

**$\alpha$ -Aminomethylpyrazine (IV).**—A mixture of 15 g (0.06 mole) of V and 600 ml of 5 N NaOH was refluxed 1 hr. The solution was cooled to 10° and extracted with fifteen 50-ml portions of CHCl<sub>3</sub>. The CHCl<sub>3</sub> extracts were combined and dried (Na<sub>2</sub>SO<sub>4</sub>) and the CHCl<sub>3</sub> was removed under vacuum. The residual oil was distilled at 87–88° (3 mm) to yield 4.43 g (64.7%) of a colorless liquid, which rapidly turned yellow on standing in the air. *Anal.* (C<sub>5</sub>H<sub>7</sub>N<sub>3</sub>) C, H, N.

**Reaction of  $\alpha$ -Aminomethylpyrazine (IV) with Ethylene Oxide.**—To 30.01 g (0.028 mole) of IV cooled to 0° was added 2.73 g (0.062 mole) of liquid ethylene oxide. The reaction flask was sealed and the contents were allowed to stand at room temperature for 24 hr. The brown viscous oil was fractionally distilled to yield four fractions: fraction 1, bp 80–81° (3 mm), was 1.09 g of IV; fraction 2, bp 102–104° (0.1 mm), afforded 1.14 g (42.3%, calculated on the basis of reclaimed IV) of II; fraction 3, a crude intermediate fraction (0.22 g), bp 104–140° (0.1 mm); fraction 4, bp 140–141° (0.1 mm), yielded 0.93 g (26.8%, calculated on the basis of reclaimed IV) of III.

**2-(2-Chloroethyl)aminomethylpyrazine Dihydrochloride (VI).**—To 0.86 g (0.0037 mole) of II·2HCl was added 5 ml of SOCl<sub>2</sub>. After standing at 40° for 15 hr, the reaction mixture was allowed to cool, treated with 50 ml of Et<sub>2</sub>O, and filtered. The solid was dissolved in MeOH and treated with decolorizing carbon, and the hydrochloride precipitated with Me<sub>2</sub>CO; it consisted of 0.74 g (79.5%) of a light green solid which did not melt below 340°. White flakes, melting above 340° dec, were obtained by recrystallization from MeOH–Me<sub>2</sub>CO. *Anal.* (C<sub>7</sub>H<sub>12</sub>Cl<sub>2</sub>N<sub>4</sub>) C, H, Cl, N.

The compound proved to be inactive<sup>1</sup> (T/C = 91% at 12 mg/kg) against the 5WA Walker 256 animal tumor screen.

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## 9-(2-Deoxycellobiosyl)adenine<sup>1</sup>

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A number of nucleosides have been reported which are derived from disaccharide sugars. The synthesis of these nucleosides was achieved by coupling of the acetylated bromides or chlorides of lactose,<sup>2,3</sup> cellobiose,<sup>3,4</sup> and maltose<sup>4</sup> with the heavy metal salts of purines or pyrimidines. In one case, that of melibiose,<sup>5</sup> it was found advantageous to use benzoyl blocking groups instead of acetyl groups in order to protect the 1→6 bond of this disaccharide from cleavage during bromination. Especially exciting from a medicinal viewpoint has been the discovery that the antibiotic, ampicillin, is a nucleosidic substance containing a disaccharide moiety.<sup>6</sup> The present report describes the first synthesis of a 2-deoxy disaccharide nucleoside, 9-(2-deoxycellobiosyl)adenine [9-(4-*O*- $\beta$ -D-glucopyranosyl-2-deoxy-D-arabino-hexopyranosyl)adenine]. The synthetic route used was based on the one reported by Davoll and Lythgoe<sup>7</sup> for the preparation of 7-(2-deoxy-D-ribosepyranosyl)theophylline from diacetyl-D-arabinal.

### Experimental Section

Hexa-*O*-acetylcellobial<sup>8</sup> [3.0 g, 5.35 mmoles, mp 132°, [ $\alpha$ ]<sub>D</sub><sup>20</sup> –21° (c 1.4, CHCl<sub>3</sub>)] was dissolved in 30 ml of Na-dried C<sub>6</sub>H<sub>6</sub>

(1) Supported by Grant No. T-442 from the American Cancer Society.  
(2) M. L. Wolfrom, P. McWain, F. Shafizadeh, and A. Thompson, *J. Am. Chem. Soc.*, **81**, 6080 (1959).

(3) C. Stevens and P. Blumbergs, *J. Org. Chem.*, **30**, 2723 (1965).

(4) M. L. Wolfrom, P. McWain, and A. Thompson, *J. Am. Chem. Soc.*, **82**, 4353 (1960).

(5) L. M. Lerner, *J. Org. Chem.*, **32**, 3063 (1967).

(6) C. Stevens, K. Nagarajan, and T. H. Haskell, *ibid.*, **27**, 2991 (1962).

(7) J. Davoll and B. Lythgoe, *J. Chem. Soc.*, 2526 (1949).

(8) Elementary analyses were performed by the Spang Microanalytical Laboratory. Melting points are corrected.

(9) W. N. Haworth, E. L. Hirst, L. Streight, H. H. Thomas, and J. Webb, *J. Chem. Soc.*, 636 (1930).

and the solution was chilled in an ice bath. Dry HCl gas was passed into the solution for 0.5 hr. C<sub>6</sub>H<sub>6</sub> was evaporated at a bath temperature of 30° and fresh, dry C<sub>6</sub>H<sub>6</sub> was added and evaporated several times in order to remove traces of HCl. The residual syrup was dissolved in 7.5 ml of dry xylene and added to an azeotropically dried refluxing mixture of 6-benzamidochloromereuripurine<sup>10</sup> (2.53 g, 5.35 mmoles), 2.5 g of Celite-545, 5 g of Molecular Sieve 4A, and 27.5 ml of xylene. The mixture was refluxed for 1 hr, the solids were removed by filtration, and the filter cake was washed with 100 ml of warm CHCl<sub>3</sub>. The solvents were removed by evaporation, the residue was dissolved in 125 ml of CHCl<sub>3</sub>, and the CHCl<sub>3</sub> solution was washed twice with 100-ml portions of 30% aqueous KI and once with 200 ml of H<sub>2</sub>O. The solution was dried (MgSO<sub>4</sub>) and after evaporation of the CHCl<sub>3</sub> a dark foam was obtained which weighed 4.8 g.

The foam was dissolved in CHCl<sub>3</sub> and applied to the top of a column containing 50 g of silicic acid (Mallinckrodt, 100 mesh, activated at 100° for 24 hr). CHCl<sub>3</sub> (375 ml) was passed through the column and discarded. Elution with 300 ml of CHCl<sub>3</sub>–MeOH (99:1 v/v) followed by 300 ml of a 97:3 v/v mixture of the same solvents yielded 3.66 g of a clear, slightly yellow syrup which was not homogeneous when chromatographed on the plates.<sup>11</sup> The blocking groups were removed by refluxing for 1 hr in 90 ml of 0.1 N methanolic NaOCH<sub>3</sub> solution. The solution was neutralized (AcOH) and evaporated to dryness. The gummy residue was dissolved in hot MeOH with the aid of a few drops of H<sub>2</sub>O. Acetone was added to incipient turbidity, heat from a steam bath was applied to just clarify it, and the flask was placed in a refrigerator for several days. A tan material weighing 650 mg was obtained, mp 165–170°, [ $\alpha$ ]<sub>D</sub><sup>20</sup> +53° (c 0.76, H<sub>2</sub>O). Recrystallization from the same solvent mixture with a prior charcoal (Dareco G-60) treatment gave a white solid. One more recrystallization, this time from *n*-BuOH–H<sub>2</sub>O, for 3 days in the refrigerator yielded the analytical sample as clear, colorless crystals, mp 175–179° (to an extremely viscous liquid): [ $\alpha$ ]<sub>D</sub><sup>20</sup> +41° (c 0.61, H<sub>2</sub>O); uv spectrum,  $\lambda_{max}^{25}$  257 m $\mu$  ( $\epsilon$  13,040),  $\lambda_{max}^{30}$  259 m $\mu$  ( $\epsilon$  13,250),  $\lambda_{max}^{35}$  259 m $\mu$  ( $\epsilon$  13,650). This material migrated as homogeneous spots on the plates,<sup>11</sup> *R<sub>f</sub>* 1.26 in 5% aqueous Na<sub>2</sub>HPO<sub>4</sub> and 0.23 in *n*-BuOH–H<sub>2</sub>O (86:14 v/v).

*Anal.* Calcd for C<sub>17</sub>H<sub>28</sub>N<sub>6</sub>O<sub>9</sub>: C, 46.05; H, 5.68; N, 15.65. Found: C, 46.34; H, 5.79; N, 15.65.

An attempt was made to elucidate the configuration of this nucleoside by nmr spectroscopy,<sup>12</sup> but the results were not conclusive and the configuration remains undesignated.

(10) J. Davoll and B. A. Lowy, *J. Am. Chem. Soc.*, **73**, 1650 (1951).

(11) The plates were prepared from silica gel HF (E. Merck, AG, Darmstadt) as 0.25 mm thick layers. Spots were visualized with an ultraviolet lamp and the homogeneity of the material was checked by the chromic acid charring method. *R<sub>f</sub>* 1.00 (of adenine).

(12) Obtained by Dr. Harry Agabizian of the Baron Consulting Co.

## Quinazolines and 1,4-Benzodiazepines. XXXIX.<sup>1</sup>

### The Synthesis of Dihydroimidazo- and Tetrahydropyrimido[1,2-*a*][1,4]benzodiazepines

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Pursuant to our interest in the pharmacological activity of new 1,4-benzodiazepines<sup>1</sup> and specifically of aminoalkyl-substituted benzodiazepines,<sup>2</sup> we have prepared some tetrahydropyrimido-[1,2-*a*][1,4]benzodiazepines<sup>3,4</sup> (3) (Table I) and 8-chloro-6-(2-fluorophenyl)-1,2-dihydro-4H-imidazo[1,2-*a*][1,4]benzodiazepine (4).

(1) Paper XXXVIII: M. E. Derieg, R. I. Fryer, and L. H. Sternbach, *J. Chem. Soc.*, in press.

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(3) R. I. Fryer, B. Brust, J. V. Earley, and L. H. Sternbach, *ibid.*, **7**, 386 (1964).

(4) See for example, G. I. Glover, R. B. Smith, and H. Rapoport, *J. Am. Chem. Soc.*, **87**, 2003 (1965).