

^a These compounds were titrated with **KI** to determine active oxygen and all gave 97-99.3% of the calculated values. All compounds were analyzed for C, H, N, and the analytical results obtained were within $\pm 0.4\%$ of the theoretical values. ^b We wish to thank Ed Hoff for C, H, N analyses.

Experimental Section

The oxaziridines were prepared by the peracetic $acid^2$ or *m*-chloroperbenzoic $acid^4$ oxidation of the corresponding infines. The oxaziridines were isolated by distillation at reduced pressure and in some cases additional purification was obtained by chromatographing with a neutral alumina column.

(4) R. G. Pews, J. Org. Chem., 32, 1628 (1967).

Possible Anticonvulsant Thiazolo[3,2-a]benzimidazole Mannich Bases. XI¹

J. M. Singh²

School of Chemistry, Meerut College, Meerut, India

Received January 28, 1969

In view of the potent pharmacodynamic activity³ of a large number of thiazole Mannich bases, additional thiazolo[3,2-*a*]benzimidazole Mannich bases were synthesized. These compounds have been tested for anticonvulsant activity (Table I).

Experimental Section

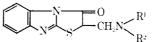
Thiazolo[3,2-*a*] benzimidazol-3(2H)-one.--2-Carbethoxymethylthiobenzimidazole⁴ (5 g) in *o*-PhCl₂ (20 ml) was refluxed for 1 hr while removing EtOH. The mixture became quite dark; colored crystals (1 g) separated and were recrystallized (EtOH), mp 179-180°.

Diethylaminothiazolo[3,2-*a*]**benzimidazol-3**(2**H**)-**one**. A mixture of thiazolo[3,2-*a*]**benzimidazol-3**(2**H**)-one (2 g), Et₂NII (2 ml), AcOH (10 ml), and CH₂O (2 ml) was heated on a steam bath for 7 hr. After cooling, H₂O (20 ml) was added and the solution was neutralized with saturated aqueous K₂CO₃. The base was filtered, washed (H₂O), and recrystallized (EtOH), yield $42\frac{V_{C0}}{C0}$ mp 210°. Anal. (C₄H₁₇NSO) N, S.

The above procedure was followed to prepare the other compounds.

TABLE I

MANNICH BASES DERIVED FROM THIAZOLO[3,2-a]BENZIMIDAZOL-3(2H)-ONE"



N 0.	N(R1R2)	Formula	Mp. °C	Yield, %	$\Lambda e tivity^{L}$	LD ₅₆ (toxicity). mg/kg
1	$\mathrm{Et}_{2}\mathbf{N}$	$C_{14}H_{17}N_3SO$	210	42	++	250
2	Me_2N	$C_{12}H_{17}N_3SO$	190	45	+	390
3	$\mathrm{Ph}_{2}\mathbf{N}$	$C_{22}H_{17}N_3SO$	175	50	+++	350
4	PhNEt	$C_{18}H_{17}N_3SO$	250	40	++	290
5	$n-\Pr_2 N$	$\mathrm{C}_{16}\mathrm{H}_{21}\mathrm{N}_3\mathrm{SO}$	240	45	++	370
6	n -B n_2N	$C_{18}H_{25}N_3SO$	250	50	++++	400
ī	sec-Bu ₂ N	$C_{18}H_{25}N_3SO$	210	45	+++	500
8	Piperidino	C ₁₅ H ₁₇ N ₃ SO	215	40	+ + + +	450
9	Morpholina	$\mathrm{C}_{14}\mathrm{H}_{15}\mathrm{N}_{3}\mathrm{SO}_{2}$	230	41	+++++	480

"All new compounds were analyzed for N and S; the analytical values were within $\pm 0.4\%$ of the calculated values. ^b Mice were used for the experiments for anticonvulsant activity following the method by T. J. Putnam and H. H. Merritt, *Science*, **85**, 525 (1937). +++++ = convulsive threshold elevated more than 60 mA, ++++ raised by 60 mA, +++ = raised by 40 mA, ++ = raised by 10–15 mA, 2 hr after treatment.

(I) Part X: J. M. Singh, J. Med. Chem., in press.

(3) F. F. Blicks, Ann. Rev. Biochem., 13, 549 (1943).

(4) 3. 11. Van Allan, J. Ocg. Chem., 21, 24 (1956).

⁽²⁾ Address inquiries to Defence Science Laboratory, Delhi-6, India.