# Comment on "Excess Enthalpies of Binary and Ternary Mixtures Containing Dibutyl Ether, Cyclohexane, and 1-Butanol at 298.15 K" 

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Trefer to the article "Excess Enthalpies of Binary and Ternary
—Mixtures Containing Dibutyl Ether, Cyclohexane, and 1Butanol at 298.15 K" by Aguilar et al. in J. Chem. Eng. Data, 2009, 54, 1672-1679 and Corrections in J. Chem. Eng. Data, 2009, 54, 2341-2342.

I used the Redlich-Kister eq 1 with constants given in Table 4 of the original paper, to calculate the excess enthalpy of a binary $\left(H^{\mathrm{E}}\right)$. I found that eq 1 is incorrect. It should read as below.

$$
\begin{equation*}
H^{\mathrm{E}}=x(1-x) \cdot\left[\sum_{i=0}^{n} A_{i}(2 x-1)^{i}\right] \tag{1}
\end{equation*}
$$

With the use of the above eq 1 and the constants given in Table 4, the values of $H_{12}^{\mathrm{E}}, H_{13}^{\mathrm{E}}$, and $H_{23}^{\mathrm{E}}$ match with values listed in Tables 3 (original paper) and 5 (correction).

For the calculation of $H_{123}^{\mathrm{E}}$, I used eqs 6 and 8 with constants, given in Table 6 (correction). Values so calculated did not match with values, given in Table 5 (correction). I therefore used the following equation for the calculation of $H_{123}^{\mathrm{E}}$.

$$
\begin{align*}
H_{123}^{\mathrm{E}}= & \left(x_{1}+x_{2}\right) H_{12}^{\mathrm{E}}+\left(x_{2}+x_{3}\right) H_{23}^{\mathrm{E}} \\
& +\left(x_{1}+x_{3}\right) H_{13}^{\mathrm{E}} \pm x_{1} x_{2} x_{3} \Delta H_{123}^{\mathrm{E}} \tag{6}
\end{align*}
$$

In the above eq 6, if the excess enthalpy of the ternary system $\left(H_{123}^{\mathrm{E}}\right)$ is endothermic, the last term will have negative sign and vice versa. I used the above eq 6 and calculated $H_{123}^{\mathrm{E}}$ for the following two compositions in Table 1.

Table 1

| $x_{1}$ | $x_{2}$ | $x_{3}$ | calculated $H^{\mathrm{E}}$ for binary |  |  | $\Delta H_{123}^{\mathrm{E}}$ | calculated | reported |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | $H_{12}^{\mathrm{E}}$ | $H_{23}^{\mathrm{E}}$ | $H_{13}^{\mathrm{E}}$ |  |  |  |
|  |  |  |  | $\mathrm{J} \cdot \mathrm{mol}^{-1}$ |  |  | $H_{123}^{\mathrm{E}}$ using above eq 6 | $H_{123}^{\mathrm{E}}$ in corrected Table 5 |
| 0.06 | 0.6996 | 0.24 | 111.2 | 602.2 | 444.1 | 9621.9 | 686.9 | 690.7 |
| 0.16 | 0.6002 | 0.2398 | 220.4 | 619.1 | 764.3 | 9473.1 | 774.7 | 803.8 |

I found that calculated values of $H_{123}^{\mathrm{E}}$ match well with those reported in the corrected Table 5.

I did not check the other four correlations for calculating excess enthalpies of binaries $\left(H^{\mathrm{E}}\right)$. Also I did not check the calculations of $\Delta H_{123}^{\mathrm{E}}$ with eq 7 .

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