

Experimental Section⁸**Transformation of 2 α -Methyl-19-nortestosterone by *A. tamarit*.**

—To each of 14 erlenmeyer flasks, each of which contained 100 ml of a 3% Difco malt extract solution and a 48-hr growth of *A. tamarit*, was added 75.0 mg of 2 α -methyl-19-nortestosterone⁹ in 0.4 ml of DMF. After an additional 72 hr of incubation on a rotary shaker at 28°, each reaction mixture was extracted with CH₂Cl₂. The organic layers were combined, dried (MgSO₄), and evaporated to a dry residue (1034 mg). A portion of the latter (1005 mg) was chromatographed on a 90-g column of silica gel H with EtOAc as the eluent. Tlc of the eluent fractions indicated that three major fractions were obtained. Fraction 1 (150 mg) was starting material, 2 α -methyl-19-nortestosterone. Fraction 2, after recrystallization from Me₂CO–hexane, produced 675 mg (68% yield) of 2 α -methyl-19-nortestolactone: mp 191–192.5°; $\nu_{\text{max}}^{\text{KBr}}$ 1725, 1670, and 1620 cm⁻¹; nmr (CDCl₃), δ 1.10 (3 H), 1.38 (3 H), and 5.83 (1 H); $[\alpha]_{\text{D}}^{25}$ +15° in CHCl₃. Anal. Calcd for C₁₉H₂₆O₃: C, 75.46; H, 8.67. Found: C, 75.45; H, 8.64.

Fraction 3 (180 mg) consisted of mixtures of the above steroids and trace amounts of other compounds which are probably additional oxidative metabolites of 2 α -methyl-19-nortestosterone, contaminated with CH₂Cl₂-soluble cellular material.

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(8) All melting points were determined by a Kofler apparatus and are corrected.

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Application of 1,3-Di(4-piperidyl)propane in the Mannich Reaction. Synthesis of β -Amino Ketones

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β -Amino ketones (Mannich bases) have been reported to possess antispasmodic,¹ analgetic,² local anesthetic,^{3–6} and antibacterial^{7–10} activity. In a recent communication¹¹ from this laboratory we described the synthesis of a series of β -amino ketones derived from 1-(N- β -hydroxyethyl-4-piperidyl)-3-(4-piperidyl)propane. Several of these compounds have exhibited antibacterial and antiviral activity.¹² Ready availability of 1,3-di(4-piperidyl)propane (4-DI-PIP) prompted us to prepare β -amino ketones of this novel secondary amine for biological screening.

Screening Results.—The compounds of Table I were screened *in vitro* against four organisms: *Pseudomonas aeruginosa*, *Staphylococcus aureus*, *Mycobacterium smegmatis*, and *Klebsiella pneumoniae*. Filter paper disks (6.35-mm diameter) saturated with the solution (20 mg/ml) of the test compound were placed on the agar. After 72 hr of incubation the zones of inhibition around the disks were measured. The results are reported in Table II.

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(12) R. S. Varma and W. L. Nobles, unpublished work.

TABLE I

No.	R	Mp, °C ^a	Yield, % ^b	Formula ^g
1	2-Thenyl	212–215	55	C ₂₇ H ₄₀ Cl ₂ N ₂ O ₂ S ^c
2	4-Ethoxyphenyl	184–185	44	C ₂₈ H ₃₂ Cl ₂ N ₂ O ₄ ^{c,f}
3	4-Hydroxyphenyl	245	43	C ₂₁ H ₂₄ Cl ₂ N ₂ O ^d
4	4-Nitrophenyl	200–203	35	C ₂₁ H ₁₇ Cl ₂ N ₂ O ^e
5	4-Chlorophenyl	209–212	81	C ₂₁ H ₁₇ Cl ₃ N ₂ O ^e
6	4-Bromophenyl	212–215	69	C ₂₁ H ₁₇ Br ₂ Cl ₂ N ₂ O ^e
7	4-Fluorophenyl	190–194	42	C ₂₁ H ₁₇ Cl ₂ F ₂ N ₂ O ₂ · 1.5H ₂ O ^e
8	4-Methylphenyl	195–197	65	C ₂₅ H ₂₈ Cl ₂ N ₂ O ₂ · H ₂ O ^e
9	3-Nitrophenyl	180–183	49	C ₂₁ H ₁₇ Cl ₂ N ₂ O ₆ · H ₂ O ^e
10	2-Hydroxyphenyl	211–212	28	C ₂₁ H ₁₉ Cl ₂ N ₂ O ₄ · H ₂ O ^e
11	Phenyl	210–212	48	C ₂₁ H ₁₉ Cl ₂ N ₂ O ₂ · 0.5H ₂ O ^e

^a All compounds melt with decomposition. ^b Yields are of the product obtained after the first crystallization. ^c Prepared by method C. ^d Prepared by method A. ^e Prepared by method B. ^f Recrystallized from EtOH; the other compounds were recrystallized from EtOH–Me₂CO–H₂O. ^g All compounds were analyzed for C, H, N. Infrared absorption bands for NH⁺ and C=O were as expected.

TABLE II

No.	<i>In Vitro</i> ANTIBACTERIAL ACTIVITY OF β -AMINO KETONES			
	Microbial spectrum ^a			
	<i>S. aureus</i> K257	<i>P. aeruginosa</i>	<i>K. pneumoniae</i> ATCC 8052	<i>M. smegmatis</i>
1	+	+	–	+
2	+	+	–	+
3	–	–	–	–
4	+	+	+	–
5	–	+	+	+
6	+	–	+	+
7	–	–	–	+
8	+	+	+	+
9	–	+	+	–
10	+	+	+	+
11	–	–	–	–

^a A negative sign indicates no observable activity.

Experimental Section¹³

1,3-Di(4-piperidyl)propane Dihydrochloride.—4-DI-PIP (42 g) was suspended in 100 ml of EtOH. Concentrated HCl (40 ml) was added dropwise with cooling and stirring. After the additions were completed, Me₂CO (200 ml) was introduced into the reaction vessel. The reaction mixture on refrigeration overnight furnished the desired salt in nearly quantitative yield. The salt was recrystallized (EtOH–Me₂CO); mp 262–264° dec. Anal. (C₁₃H₂₃Cl₂N₂) C, H, N.

β -Amino Ketone Dihydrochlorides. Method A.—A mixture of 0.04 mole of the appropriate ketone, 0.02 mole of 4-DI-PIP dihydrochloride, 1.8 g of paraformaldehyde, and 50 ml of EtOH containing 2 drops of concentrated HCl was refluxed for 5 hr. The warm solution was poured into Me₂CO (100 ml). Overnight refrigeration of the contents yielded the desired product.

Method B.—Concentrated HCl (4 ml) was added dropwise to a cooled suspension of 4-DI-PIP (4.2 g, 0.02 mole) in 10 ml of EtOH with shaking. Aqueous formaldehyde (37% 6 ml) was then introduced into the reaction vessel followed by the appropriate ketone (0.04 mole). The resulting reaction mixture was heated at 90–100° for 6 hr. During this time the entire mixture went into the solution. In a few cases an amorphous solid product separated at the end of this period. The contents were then diluted with Me₂CO (100 ml) and refrigerated overnight or until a solid product separated.

Method C was similar to that of B except that paraformaldehyde was used in place of formalin.

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(13) All melting points were taken on a Thomas-Hoover capillary melting point apparatus and are uncorrected. Microanalyses were performed by Galbraith Laboratories, Inc., Knoxville, Tenn. Infrared spectra were recorded in Nujol mull on a Perkin-Elmer Model 137 Infracord spectrophotometer and were as expected. Where analyses are indicated only by symbols of the elements analytical results obtained for these elements were within $\pm 0.4\%$ of the theoretical values.