Corrections

Excess Gibbs Free Energies at Eight Temperatures and Excess Enthalpies and Volumes at T = 298.15 K for Butanenitrile + 2-Butanol. Rosa Garriga, Francisco Sánchez, Pascual Pérez, and Mariano Gracia, *J. Chem. Eng. Data* **1997**, *42*, 78−83.

Recently we found a mistake in V^E measurements for {(1 − x)butanenitrile + x2-butanol}, which is important to correct in order to allow others to take advantage of our data.

The V^E values that appear in the original Table 4 must be replaced by:

x	V^E/cm^3·mol^{-1}	x	V^E/cm^3·mol^{-1}
0.0285	0.046	0.5383	0.348
0.0778	0.122	0.7142	0.288
0.1850	0.239	0.7171	0.286
0.3208	0.318	0.8350	0.206
0.3633	0.334	0.8470	0.195
0.4757	0.351	0.9626	0.068
0.5205	0.349		

$$V^E/\text{cm}^3\cdot\text{mol}^{-1} = x(1-x)\{1.396 + 0.078(1-2x) + 0.339(1-2x)^2\};\ \sigma(V^E) = 0.004$$

Consequently, Figure 3b must be modified, and the second to last sentence would be "V^E is also lower for 1-butanol than for 2-butanol".

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Vapor−Liquid Equilibria of Binary Mixtures Cyclopentane + Isopropyl Acetate, Isopropyl Acetate + Hexane, and Cyclopentane + Methyl Methacrylate at 101.3 kPa. Kong-Wei Cheng, Jia-Yuh Chen, Muoi Tang, and Yan-Ping Chen, *J. Chem. Eng. Data* **1997**, *42*, 754−757.

The unit of the binary parameters A_{12} and A_{21} listed in Table 7 should be (cal mol^{-1}). It was misprinted as (J mol^{-1}) in the original article.

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Phase Equilibria in the Systems Oxolane + Octane and Methyl 1,1-Dimethylethyl Ether + Hex-1-ene. Jaime Wisniak, Eti Fishman, and Rotem Shaulovitch, *J. Chem. Eng. Data* **1998**, *43*, 304−306.

There is a misprint in the temperatures listed in Table 2. The correct values are

T/K
391.40
388.27
381.80
377.95
372.33
367.34
360.39
352.35
348.60
347.28
346.08
344.90
343.92
342.75
341.49
339.95
338.68
338.04

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