthors is misspelled. A.K. Keshpande should read as A.K. Deshpande. The same error is made in the index, p. 63; on the front cover the name appears as A.K. Koshpande.

CORRECTION:

In the article entitled, "A Generalized Equation for Diffusion in Liquids," by Ramalingam Sitaraman, S.H. Ibrahim, and N.R. Kuloor [J. CHEM. ENG. DATA 8, 198 (1963)], there are two errors. In Equation 1 the term $(XM_s)^{1/2}$ should read $(XM_s)^{1/2}$. In equation 2 the term $L_s^{0.3}$ should read $L^{0.3}$.