

Experimental⁹

1-(2-Deoxy-5-O-trityl-β-D-lyxosyl)uracil (II).—To a cooled solution of 1.73 g. (3.68 mmoles) of 5'-O-trityl-2'-deoxyuridine¹⁰ (I) in 10 ml. of dry pyridine was added 0.57 ml. (7.36 mmoles) of methanesulfonyl chloride and the mixture held at 5° for 16 hr. Water (2.0 ml) was added and, after 0.5 hr. at room temperature, the reaction mixture was poured into 500 ml. of stirred ice-water. The off-white solid was collected and washed with generous quantities of water. The air-dried product was dissolved in 80 ml. of 50% ethanol that contained 14.5 ml. of *N* sodium hydroxide and the solution was refluxed for 4 hr. The volume was then reduced *in vacuo* to ca. 40 ml.; the reaction mixture was then chilled and carefully acidified (pH 2) with dilute hydrochloric acid. The gelatinous product was collected, washed with water, and sucked dry. The dry product was readily transformed to a crystalline solid on stirring with 50 ml. of ethanol at room temperature for 0.5 hr., wt. 1.64 g. (95% yield), m.p. 225–228°. Two recrystallizations from ethanol provided an analytical sample, m.p. 239–240°, $[\alpha]_D^{25} -14.9^\circ$ (*c* 0.93, DMF); λ_{max}^{EtOH} 262 m μ (ϵ 10,630), and λ_{min} 243 m μ (ϵ 6370).

Anal. Calcd. for C₂₃H₂₆N₂O₅: C, 71.47; H, 5.57; N, 5.96. Found: C, 71.30; H, 5.73; N, 5.69.

1-(2-Deoxy-β-D-lyxofuranosyl)uracil (III).—A solution of 1.45 g. (3.08 mmoles) of II in 10 ml. of 80% acetic acid was refluxed for 10 min. The reaction mixture was evaporated to dryness *in vacuo* and the residue was evaporated from three 10-ml. portions of ethanol. The product crystallized from methanol-ethyl acetate, 0.58 g. (two crops, 83% yield), m.p. 163–165°. A second recrystallization from methanol failed to alter the melting point; $[\alpha]_D^{25} +58.2^\circ$ (*c* 0.55, ethanol); in H₂O, λ_{max} 263 m μ (ϵ 10,620), and λ_{min} 232 m μ (ϵ 2760); 0.1 *N* HCl, λ_{max} 262 m μ (ϵ 9610), and λ_{min} 231 m μ (ϵ 2220); 0.1 *N* NaOH, λ_{max} 262 m μ (ϵ 7070), and λ_{min} 241 m μ (ϵ 4520).

Anal. Calcd. for C₉H₁₂N₂O₅: C, 47.36; H, 5.30; N, 12.28. Found: C, 47.22; H, 5.40; N, 12.09.

1-(2-Deoxy-β-D-lyxofuranosyl)-5-iodouracil (IV).¹¹—A mixture of 0.25 g. (1.1 mmoles) of III, 0.25 g. (1 mmole) of iodine, 2.5 ml. of *N* nitric acid, and 1.3 ml. of chloroform was refluxed for 2 hr. On cooling, a colorless crystalline solid was deposited. The product was collected, washed free of iodine with ether, and recrystallized from water, 0.23 g. (57% yield), m.p. 180–181° dec., $[\alpha]_D^{25} -4.9^\circ$ (*c* 0.81, ethanol); in H₂O, λ_{max} 289 m μ (ϵ 6530) and λ_{min} 247 m μ (ϵ 970); 0.1 *N* HCl, λ_{max} 288 (ϵ 4330), and 255 m μ (ϵ 2010); 0.1 *N* NaOH, λ_{max} 278 m μ (ϵ 5310) and λ_{min} 254 m μ (ϵ 2830).

Anal. Calcd. for C₉H₁₁IN₂O₅: C, 30.52; H, 3.13; I, 35.84; N, 7.91. Found: C, 30.28; H, 2.88; I, 35.70; N, 8.09.

(9) Melting points are corrected. Ultraviolet spectra were recorded by a Cary Model 11 spectrophotometer. Analyses were performed by Micro-Tech Laboratories, Skokie, Ill.

(10) J. Smrt and F. Sorm, *Collection Czech. Chem. Commun.*, **25**, 553 (1960); *Chem. Abstr.*, **54**, 12145 (1960).

(11) This procedure is identical with that described by W. H. Prusoff, *Biochim. Biophys. Acta*, **32**, 295 (1959), for the preparation of 5-iodo-2'-deoxyuridine.

5-Aryl-1,3-dihydro-2H-1,3,4-benzotriazepin-2-ones

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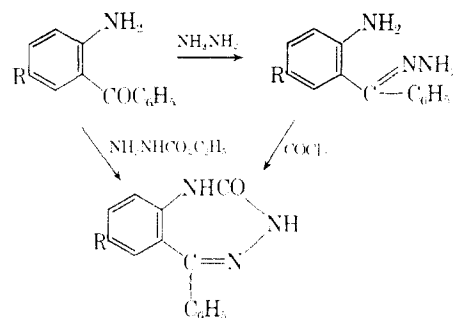
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Because of our interest in 1,4-benzodiazepin-2-ones¹ and 3,1,4-benzoxadiazepin-2(1H)ones,² it seemed desirable to prepare some aza analogs of these ring systems. Accordingly, several 5-aryl-1,3-dihydro-2H-1,3,4-benzotriazepin-2-ones have been prepared by two related methods, as shown in the reaction scheme.

(1) S. C. Bell, T. S. Sulkowski, C. Gochman, and S. J. Childress, *J. Org. Chem.*, **27**, 562 (1962).

(2) T. S. Sulkowski and S. J. Childress, *ibid.*, **27**, 4424 (1962).



Experimental

5-Phenyl-1,3-dihydro-2H-1,3,4-benzotriazepin-2-one.—A solution of 12 g. of 12.5% phosgene in benzene was added dropwise to a cooled solution of 3.2 g. of 2-aminobenzophenone hydrazone and 5 ml. of triethylamine in 50 ml. of benzene. After addition was completed, the mixture was stirred at room temperature for 1 hr. After separating the triethylamine hydrochloride by filtration, the solution was evaporated to dryness *in vacuo*. Recrystallization of the residue from ethanol afforded 1.4 g. of product, m.p. 238°.

Anal. Calcd. for C₁₄H₁₁N₃O: C, 70.87; H, 4.68; N, 17.70. Found: C, 70.69; H, 4.53; N, 18.04.

7-Methyl-5-phenyl-1,3-dihydro-2H-1,3,4-benzotriazepin-2-one, m.p. 253–255°, was prepared similarly from 2-amino-5-methylbenzophenone hydrazone in a yield of 50%.

Anal. Calcd. for C₁₅H₁₃N₃O: C, 71.70; H, 5.21; N, 16.72. Found: C, 71.61; H, 5.43; N, 16.64.

7-Chloro-5-phenyl-1,3-dihydro-2H-1,3,4-benzotriazepin-2-one.—A mixture of 5 g. of 2-amino-5-chlorobenzophenone and 5 ml. of ethyl hydrazinecarboxylate was heated at 190° for 1 hr. The mixture was cooled and dissolved in 75 ml. of ethanol. On standing there was obtained 1.6 g. of product, m.p. 246–248°.

Anal. Calcd. for C₁₄H₁₀ClN₃O: C, 61.89; H, 3.72; Cl, 13.05; N, 15.47. Found: C, 61.65; H, 3.72; Cl, 13.27; N, 15.18.

Concentration of the mother liquor afforded 0.5 g. of 2-amino-5-chlorobenzophenone hydrazone ethyl carboxylate, m.p. 209°.

Anal. Calcd. for C₁₆H₁₆ClN₃O₂: C, 60.47; H, 5.07; Cl, 11.16; N, 13.23. Found: C, 60.16; H, 5.07; Cl, 11.25; N, 13.04.

Quinazolines and 1,4-Benzodiazepines. XIX.¹ N-Alkyl Derivatives of Substituted 1,3,4,5-Tetrahydro-5-phenyl-2H-1,4-benzodiazepin-2-ones

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As a continuation of our investigation on psychotherapeutic agents of the 1,4-benzodiazepine class of compounds, we have prepared a series of 1,3,4,5-tetrahydro-5-phenyl-2H-1,4-benzodiazepin-2-ones and from these compounds a number of N-alkyl derivatives. For the sake of simplicity, the Experimental section of this paper will concern itself with the chemistry of only one of these compounds, namely, 7-chloro-5-(2-fluorophenyl)-1,3,4,5-tetrahydro-2H-1,4-benzodiazepin-2-one and its N-methyl derivatives. As shown in the Experimental, because of the difference in basicity between the two nitrogen atoms, we found it possible to alkylate the 1-nitrogen independently of the 4-nitrogen and *vice versa*. All other compounds and derivatives prepared by the same procedures will be found in Table III.

(1) Paper XVIII: I. H. Sternbach, E. Reeder, A. Stempel, and A. I. Rachlin, *J. Org. Chem.*, **29**, 332 (1964).