Corrections

For the paper, "Heat Capacities of Liquid Hydrocarbons. Estimation of Heat Capacities at Constant Pressure as a Temperature Function, Using Additivity Rules", by M. Luria and S. W. Benson (*J. Chem. Eng. Data* **1977**, *22*, 90), the corrections are as follows: in Table I, the heading should read $+DT^3$ not $-DT^3$, in Table I, column item A, the 6th number down should be 8.153 not 8.754, in Table I, column item B, the 8th number down should be -1.568E-2 not -1568-1, in Table II, column item D, the 3rd number from the bottom should be 3.167E-6 not 3.167E-F.

For the paper, "Vaporization Thermodynamics of Lanthanide Trihalides", by C. E. Myers and D. T. Graves (*J. Chem. Eng. Data* **1977**, *22*, 440), the beginning of the last paragraph on p 443 should read as follows: Hastie and co-workers (*15*) have observed the $La_2Cl_5^+$ ion above $LaCl_3(s)$ in the range 1000–1100 K and, using the pressure data from Shimazaki and Niwa (*34*) in a second-law treatment, calculate $\Delta H = -48$ kcal/mol for formation of the dimer from the monomer in the gaseous state at 1030 K. They also observed the ions $Eu_2Cl_5^+$ and $Lu_2Cl_5^+$ above a liquid mixture of $EuCl_3$ and $LuCl_3$ and calculate the respective enthalpies of formation of the dimer from the monomer at 1000 K to be -32 and -47 kcal/mol.