

Notes

Chemotherapeutic Nitroheterocycles. 2.^{1a,b}
2-(5-Nitro-2-furyl)pyrimidines with
Basic Substituents

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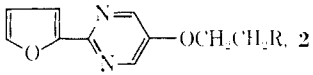








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It was shown in a previous paper^{1a} that 5-substituted 2-(5-nitro-2-furyl)pyrimidines possess a high *in vitro* activity against *Trichomonas vaginalis* and an *in vivo* activity could be demonstrated. A continuous search

reaction of 2-(2-furyl)-5-(2-chloroethoxy)pyrimidine (**1a**) or of the tosylate of 2-(2-furyl)-5-(2-hydroxyethoxy)pyrimidine (**1b**) with excess secondary amine in EtOH and subsequent nitration of the 2-(2-furyl)-5-(2-aminoethoxy)pyrimidine derivatives (**2**) with HNO₃-H₂SO₄ according to Scheme I.

The compounds were isolated as free bases; only in the case that no crystalline material could be obtained the HCl or HSO₄ salts were prepared. All synthesized intermediates **2** and nitro compounds **3** are summarized in Table I. The analytical data and physical measurements (uv, ir, nmr, or titrations) are in accordance with the given structures and were determined for all compounds.

TABLE I

Compd	R	Crystn solvent	Mp, °C	Yield, %	Formula	Analysis	Crystn solvent	Mp, °C	Yield, %	Formula	Analysis	
												
a	NMe ₂	<i>a</i>	81-83	57	C ₁₂ H ₁₃ N ₃ O ₂	<i>b</i>	EtOH	147-148	37	C ₁₂ H ₁₃ N ₃ O ₄	N	
b	NEt ₂	<i>a</i>	47-49	51	C ₁₄ H ₁₉ N ₃ O ₂	C, H, N	EtOH	137-141	40	C ₁₄ H ₁₈ N ₄ O ₄	C, H, N	
c	N-(<i>n</i> -Bu) ₂	<i>i</i> -PrOH	168-173	44	C ₁₈ H ₂₅ ClN ₃ O ₂ ^c	C, H, Cl, N	<i>i</i> -PrOH-H ₂ O	83-85	45	C ₁₈ H ₂₆ N ₄ O ₄	C, H, N	
d		Petroleum ether-C ₆ H ₆	91-92	83	C ₁₄ H ₁₇ N ₃ O ₂	C, H, N	EtOH-2-methoxyethanol	165-167	54	C ₁₄ H ₁₅ N ₄ O ₈ ^d	C, H, N, S	
e		<i>a</i>	91-92	66	C ₁₅ H ₁₉ N ₃ O ₂	<i>b</i>	EtOH	135-136	36	C ₁₅ H ₁₈ N ₄ O ₄	C, H, N	
f		<i>a</i>	83-84	75	C ₁₆ H ₂₁ N ₃ O ₂	<i>b</i>	EtOH	144-145	57	C ₁₆ H ₁₉ N ₄ O ₄	C, H, N	
g		<i>a</i>	78-79	60	C ₁₆ H ₂₁ N ₃ O ₂	<i>b</i>	EtOH	134-136	31	C ₁₆ H ₂₄ N ₄ O ₈ ^d	C, H, N, S	
h		Petroleum ether-C ₆ H ₆	109-112	69	C ₁₈ H ₂₃ N ₃ O ₂	<i>b</i>	MeOH	252 dec	51	C ₁₈ H ₂₃ ClN ₄ O ₄ ^e	C, H, N, Cl	
i		Petroleum ether-C ₆ H ₆	94-95	75	C ₁₄ H ₁₇ N ₃ O ₃	<i>b</i>	MeOH-H ₂ O	200-201	48	C ₁₄ H ₁₈ N ₄ O ₈ ^d	C, H, N, S	
j		Petroleum ether-C ₆ H ₆	76-80	64	C ₁₅ H ₂₀ N ₄ O ₂	<i>b</i>	EtOH	144-145	39	C ₁₅ H ₁₉ N ₃ O ₄	C, H, N	
k		Petroleum ether-C ₆ H ₆	73-74	53	C ₁₇ H ₂₄ N ₄ O ₂	<i>b</i>	MeOH	257 dec	22	C ₁₇ H ₂₅ Cl ₂ N ₂ O ₄ ^e	C, H, N, Cl	

^a Crude material, which was used without purification. ^b These compounds were used without analytical measurements and were characterized by their nitration products **3**. ^c Hydrochloride. ^d Hydrogen sulfate. ^e Dihydrochloride.

for derivatives with better biological potency led us to the syntheses of derivatives with basically substituted side chains in position 5 of the pyrimidine nucleus. This paper refers to the synthesis and biological evaluation of the new nitrofurylpyrimidines **3**.

Chemistry.—The compounds were synthesized by

Biological Results.—The compounds were screened *in vitro* against Gram-positive and Gram-negative bacteria, fungi, and protozoa. The activity against a selection of microorganisms is shown in Table II. It can be seen that only in a few cases does a pronounced activity occur against Gram-positive and Gram-negative bacteria and in selected cases a slight activity against *Candida albicans*. But the activity against *T. vaginalis* is impressive; all compounds show a better activity than metronidazole which has a MIC of

(1) (a) Paper 1: R. Albrecht, K. Gutsche, H.-J. Kessler, and E. Schroder: *J. Med. Chem.*, **13**, 733 (1970); (b) a preliminary report of part of this work has been presented at the 6th International Congress of Chemotherapy, Tokyo, Aug. 1969; (c) to whom inquiries should be addressed.

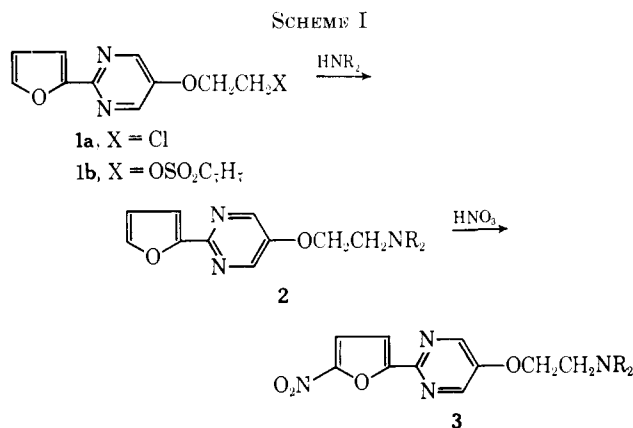


TABLE II

In Vitro ANTIMICROBIAL DATA (TUBE DILUTION TEST; MIC VALUES IN $\mu\text{g/ml}$)

Compd	<i>S. aureus</i> ^a	<i>E. coli</i> ^b	<i>P. vulgaris</i> ^c	<i>K. pneumoniae</i> ^d	<i>C. albicans</i>	<i>T. vaginalis</i>
2a	12	1.6	6.2	3.1	100	0.39
2b	25	6.2		12		0.19
2c						0.78
2d	25	12	25	25		0.19
2e	25	3.1		12	25	0.39
2f		3.1		25		0.19
2g	50	6.2	50	50		0.78
2h	50	12				0.19
2i	12	1.6	50	25		0.78
2j	3.1	1.6	12	6.2	100	0.78
2k	25	12				0.10
Metronidazole						1.6

^a *Staphylococcus aureus*. ^b *Escherichia coli*. ^c *Proteus vulgaris*. ^d *Klebsiella pneumoniae*.

1.6 $\mu\text{g/ml}$ in our test. The most active compound *in vitro* is 2-(5-nitro-2-furyl)-5-(2-*N*-*n*-propylpiperazinoethoxy)pyrimidine dihydrochloride (**3k**) with an MIC of 0.10 $\mu\text{g/ml}$. Some of the compounds were investigated *in vivo* against a subcutaneous *T. vaginalis* infection in mice and proved to be active by oral application. Preliminary results indicate that *in vivo* activity of 2-(5-nitro-2-furyl)-5-(2-pyrrolidinoethoxy)pyrimidine (**3d**) is comparable to metronidazole.

Experimental Section²

2-(2-Furyl)-5-(2-*p*-toluenesulfonyloxyethoxy)pyrimidine (1b).—2-(2-Furyl)-5-(2-hydroxyethoxy)pyrimidine (10.3 g, 50 mmol) was dissolved in 100 ml of pyridine, 9.53 g (50 mmol) of *p*-TsCl was added and the mixture was stirred for 1.5 hr. The solution was poured into H₂O, the insoluble product was filtered and recrystd from PhMe, mp 154–155°; yield 8.6 g (48%). *Anal.* (C₁₇H₁₆N₂O₅S) N, S.

2-(2-Furyl)-5-(2-diethylaminoethoxy)pyrimidine (2b).—2-(2-Furyl)-5-(2-chloroethoxy)pyrimidine (7.85 g, 38.3 mmol) and 10.42 g (0.142 mol) of Et₂NH was heated in 80 ml EtOH at 70–80° for 18 hr. The solvent and excess amine were removed by distillation, the residue was mixed with H₂O and extracted with CH₂Cl₂. The solution was dried (K₂CO₃) and CH₂Cl₂ was removed, yielding a crystalline compound, mp 47–49°; yield 5.1 g (51%). *Anal.* (C₁₄H₁₈N₂O₂) C, H, N.

2-(5-Nitro-2-furyl)-5-(2-diethylaminoethoxy)pyrimidine (3b)—2-(2-Furyl)-5-(2-diethylaminoethoxy)pyrimidine (5.0 g, 19.1

mmol) was suspended in 25 ml of concd H₂SO₄ and 1.5 ml of HNO₃ (*d* = 1.48) was added dropwise at 5°. After stirring for 0.5 hr the mixture was poured on ice, the aq solution was made alkaline with aq NH₃ and the solid product was filtered, washed with H₂O, and recrystallized from EtOH, mp 137–141°; yield 2.0 g (34%). *Anal.* (C₁₄H₁₈N₄O₄) C, H, N.

2-(2-Furyl)-5-(2-di-*n*-butylaminoethoxy)pyrimidine·HCl.—2-(2-Furyl)-5-(2-*p*-toluenesulfonyloxyethoxy)pyrimidine (2.08 g, 5.77 mmol) and 2.33 g (18.1 mmol) of *n*-Bu₂NH in 60 ml of EtOH was refluxed for 20 hr. EtOH and excessive Bu₂NH were removed by distillation, the residue was dissolved in H₂O and the aq solution was extracted with CH₂Cl₂. The solution was dried (Na₂SO₄) and the solvent removed. The residue was dissolved in *i*-PrOH–Et₂O and HCl in Et₂O was added. The solid hydrochloride was filtered off and recrystd from *i*-PrOH, mp 168–173°; yield 0.90 g (44%). *Anal.* (C₁₈H₂₈ClN₂O₂) C, H, Cl, N.

2-(5-Nitro-2-furyl)-5-(2-di-*n*-butylaminoethoxy)pyrimidine (3c).—2-(2-Furyl)-5-(2-di-*n*-butylaminoethoxy)pyrimidine·HCl (0.90 g, 2.54 mmol) was suspended in 3 ml of concd H₂SO₄ and 0.255 ml of HNO₃ (*d* = 1.48) was added at 5°. After stirring for 0.5 hr the mixture was poured onto ice, the resulting aq mixture was neutralized with concd NH₃ and the insoluble material recrystallized from *i*-PrOH–H₂O, mp 83–85°; yield 413 mg (45%). *Anal.* (C₁₈H₂₈N₄O₄) C, H, N.

Reassignment of the Absolute Configuration of 3-Acetoxyquinuclidine Methiodide and the Absolute Configuration of Receptor-Bound Acetylcholine

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The resolution and absolute configurations of (+)- and (–)-3-acetoxyquinuclidine methiodide have recently been reported.¹ It was noted that only one of the optical enantiomers is a substrate of acetylcholinesterase (AChE) and is a muscarinic agonist.¹ This enantiomer was assigned the (*S*) configuration by the method of asymmetric sulfoxide synthesis² as applied to (+)-3-quinuclidinol. This assignment led to the conclusion that the bridgehead carbon atom of the biologically active enantiomer occupies a position equivalent to the methyl substituent of (*S*)-(+)- β -methylacetylcholine (I), the enantiomer which is a substrate of AChE and possesses muscarinic activity equivalent to that of acetylcholine (ACh).¹ In the meantime, the crystal structures of the α - and β -methyl-ACh isomers have been analyzed by Chothia and Pauling,^{3,4} and their conformation was discussed in relation to hydrolysis by AChE.⁵ The results led us to question the configurational assignment of (+)-3-quinuclidinol¹ and after reexamination of the experimental data it is evident that the dextrorotatory enantiomer possesses the (*S*) rather than the (*R*) configuration. It follows that the previous analysis of results¹ requires reinterpretation.

According to the rules governing asymmetric sulfoxide synthesis,² alcohols corresponding to stereo-

(1) J. B. Robinson, B. Belleau, and B. Cox, *J. Med. Chem.*, **12**, 848 (1969).

(2) M. M. Green, M. Axelrod, and M. Mislow, *J. Amer. Chem. Soc.*, **88**, 861 (1966).

(3) C. Chothia and P. Pauling, *Chem. Commun.*, 626 (1969).

(4) C. Chothia and P. Pauling, *ibid.*, 746 (1969).

(5) C. Chothia and P. Pauling, *Nature*, **223**, 919 (1969).

(2) Melting points are uncorrected and taken on a Tottoli melting point apparatus (Fa. W. Büchi, Switzerland). Where analytical results are indicated only by symbols of the elements or functions, values found for those elements or functions were within $\pm 0.4\%$ of the calculated values.