Correction

Thermodynamic Properties of Binary Mixtures of *p*-Xylene with Cyclohexane, Heptane, Octane, and *N*-Methyl-2-pyrrolidone at Several Temperatures. Changsheng Yang, Wei Xu, and Peisheng Ma* *J. Chem. Eng. Data* **2004**, *49*, 1794−1801.

In this paper, there are two errors in Table 2. For the mixture *p*-xylene + heptane, the compositions at 298.15 K, 313.15 K, 333.15 K, and 343.15 K are incorrect and some of the density values are incorrect. The correct values are given below in Table 2.

Table 2. Densities, ρ, Viscosities, η, Excess Molar Volumes, V^E, and Viscosity Deviations, $\Delta\eta$, for the Binary Mixtures from 298.15 K to 353.15 K

x_1	$\rho/\text{g·cm}^{-3}$	$\eta/\text{mP·s}$	$V^E/\text{cm}^3\text{·mol}^{-1}$	$\Delta\eta/\text{mP·s}$	x_1	$\rho/\text{g·cm}^{-3}$	$\eta/\text{mP·s}$	$V^E/\text{cm}^3\text{·mol}^{-1}$	$\Delta\eta/\text{mP·s}$
				$(x)p$-Xylene + $(1-x_1)$Heptane					
				T = 298.15 K					
0.0000	0.67990	0.4009	0.0000	0.000	0.5861	0.77651	0.4807	−0.0889	−0.046
0.0948	0.69432	0.4074	−0.0232	−0.014	0.6877	0.79516	0.5047	−0.0761	−0.044
0.1911	0.70939	0.4176	−0.0383	−0.024	0.7906	0.81474	0.5331	−0.0619	−0.038
0.2880	0.72504	0.4295	−0.0522	−0.033	0.8949	0.83524	0.5714	−0.0315	−0.022
0.3865	0.74150	0.4436	−0.0703	−0.041	1.0000	0.85670	0.6162	0.0000	0.000
0.4856	0.75845	0.4598	−0.0724	−0.046					
				T = 303.15 K					
0.0000	0.67565	0.3750	0.0000	0.0000	0.5861	0.77219	0.4496	−0.0962	−0.041
0.0948	0.69005	0.3820	−0.0243	−0.012	0.6877	0.79083	0.4716	−0.0824	−0.039
0.1911	0.70512	0.3912	−0.0439	−0.021	0.7906	0.81042	0.4979	−0.0696	−0.032
0.2880	0.72075	0.4019	−0.0577	−0.030	0.8949	0.83091	0.5308	−0.0364	−0.020
0.3865	0.73720	0.4154	−0.0771	−0.036	1.0000	0.85235	0.5715	0.0000	0.000
0.4856	0.75424	0.4304	−0.0816	−0.040					
				T = 313.15 K					
0.0000	0.66708	0.3373	0.0000	0.000	0.5861	0.76352	0.4034	−0.1136	−0.034
0.0948	0.68145	0.3440	−0.0280	−0.010	0.6877	0.78215	0.4234	−0.0981	−0.031
0.1911	0.69652	0.3523	−0.0563	−0.018	0.7906	0.80171	0.4440	−0.0787	−0.028
0.2880	0.71212	0.3623	−0.0710	−0.024	0.8949	0.82221	0.4729	−0.0435	−0.017
0.3865	0.72857	0.3740	−0.0960	−0.029	1.0000	0.84364	0.5080	0.0000	0.000
0.4856'	0.74557	0.3865	−0.0966	−0.034					
				T = 323.15K					
0.0000	0.65837	0.3074	0.0000	0.000	0.5861	0.75477	0.3683	−0.1337	−0.029
0.0948	0.67273	0.3136	−0.0349	−0.008	0.6877	0.77341	0.3835	−0.1177	−0.028
0.1911	0.68779	0.3211	−0.0687	−0.016	0.7906	0.79297	0.4041	−0.0940	−0.024
0.2880	0.70338	0.3304	−0.0869	−0.021	0.8949	0.81345	0.4285	−0.0490	−0.015
0.3865	0.71983	0.3408	−0.1160	−0.026	1.0000	0.83490	0.4599	0.0000	0.000
0.4856	0.73682	0.3529	−0.1164	−0.029					
				T = 333.15 K					
0.0000	0.64953	0.2793	0.0000	0.000	0.5861	0.74592	0.3345	−0.1558	−0.033
0.0948	0.66388	0.2848	−0.0417	−0.009	0.6877	0.76457	0.3537	−0.1379	−0.029
0.1911	0.67895	0.2922	−0.0850	−0.016	0.7906	0.78415	0.3712	−0.1117	−0.027
0.2880	0.69452	0.2997	−0.1038	−0.023	0.8949	0.80464	0.3973	−0.0597	−0.017
0.3865	0.71098	0.3097	−0.1384	−0.028	1.0000	0.82609	0.4298	0.0000	0.000
0.4856	0.72797	0.3201	−0.1394	−0.032					
				T = 343.15 K					
0.0000	0.64047	0.2541	0.0000	0.000	0.5861	0.73691	0.3030	−0.1815	−0.024
0.0948	0.65483	0.2612	−0.0524	−0.005	0.6877	0.75557	0.3166	−0.1592	−0.023
0.1911	0.66989	0.2658	−0.0996	−0.012	0.7906	0.77517	0.3317	−0.1278	−0.021
0.2880	0.68548	0.2730	−0.1263	−0.017	0.8949	0.79569	0.3498	−0.0691	−0.015
0.3865	0.70194	0.2818	−0.1626	−0.020	1.0000	0.81717	0.3783	0.0000	0.000
0.4856	0.71894	0.2915	−0.1643	−0.023					