

## 4-Oxo-1,2,3,4-tetrahydroquinazolines. II.<sup>1</sup>

### Synthesis of 1-Alkyl- and 1-[2-(Disubstituted amino)ethyl]-2-methyl-3-aryl-4-oxo-1,2,3,4-tetrahydroquinazolines

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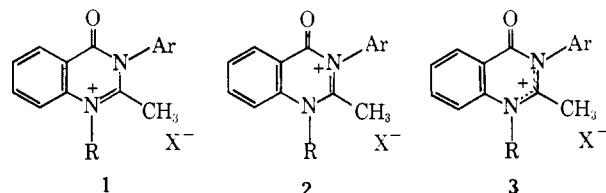
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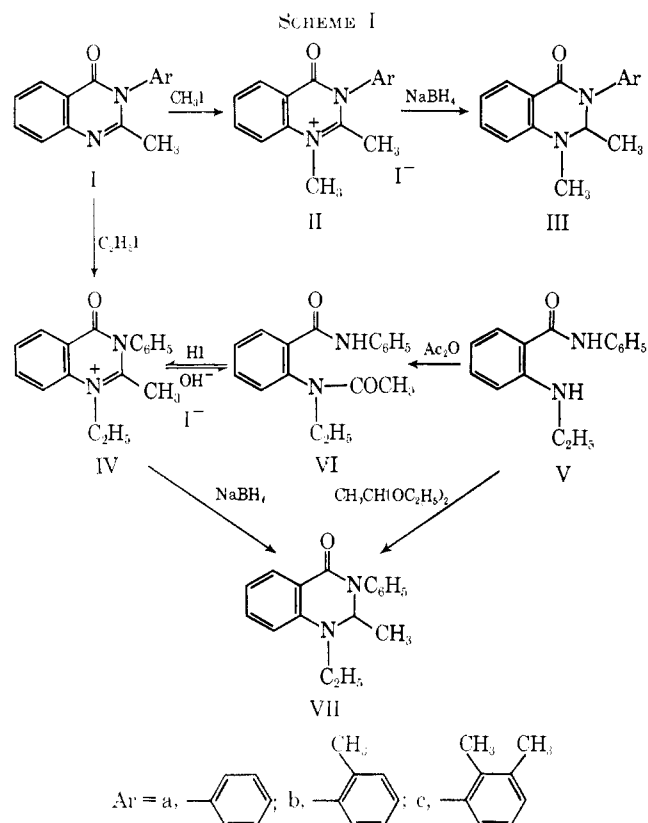
Reaction of 2-(N-ethylacetamido)benzanilide with III in EtOH afforded 1-ethyl-2-methyl-3-phenyl-4-oxo-dihydroquinazolium iodide in quantitative yield. This conversion was reversible under mild alkaline conditions. 1-Ethyl-2-methyl-3-phenyl-4-oxo-1,2,3,4-tetrahydroquinazolinone was obtained by NaBH<sub>4</sub> reduction of the salt in good yield. Nine derivatives of 1-[2-(disubstituted amino)ethyl]-2-methyl-3-aryl-4-oxo-1,2,3,4-tetrahydroquinazolines were synthesized from N-substituted anthranilic acid derivatives *via* 1-[2-(disubstituted amino)ethyl]-2-methyl-3-aryl-4-oxodihydroquinazolium compounds.

According to Hauptmann,<sup>2</sup> 1-dialkylaminoalkyl-2-aryl-4-oxo-1,2,3,4-tetrahydroquinazolines were synthesized, and the pharmacological studies of these compounds were reported by Mutačević and co-workers.<sup>3</sup> In a previous paper,<sup>1</sup> we had found that the reduction of 2-methyl-3-aryl-4(3H)-quinazolinone hydrochloride with NaBH<sub>4</sub> afforded the corresponding 4-oxo-1,2,3,4-tetrahydroquinazolines in good yield, and also reported the synthesis of the 1-acyl derivatives. This paper describes the synthesis and some pharmacological properties of 1-alkyl- and 1-[2-(disubstituted amino)ethyl]-2-methyl-3-aryl-4-oxo-1,2,3,4-tetrahydroquinazolines (XIV).

Methylation of 2-methyl-3-aryl-4(3H)-quinazolinone [I, Ar = C<sub>6</sub>H<sub>5</sub>, 2-CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>, and 2,3-(CH<sub>3</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>] with methyl iodide was carried out by a modified Bogert<sup>4</sup> procedure to give 1,2-dimethyl-3-aryl-4-oxodihydroquinazolium iodide (II),<sup>5</sup> which was converted



to 1,2-dimethyl-3-aryl-4-oxo-1,2,3,4-tetrahydroquinazolines (III) by NaBH<sub>4</sub> reduction in good yield. However reaction of I (Ar = C<sub>6</sub>H<sub>5</sub>) with EtI in a sealed tube to 1-ethyl-2-methyl-3-phenyl-4-oxo-3,4-dihydroquinazolium iodide (IV) did not go too well. A more possible approach to IV appeared to be the sequence V → VI → IV as shown in Scheme I. Acetylation of 2-ethylaminobenzanilide (V) gave the N-acetyl derivatives (VI), and reaction of VI with HI in ethanol gave IV in quantitative yield. This new type of cyclization is a useful method of synthesis of 1-substituted 2-methyl-3-aryl-4-oxodihydroquinazolium compounds. IV readily cleaved to VI under mild alkaline condition. This conversion was found to be reversible. The 4-oxo-



dihydroquinazolium iodides (II and IV) showed typical absorption bands at 1725, 1620, 1560 cm<sup>-1</sup>. Reduction of IV with NaBH<sub>4</sub> gave 1-ethyl-2-methyl-3-phenyl-4-oxo-1,2,3,4-tetrahydroquinazolinone (VII) in excellent yield.

The structure of VII was confirmed by an independent synthesis. Reaction of V with acetaldehyde diethyl acetal in the presence of sulfuric acid gave an oily mixture, from which VII was isolated in about 33% yield from chromatography over alumina.

The new cyclization of 2-acetamidobenzanilide derivatives was applied to a synthesis of 1-[2-(disubstituted amino)ethyl]-2-methyl-3-phenyl-4-oxo-1,2,3,4-tetrahydroquinazolines (XIV, N(R)<sub>2</sub> = N(CH<sub>3</sub>)<sub>2</sub>, N(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>, -N<img alt="cyclohexane ring" style="vertical-align: middle;"/>). N-Substituted anthranilic acid derivatives (IX) were prepared from 2-chlorobenzoic acid (VIII) by a procedure modified from that of Hauptmann.<sup>2</sup> Acetylation of IX gave the N-acetyl

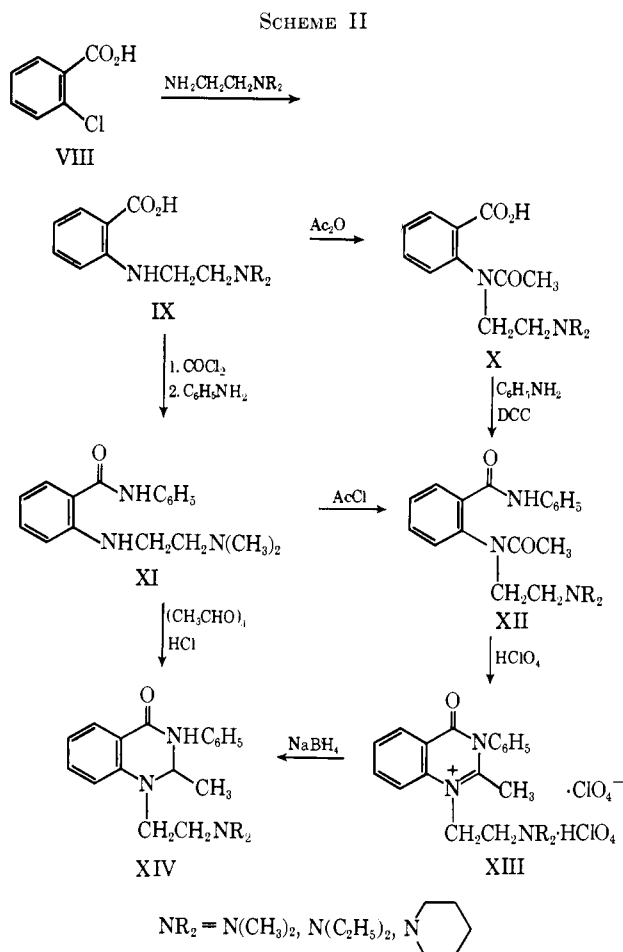
(1) Part I: K. Okumura, T. Oine, Y. Yamada, G. Hayashi, and M. Nakama, *J. Med. Chem.*, **11**, 348 (1968).

(2) K. H. Hauptmann, *Arzneimittel-Forsch.*, **15**, 610 (1965).

(3) G. Mutačević, H. Sčitzer, and H. Wick, *ibid.*, **15**, 613 (1965).

(4) M. T. Bogert and C. A. Geiger, *J. Am. Chem. Soc.*, **34**, 683 (1912).

(5) It is not clear whether 4-oxodihydroquinazolium compounds possess the 3,4-dihydro structure (1), the 1,4-dihydro structure (2), or the resonance structure (3). However in this paper, 1 is expediently used for the 4-oxodihydroquinazolium compounds.



derivatives X. Condensation of X and aniline by the use of *N,N'*-dicyclohexylcarbodiimide afforded the oily anilides XII, which were converted to 4-oxodihydroquinazolinium compounds (XIII) by the addition of excess 66% aqueous perchloric acid (Scheme II). The ir spectra of XIII agreed with that of II and IV at 1500–1800  $\text{cm}^{-1}$ . Reduction of XIII with  $\text{NaBH}_4$  gave expected XIV in good yield. The other derivatives shown in Tables I and II were prepared similarly. XIV [ $\text{R}_2 = (\text{CH}_3)_2$ ] was synthesized by the following method. The anilide derivative XI was obtained from IX [ $\text{R}_2 = (\text{CH}_3)_2$ ] in an unsatisfactory yield through the isatoic anhydride derivative. Reaction of XI and paraldehyde in the presence of dry HCl in EtOH afforded an oily mixture, from which XIV [ $\text{R}_2 = (\text{CH}_3)_2$ ] was isolated by distillation after chromatography over alumina (yield 32%). The ir spectra of this product and a sample obtained by  $\text{NaBH}_4$  reduction of XIII [ $\text{R}_2 = (\text{CH}_3)_2$ ] were superimposable.

## Results

Results of pharmacological observations are summarized in Table III. It was previously mentioned<sup>1</sup> that the analgetic activities of these derivatives were affected by the relation between substituents at positions 1 and 3 of the quinazolinone. The analgetic activities of **17** and **18** which have a diethylaminoethyl substituent at position 1 were almost as potent as that of aminopyrine intraperitoneally and twice as active as that of aminopyrine orally. However, by intraperitoneal injection, the activities of **17** and **18** were not resolvable, since the  $\text{ED}_{50}$  values were about  $1/2.2$  of the  $\text{LD}_{50}$  values.

The other compounds which have methyl, dimethylaminoethyl, and piperidinoethyl substituents at position 1 showed rather weak analgetic activity. Prolongation of thiopental sleeping time was found in all test compounds. In **21** the highest potency was observed in a carrageenin-induced edema test; **12**, **14**, and **18** showed higher antiinflammatory activity. In particular, the activities in **14** and **21** were higher than that of phenylbutazone. Antihistaminic effects in these compounds were very weak. At the maximum tolerated dose, many compounds produced a slight increase of spontaneous activity.

## Experimental Section<sup>6</sup>

**Pharmacological Methods.** (1) **Analgetic Activity.**—Analgetic activities were estimated by the tail pinch method in mice.<sup>7</sup>  $\text{ED}_{50}$  values were determined by the number of mice that responded analgetically in each dose.

(2) **Potentiation of the Anesthetic Effect of Barbiturates.**—The test compounds were injected intraperitoneally at 100 mg/kg. Thirty minutes later, thiopental sodium was injected intravenously in a dose of 25 mg/kg. Prolongation of the sleeping time was compared with that of control animals.

(3) **Antiinflammatory Effect.**—Male Wistar rats (about 150 g) were used. Antiinflammatory effects were determined by the inhibition of the test compounds on the carrageenin-induced edema.<sup>9</sup>

(4) **Antihistaminic Activity.**—Guinea pig intestine was isolated. Antihistaminic activities were assayed according to Magnus. The test compounds were added to the Magnus bath at  $10^{-6}$  g/ml 3 min prior to addition of histamine ( $5 \times 10^{-8}$  g/ml).

(5) **Acute Toxicity and Behavioral Observation.**—Adult male dd strain mice weighing  $20 \pm 1$  g were used. The test compounds were dissolved in saline or suspended in 0.5% CMC solution and administered intraperitoneally or orally. The behavioral changes of the animals were observed and  $\text{LD}_{50}$  values were calculated by the Weil method<sup>8</sup> according to the number of dead animals on day 4.

**Chemical Methods.** **1,2-Dimethyl-3-aryl-4-oxodihydroquinazolinium iodides (II)** were prepared by a modification of the procedure of Bogert.<sup>3</sup>

**a, Ar = 2- $\text{CH}_3\text{C}_6\text{H}_4$ .**—A mixture of 2-methyl-3-(2-tolyl)-4(3H)-quinazolinone (I, Ar = 2- $\text{CH}_3\text{C}_6\text{H}_4$ ) (1.25 g), MeI (2.1 g), and dry  $\text{C}_6\text{H}_6$  (2 ml) was heated in a sealed tube at 110–120° for 7 hr and cooled. The separated crystalline product was collected. Recrystallization from MeOH gave pale yellow needles, mp 228–229° dec, yield 0.9 g (45%). *Anal.* ( $\text{C}_{17}\text{H}_{17}\text{IN}_2\text{O}$ ) C, H, N.

**b, Ar = 2,3-( $\text{CH}_3$ ) $_2\text{C}_6\text{H}_3$ .**—Methylation of I [Ar = 2,3-( $\text{CH}_3$ ) $_2\text{C}_6\text{H}_3$ ] (5.3 g) was carried out as above, giving colorless needles from MeOH, mp 240–241° dec, yield 4.8 g (59%). *Anal.* ( $\text{C}_{18}\text{H}_{19}\text{IN}_2\text{O}$ ) C, H, N.

**1,2-Dimethyl-3-aryl-4-oxo-1,2,3,4-tetrahydroquinazolines (III).**  
**a, Ar = 2- $\text{CH}_3\text{C}_6\text{H}_4$ .**—A solution of  $\text{NaBH}_4$  (0.23 g) in EtOH (25 ml) was added to a stirred suspension of II (Ar = 2- $\text{CH}_3\text{C}_6\text{H}_4$ ) (1.95 g) in EtOH (20 ml) during 30 min in an ice bath. The mixture became clear and stirring was continued for 2.5 hr at room temperature. The solvent was distilled under reduced pressure. To the residue was added  $\text{H}_2\text{O}$  and the separated oil was extracted with  $\text{Et}_2\text{O}$ . The dried extract was evaporated to give a crystalline residue, which was recrystallized from EtOH-hexane to give colorless prisms, mp 106–107°, yield 1.2 g (90%) (Table II, **11**).

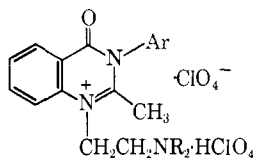
**b, Ar = 2,3-( $\text{CH}_3$ ) $_2\text{C}_6\text{H}_3$ .**—Compound II [Ar = 2,3-( $\text{CH}_3$ ) $_2\text{C}_6\text{H}_3$ ] (3.0 g) was reduced with  $\text{NaBH}_4$  (0.34 g) as above, yielding 2.0 g (95%), mp 137–138° (Table II, **12**).

(6) Melting points were uncorrected and were determined in open capillaries in an oil bath. Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within  $\pm 0.4\%$  of the theoretical values.

(7) H. Takagi, T. Inukai, and M. Nakama, *Japan. J. Pharmacol.*, **16**, 287 (1966).

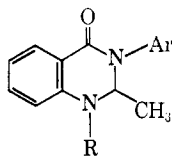
(8) C. S. Weil, *J. Biometric Soc.*, **8**, 249 (1952).

(9) K. Shimamoto and O. Kanemitsu, *Folia Pharmacol. Japon.*, **56**, 575 (1960).

TABLE I  
 1-[2-(DISUBSTITUTED AMINO)ETHYL]-2-METHYL-3-ARYL-4-OXODIHYDROQUINAZOLIUM COMPOUNDS


No.	Ar	-NR <sub>2</sub>	Yield, % from X	Mp, °C dec	Recrystn solvent <sup>c</sup>	Formula <sup>b</sup>
1	C <sub>6</sub> H <sub>5</sub>	-N(CH <sub>3</sub> ) <sub>2</sub>	85	255-256	A	C <sub>19</sub> H <sub>23</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub>
2	2-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	-N(CH <sub>3</sub> ) <sub>2</sub>	51	259-260	B	C <sub>26</sub> H <sub>27</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub>
3	2,3-(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	-N(CH <sub>3</sub> ) <sub>2</sub>	58	258-260	C	C <sub>21</sub> H <sub>27</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub>
4	C <sub>6</sub> H <sub>5</sub>	-N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>	53	248-249	C	C <sub>21</sub> H <sub>27</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub>
5	2-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	-N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>	50	193-195	D	C <sub>22</sub> H <sub>29</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub> · H <sub>2</sub> O
6	2,3-(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	-N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>	38	220-222	D	C <sub>23</sub> H <sub>31</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub>
7	C <sub>6</sub> H <sub>5</sub>		58	249-250	A	C <sub>22</sub> H <sub>27</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub>
8	2-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>		30	253-255	A	C <sub>23</sub> H <sub>29</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub> <sup>e</sup>
9	2,3-(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>		54	233-235	A	C <sub>21</sub> H <sub>21</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub> <sup>d</sup>

<sup>a</sup> A, DMF-EtOH; B, Me<sub>2</sub>CO-Et<sub>2</sub>O; C, DMF-Me<sub>2</sub>CO-Et<sub>2</sub>O; D, MeOH. <sup>b</sup> All compounds were analyzed for C, H, N. <sup>c</sup> C: calcd, 49.11; found, 49.85. <sup>d</sup> C: calcd, 50.00; found, 50.69.

 TABLE II  
 1-SUBSTITUTED 2-METHYL-3-ARYL-4-OXO-1,2,3,4-TETRAHYDROQUINAZOLINES


No.	Ar	R	Yield, %	Salt	Mp, °C	Recrystn solvent <sup>f</sup>	Formula <sup>g</sup>
10	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	87	<i>a</i>	<i>b</i>	...	C <sub>16</sub> H <sub>16</sub> N <sub>2</sub> O
11	2-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	CH <sub>3</sub>	90	<i>a</i>	106-107	A	C <sub>17</sub> H <sub>18</sub> N <sub>2</sub> O
12	2,3-(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	CH <sub>3</sub>	95	<i>a</i>	137-138	A	C <sub>18</sub> H <sub>24</sub> N <sub>2</sub> O
13	C <sub>6</sub> H <sub>5</sub>	C <sub>2</sub> H <sub>5</sub>	87	<i>a</i>	77-78	B	C <sub>17</sub> H <sub>18</sub> N <sub>2</sub> O
14	C <sub>6</sub> H <sub>5</sub>	CH <sub>2</sub> CH <sub>2</sub> N(CH <sub>3</sub> ) <sub>2</sub>	66	<i>a</i>	<i>c</i>	...	C <sub>19</sub> H <sub>23</sub> N <sub>3</sub> O <sup>b</sup>
15	2-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	CH <sub>2</sub> CH <sub>2</sub> N(CH <sub>3</sub> ) <sub>2</sub>	96	Picrolonate	209-210 <sup>d</sup>	A	C <sub>34</sub> H <sub>39</sub> N <sub>7</sub> O <sub>6</sub> <sup>e</sup>
16	2,3-(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	CH <sub>2</sub> CH <sub>2</sub> N(CH <sub>3</sub> ) <sub>2</sub>	65	<i>a</i>	102-103	E	C <sub>21</sub> H <sub>22</sub> N <sub>3</sub> O
17	C <sub>6</sub> H <sub>5</sub>	CH <sub>2</sub> CH <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>	87	(COOH) <sub>2</sub>	182-183 <sup>d</sup>	D	C <sub>23</sub> H <sub>29</sub> N <sub>3</sub> O <sub>5</sub>
18	2-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	CH <sub>2</sub> CH <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>	79	Picrate	158-160 <sup>d</sup>	D	C <sub>28</sub> H <sub>32</sub> N <sub>6</sub> O <sub>5</sub>
19	2,3-(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	CH <sub>2</sub> CH <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>	76	<i>a</i>	110-111	E	C <sub>23</sub> H <sub>31</sub> N <sub>3</sub> O
20	C <sub>6</sub> H <sub>5</sub>	CH <sub>2</sub> CH <sub>2</sub>	68	HCl	242-244 <sup>e</sup>	A	C <sub>22</sub> H <sub>28</sub> ClN <sub>3</sub> O
21	2-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	CH <sub>2</sub> CH <sub>2</sub>	61	HCl	264-266 <sup>d</sup>	D	C <sub>23</sub> H <sub>34</sub> ClN <sub>3</sub> O
22	2,3-(CH <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	CH <sub>2</sub> CH <sub>2</sub>	64	HCl	250-252 <sup>d,e</sup>	F	C <sub>21</sub> H <sub>32</sub> ClN <sub>3</sub> O

<sup>a</sup> Free base. <sup>b</sup> Bp 186-190° (0.5 mm). <sup>c</sup> Bp 250-255° (0.2 mm). <sup>d</sup> Decomposition. <sup>e</sup> Free base melts at 130-131°. <sup>f</sup> A, EtOH; B, hexane-Et<sub>2</sub>O; C, Me<sub>2</sub>CO; D, MeOH; E, hexane; F, EtOH-Et<sub>2</sub>O. <sup>g</sup> All compounds were analyzed for C, H, N. <sup>h</sup> C: calcd, 73.75; found, 73.29. <sup>i</sup> C: calcd, 61.23; found, 61.72.

**2-(N-Ethylacetamido)benzanilide (VI).**—A mixture of 2-ethylaminobenzanilide (V)<sup>1</sup> (2.0 g) and Ac<sub>2</sub>O (6 g) was stirred for 2 hr at room temperature and for an additional 30 min at 60°. Excess Ac<sub>2</sub>O was distilled under reduced pressure. The residue was treated with H<sub>2</sub>O, extracted with CHCl<sub>3</sub>, and washed with 10% Na<sub>2</sub>CO<sub>3</sub>. Distillation of the dried extract gave a crystalline residue, which was recrystallized from C<sub>6</sub>H<sub>6</sub> to give colorless prisms, mp 146-148°, yield 2.3 g (98%). *Anal.* (C<sub>17</sub>H<sub>18</sub>N<sub>2</sub>O<sub>2</sub>) C, H, N.

**1-Ethyl-2-methyl-3-phenyl-4-oxodihydroquinazolium Iodide (IV).** **A.**—A mixture of I (Ar = C<sub>6</sub>H<sub>5</sub>) (4.8 g), EtI (12.5 g), and C<sub>6</sub>H<sub>6</sub> (5 ml) was heated in a sealed tube at 115-125° for 20 hr and cooled. The separated crystalline product was collected by filtration. Recrystallization from MeOH (150 ml) gave IV as colorless needles, mp 236-239° dec, yield 0.9 g (12%). An analytical sample was recrystallized twice from MeOH; mp 241-242° (dec); ir,  $\nu_{\text{max}}^{\text{solid}}$  (cm<sup>-1</sup>) 1720, 1620, 1552. *Anal.* (C<sub>7</sub>H<sub>17</sub>IN<sub>2</sub>O) C, H, N.

I-III (1.7 g) was obtained from the mother liquor of recrystallization; mp 218-220° dec; ir,  $\nu_{\text{max}}^{\text{solid}}$  (cm<sup>-1</sup>) 1730, 1652, 1575, 1545. I (Ar = C<sub>6</sub>H<sub>5</sub>) (1.9 g, 40%) was recovered from the mother liquor of the reaction mixture.

**B.**—To a solution of VI (1.2 g) in EtOH (10 ml) was added 57% HI (1.2 g). A crystalline product appeared soon after the addition. The mixture was allowed to stand for 3 hr at room temperature. The crystalline product was filtered and recrystallized from MeOH to give pale yellow needles, mp 241-242° dec, yield 1.6 g (96%). Ir spectra of the product and a sample obtained by method A were superimposable.

**Reaction of IV with Aqueous Alkaline.**—To a suspension of IV (0.2 g) in a mixture of H<sub>2</sub>O (10 ml) and C<sub>6</sub>H<sub>6</sub> (20 ml) was added 10% K<sub>2</sub>CO<sub>3</sub> (1.5 ml), and the mixture was shaken until the crystals disappeared. The C<sub>6</sub>H<sub>6</sub> layer was separated, dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated to give a crystalline residue (130 mg), mp 139-142°. Recrystallization from C<sub>6</sub>H<sub>6</sub> afforded colorless prisms, mp 146-147°, yield 0.11 g (76.5%). The product

TABLE III  
 SUMMARY OF PHARMACOLOGICAL OBSERVATIONS

Compd	Salt	Max tolerated dose, mg/kg		Behavior in max tolerated dose <sup>a</sup>	Eff %		ED <sub>50</sub> , mg/kg	Prolonged % of sleeping time by barbiturate 100 mg/kg ip	Anti-inflam act., % inhib of carrageenin abscess 100 mg/kg po	Anti-histamic effect, % inhib of histamine spasm 10 <sup>-5</sup> g/ml	LD <sub>50</sub> , mg/kg
		po	ip		66 mg/kg ip	100 mg/kg ip					
11	Free	400	400	a	...	...	...	119	...	b	...
12	Free	400	200	Unchanged	20	40	...	120	20	b	...
14	HCl	400	200	a, c, d	20	40	...	83	22	70	...
15	Free	400	200	a, c	...	...	...	50	...	b	...
16	HCl	400	200	c, d	0	20	...	90	5	60	...
17	Oxalate	400	100	b, c	40	80	80.3 (57.5-134.0) ip 195.0 (145.0-256.3) po	70	10	65	180.0 (157.2-221.0) ip 677.6 (524.3-857.7) po
18	Oxalate	400	100	b, c	40	80	70.6 (41.9-118.5) ip 168.1 (42.7-334.9) po	140	20	80	155.5 (127.3-180.4) ip 620.4 (520.5-750.7) po
19	Oxalate	200	100	b, c	0	20	...	126	10	80	...
20	HCl	400	200	b	0	20	...	105	5	95	...
21	HCl	400	200	Unchanged	0	0	...	140	27	50	...
22	HCl	200	100	b, d	0	0	...	300	7	50	...
Aminopyrine	...	400	100	a, e	...	...	77.9 (57.1-106.3) ip 356.4 (246.1-491.1) po	...	13	...	315.6 (241.1-413.2) ip 1505.8 (1087.8-1816.8) po
Chlorphenylamine	...	100	50	a, e	...	...	...	...	...	100	...
Diphenhydramin	...	100	50	a	...	...	...	...	...	100	...
Phenylbutazone	...	400	200	Unchanged	...	...	...	...	20	...	...

<sup>a</sup> Behavior: a, decreased spontaneous activity; b, increased spontaneous activity; c, tremors; d, staggering gate; e, muscle relaxation. <sup>b</sup> Insoluble in Tyrode solution.

was confirmed to be identical with VI (melting point and ir spectra).


**1-Ethyl-2-methyl-3-phenyl-4-oxo-1,2,3,4-tetrahydroquinazoline (VII).** A.—Compound IV (1.0 g) was reduced with NaBH<sub>4</sub> (0.11 g) in EtOH. Recrystallization from Et<sub>2</sub>O-hexane (1:2) afforded colorless prisms, mp 77-78°, yield 0.59 g (87%). Analytical data are listed in Table II (4).

**B.**—A mixture of V (3.0 g), acetaldehyde diethyl acetal (2.5 g), concentrated H<sub>2</sub>SO<sub>4</sub> (1 drop), and absolute EtOH (30 ml) was refluxed for 4.5 hr. The solvent was evaporated and the residue was dissolved in Et<sub>2</sub>O and washed (5% NaHCO<sub>3</sub>, saturated NaCl). The dried ethereal solution was distilled to give a colorless oil (3.3 g). The residue was dissolved in absolute hexane and chromatographed over Al<sub>2</sub>O<sub>3</sub>. The eluate with hexane-C<sub>6</sub>H<sub>6</sub> (3:2) gave a colorless oil (1.3 g), which crystallized after standing in a refrigerator. Recrystallization from Et<sub>2</sub>O-hexane (1:2) gave colorless prisms, mp 75-77°, yield 1.1 g (33%). Admixture with a sample obtained by method A did not depress its melting point.

**N-[2-(Disubstituted aminoethyl)anthranilic acids (IX)]** were prepared by a modification of Hauptmann's method.<sup>2</sup>

**a, R<sub>2</sub> = (CH<sub>3</sub>)<sub>2</sub>.**—2-Chlorobenzoic acid (35.5 g) was dissolved in a solution of KOH (13 g) in H<sub>2</sub>O (23 ml), and 1,1-dimethylethylenediamine (22 g), K<sub>2</sub>CO<sub>3</sub> (15.8 g), and Cu powder (5 g) were added. The mixture was refluxed for 72 hr, cooled, and acidified with 10% HCl (170 ml). The separated semisolid was extracted with Et<sub>2</sub>O, and the acidic aqueous layer was placed on a column of Amberlite IR-120 (1000 ml of acid form). The column was washed with H<sub>2</sub>O and eluted with 5% NH<sub>4</sub>OH (5 l). The eluate was evaporated to dryness under reduced pressure. The residue was crystallized by trituration in MeOH and recrystallization from MeOH to give IX [R<sub>2</sub> = (CH<sub>3</sub>)<sub>2</sub>] as colorless needles, mp 180-182° dec, yield 15.4 g (33%). *Anal.* (C<sub>11</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub>) C, H, N.

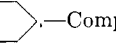
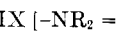
**b, R<sub>2</sub> = (C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>.**—VIII (108.6 g) and 1,1-diethylethylenediamine (88.5 g) were worked up as for a. Recrystallization from 96% EtOH gave colorless prisms, mp 161-163° dec, yield 57.8 g (31%). *Anal.* (C<sub>13</sub>H<sub>20</sub>N<sub>2</sub>O<sub>2</sub>) C, H, N.

**c, -NR<sub>2</sub> = -N**  **was similarly prepared from VIII (108.6 g) and N-(2-aminoethyl)piperidine (87.6 g). Recrystallization from 96% EtOH afforded prisms, mp 195-197° dec, yield 51 g (30%).** *Anal.* (C<sub>14</sub>H<sub>20</sub>N<sub>2</sub>O<sub>2</sub>) C, H, N.

**N-Acetyl-N-[2-(disubstituted amino)ethyl]anthranilic Acids (X).** **a, R<sub>2</sub> = (CH<sub>3</sub>)<sub>2</sub>.**—A mixture of IX [R<sub>2</sub> = (CH<sub>3</sub>)<sub>2</sub>] (3.0 g) and Ac<sub>2</sub>O (30 g) was stirred for 4 hr at room temperature. Excess Ac<sub>2</sub>O and AcOH were distilled under reduced pressure. The residue was crystallized by the addition of Et<sub>2</sub>O-Me<sub>2</sub>CO. Recrystallization from Me<sub>2</sub>CO afforded colorless needles of hy-

drated X [R<sub>2</sub> = (CH<sub>3</sub>)<sub>2</sub>], mp 95-97°, yield 25.1 g (65%). *Anal.* (C<sub>13</sub>H<sub>18</sub>N<sub>2</sub>O<sub>3</sub>·H<sub>2</sub>O) C, H, N.

**b, R<sub>2</sub> = (C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>.**—Acetylation of IX [R<sub>2</sub> = (C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>] (25 g) was similarly worked up. Recrystallization from Me<sub>2</sub>CO gave acetyl derivative of X [R<sub>2</sub> = (C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>] as colorless prisms, mp 102-103°, yield 32.3 g (90%). *Anal.* (C<sub>15</sub>H<sub>22</sub>N<sub>2</sub>O<sub>3</sub>·AcOH) C, H, N.

**c, -N(R)<sub>2</sub> = -N**  **—Compound IX [-NR<sub>2</sub> = -N**   **(30 g) was acetylated with Ac<sub>2</sub>O (60 g) as above. The product was recrystallized from THF to give colorless prisms, mp 150-152°, yield 29.5 g (84%).** *Anal.* (C<sub>15</sub>H<sub>22</sub>N<sub>2</sub>O<sub>3</sub>) C, H, N.

**2-(2-Dimethylaminoethyl)aminobenzanilide (XI).**—A stirred solution of IX [R<sub>2</sub> = (CH<sub>3</sub>)<sub>2</sub>] (6.2 g) in H<sub>2</sub>O (60 ml) was treated with COCl<sub>2</sub> at 15-20°. Gelatinous precipitates separated from the mixture. After 1 hr, a solution of NaOAc·3H<sub>2</sub>O (8.4 g) in H<sub>2</sub>O (10 ml) was added to the mixture and the treatment with COCl<sub>2</sub> continued for an additional 4 hr. The mixture was evaporated to dryness under reduced pressure at room temperature. Aniline (10 g) was added to the residue and the mixture was heated at 95-100° for 2 hr and cooled. The brown reaction mixture dissolved in H<sub>2</sub>O (20 ml) and was made alkaline with K<sub>2</sub>CO<sub>3</sub>, and the separated oil was extracted with Et<sub>2</sub>O. The dried extract was evaporated to give a brown oil, from which excess aniline was distilled off under reduced pressure. The residue was crystallized by the addition of hexane. Recrystallization from C<sub>6</sub>H<sub>6</sub>-hexane (1:3) afforded colorless plates, mp 116-118°, yield 1.75 g (21%). *Anal.* (C<sub>17</sub>H<sub>21</sub>N<sub>3</sub>O) C, H, N.

**1-(2-Dimethylaminoethyl)-2-methyl-3-phenyl-4-oxodihydroquinazolinium Diperchlorate (XIII, R<sub>2</sub> = (CH<sub>3</sub>)<sub>2</sub>).** **A.**—To a solution of X [R<sub>2</sub> = (CH<sub>3</sub>)<sub>2</sub>] (5.0 g) and aniline (2.5 g) in dry THF was added N,N'-dicyclohexylcarbodiimide (9.0 g). The mixture was stirred for 20 hr at room temperature and filtered to separate colorless precipitates. The filtrate was evaporated to give an oily residue. The residue was dissolved in 10% HCl and filtered, and the acidic filtrate was washed with Et<sub>2</sub>O to remove unreacted materials. The aqueous layer was made alkaline with K<sub>2</sub>CO<sub>3</sub> and the separated oil was extracted with CHCl<sub>3</sub>. Distillation of the dried extract afforded an oily residue (ca. 6 g). A solution of the residue (0.5 g) in Et<sub>2</sub>O (20 ml) was adjusted at pH 6.4-6.6 by the addition of 60% HClO<sub>4</sub> to give a gummy precipitate, which was crystallized by treating with Me<sub>2</sub>CO. Recrystallization from EtOH gave perchlorate of 2-N-(2-dimethylaminoethyl)acetamidobenzanilide [XII, R<sub>2</sub> = (CH<sub>3</sub>)<sub>2</sub>] as colorless needles: mp 175-179° dec; ir, ν<sub>max</sub><sup>101</sup> (cm<sup>-1</sup>) 3300, 3100, 1675, 1650, 1600, 1540. *Anal.* (C<sub>19</sub>H<sub>24</sub>ClN<sub>3</sub>O<sub>8</sub>) C, H, N.

The remaining residue (ca. 5.5 g) was dissolved in absolute EtOH (40 ml) and treated with excess 60% HClO<sub>4</sub> to give a crystalline product, which was very insoluble in organic solvents.

Recrystallization from DMF-EtOH afforded XIII [ $R_2 = (CH_3)_2$ ] as colorless needles: mp 255–256° dec; yield 7.4 g (85%); ir,  $\nu_{max}^{NaCl}$  ( $cm^{-1}$ ) 3130, 1720, 1625, 1555. Analytical data are listed in Table I.

1-[2-(Disubstituted amino)ethyl]-2-methyl-3-aryl-4-oxodihydroquinazolinium diperchlorates were prepared in a similar way, without isolation of XII. The products are shown in Table I (2–9).

**B.**—AcCl (0.15 g) was added to a mixture of XIII (0.28 g), powdered  $K_2CO_3$  (0.2 g), and dry  $C_6H_6$  (3 ml). The mixture was stirred overnight at room temperature, treated with  $H_2O$  (5 ml), and made alkaline with 10%  $K_2CO_3$ . The benzene layer was separated, washed with  $H_2O$ , and dried ( $K_2CO_3$ ). Distillation of the solvent gave an oily residue (0.3 g), which was dissolved in  $Et_2O$  and treated with excess 60%  $HClO_4$ . The crude crystalline product and EtOH (3 ml) were refluxed for 10 min and cooled. The separated crystals were collected, mp 255–256° dec, yield 0.35 g. Ir spectra of the product and a sample obtained by method A were superimposable.

**1-(2-Dimethylaminoethyl)-2-methyl-3-phenyl-4-oxo-1,2,3,4-tetrahydroquinazoline [XIV,  $R_2 = (CH_3)_2$ ].** **A.**—To a stirred suspension of XII [ $R_2 = (CH_3)_2$ ] (5.0 g) in EtOH (50 ml) was added a solution of  $NaBH_4$  (0.8 g) in EtOH (90 ml) at  $-5$  to  $-2^\circ$  over a period of 3 hr. The mixture became clear after the addition and the solvent was distilled under reduced pressure. To the residue was added  $H_2O$  and separated oil was extracted with  $Et_2O$ . The dried extract was evaporated and the residue was distilled at 250–255° (0.2 mm) to give a viscous oil, yield 2.0 g (66%). *Anal.* ( $C_{15}H_{23}N_3O$ ) H, N; C: calcd, 73.75; found, 73.29.

The **oxalate** yielded colorless plates (from  $Me_2CO$ ), mp 86–89° dec. Analytical data are listed in Table II (14).

1-[2-(Disubstituted amino)ethyl]-2-methyl-3-phenyl-4-oxo-1,2,3,4-tetrahydroquinazolines were prepared by the same way as above. The products are shown in Table II (15–22).

**B.**—A solution of XIII (1.4 g) and paraldehyde (0.66 g) in absolute EtOH (20 ml) was saturated with dry HCl at 0–5° and the mixture was allowed to stand at room temperature overnight. The solvent was distilled under reduced pressure. The residue was made alkaline with concentrated  $NH_4OH$  and the separated oil was extracted with  $Et_2O$ . Distillation of the extract gave an oily residue (1.5 g), which was dissolved in dry  $C_6H_6$  and chromatographed over  $Al_2O_3$ . The  $C_6H_6$ -MeOH (98:2) eluate afforded a colorless viscous oil, which was distilled, bp 280° (bath temperature) (0.2 mm), yield 0.5 g (32%). Ir spectra of the product and a sample obtained by method A were superimposable.

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## Derivatives of 3-Piperidinol as Central Stimulants

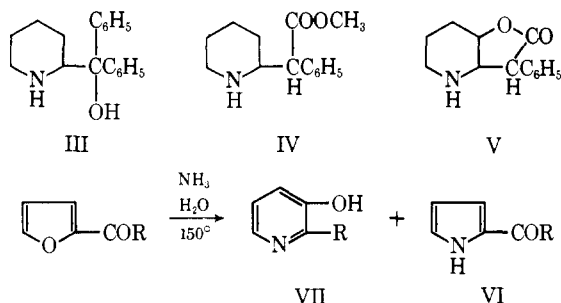
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A number of 2-alkyl-3-pyridinols, prepared by the ammonolysis of alkyl 2-furyl ketones, were hydrogenated to 3-piperidinols. Several of these latter compounds and their N-substituted derivatives were potent central stimulants. Two of the most active compounds were resolved and the activity was found predominantly in one optical isomer in each case.

The clinical usefulness of pipradrol (III) and methylphenidate (IV) as central stimulants prompted the preparation of the 2-alkyl-3-piperidinols I (Table I). These compounds were obtained by hydrogenating the corresponding 3-pyridinols II (Table II), followed by alkylation or acylation where necessary. The 3-pyridinols II (Table II), where X = H, were prepared by the ammonolysis and concomitant ring closure of 2-furyl ketones to a mixture of the desired II and an alkyl 2-pyrryl ketone VI.



Ammonia or ammonium salts in various solvent combinations have been used for this reaction,<sup>1</sup> but in the present study ammonia in aqueous methanol gave the best yields. In some instances the reaction failed.

For example, 2-furoylphenylacetone nitrile cleaved to furamide; 2-furyl 9-fluorenyl ketone gave fluorene quantitatively, and 2-furyl 9-xanthenyl ketone gave xanthene and a small amount of the desired 3-pyridinol. A 5-methyl group in the furan ring reduced the yield of pyridinol to less than 5%. No effort was made to isolate the by-product 2-pyrryl ketones.

The diarylmethyl 2-furyl ketones (Table III) were obtained in good yield from ethyl furoate and a diphenylmethane with  $NaNH_2$  or  $KNH_2$  in liquid ammonia. Several attempts to acylate furan with diphenylacetyl chloride and stannic chloride gave only traces of ketone, although this method<sup>2</sup> has been used successfully for similar compounds.

The hydrogenation of the 3-pyridinols to the 3-piperidinols was difficult. With Pt hydrogenation of the bases in acetic acid or the hydrochloride salts in ethanol gave complex mixtures containing most of the possible combinations of phenyl, cyclohexyl, pyridine,

(1) (a) H. Leditschke, *Angew. Chem.*, **61**, 446 (1949); (b) *Chem. Ber.*, **85**, 202 (1952); (c) *Ibid.*, **86**, 123 (1953); (d) A. P. Dunlop and S. Swadesh, U.S. Patent 2,636,882 (March 16, 1954); (e) W. Gruber, *Can. J. Chem.*, **31**, 561 (1953); (f) *Chem. Ber.*, **88**, 178 (1955); (g) B. R. Baker and E. J. McEvoy, *J. Org. Chem.*, **20**, 136 (1955); (h) H. Rapaport and E. Volcheck, *J. Am. Chem. Soc.*, **78**, 2451 (1956).

(2) Yu. Goldfarb and L. Smorgonskii, *J. Gen. Chem. USSR*, **8**, 1523 (1938).