

Density Functional Theory Calculation on Polychlorinated Anthraquinones: Their Gas Phase Thermodynamic Function and Implication of the Cl Substituted Position.

Chen Wang, Hong X. Liu, Zhe Y. Fang, and Zun Y. Wang,* *J. Chem. Eng. Data* DOI: 10.1021/je9003564.

Because the molecular symmetry was missed in the calculation process, the values of je-2009-003564 have been corrected considering molecular symmetry. All molecules were calculated at the B3LYP/6-31G* level, and the vibrational scaling factor is 0.96.

Eq 13 should be replaced by:

$$\begin{aligned} S^\theta/J \cdot \text{mol}^{-1} \cdot \text{K}^{-1} &= 448.90 + 29.18N \\ R^2 &= 0.980 \quad SE = 6.40 \quad F = 3657.873 \end{aligned} \quad (13)$$

Equations 14 and 15 should be replaced by:

$$\begin{aligned} \Delta_r H^\theta/kJ \cdot \text{mol}^{-1} &= -118.218 + 1.424N_\alpha - 26.971N_\beta + \\ &14.712N_o + 3.212N_m + 4.173N_p \\ R^2 &= 0.988 \quad SE = 2.352 \quad F = 1139.471 \end{aligned} \quad (14)$$

$$\begin{aligned} \Delta_r G^\theta/kJ \cdot \text{mol}^{-1} &= -8.886 + 5.733N_\alpha - 21.994N_\beta + \\ &15.329N_o + 3.321N_m + 5.549N_p \\ R^2 &= 0.990 \quad SE = 2.398 \quad F = 1352.333 \end{aligned} \quad (15)$$

JE100011N

10.1021/je100011n

Published on Web 02/16/2010

Table 5. Thermodynamic Parameters of PCAQNs Computed at B3LYP/6-31G* Level and N_{PCS}^a

molecule	group	symmetry number	S^0	$\Delta_f H^0$	$\Delta_f G^0$	$\Delta_f G^0_{\text{R}}$	$C_{p,m}/\text{J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$											
			$\text{J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$	$\text{kJ}\cdot\text{mol}^{-1}$	$\text{kJ}\cdot\text{mol}^{-1}$	$\text{kJ}\cdot\text{mol}^{-1}$	constant	$10^{-3} T$	$10^5 T^{-1}$	$10^7 T^{-2}$	R^2	SE	N_α	N_β	N_o	N_m	N_p	$N_{1,8}$
AQN	D_{2h}	4	430.53	-119.40	-6.26		565.63	43.76	-1.60	1.47	1.000	1.34	0	0	0	0	0	0
mono-CAQN																		
1	C_1	1	476.54	-117.93	-4.31	27.56	578.77	36.17	-1.57	1.42	1.000	1.28	1	0	0	0	0	0
2	C_s	1	473.08	-146.53	-31.87	0.00	578.09	36.55	-1.57	1.42	1.000	1.29	0	1	0	0	0	0
di-CAQN																		
1,2	C_1	1	507.44	-129.25	-10.63	40.86	591.84	28.76	-1.54	1.37	1.000	1.23	1	1	1	0	0	0
1,3	C_1	1	507.10	-142.24	-23.52	27.96	591.39	28.93	-1.53	1.37	1.000	1.23	1	1	0	1	0	0
1,4	C_s	1	514.88	-108.29	8.11	59.59	591.78	28.65	-1.53	1.36	1.000	1.23	2	0	0	0	1	0
1,5	C_2	2	517.75	-116.80	-1.26	50.23	591.84	28.65	-1.53	1.36	1.000	1.23	2	0	0	0	0	0
1,6	C_1	1	507.32	-144.56	-25.90	25.58	591.18	29.00	-1.53	1.36	1.000	1.24	1	1	0	0	0	0
1,7	C_1	1	507.20	-144.71	-26.02	25.46	591.28	28.94	-1.53	1.36	1.000	1.23	1	1	0	0	0	0
1,8	C_s	1	505.21	-114.53	4.75	56.24	591.44	28.84	-1.53	1.36	1.000	1.23	2	0	0	0	0	1
2,3	C_{2v}	2	496.03	-160.75	-38.73	12.75	590.48	29.61	-1.53	1.37	1.000	1.25	0	2	1	0	0	0
2,6	C_{2h}	2	497.50	-173.07	-51.49	0.00	590.49	29.39	-1.53	1.36	1.000	1.24	0	2	0	0	0	0
2,7	C_{2v}	2	497.60	-172.94	-51.39	0.10	590.55	29.34	-1.53	1.36	1.000	1.24	0	2	0	0	0	0
tri-CAQN																		
1,2,3	C_1	1	537.04	-140.20	-16.20	45.01	604.81	21.51	-1.51	1.32	1.000	1.19	1	2	2	1	0	0
1,2,4	C_1	1	540.64	-116.93	5.99	67.21	604.71	21.37	-1.50	1.31	1.000	1.18	2	1	1	1	1	0
1,2,5	C_1	1	541.94	-128.05	-5.52	55.70	604.78	21.30	-1.50	1.31	1.000	1.18	2	1	1	0	0	0
1,2,6	C_1	1	537.84	-155.40	-31.64	29.58	604.21	21.63	-1.50	1.31	1.000	1.19	1	2	1	0	0	0
1,2,7	C_1	1	538.48	-155.42	-31.86	29.36	604.39	21.52	-1.50	1.31	1.000	1.19	1	2	1	0	0	0
1,2,8	C_1	1	533.09	-126.27	-1.09	60.13	604.24	21.61	-1.50	1.31	1.000	1.18	2	1	1	0	0	1
1,3,5	C_1	1	544.98	-140.87	-19.24	41.97	604.59	21.35	-1.50	1.31	1.000	1.18	2	1	0	1	0	0
1,3,6	C_1	1	537.80	-168.13	-44.36	16.86	603.94	21.69	-1.50	1.31	1.000	1.19	1	2	0	1	0	0
1,3,7	C_1	1	537.76	-168.42	-44.63	16.59	604.04	21.62	-1.50	1.31	1.000	1.18	1	2	0	1	0	0
1,3,8	C_1	1	535.83	-138.27	-13.92	47.30	604.08	21.62	-1.50	1.31	1.000	1.18	2	1	0	1	0	1
1,4,5	C_1	1	535.25	-109.56	14.97	76.19	604.02	21.58	-1.49	1.30	1.000	1.18	3	0	0	0	1	1
1,4,6	C_1	1	546.90	-134.65	-13.60	47.62	604.33	21.40	-1.50	1.31	1.000	1.18	2	1	0	0	1	0
2,3,5	C_1	1	536.06	-158.39	-34.10	27.12	603.65	22.03	-1.50	1.31	1.000	1.20	1	2	1	0	0	0
2,3,6	C_1	1	532.59	-186.54	-61.22	0.00	603.01	22.39	-1.50	1.31	1.000	1.20	0	3	1	0	0	0
TCAQN																		
1,2,3,4	C_s	1	566.58	-110.07	19.32	84.91	618.85	13.53	-1.47	1.27	1.000	1.13	2	2	3	2	1	0
1,2,3,5	C_1	1	585.04	-138.77	-14.88	50.71	617.96	13.97	-1.47	1.27	1.000	1.14	2	2	2	1	0	0
1,2,3,6	C_1	1	567.06	-165.80	-36.55	29.04	617.37	14.27	-1.47	1.27	1.000	1.15	1	3	2	1	0	0
1,2,3,7	C_1	1	566.65	-166.03	-36.66	28.93	617.33	14.29	-1.47	1.27	1.000	1.14	1	3	2	1	0	0
1,2,3,8	C_1	1	561.96	-136.74	-5.97	59.62	617.23	14.36	-1.47	1.26	1.000	1.14	2	2	2	1	0	1
1,2,4,5	C_1	1	564.33	-118.27	11.80	77.38	616.97	14.32	-1.46	1.26	1.000	1.13	3	1	1	1	1	1
1,2,4,6	C_1	1	572.17	-142.72	-14.98	50.60	617.38	14.08	-1.47	1.26	1.000	1.13	2	2	1	1	1	0
1,2,4,7	C_1	1	572.17	-142.61	-14.88	50.70	617.33	14.10	-1.46	1.26	1.000	1.13	2	2	1	1	1	0
1,2,4,8	C_1	1	564.05	-118.85	11.30	76.88	616.94	14.35	-1.46	1.26	1.000	1.13	3	1	1	1	1	1
1,2,5,6	C_2	2	563.81	-139.00	-8.78	56.80	617.62	14.07	-1.47	1.26	1.000	1.13	2	2	2	0	0	0
1,2,5,7	C_1	1	582.41	-151.51	-26.84	38.74	617.61	13.99	-1.47	1.26	1.000	1.13	2	2	1	1	0	0
1,2,5,8	C_1	1	564.65	-121.28	8.69	74.27	616.67	14.46	-1.46	1.25	1.000	1.13	3	1	1	0	1	1
1,2,6,7	C_1	1	566.88	-168.81	-39.50	26.08	616.84	14.56	-1.47	1.26	1.000	1.15	1	3	2	0	0	0
1,2,6,8	C_1	1	563.75	-149.46	-19.22	46.36	617.02	14.29	-1.46	1.26	1.000	1.14	2	2	1	1	0	1
1,2,7,8	C_s	1	561.89	-137.65	-6.86	58.72	617.10	14.34	-1.47	1.26	1.000	1.14	2	2	2	0	0	1
1,3,5,7	C_2	2	569.82	-164.31	-35.88	29.70	617.24	14.12	-1.47	1.26	1.000	1.13	2	2	0	2	0	0
1,3,5,8	C_1	1	566.85	-132.45	-3.14	62.45	616.64	14.37	-1.46	1.25	1.000	1.13	3	1	0	1	1	1
1,3,6,7	C_1	1	566.30	-181.67	-52.19	13.39	616.35	14.76	-1.46	1.26	1.000	1.15	1	3	1	1	0	0
1,3,6,8	C_s	1	566.85	-161.16	-31.84	33.74	616.70	14.40	-1.46	1.26	1.000	1.14	2	2	0	2	0	1
1,4,5,8	C_{2v}	2	559.11	-107.25	24.38	89.96	616.07	14.66	-1.45	1.25	1.000	1.13	4	0	0	0	2	2
1,4,6,7	C_s	1	579.53	-148.09	-22.56	43.03	616.78	14.46	-1.46	1.26	1.000	1.14	2	2	1	0	1	0
2,3,6,7	D_{2h}	4	549.97	-199.93	-65.58	0.00	615.50	15.40	-1.47	1.26	1.000	1.16	0	4	2	0	0	0
penta-CAQN																		
1,2,3,4,5	C_1	1	591.92	-112.67	23.38	67.52	630.91	6.61	-1.44	1.22	1.000	1.08	3	2	3	2	1	1
1,2,3,4,6	C_1	1	597.28	-135.49	-1.04	43.10	631.45	6.26	-1.44	1.22	1.000	1.08	2	3	3	2	1	0
1,2,3,5,6	C_1	1	601.51	-149.35	-16.17	27.97	630.90	6.65	-1.44	1.22	1.000	1.09	2	3	3	1	0	0
1,2,3,5,7	C_1	1	608.75	-161.89	-30.86	13.28	630.72	6.68	-1.44	1.22	1.000	1.09	2	3	2	2	0	0
1,2,3,5,8	C_1	1	592.81	-131.48	4.30	48.44	629.79	7.13	-1.43	1.21	1.000	1.09	3	2	2	1	1	1
1,2,3,6,7	C_1	1	595.68	-179.06	-44.14	0.00	629.89	7.28	-1.44	1.22	1.000	1.10	1	4	3	1	0	0
1,2,3,6,8	C_1	1	592.97	-159.32	-23.58	20.56	629.93	7.11	-1.43	1.22	1.000	1.09	2	3	2	2	0	1
1,2,3,7,8	C_1	1	590.29	-147.84	-11.31	32.83	630.13	7.07	-1.44	1.22	1.000	1.10	2	3	3	1	0	1
1,2,4,5,6	C_1	1	593.86	-129.78	5.68	49.82	629.80	7.08	-1.43	1.21	1.000	1.09	3	2	2	1	1	1
1,2,4,5,7	C_1	1	595.15	-140.92	-5.84	38.31	629.56	7.14	-1.43	1.21	1.000	1.08	3	2	1	2	1	1
1,2,4,5,8	C_1	1	593.45	-116.78	18.81	62.95	629.01	7.40	-1.42	1.20	1.000	1.08	4	1	1	1	2	2
1,2,4,6,7	C_1	1	601.87	-155.72	-22.64	21.50	629.91	7.08	-1.43	1.21	1.000	1.09	2	3	2	1	1	0
1,2,4,6,8	C_1	1	594.94	-141.49	-6.35	37.79	629.62	7.09	-1.43	1.21	1.000	1.08	3	2	1			

Table 5. Continued

molecule	group	symmetry number	S^θ	$\Delta_f H^\theta$	$\Delta_f G^\theta$	$\Delta_f G^\theta_R$	$C_{p,m}/J \cdot \text{mol}^{-1} \cdot \text{K}^{-1}$											
			$\text{J} \cdot \text{mol}^{-1} \cdot \text{K}^{-1}$	$\text{kJ} \cdot \text{mol}^{-1}$	$\text{kJ} \cdot \text{mol}^{-1}$	$\text{kJ} \cdot \text{mol}^{-1}$	constant	$10^{-3} T$	$10^5 T^{-1}$	$10^7 T^{-2}$	R^2	SE	N_α	N_β	N_o	N_m	N_p	$N_{1,8}$
1,2,3,4,6,7	C_s	1	627.52	-148.21	-8.58	10.31	644.00	-0.74	-1.41	1.17	1.000	1.04	2	4	4	2	1	0
1,2,3,5,6,7	C_2	2	624.17	-159.52	-18.88	0.00	643.99	-0.65	-1.41	1.17	1.000	1.05	2	4	4	2	0	0
1,2,3,5,6,8	C_1	1	622.19	-139.63	1.60	20.48	642.86	-0.21	-1.40	1.16	1.000	1.04	3	3	3	2	1	1
1,2,3,5,7,8	C_1	1	622.22	-140.16	1.06	19.94	642.85	-0.19	-1.40	1.16	1.000	1.05	3	3	3	2	1	1
1,2,3,6,7,8	C_s	1	619.63	-157.45	-15.46	3.42	643.16	-0.20	-1.41	1.17	1.000	1.05	2	4	4	2	0	1
1,2,4,5,6,8	C_2	2	616.80	-125.79	17.04	35.92	641.99	0.13	-1.39	1.16	1.000	1.04	4	2	2	2	2	2
1,2,4,5,7,8	C_s	1	622.60	-125.79	15.32	34.20	641.89	0.20	-1.39	1.16	1.000	1.04	4	2	2	2	2	2
hepta-CAQN																		
1,2,3,4,5,6,7	C_1	1	649.21	-133.77	13.60	0.00	656.70	-7.84	-1.37	1.12	1.000	1.00	3	4	5	3	1	1
1,2,3,4,5,6,8	C_1	1	649.05	-120.75	26.67	13.07	655.71	-7.43	-1.37	1.11	1.000	1.00	4	3	4	3	2	2
OCAQN																		
1,2,3,4,5,6,7,8	C_{2v}	2	670.78	-115.44	39.71		669.53	-15.07	-1.34	1.07	1.000	0.95	4	4	6	4	2	2

^a V_m is the molecular volume. S^θ is standard entropy. $\Delta_f H^\theta$ is the standard enthalpy of formation of the compound. $\Delta_f G^\theta$ is the standard Gibbs energy of formation of the compound. $\Delta_f G^\theta_R$ is the relative magnitude of the standard Gibbs energy of formation. $C_{p,m}^\theta$ is the molar heat capacity at constant pressure. N is the number of Cl atom substitutions, and the subscript PCS indicates the positions.

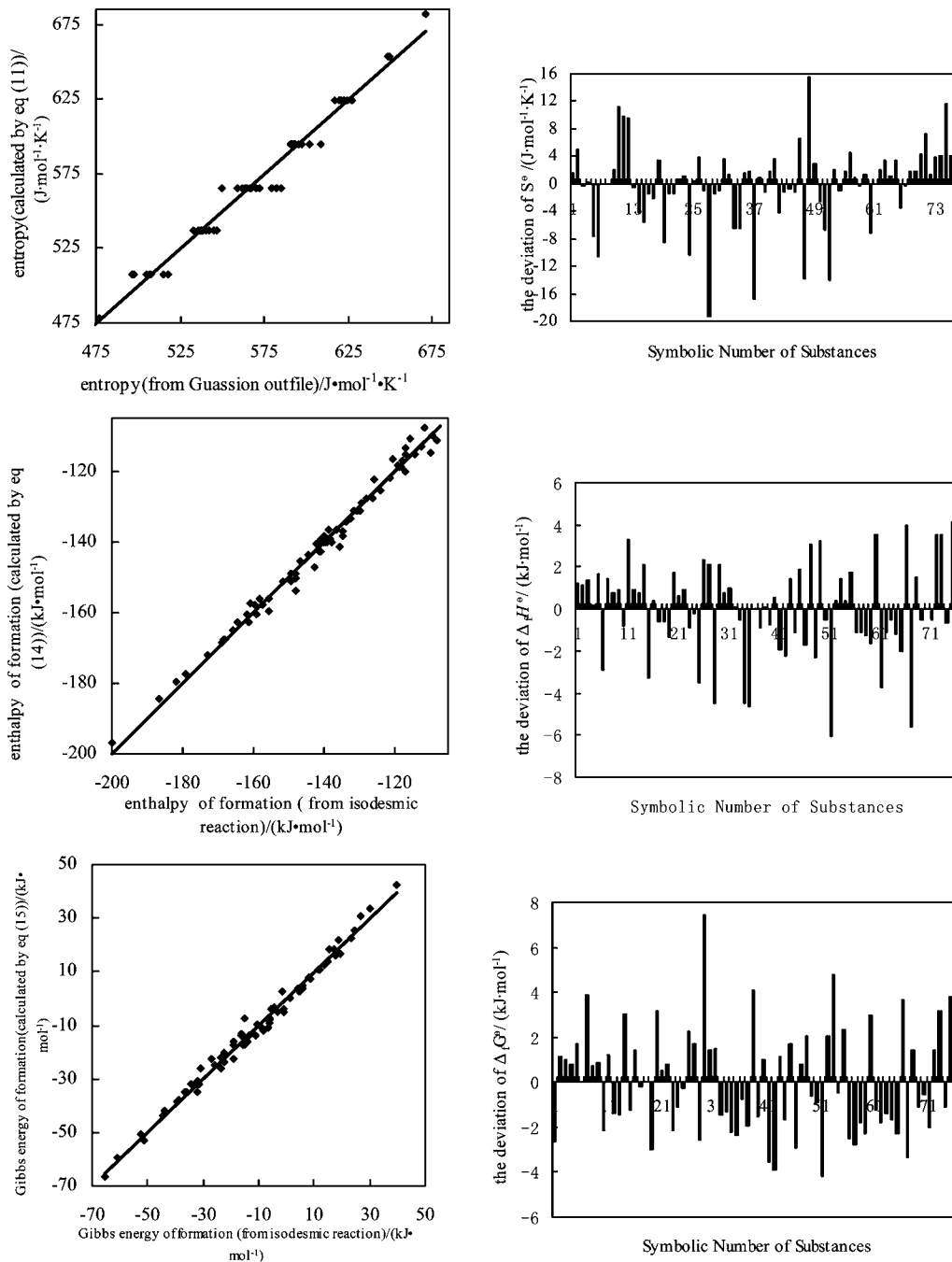


Figure 2. Plots of the values obtained from the correlations versus the corresponding DFT results.