

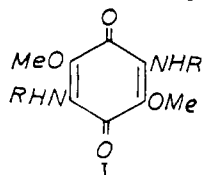
Preparation of 2,5-Bis(alkylamino)-3,6-dimethoxy-*p*-benzoquinones

HERBERT S. VERTER, Inter American University of Puerto Rico, San German, P. R.

ROBERT DOMINIC, Central Michigan University, Mt. Pleasant, Mich.

The properties and method of preparation of several 2,5-bis(alkylamino)-3,6-dimethoxy-*p*-benzoquinones are described.

REACTION of tetramethoxyquinone with primary amines yielded 2,5-bis(alkylamino)-3,6-dimethoxy-*p*-benzoquinones of general formula I. The scope of the reaction has



been previously discussed (2). Table I summarizes the preparation and properties of several other members of this series. The reactions proceeded smoothly in methanol at reflux temperature. Approximately 30 ml. of methanol per

the color of these quinones, has been commented on in detail elsewhere (1, 3). Ultraviolet absorption maxima of quinones derived from aromatic amines occur at longer wavelength than the maxima of quinones derived from aliphatic amines. The spectra of the aliphatic aminoquinones show absorptions at 220 μ ($\epsilon = 25,000$) and 369 μ ($\epsilon = 20,000$), while the spectra of the aromatic aminoquinones show absorptions at 260 μ ($\epsilon = 25,000$) and 400 μ ($\epsilon = 17,000$). The infrared spectra of these and related compounds (1, 2) show absorption at 3.0 microns (N—H) and at 6.1 and 6.4 microns (quinone). Infrared spectra were obtained on a Perkin-Elmer Infracord, Model 137; ultraviolet spectra were obtained on a Bausch and Lomb Spectronic 505.

Table I. Characterization of 2,5-Bis(alkylamino)-3,6-dimethoxy-*p*-benzoquinones

Amine	M.P., ° C.	Formula	Calcd., %		Found, %		Color	Reflux, Hr.	Crystal. Solvent	Yield, %
			C	H	C	H				
Furfuryl	147	C ₁₈ H ₁₈ N ₂ O ₆	60.3	5.1	60.2	5.2	Black	10	EtOH	78
Cyclohexyl	165-190 ^a	C ₂₀ H ₃₀ N ₂ O ₄	66.3	8.3	66.2	8.4	Green	1	EtOH	90
Benzyl	213	C ₂₂ H ₂₂ N ₂ O ₄	69.8	5.9	69.5	6.0	Black	0.1	CHCl ₃	90
α -Phenylethyl	183	C ₂₄ H ₂₆ N ₂ O ₄	70.9	6.4	70.9	6.5	Black	21	Acetone	88
β -Phenylethyl	192	C ₂₄ H ₂₆ N ₂ O ₄	70.9	6.4	70.7	6.4	Violet	0.3	C ₆ H ₆	92
<i>o</i> -Toluidine	235	C ₂₂ H ₂₂ N ₂ O ₄	69.8	5.9	69.8	5.9	Green	144	EtOH	14
<i>m</i> -Toluidine	250	C ₂₂ H ₂₂ N ₂ O ₄	69.8	5.9	69.7	5.7	Black	20	CH ₂ Cl ₂	85
<i>o</i> -Ethylaniline	196	C ₂₄ H ₂₆ N ₂ O ₄	70.9	6.4	70.8	6.6	Green	117	EtOH	11
<i>p</i> -Ethylaniline	203	C ₂₄ H ₂₆ N ₂ O ₄	70.9	6.4	71.0	6.6	Green	21	EtOH	86
<i>o</i> -Phenetidine	183	C ₂₄ H ₂₆ N ₂ O ₆	65.7	6.0	65.9	6.0	Black	53	EtOH	50
<i>p</i> -Phenetidine	263	C ₂₄ H ₂₆ N ₂ O ₆	65.7	6.0	65.5	5.8	Black	15	Dioxane	93
2,3-Dimethylaniline	254	C ₂₄ H ₂₆ N ₂ O ₄	70.9	6.4	71.1	6.5	Green	87	C ₆ H ₆	18

^a Decomposes.

gram of tetramethoxyquinone was used. Usually a ratio of 4 moles of amine to 1 mole of tetramethoxyquinone was sufficient to produce complete reaction. Purity of the compounds was established by thin-layer chromatography and by elemental analysis. The compounds moved as a single spot on silica, and their analysis was within the acceptable range.

The auxochromic effect of amino groups, which deepens

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Tetrachloroethyl and Trichlorovinyl Thiolsulfonate Esters

JAMES P. WEIDNER, SEYMOUR S. BLOCK, and WALLACE S. BREY

University of Florida, Gainesville, Fla.

Trichlorovinyl and tetrachloroethyl thiolsulfonates have been prepared by treating sodium and zinc sulfinates with tetrachloroethanesulfonyl chloride. The resulting compounds are generally mixtures of the trichlorovinyl and tetrachloroethyl forms and are purified with difficulty by thin-layer chromatography. Infrared and NMR spectra are given.

TRICHLOROMETHYL thiolsulfonates (1, 2) are the only thiolsulfonate esters that have been reported with halogen substitution in the *S*-alkyl grouping. Tetrachloroethanesulfonyl chloride was shown to react with various nucleophiles—e.g., sodium phthalimide and sodium cyanide

(3)—and has been employed here to produce new halogen-substituted *S*-alkyl thiolsulfonates.

In the above report (3) of the reaction of tetrachloroethanesulfonyl chloride with salts of phthalimide or tetrahydrophthalimide, it was noted that *N*-trichlorovinylthio-