

B. 1,4-Bis(2-indol-3-ylethyl)piperidine.—A mixture of 4-[2-(indol-3-yl)ethyl]piperidine² (15 g) and finely powdered Na₂CO₃·H₂O (15 g) in *i*-PrOH (100 ml) was stirred and refluxed while 3-(2-bromoethyl)indole (15 g) in *i*-PrOH (100 ml) was added dropwise during 1 hr. Stirring and refluxing were continued for 18 hr, then the mixture was evaporated to dryness *in vacuo*. The residue was triturated under CHCl₃ several times and an insoluble white solid was filtered off. Evaporation of the filtrate and recrystallization of the residue from AcMe-H₂O gave 13.5 g (53%) of product, mp 145–147°.

C. 1-(2-Indol-3-ylethyl)-4-[2-(2-methylindol-3-yl)ethyl]piperidine.—A mixture of 4-[2-(2-methylindol-3-yl)ethyl]pyridine (23.6 g) and 3-(2-bromoethyl)indole (22.4 g) in absolute EtOH (500 ml) was kept at room temperature in the dark for 1 week. The resultant solution of quaternary salt was hydrogenated at 50° and 28.12 kg/cm² in the presence of PtO₂ (2.0 g) for 18 hr. The product precipitated and was filtered off along with the catalyst. Separation from catalyst was achieved by stirring with AcNMe₂ and filtering. H₂O was added to the filtrate until crystallization commenced to give 15.9 g (35%) of product hydrobromide, mp 242–243°.

D. 1,4-Bis[2-(1-methylindol-3-yl)ethyl]piperidine.—1,4-Bis(2-indol-3-ylethyl)piperidine (2.0 g) was added portionwise to NaNH₂ in liquid NH₃ prepared from 248 mg of Na. After stirring for 1 hr, MeI (1.56 g) in Et₂O (10 ml) was added dropwise and stirring was continued for 2 hr. NH₃ was allowed to evaporate and the residue was stirred with Et₂O (100 ml) and H₂O (100 ml). Evaporation of the dried (MgSO₄) ether layer and recrystallization of the solid residue from AcMe-H₂O provided colorless needles (1.2 g), mp 123–124°.

Derivatives of 9-Thioxanthenecarbonitrile

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Since the discovery of methadone, the analgetic first synthesized by Boekmühl and Ehrhart,¹ the medicinal chemist has found 3,3-diphenylpropylamines in general to be a fertile source of diverse biological activities.² Our brief excursion into this area was directed toward amalgamation of the analgetic properties of the diphenylpropylamines with the antianxiety properties of the thioxanthen-9-alkylamines by attempting to prepare the thioxanthen analog (**9**) of methadone and some of its congeners.

Application of the classic methadone synthetic scheme¹ to the thioxanthen ring system embraced two problems: (1) preparation of 9-thioxanthenecarbonitrile (**3**) in adequate yield for use as an intermediate, and (2) conversion of a crowded nitrile to the corresponding ethyl ketone.

Preparation of **3a** from thioxanthen-9-ol (**1a**), obtained by NaBH₄ reduction of thioxanthen-9-one,³ was accomplished in 61% over-all yield by treating intermediate 9-chlorothioxanthen (**2a**)⁴ with CuCN. (Scheme I). Conversion of **1a** to other reactive intermediates, *e.g.*, bromide, tosylate, or mesylate, in each case produced markedly inferior results, even when a variety of CN⁻ sources was employed. This is not unreasonable in the light of our experience with chloride **2a** which is rapidly converted to **1a** and thioxanthen-9-

one in the presence of air. Thus the yield of nitrile depends upon the care exercised in isolation of the intermediate.

Alkylation of the Na derivative of **3a** with 2-chloro-N,N-dimethylethylamine proceeds smoothly to **4a**, which reacts with EtMgBr at 0°. The ketimine is completely hydrolyzed to the corresponding ketone (**5a**) after 30 min at reflux in dilute aqueous acid. By the same scheme 2-chlorothioxanthen-9-ol (**1b**)⁵ is converted to **5b**. In each case temperature control during the Grignard addition is critical, since the Mg derivative of the ketimine is unstable above 25°. In fact, if **4b** and the Grignard reagent are heated to 100° before hydrolysis, no **5b** can be detected. The major product, **6**, apparently arises by fragmentation of the Mg adduct driven by the propensity of the thioxanthen system to delocalize anionic charge at C-9.

Alkylation of the Na derivative of **3a** with 2-(dimethylamino)isopropyl chloride provides the two expected isomers **7** and **8**, which can be separated on alumina. Structural assignments are based upon several bits of data: (1) relative chemical shifts of the N-Me and C-Me protons based on models reported by Casey,⁶ (2) isomer **7** runs faster than **8** on both alumina and silica gel, and (3) relative reactivity toward the Grignard reagent. Neither isomer reacts with EtMgBr at room temperature. Isomer **8** is consumed only at temperatures above 100°, but after aqueous acid treatment no ketonic material can be detected by ir analysis of the total product. Isomer **7** reacts at ca. 50–60° with similar results. Apparently steric hindrance by the α-Me is sufficient to preclude attack of the reagent at a temperature compatible with adduct stability. Further studies indicate that variation of the solvent and/or the halogen has no influence on the temperature requirement. The utility of EtLi has not yet been explored.

The intermediates **4–8** were devoid of activity at 10 mg/kg (administered subcutaneously) when subjected to three well-known laboratory methods for evaluating analgetic activity: the hot plate method,⁷ the tail flick method,⁸ and the phenylbenzoquinone writhing method.⁹

Experimental Section

Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within ±0.4% of the theoretical values. Melting points (Thomas-Hoover capillary melting point apparatus) are uncorrected. The ir spectra were measured with a Perkin-Elmer Model 21 spectrophotometer, uv spectra with a Cary Model 11 recording spectrometer, and nmr spectra with a Varian A-60 spectrometer with Me₄Si as an internal standard.

9-Thioxanthenecarbonitrile (3a).—A suspension of thioxanthen-9-ol³ (21.4 g, 0.100 mole) in 200 ml of anhydrous Et₂O under N₂ was treated carefully with SOCl₂ (8.0 ml) with stirring at 0–5°. The alcohol dissolved before a white solid began to separate. After 1 hr at 25°, the solvent was removed *in vacuo*, 100 ml of anhydrous C₆H₆ added, and the solution again evaporated. All manipulations were performed with rigorous exclusion of air. Anhydrous C₆H₆ (100 ml) and CuCN powder

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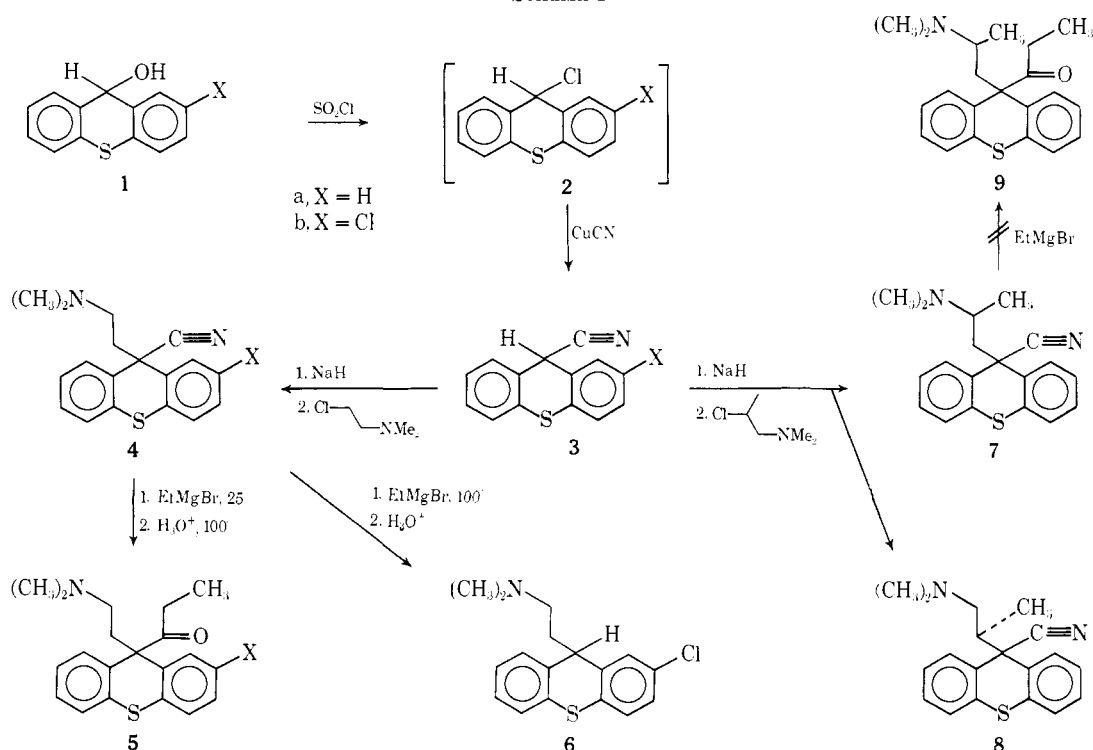
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SCHEME I



(17.9 g, 0.20 mole) were added and the suspension was stirred at reflux under N_2 for 4 hr. Before cooling the mixture was filtered, the solids were washed with hot C_6H_6 , and the filtrate was evaporated to yield a crystalline residue (20.6 g), mp 85–89°, containing thioxanthene and thioxanthene-9-one, as well as the desired nitrile (tlc). One recrystallization from hexane removed thioxanthene and one recrystallization from *i*-PrOH removed most of the thioxanthene-9-one to provide 13.6 g (61%) of **3a**, mp 92.5–95°, which was pure enough to be used as a synthetic intermediate. Three additional recrystallizations from *i*-PrOH provided pure **3a**: mp 97–98°, ir (KBr) $\text{C}\equiv\text{N}$ stretching either weak or absent; uv max (EtOH) 262 μ (log ϵ 4.0). *Anal.* ($\text{C}_{14}\text{H}_9\text{NS}$) C, H, N.

2-Chloro-9-thioxanthene-carbonitrile (3b) was prepared by the same procedure in 64% yield, mp 134–136° (Me_2CO -hexane), ir (KBr) 4.49 μ ($\text{C}\equiv\text{N}$, very weak), uv max (EtOH) 266 μ (log ϵ 4.1). *Anal.* ($\text{C}_{14}\text{H}_8\text{ClNS}$) C, H, N.

Method A. 9-[2-(Dimethylamino)ethyl]-9-thioxanthene-carbonitrile (4a).—A solution of **3a** (11.2 g, 0.050 mole) in 50 ml of anhydrous DMF under N_2 was treated with 49% NaH in oil (2.5 g, 0.050 mole). After ca. 10 min H_2 evolution subsided and 2-chloro-*N,N*-dimethylethylamine (8.5 g, 0.070 mole) was added to the deep red carbanion solution. The mixture was stirred at 95° for 15 min during which time the color discharged. After dilution by 2 vol of H_2O , the products were extracted into EtOAc. The EtOAc extracts were washed twice with H_2O , then extracted by three 30-ml portions of 1 *N* aqueous HCl. The combined acidic extracts were treated with 10 *N* aqueous NaOH to pH 11 and the product was extracted with EtOAc.

The residue dissolved in Et_2O afforded a crystalline salt upon treatment with HCl- Et_2O . One recrystallization from *i*-PrOH provided 10.0 g (60%) of pure **4a**·HCl, mp 232–234°, uv max (EtOH) 265 μ (log ϵ 4.1). *Anal.* ($\text{C}_{15}\text{H}_{18}\text{N}_2\text{S}\cdot\text{HCl}$) C, H, N.

2-Chloro-9-[2-(dimethylamino)ethyl]-9-thioxanthene-carbonitrile (4b) was prepared in 73% yield from **3b** via method A: mp 71.5–73° (hexane). The hydrochloride crystallized from *i*-PrOH: mp 260–261.5°, uv max (EtOH) 270 μ (log ϵ 4.1). *Anal.* ($\text{C}_{15}\text{H}_{17}\text{ClN}_2\text{S}\cdot\text{HCl}$) C, H, N.

1-[9-[2-(Dimethylamino)ethyl]thioxanthene-9-yl]-1-propanone (5a).—A solution of **4a** (4.1 g, 0.014 mole) in 60 ml of anhydrous C_6H_6 was added to a solution of EtMgBr (from 3.3 g of EtBr, 0.73 g of Mg, and 15 ml of anhydrous Et_2O) under N_2 at such a rate so as to maintain the reaction temperature below 5°. After stirring at 25° for 18 hr, the pale green suspension was treated with 30 ml of 3 *N* aqueous HCl and heated for 0.5 hr on a steam

bath while the Et_2O and C_6H_6 were allowed to distil away from the reaction mixture. The mixture was cooled and the oily hydrochloride was extracted into CHCl_3 . Two recrystallizations from C_6H_6 -hexane afforded 3.05 g (50%) of pure **5a**·HCl,¹⁰ mp 178–180°, ir (KBr), 5.81 μ ($\text{C}=\text{O}$), uv max (EtOH) 274 μ (log ϵ 4.0). *Anal.* ($\text{C}_{20}\text{H}_{23}\text{NOS}\cdot\text{HCl}$) C, H, N.

1-(2-Chloro-9-[2-(dimethylamino)ethyl]thioxanthene-9-yl)-1-propanone (5b) was prepared in 41% yield from **4b** by the same procedure but was isolated as the maleate salt: mp 188–190°, ir (KBr) 5.85 μ ($\text{C}=\text{O}$), uv max (EtOH) 277 μ (log ϵ 4.1). *Anal.* ($\text{C}_{20}\text{H}_{22}\text{ClNOS}\cdot\text{C}_4\text{H}_4\text{O}_4$) C, H, N.

2-Chloro-*N,N*-dimethylthioxanthene-9-ethylamine (6), the major product if the Grignard mixture is heated to 100° before addition of the aqueous HCl, was isolated as the hydrochloride, mp 257–258.5°, uv max (EtOH) 272 μ (log ϵ 4.1). *Anal.* ($\text{C}_{17}\text{H}_{18}\text{ClNS}\cdot\text{HCl}$) C, H, N.

9-[2-(Dimethylamino)propyl]-9-thioxanthene-carbonitrile (7) was prepared from 11.2 g of **3a** and 2-chloro-*N,N*-dimethylpropylamine by method A. The crude residue (14.8 g) derived from the EtOAc extract was dissolved in hexane and chromatographed on 300 g of neutral Al_2O_3 (Woelm, activity I). Elution with 2% Et_2O -hexane provided 3.0 g of an oil, which was greatly enriched (ca. 9:1) in the less polar (LP) isomer. Subsequent elution by Et_2O yielded 8.7 g of a mixture slightly enriched (ca. 6:4) in the more polar (MP) isomer. The 90% LP fraction was converted to the hydrochloride salt in Et_2O and twice recrystallized from EtOH to yield 1.2 g of pure **7**·HCl: mp 276–277° dec; nmr (CDCl_3) δ 2.42 (broad singlet, $^+\text{NCH}_3$), 1.20 (d, 3 H, $J = 6$ cps, CCH_3). *Anal.* ($\text{C}_{19}\text{H}_{20}\text{N}_2\text{S}\cdot\text{HCl}$) C, H, N.

9-[1-(Dimethylamino)-2-propyl]-9-thioxanthene-carbonitrile (8) was obtained by rechromatographing the above Et_2O eluate on 90 g of neutral Al_2O_3 , eluting with C_6H_6 to remove all LP isomer, then eluting the MP isomer with Me_2CO . The Me_2CO eluate was treated with HCl and the insoluble salt was recrystallized from EtOH to afford 1.4 g of pure **8**·HCl: mp 241.5–243° dec; nmr (CDCl_3) δ 2.46 (sharp singlet, $^+\text{NCH}_3$), 1.29 (d, 3 H, $J = 7$ cps, CCH_3). *Anal.* ($\text{C}_{19}\text{H}_{20}\text{N}_2\text{S}\cdot\text{HCl}$) C, H, N.

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(10) Compound **5a** previously reported is the maleate salt: C. L. Zirkle, U. S. Patent 3,095,425 (1963); *Chem. Abstr.*, **59**, 12842g (1963).