

ogous transformations of the 4 series described above. The bromomethyl ketone hydrobromide was obtained in 61% yield, mp 244–249° dec. *Anal.* (C₁₁H₆BrCl₂NO·HBr) C, H, Br, Cl, N.

6,8-Dichloro-5-quinolyethylene Oxide.—The bromomethyl ketone hydrobromide was reduced by NaBH₄ in MeOH suspension (as in the 4 series) to give the oxide (78%) which, after recrystallization from EtOH, had mp 146–147°. *Anal.* (C₁₁H₇Cl₂NO) C, H, Cl, N.

α-(Di-*n*-butylaminomethyl)-6,8-dichloro-5-quinolinemethanol Hydrochloride (II·HCl).—The oxide precursor and *n*-Bu₂NH reacted under the same conditions used in the 4 series to give an oil that was converted into the HCl salt by the action of ethereal HCl. The yield of product, mp 162–164°, was quantitative. *Anal.* (C₁₉H₂₆Cl₂N₂O·HCl) C, H, Cl, N.

α-(Di-*n*-butylaminomethyl)-2-phenyl-4',6,8-trichloro-5-quinolinemethanol Hydrochloride (III·HCl).—The 4-chlorophenylation reaction with II·HCl was carried out using a procedure similar to that described¹⁹ for the phenylation of certain quinolinemethanols. 4-Chlorophenyllithium was prepared immediately before use by a literature procedure²⁵ and was used in a tenfold excess. The crude product was converted into the HCl salt by ethereal HCl and the salt was recrystallized from *i*-PrOH to give a 35% yield of a colorless powder, mp 258–260°. *Anal.* (C₂₅H₂₇Cl₃N₂O·HCl) C, H, Cl, N.

Acknowledgment.—We wish to thank Dr. Francis J. Bullock, Professor Robert E. Lyle, and Dr. Richard E. Strube for their many helpful suggestions made during the course of this work.

(25) H. Gilman and S. M. Spatz, *J. Amer. Chem. Soc.*, **66**, 621 (1944).

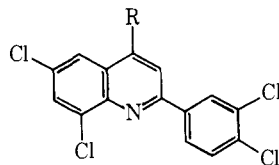
Antimalarials. Quinolinemethanol Derivatives

TARA SINGH AND JOHN H. BIEL

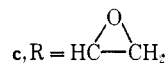
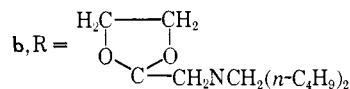
Research Laboratories, Aldrich Chemical Company, Inc., Milwaukee, Wisconsin 52333

Received October 15, 1969

6,8-Dichloro-2-(3,4-dichlorophenyl)-4-(α-di-*n*-butylaminomethyl)quinolinemethanol (**1a**)¹ is a very active antimalarial compound among 4-quinolinemethanols,²

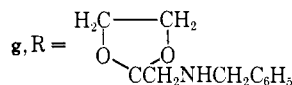
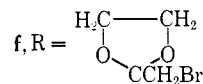


1a, R = CHOCH₂N(*n*-C₄H₉)₂



d, R = COCH₂Br

e, R = CHOCH₂NHCH₂C₆H₅·HCl



(1) R. E. Lutz, *et al.*, *J. Amer. Chem. Soc.*, **68**, 1827 (1946).

(2) W. E. Roth and D. P. Jacobs, Paper No. 37, Division of Medicinal Chemistry, 154th Meeting of the American Chemical Society, September, 1967, Chicago, Ill.

but is highly phototoxic, and therefore cannot be used for the treatment of malaria. Several approaches have been explored to prepare less phototoxic 4-quinolinemethanols³ without reducing their antimalarial activity.

We started to make **1b** with a view that the modified environments at this carbon atom might decrease the phototoxicity.

By the reaction of **1d** with *n*-Bu₂NH, the desired di-*n*-butylaminoketone could not be obtained. When **1d** was converted into the dioxyethylene compound **1f**, it failed to react with di-*n*-butylamine even when heated in a sealed tube at 180° for 24 hr. Thus, the comparison of antimalarial activity between **1a** and **1b** could not be made. When **1c** was treated with benzylamine **1e** was obtained. The dioxyethylene bromo compound **1f** reacted with benzylamine smoothly to give the target compound **1g** which could be compared in its antimalarial activity and phototoxicity with **1e** to test the hypothesis we started with.

Biological Tests.—The compounds were tested for their antimalarial activity against *Plasmodium berghei* in mice by Dr. L. Rane according to the procedure already published.⁴ **1e** showed activity at 40 mg/kg, cured 4 mice at 160 mg and all 5 mice at 320 mg with no toxic deaths. The dioxyethylene derivative **1g** was inactive even at a dose of 640 mg/kg. When tested for phototoxicity in mouse (ip), **1e** was approximately 9 times more phototoxic than **1g**.⁵

Experimental Section

6,8-Dichloro-2-(3,4-dichlorophenyl)-4-(α-benzylaminomethyl)quinolinemethanol·HCl (1e) was prepared in 77.6% yield by the procedure of Lutz *et al.*¹ It was crystallized from MeOH-Et₂O, mp 250–254°. *Anal.* (C₂₄H₁₁Cl₅N₂O) C, H, Cl, N.

6,8-Dichloro-2-(3,4-dichlorophenyl)-4-(2-bromo-1,1-ethylenedioxyethyl)quinoline (1f) was prepared from **1d**⁶ in 64.0% yield by the procedure of Takahashi and Tanabe.⁷ It was crystallized several times from C₆H₆, mp 212–214°. *Anal.* (C₁₉H₁₂BrCl₄NO₂) C, H, Br, Cl, N.

6,8-Dichloro-2-(3,4-dichlorophenyl)-4-(2-benzylamino-1,1-ethylenedioxyethyl)quinoline (1g).—A mixture of **1f** (1.0 g), benzylamine (10 ml), ethoxyethanol (10 ml), and a crystal of I₂ was refluxed for 24 hr. Solvent and excess benzylamine were removed *in vacuo* and the residue was triturated with 10% NaOH and extracted (C₆H₆). The extract was dried (K₂CO₃), filtered, and concentrated to give 790 mg (67.0%) of crude product which, after two crystallizations from C₆H₆, melted at 159–161°. *Anal.* (C₂₆H₂₀Cl₄N₂O₂) C, H, N.

Acknowledgment.—This work was supported by U. S. Army Medical Research and Development Command under the Research Contract No. DA-49-193-MD-2869 and is Contribution No. 703 from the Army Research Program on Malaria.

(3) E. R. Atkinson and A. J. Puttick, Paper No. 42, Division of Medicinal Chemistry, 156th Meeting of the American Chemical Society, September 1968, Atlantic City, N. J.

(4) T. S. Osdene, P. B. Russell, and L. Rane, *J. Med. Chem.*, **10**, 431 (1967).

(5) Col. William E. Rothe, VC, Division of Medicinal Chemistry, WRAIR, Walter Reed Army Medical Center, Washington, D. C. 20012, private communications.

(6) R. E. Lutz, *et al.*, *J. Amer. Chem. Soc.*, **68**, 1820 (1946).

(7) M. Takahashi and R. Tanabe, *Chem. Pharm. Bull.*, **15**, 793 (1967).