Correction

Hydroxyl Ammonium Ionic Liquids: Synthesis, Properties, and Solubility of SO₂. Xiao Liang Yuan, Suo Jiang Zhang,* and Xing Mei Lu *J. Chem. Eng. Data* **2007**, *52*, 596–599.

Page 597. The units for the *x*-axis of Figures 2 and 4 should be t/min not t/s, and "(F)" in the caption of Figure 4 should be deleted.

Table 1. Properties of the Synthesized ILs

| ionic liquids | $T_{ m m}/{ m K}$ | $T_{ m g}/{ m K}$ |
|--------------------------------------|-------------------|-------------------|
| 2-hydroxyethylammonium formate | | 188.1 |
| 2-hydroxyethylammonium acetate | | 197.2 |
| 2-hydroxyethylammonium lactate | | 209.2 |
| tri-(2-hydroxyethyl)ammonium formate | 334.9 | |
| tri-(2-hydroxyethyl)ammonium acetate | 294.9 | 194.3 |
| tri-(2-hydroxyethyl)ammonium lactate | | 213.8 |
| 2-(2-hydroxyethoxy)ammonium formate | | 194.4 |
| 2-(2-hydroxyethoxy)ammonium acetate | | 208.5 |
| 2-(2-hydroxyethoxy)ammonium lactate | | 213.2 |

Page 598. In Table 1, the glass transition temperatures and melting temperatures of the ionic liquids contained some incorrect values. The correct values are given here.

Page 598. The last sentence of the paragraph *Properties of ILs* (For example, the melting temperature of tri-(2-hydroxy ethyl)-ammonium lactate is 217.6 K and its decomposition temperature is 547.8 K.) should be deleted.

The experimental data were not given in the original paper. Both the experimental data and DSC traces are given in the Supporting Information associated with this correction.

Acknowledgment

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Supporting Information Available:

Experimental data and DSC traces. This material is available free of charge via the Internet at http://pubs.acs.org.

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